



Designation: **D7419—13 D7419 – 18**

# Standard Test Method for Determination of Total Aromatics and Total Saturates in Lube Basestocks by High Performance Liquid Chromatography (HPLC) with Refractive Index Detection<sup>1</sup>

This standard is issued under the fixed designation D7419; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reappraisal. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reappraisal.

## 1. Scope\*

1.1 This test method covers the determination of total aromatics and total saturates in additive-free lube basestocks using high performance liquid chromatography (HPLC) with refractive index (RI) detection. This test method is applicable to samples containing total saturates in the concentration range of 74.9 % to 100.0 % by mass and aromatics in the concentration range of 0.2 to 46 mass % 0.0 % to 25.1 % by mass. The precision is expressed in terms of the total saturates.

1.1.1 Polar compounds, if present, are combined with the total aromatics. Precision was determined for basestocks with polars content  $< 1.0$  mass %  $< 1.0$  % by mass.

1.2 This test method includes a relative bias section for total saturates in basestocks based on a Practice **D6708** accuracy assessment between Test Method D7419 and Test Method **D2007**. The derived correlation equation is only applicable for basestocks in the total saturates concentration range from 75.0 % to 100.0 % by mass as measured by Test Method D7419.

1.2.1 The applicable range for total saturates by Test Method **D2007** is from 71.0 % to 99.0 % by mass as reported by Test Method **D2007**.

1.3 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.*

1.5 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

## 2. Referenced Documents

2.1 *ASTM Standards:*<sup>2</sup>

[D2007 Test Method for Characteristic Groups in Rubber Extender and Processing Oils and Other Petroleum-Derived Oils by the Clay-Gel Absorption Chromatographic Method](#)

[D4057 Practice for Manual Sampling of Petroleum and Petroleum Products](#)

[D4177 Practice for Automatic Sampling of Petroleum and Petroleum Products](#)

[D6299 Practice for Applying Statistical Quality Assurance and Control Charting Techniques to Evaluate Analytical Measurement System Performance](#)

[D6300 Practice for Determination of Precision and Bias Data for Use in Test Methods for Petroleum Products and Lubricants](#)

[D6708 Practice for Statistical Assessment and Improvement of Expected Agreement Between Two Test Methods that Purport to Measure the Same Property of a Material](#)

## 3. Terminology

3.1 *Definitions:*

<sup>1</sup> This test method is under the jurisdiction of ASTM Committee **D02** on Petroleum Products, Liquid Fuels, and Lubricants and is the direct responsibility of Subcommittee **D02.04.0C** on Liquid Chromatography.

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<sup>2</sup> For referenced ASTM standards, visit the ASTM website, [www.astm.org](http://www.astm.org), or contact ASTM Customer Service at [service@astm.org](mailto:service@astm.org). For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

\*A Summary of Changes section appears at the end of this standard

3.1.1 *aromatics, n—in high performance liquid chromatography*, aromatic hydrocarbon components, minus polar material, that ~~has~~have a longer retention time than saturates on the specified polar columns, but can be removed as a single peak by backflushing the columns with heptane.

3.1.1.1 *Discussion—*

Generally, aromatic hydrocarbons contain ~~one~~ to four rings.

3.1.2 *backflush, v—*elution of the HPLC mobile phase in the backward or reverse direction from the silica gel column towards the cyano column.

3.1.2.1 *Discussion—*

In this test method, it is used to elute the total aromatics plus polars as one sharp component.

3.1.3 *foreflush, v—*elution of HPLC mobile phase in the forward direction.

3.1.3.1 *Discussion—*

In this test method, the sample enters the cyano column first followed by elution through the silica gel column.

3.1.4 *polars, n—in high performance liquid chromatography*, components that may contain organically bonded nitrogen, oxygen, and oxidized sulfur components and are more strongly retained than aromatic hydrocarbons.

3.1.4.1 *Discussion—*

In this HPLC method, polars are backflushed with the aromatics and the two cannot be distinguished. Generally present in very small amounts, such as ~~< 1 mass %~~ < 1 % by mass.

