



Designation: ~~E2551–13~~ E2551 – 18

Standard Test Method ~~Methods~~ for Humidity Calibration (or Conformance) of Humidity Generators for Use with Thermogravimetric Analyzers¹

This standard is issued under the fixed designation E2551; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 ~~This~~ These test method describes methods describe the humidity calibration (or conformance) of humidity generators for use with thermogravimetric analyzers and other thermal analysis apparatus. The humidity range covered is ~~5 to 95 % RH~~ 5 to 95 % RH and the temperature range is ~~0 °C to 80 °C~~ 0 °C to 80 °C.

1.2 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

~~1.3 There are no ISO equivalents to this standard.~~

~~1.3 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.~~

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~~1.4 This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.~~

2. Referenced Documents

2.1 ASTM Standards:²

[D1193 Specification for Reagent Water](#)

[E177 Practice for Use of the Terms Precision and Bias in ASTM Test Methods](#)

[E473 Terminology Relating to Thermal Analysis and Rheology](#)

[E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method](#) /astm-e2551-18

[E1142 Terminology Relating to Thermophysical Properties](#)

¹ ~~This~~ These test method is methods are under the jurisdiction of ASTM Committee [E37](#) on Thermal Measurements and ~~is~~ are the direct responsibility of Subcommittee [E37.10](#) on Fundamental, Statistical and Mechanical Properties.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

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3. Terminology

3.1 Specific technical terms used in this standard are defined in Terminologies E473 and E1142. These terms include *thermal curve*, *first-deviation-from-baseline*, *isohume*, *relative humidity*, *thermal curve*, and *thermogravimetric analysis*.

3.2 *Definitions of Terms Specific to This Standard:*

3.2.1 *relative humidity, n*—the ratio of actual partial pressure of water to the saturated water vapor pressure at the same temperature, expressed as a percentage.

3.2.1 *water activity, n*—the ratio of actual partial pressure of water to the saturated water vapor pressure at the same temperature, expressed as a decimal fraction.

3.2.1.1 *Discussion*—

Water activity is also known as relative pressure in some applications areas.

3.2.3 *first-deviation-from-baseline, n*—the relative humidity or water activity at which a deflection from the established baseline is first observed.

4. Summary of Test Method

4.1 Humidity generators are devices aimed at producing a specific level of humidity in the purge gas used by thermogravimetric analyzers or other thermal analysis apparatus. The requested humidity levels may be held constant (~~isohum~~)(isohume) or increased or decreased in a continuous or stepped fashion.

4.2 The humidified purge gas is submitted to a thermogravimetric analyzer in which the weight of a hygroscopic material is observed. The relative humidity (or activity) of the moisture in the purge gas is stepped or scanned through a humidity range. At a fixed humidity of the purge gas, the test specimen deliquesces and gains weight. In Test Method A, the humidity of the onset of this weight gain is taken as the humidity calibration point. In Test Methods B and C, the rate of weight change is zero at the humidity calibration point.

5. Significance and Use

5.1 ~~This~~These test method ~~calibrates~~methods calibrate or demonstratesdemonstrate conformity of the humidity level in a purge gas generated by a humidity generator at a fixed temperature. Such calibration or demonstration of conformity may be required by quality initiatives.

5.2 Conformance demonstrates that the humidified purge gas is within some established limits.

5.3 Calibration provides an offset and or slope value that may be used for establishing the relative humidity scale of the apparatus.

6. Interferences

6.1 Temperature regulation of any solution-head space environment to within $\pm 0.1^\circ\text{C}$ – $\pm 0.1^\circ\text{C}$ is essential for realizing generated relative humidity values stable to within $\pm 1\%$ RH (expected).

7. Apparatus

7.1 The humidity generator that is the focus of this standard may be an accessory providing a humidified purge gas to some other thermal analysis apparatus (typically a thermogravimetric analyzer) or it may be part of a self-contained instrument that includes both the humidity generator and the thermal analysis apparatus. In the former case, some of the components described below may be redundant.

