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Standard Practice for Conducting and Evaluating Laboratory Corrosion Tests in Soils¹

This standard is issued under the fixed designation G162; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This practice covers procedures for conducting laboratory corrosion tests in soils to evaluate the corrosive attack on engineering materials. The test is conducted under laboratory ambient temperature unless the effect of temperature is being evaluated. This practice does not include provisions for microbiological influenced corrosion (MIC) testing, nor its influence on results.

1.2 This practice covers specimen selection and preparation, test environments, <u>evaluation</u>, and <u>evaluation</u> of test results.

1.3 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.4 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety safety, health, and health environmental practices and determine the applicability of regulatory limitations prior to use.

1.5 This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.

2. Referenced Documents

2.1 ASTM Standards:²

D698 Test Methods for Laboratory Compaction Characteristics of Soil Using Standard Effort (12,400 ft-lbf/ft³ (600 kN-m/m³)) D1193 Specification for Reagent Water

D1654 Test Method for Evaluation of Painted or Coated Specimens Subjected to Corrosive Environments

D2570 Test Method for Simulated Service Corrosion Testing of Engine Coolants 5b-78d7471ae4eb/astm-g162-18

- G1 Practice for Preparing, Cleaning, and Evaluating Corrosion Test Specimens
- G3 Practice for Conventions Applicable to Electrochemical Measurements in Corrosion Testing
- G4 Guide for Conducting Corrosion Tests in Field Applications

G16 Guide for Applying Statistics to Analysis of Corrosion Data

- G31 Guide for Laboratory Immersion Corrosion Testing of Metals
- G46 Guide for Examination and Evaluation of Pitting Corrosion
- G51 Test Method for Measuring pH of Soil for Use in Corrosion Testing
- G57 Test Method for Field Measurement of Soil Resistivity Using the Wenner Four-Electrode Method

G71 Guide for Conducting and Evaluating Galvanic Corrosion Tests in Electrolytes

G102 Practice for Calculation of Corrosion Rates and Related Information from Electrochemical Measurements

G187 Test Method for Measurement of Soil Resistivity Using the Two-Electrode Soil Box Method

G193 Terminology and Acronyms Relating to Corrosion

G200 Test Method for Measurement of Oxidation-Reduction Potential (ORP) of Soil

G215 Guide for Electrode Potential Measurement

¹ This practice is under the jurisdiction of ASTM Committee G01 on Corrosion of Metals and is the direct responsibility of Subcommittee G01.10 on Corrosion in Soils. Current edition approved Feb. 1, 2010Oct. 1, 2018. Published March 2010November 2018. Originally approved in 1999. Last previous edition approved in 20042010 as G162-99(2004).G162 - 99 (2010). DOI: 10.1520/G0162-99R10.10.1520/G0162-18.

² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

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3. Terminology

3.1 Definitions:

3.1.1 purified water, n-water that meets Specification D1193 Type IV requirements.

3.2 For other definitions of terms used in this guide, refer to NACE/ASTM G193 (Standard Terminology and Acronyms Relating to Corrosion).

4. Significance and Use

4.1 This practice provides a controlled corrosive environment that has been utilized to produce relative corrosion information.

4.2 The primary application of the data from this practice is to evaluate metallic materials for use in soil environments.

4.3 This practice may not duplicate all field conditions and variables such as stray currents, microbiologically influenced corrosion, non-homogeneous conditions, and long cell corrosion. The reproducibility of results in the practice is highly dependent on the type of specimen tested and the evaluation criteria selected as well as the control of the operating variables. In any testing program, sufficient replicates should be included to establish the variability of the results.

4.4 Structures and components may be made of several different metals; therefore, the practice may be used to evaluate galvanic corrosion effects in soils (see Guide G71).

4.5 Structures and components may be coated with sacrificial or noble metal coatings, which may be scratched or otherwise rendered discontinuous (for example, no coating on the edges of metal strips cut from a wide sheet). This test is useful to evaluate the effect of defective metallic coatings.

4.6 Structures and components may be coated or jacketed with organic materials (for example, paints and plastics), and these coatings and jackets may be rendered discontinuous. The test is useful to evaluate the effect of defective or incompletely covering coatings and jackets.

4.7 The corrosivity of soils strongly depends on soluble salt content (related parameters are <u>chemistry and</u> soil resistivity, see Test <u>Method Methods</u> G57; and <u>G187chemistry</u>);), acidity or alkalinity (measured by soil pH, see Test Method G51), and oxygen content (loose, for example, sand, or compact, for example, clay, soils are extreme examples). <u>examples</u>, see Test Method <u>G200 – oxidation-reduction potential</u>). The manufacturer, supplier, or user, or combination thereof, should establish the nature of the expected soil environment(s) and select the test environment(s) accordingly. Multiple types of soil can be used to determine the effect of this variable.

5. Test Apparatus and Conditions

5.1 *Container*—The container for the soil shall be made from a material that is not affected by the soil environment and that does not affect the soil. Container materials, such as glass, plastic, or corrosion-resistant metal or alloy, can be used; however, electrically conductive containers must be electrochemically isolated from the specimens. The size of the container is determined by the volume of soil required for the test. A minimum of 40 cm³ should be used for each 1 cm² of exposed metal surface area (see Fig. 1).

5.2 *Soil Environment*—The container is filled with a soil sample of choice. A soil sample from a specific outdoor location may be retrieved for the test, or a soil sample may be prepared with a specific property and chemistry. If necessary, physical and chemical characteristics of the soil may be determined.

5.2.1 A field soil sample may be utilized for purposes of conducting a soil corrosion test in a specific environment.

5.2.2 Laboratory soil samples may be prepared by using washed sand, (that is, No. 2 silica sand) clean clay (that is, bentonite) or other uniform known media.

5.2.3 *Soil Chemistry*—The field soil sample and the laboratory soil sample are saturated with a known electrolyte chosen for the test. Typically, the electrolyte is added to the soil of choice in the container. A typical electrolyte for use with washed sand is ASTM corrosive water (see Test Method D2570). With field soil samples, deionized or distilled water (see Test Method purified

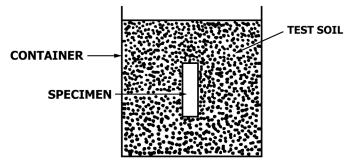


FIG. 1 Apparatus for Conducting Laboratory Corrosion Tests in Soils