3.1.5 *saturates, n—*hydrocarbon components that are not retained strongly by the specified polar columns when heptane is used as the mobile phase.

3.1.5.1 *Discussion—*

Generally, these consist of paraffins and cycloparaffins. [ASTM D7419-18](#)

#### 4. Summary of Test Method [catalog/standards/sist/cb191260-089d-46bd-8ab9-c9fd392357a8/astm-d7419-18](#)

4.1 A known mass of sample is diluted in the mobile phase and a fixed volume of this solution is injected into a calibrated high performance liquid chromatograph. The separation column set has little affinity for the saturates while retarding the aromatic hydrocarbons and the polars. As a result of this retardation, the aromatic hydrocarbons and polars are separated from the saturates. At a predetermined time, after the elution of the saturates, the column is backflushed to elute the aromatics and polars as a single sharp band.

4.2 The column set is connected to a refractive index detector that detects the components as they elute from the column. The electronic signal from the detector is continually monitored by a data processor. The integrated signals (peak areas) from the saturates and aromatics components are corrected using a predetermined response factor and the ~~mass %~~ percent by mass saturates and aromatics plus polars are calculated.

#### 5. Significance and Use

5.1 The composition of a lubricating oil has a large effect on the characteristics and uses of the oil. The determination of saturates, aromatics, and polars is a key analysis of this composition. The characterization of the composition of lubricating oils is important in determining their interchangeability for use in blending etcetera.

#### 6. Apparatus

6.1 *High Performance Liquid Chromatograph (HPLC)*—Any HPLC capable of pumping the mobile phase at flow rates between ~~33 mL/min~~ and 5 mL/min, with a precision better than 0.5 %.

6.2 *HPLC Sample Injection System*—Capable of injecting 10  $\mu$ L (nominal) of sample solution with a repeatability of 1 % or better.

6.3 *Column System*—A column set is used. Any stainless steel HPLC column packed with silica gel stationary phase that meets the resolution and capacity requirements specified in [9.3](#) is suitable. Use a single silica column or two connected in series with

a total length of 500 mm with an internal diameter of  $7.57.5\text{ mm}$  to 10 mm and packed with  $5\text{ }\mu\text{m}$  particle size. In addition to the silica column, an HPLC column packed with cyano (CN) stationary phase is required and placed in series in front of the silica column. A CN column length of  $100100\text{ mm}$  to 250 mm with an internal diameter of  $7.77.7\text{ mm}$  to 10 mm and packed with  $55\text{ }\mu\text{m}$  to  $10\text{ }\mu\text{m}$  particle size stationary phase has been found to be satisfactory. Table 1 gives examples of column sets used in the cooperative study.

6.4 *Backflush Valve*—Automatic flow-switching valve designed for use in HPLC systems that is capable of operating at pressures up to  $2 \times 10^4\text{ kPa}$ .

6.5 *Refractive Index Detector*—Any refractive index detector may be used provided it is capable of being operated over the refractive index range from 1.3 to 1.6 or equivalent, meets the sensitivity and linearity of calibration requirement specified in the method, and has a suitable output signal for the data system. If the refractive index detector has a facility for independent temperature control, it is recommended that this be set at  $5^{\circ}\text{C} \pm 5^{\circ}\text{C}$  above the laboratory temperature.

6.5.1 *UV-Detector*—An optional but recommended UV detector set to wavelength 254 nm may be used in series with the RI detector to aid in setting and monitoring the backflush time between saturates and aromatics in lube samples.

6.6 *Computer or Computing Integrator*—Any data system can be used provided it is compatible with the refractive index detector, has a minimum sampling rate of 1 Hz, and is capable of peak area and retention time measurement. The data system shall have minimum capabilities for post-analysis data processing, such as automatic or manual baseline correction and reintegration.

6.7 *Volumetric Flasks*—Grade B or better, of 10 mL capacity.