7.2 *Humidity Generator*—The essential instrumentation required to provide the minimum humidity generator capability for ~~this method~~these test methods includes:

7.2.1 *Temperature Sensor*—*Sensor*, to provide an indication of the purge gas temperature readable to within $\pm 0.1^\circ\text{C}$ – $\pm 0.1^\circ\text{C}$.

7.2.2 *Temperature Controller*—*Controller*, capable of executing a specific temperature program by operating heaters or coolers between selected temperature limits at a rate of temperature change of $0.5^\circ\text{C}/\text{min}$ – $0.5^\circ\text{C}/\text{min}$ constant to $\pm 0.1^\circ\text{C}/\text{min}$ – $\pm 0.1^\circ\text{C}/\text{min}$ or at an isothermal temperature constant to within $\pm 0.1^\circ\text{C}$ – $\pm 0.1^\circ\text{C}$.

7.2.3 *Humidity Sensor*—*Sensor*, capable of indicating the humidity of the purge gas over the range of 5 to 95 % relative humidity (% RH)–% RH to 95 % RH readable to within $\pm 0.1\%$ RH.

7.2.4 *Humidity Controller*—*Controller*, capable of executing a specific humidity program by operating purge gas humidifiers between selected humidity limits at a rate of humidity change of 0.5% RH– 0.5% RH/min–min constant to within $\pm 0.1\%$ RH or at an ~~isohum~~isohume relative humidity to within $\pm 0.1\%$ RH.

7.2.5 *Purge Gas Flow Sensor*—*Sensor*, capable of measuring purge gas flow readable to within ± 0.1 mL/min.

7.2.6 *Purge Gas Flow Controller—Controller*, capable of controlling purge gas flow readable to within ± 0.1 mL/min.
 7.2.7 *Humidifier element—Humidifier Element*, capable of generating purge gases with relative humidity continuously over the range of 5 % RH to 95 % RH.

7.3 *Thermogravimetric Analyzer (TGA)*—The essential instrumentation required to provide the minimum thermogravimetric analytical capability for this method includes:

7.3.1 A furnace to provide uniform controlled heating or cooling of a specimen to a constant temperature or at a constant rate within the applicable temperature range of this method.

7.3.2 A temperature sensor to provide an indication of the specimen or furnace temperature to within $\pm 0.1^\circ\text{C}$, $\pm 0.1^\circ\text{C}$.

7.3.3 A continuously recording balance to measure the specimen weight with a minimum capacity of 100 mg and sensitivity of ± 10 μg .

7.3.4 A means of maintaining the specimen/container under atmospheric control at a purge rate of 10 mL/min to 200 mL/min ± 10 mL/min or 1 %, whichever is greater.

7.3.5 A temperature controller capable of executing a specific temperature program by operating the furnace between selected temperature limits at a rate of temperature change of $0.5^\circ\text{C}/\text{min}$ $\pm 0.5^\circ\text{C}/\text{min}$ constant to within $\pm 0.1^\circ\text{C}/\text{min}$ or to an isothermal temperature that is maintained constant to within $\pm 0.1^\circ\text{C}$ $\pm 0.1^\circ\text{C}$ for a minimum of 100 h.

7.3.6 Containers (pans, crucibles, etc.) that are inert to the specimen and that will remain gravimetrically stable within the temperature limits of this method.

7.3.7 Data storage capable of storage of the weight and relative humidity signals.

7.3.8 A display capable of plotting a thermal curve with weight on the ordinate (Y-axis) and relative humidity (or activity) on the abscissa (X-axis) with a sensitivity of 10 μg for weight and 0.1 % RH, respectively.

8. Reagents and Materials

8.1 One or more inorganic salts taken from Table 1 selected to provide the humidity range of interest.

8.2 Purity of Reagents—Reagent grade chemicals (or better) shall be used for preparation of all standard solutions.

8.3 Purity of Water—Reagent water produced by distillation or by ion exchange, or reverse osmosis followed by distillation shall be used (see Specification D1193).