6.8 *Autosampler Vials*—per instrument manufacturer. Vials with a capacity of  $>1.5\text{ mL}$  have been used successfully.

6.9 *Analytical Balance*—accurate to  $\pm 0.0001\text{ g}$ .

## 7. Reagents and Materials

7.1 *Heptane*, HPLC grade. If necessary, dry solvent with molecular sieves and then filter before use.

7.2 *Dichloromethane*, HPLC or UV grade. If necessary, dry solvent with molecular sieves and then filter before use.

7.3 *Octadecylbenzene*,  $\geq 97\%$  pure.

7.4 *Hexadecane*,  $\geq 98\%$  pure.

## 8. Sampling

8.1 Follow Practice D4057 or D4177, or a similar standard to obtain a representative laboratory sample of the basestock. Mix well before sampling.

## 9. Preparation of Apparatus

9.1 Set up the liquid chromatograph, injection system, columns, backflush valve, optional column oven, optional UV detector, refractive index detector, and computing integrator in accordance with the manufacturer's instructions and as depicted in Fig. 1. Insert the backflush valve so that the detector is always connected independently of the direction of flow through the column (see Fig. 1). Maintain the sample injection valve at the same temperature as the sample solution; in most cases this will be at room temperature. To minimize drifts in signal, ensure that the ambient temperature is relatively constant during analysis and calibration.

9.2 New commercial columns may be packed in water/methanol or other polar solvents. Before these columns can be used, flush them with dichloromethane followed with heptane before proceeding. Other suitable solvents that restore the required resolution may be used. If the resolution requirement is not met, the column may be reactivated by flushing it with additional dichloromethane. If the resolution still cannot be attained, it may be necessary to replace the column or purchase an appropriate column from other vendors. Si60 silica gel was found effective in yielding acceptable resolution and performance when properly conditioned. When not analyzing samples, column may be flushed with a low flow of heptane such as  $0.1\text{ mL/min}$ .

9.2.1 Adjust the flow rate of the mobile phase to a constant  $3.0\text{ mL/min}$  to  $3.5\text{ mL/min}$ , and ensure the reference cell of the refractive index detector is full of mobile phase. Fill the reference cell as instructed by the manufacturer.

TABLE 1 Examples of Operating Conditions Used in Cooperative Studies

	Lab A	Lab B	Lab C
Silica Column	Varian, 50 cm length by 7.7 mm i.d. $5\text{ }\mu\text{m}$ Si60	Varian, 50 cm by 7.7 mm Si60 (CP28526)	Phenomenex, 2 x Si60 (10 by 250 mm, $5\text{ }\mu\text{m}$ )
Cyano Column	Alltech/YMC, 100 by 10 mm $10\text{ }\mu\text{m}$	Waters/YMC, 100 by 12 mm $5\text{ }\mu\text{m}$	YMC, 10 by 100 mm $5\text{ }\mu\text{m}$
RI Detector	Agilent 1200	Hewlett Packard RI, model HP1047A	Shimadzu RID-10A
Heptane Flow (mL/min)	3.5 mL/min	3.0	3.0
Resolution	5	5-6	10.3
Injected Volume (microlitres)	10	10	10

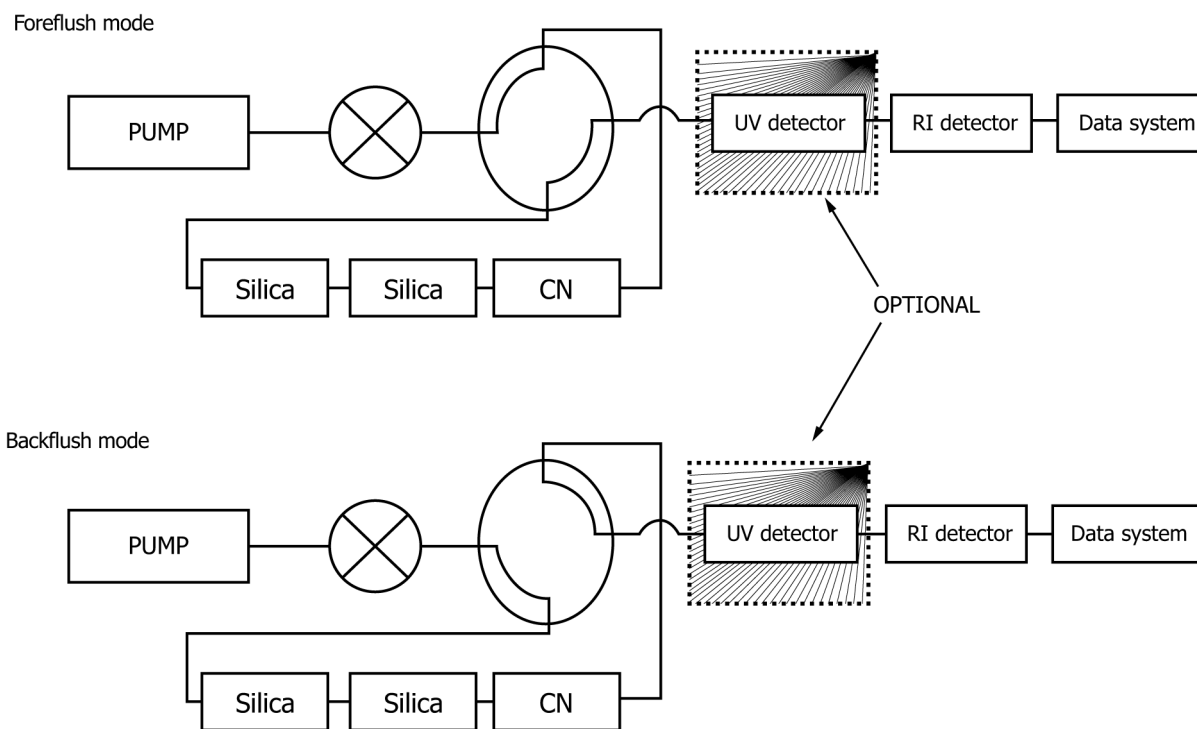


FIG. 1 Diagrammatic Representation of Liquid Chromatograph

9.2.2 To minimize drift, it is essential to make sure the reference cell of the RI detector is full of solvent. The best way to accomplish this is either (1) to flush the mobile phase through the reference cell (then isolate the reference cell to prevent evaporation of the solvent) immediately prior to analysis, or (2) to continuously make up for solvent evaporation by supplying a steady independent flow through the reference cell. The make-up flow is optimized so that reference and analytical cell mismatch due to drying-out, temperature, or pressure gradients is minimized. Typically, this can be accomplished with a make-up flow set at ~~one-tenth~~ one-tenth of the analytical flow.

9.3 Column Resolution and Capacity Factor:

9.3.1 Prepare a system performance standard (SPS) by weighing hexadecane ( $\pm 0.10 \text{ g} \pm 0.1 \text{ g}$ ) and octadecylbenzene ( $\pm 0.10 \text{ g} \pm 0.1 \text{ g}$ ) into a 10 mL volumetric flask and filling to the mark with heptane. For the preparation of standards, use the same source for the heptane as that used for the mobile phase. Ensure that the octadecylbenzene is completely dissolved in the mixture, for example, by using an ultrasonic bath.

9.3.2 When operating conditions are steady, as indicated by a stable horizontal baseline of the RI detector, inject 10  $\mu\text{L}$  of the SPS in the foreflush mode (backflush valve = OFF) and record the chromatogram using the data system. Fig. 2 gives an example chromatogram of the SPS mixture.

9.3.3 Ensure that the resolution between hexadecane and octadecylbenzene is five or greater as defined below. Calculate the resolution between hexadecane and octadecylbenzene as follows:

$$Resolution = \frac{2 \times (t_2 - t_1)}{3 \times (y_1 + y_2)} \quad (1)$$

where:

where:

- $t_1$  = retention time of the hexadecane peak in minutes,
- $t_2$  = retention time of the octadecylbenzene peak in minutes,
- $y_1$  = half-height width of the hexadecane peak in minutes, and
- $y_2$  = half-height width of the octadecylbenzene peak in minutes.