9. Hazards

9.1 Salt solutions are extremely corrosive to apparatus if spilled. Care shall be taken in their preparation and handling to prevent contact with apparatus.

10. Preparation of Apparatus

10.1 Perform any setup or calibration procedures recommend by the apparatus manufacturer in the operations manual.

TABLE 1 Humidity Fixed Points

Greenspan, L., “Humidity Fixed Points of Binary Saturated Aqueous Solutions,” *Journal of Research of the National Bureau of Standards—A. Physics and Chemistry*, Vol 81A, No. 1, 1977, pp. 89–96.

NOTE 1—Greenspan, L., “Humidity Fixed Points of Binary Saturated Aqueous Solutions,” *Journal of Research of the National Bureau of Standards—A. Physics and Chemistry*, Vol 81A, No. 1, 1977, pp. 89–96.

| Temperature (°C) | Lithium Chloride | Potassium Acetate | Magnesium Chloride | Potassium Carbonate | Magnesium Nitrate | Sodium Bromide | Strontium Chloride | Sodium Chloride | Potassium Chloride |
|------------------|------------------|-------------------|--------------------|---------------------|-------------------|----------------|--------------------|-----------------|--------------------|
| 10 | 11.3 | 23.7 | 33.5 | 43.1 | 57.4 | 62.2 | 75.66 | 75.7 | 86.8 |
| 15 | 11.3 | 23.4 | 33.3 | 43.2 | 55.9 | 60.7 | 74.13 | 75.6 | 85.9 |
| 20 | 11.3 | 23.1 | 33.1 | 43.2 | 54.4 | 59.1 | 72.52 | 75.5 | 85.1 |
| 25 | 11.3 | 22.5 | 32.8 | 43.2 | 52.9 | 57.6 | 70.85 | 75.3 | 84.3 |
| 30 | 11.3 | 21.6 | 32.4 | 43.2 | 51.4 | 56.0 | 69.12 | 75.1 | 83.6 |
| 35 | 11.3 | ... | 32.1 | ... | 49.9 | 54.6 | ... | 74.9 | 83.0 |
| 40 | 11.2 | ... | 31.6 | ... | 48.4 | 53.2 | ... | 74.7 | 82.3 |
| 45 | 11.2 | ... | 31.1 | ... | 46.9 | 52.0 | ... | 74.5 | 81.7 |
| 50 | 11.1 | ... | 30.5 | ... | 45.4 | 50.9 | ... | 74.4 | 81.2 |
| 55 | 11.0 | ... | 29.9 | ... | ... | 50.2 | ... | 74.4 | 80.7 |
| 60 | 11.0 | ... | 29.3 | ... | ... | 49.7 | ... | 74.5 | 80.3 |
| 65 | 10.9 | ... | 28.5 | ... | ... | 49.5 | ... | 74.7 | 79.9 |
| 70 | 10.8 | ... | 27.8 | ... | ... | 49.7 | ... | 75.1 | 79.5 |
| 75 | 10.6 | ... | 26.9 | ... | ... | 50.3 | ... | 75.6 | 79.2 |
| 80 | 10.5 | ... | 26.1 | ... | ... | 51.4 | ... | 76.3 | 78.9 |

10.2 *Positioning of the Temperature Sensor*—If the system employs a temperature sensor that is movable, it shall be located as close to the specimen as possible without touching it or the balance pan. In addition, it must be located in exactly the same position during calibrations as used during analytical determinations.

11. Calibration and Standardization

11.1 Calibrate the temperature display of the apparatus according to Test Method **E1582** using a heating rate of $0.5 \text{ }^{\circ}\text{C}/\text{min} \pm 0.2^{\circ}\text{C}/\text{min}$.

12. Procedure

12.1 Close the system, adjust the atmospheric flow rate of the purge gas to the selected rate, and zero (tare) the balance.

12.2 Open the system and place 5 mg to 10 mg of the reference material in the specimen container in the same position as would be placed for a test specimen. Close the system.

12.3 Measure the weight of the reference material and report its value.

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