If the resolution is less than ~~five~~ 5, verify that all system components are functioning correctly and that the chromatographic dead volume has been minimized by using low dead volume connectors, tubing, etcetera. Ensure that the mobile phase is of sufficiently high quality. Finally, regenerate or replace the column if necessary. The column may be regenerated by flushing with dichloromethane followed by heptane until the signal is relatively constant on the RI detector. If after regenerating the silica columns, the resolution is still less than 5 then replace the silica columns. Si60 was found to be an effective silica gel with proper conditioning. For a proper analysis, a resolution of at least ~~five~~ 5 is required.

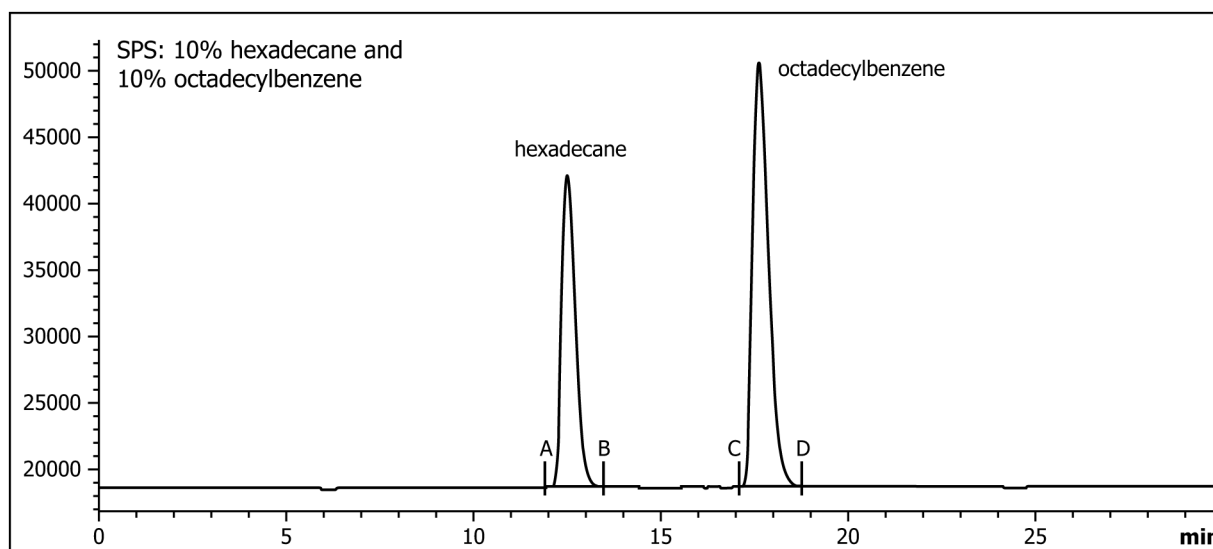


FIG. 2 Chromatogram of System Performance Standard in Foreflush Mode for Determination of Resolution, Capacity Factor, and Backflush Time

NOTE 1—Resolution loss over time may occur if a heptane mobile phase of low water content is not used. Use heptane as specified in this method. If necessary, dry the heptane with the addition of activated molecular sieves, such as MS 5A and then filter with at least 0.45 micron HPLC filter before use.

9.3.4 Calculate the capacity factor,  $k$ , for octadecylbenzene from 9.3.2 as follows:

$$\text{Capacity Factor} = k = \frac{(t_2 - t_1)}{(t_1)} \quad (2)$$

where:

where:

$t_1$  = retention time of the hexadecane peak in minutes,

$t_2$  = retention time of the octadecylbenzene peak in minutes.

$t_1$  = retention time of the hexadecane peak in minutes, and

$t_2$  = retention time of the octadecylbenzene peak in minutes.

Ensure that the capacity factor is  $> 0.4$ .

9.3.5 Using the determined retention times of the hexadecane and octadecylbenzene peaks in 9.3.2, calculate an approximate switching valve backflush time,  $B$ , in seconds, using the following equation:

$$B = t_1 + 0.1 \times (t_2 - t_1) \quad (3)$$

where:

$t_1$  = retention time of hexadecane in minutes, and

$t_2$  = retention time of octadecylbenzene in minutes.

9.4 Once the backflush time is determined, re-inject the SPS mixture with backflush in place and ensure that the backflush time as observed as a signal marker on the chromatogram occurs at the base of the eluted saturate peak. The return to baseline shall display as shown in Fig. 3, point B. This observation shall be made also for all actual lube samples analyzed. If necessary optimize, reconfirm the resolution and capacity factor, and recheck the backflush time. The use of the optional UV detector will simplify optimization of the backflush time.

9.5 Check system precision as described in 12.3.

NOTE 2—If peak area precision is poor, verify that the injection system is working optimally and that the baseline is stable (minimal drift) and noise-free.

9.6 Prepare a detection limit standard (DLS) by weighing 0.01g octadecylbenzene into a vial and adding 5.00 g hexadecane. This makes a 0.2 mass-%-0.2 % by mass aromatics standard.

9.6.1 Inject the DLS in the foreflush mode and ensure that the octadecylbenzene is detected with a signal/noise (S/N) of at least 8. Fig. 4 shows how to calculate the signal/noise.

9.6.1.1 If the octadecylbenzene is not detected, recheck the instrument, making sure the RI detector meets the manufacturer's sensitivity specifications. If necessary, increase the injection volume to 20  $\mu\text{L}$  and repeat all of the steps in Section 9. If the 20  $\mu\text{L}$

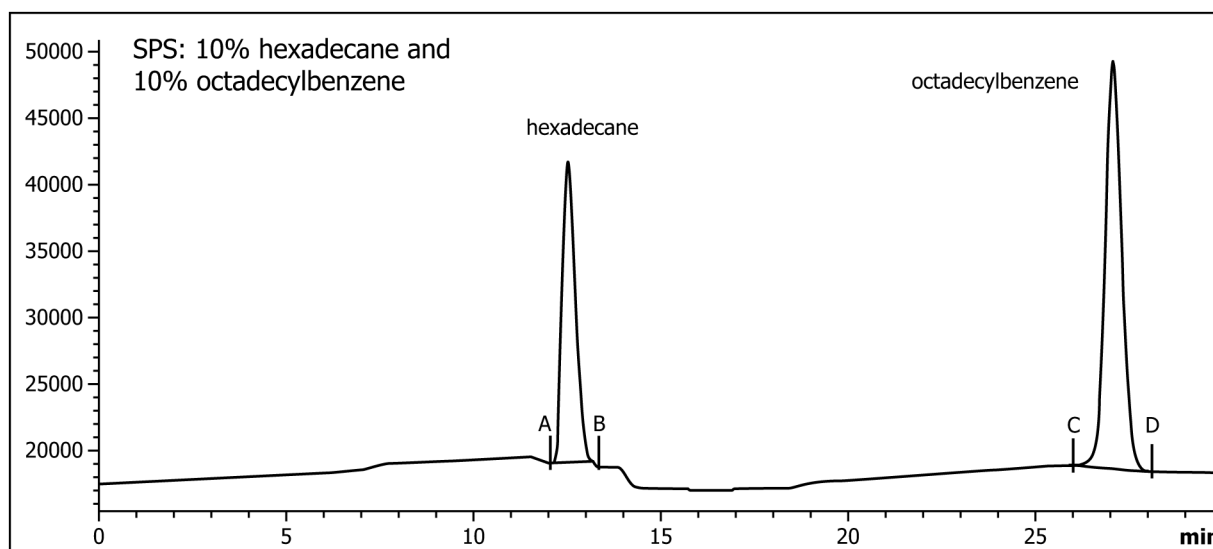


FIG. 3 Chromatogram of System Performance Standard in Backflush Mode

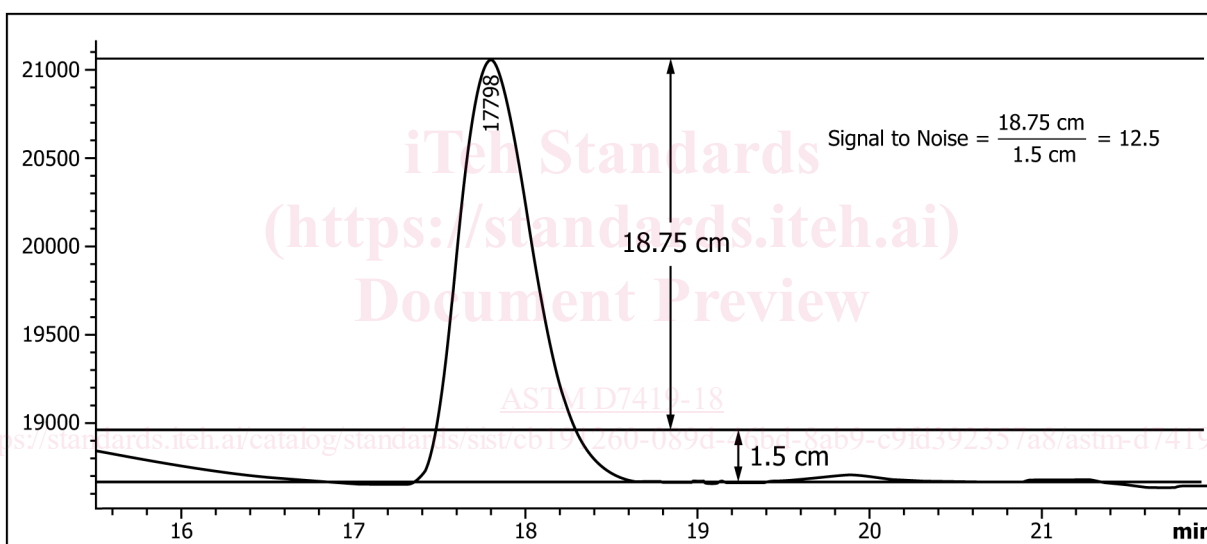


FIG. 4 Calculation of Signal/Noise Ratio for 0.2 Mass% by Mass Octadecylbenzene in Hexadecane. Peak Shown is that of Octadecylbenzene Obtained in Foreflush Mode (Saturate Peak not Shown)

injection is successful in meeting all of the specifications in the test method, then use a 20  $\mu\text{L}$  injection for all analyses. In the cooperative study, 10  $\mu\text{L}$  was adequate for all laboratories.

9.7 To perform the following step it will be necessary to calibrate the system first as described in 10.1. Verify that a minimal response is obtained at low concentrations as follows:

9.7.1 Prepare a 0.1 mass % 0.1 % by mass mixture of octadecylbenzene in hexadecane by weighing 0.01 g of octadecylbenzene in 10.0 g 10.0 g of hexadecane and analyze as a sample in the backflush mode using the cut time determined in 9.3.5.

9.7.2 Fig. 5 gives an example chromatogram response.

9.7.3 Calculate the mass % of octadecylbenzene as described in Section 13 and ensure that the results are < 0.15 mass % 0.15 % by mass.

## 10. Calibration

10.1 Prepare five calibration standards (A, B, C, D, and E) in accordance with the concentrations given in Table 2, by weighing, to the nearest 0.0001 g, the appropriate materials into 10 mL 10 mL volumetric flasks and making up to the mark with heptane.

10.2 When operating conditions are steady, as indicated by a stable horizontal baseline, inject 10  $\mu\text{L}$  of calibration standard A. For the calibration, it is necessary to use the backflush mode. Record the chromatogram, and measure the peak areas for hexadecane and octadecylbenzene.