



Designation: ~~D7649 – 10 (Reapproved 2017)~~ D7649 – 19

Standard Test Method for Determination of Trace Carbon Dioxide, Argon, Nitrogen, Oxygen and Water in Hydrogen Fuel by Jet Pulse Injection and Gas Chromatography/Mass Spectrometer Analysis¹

This standard is issued under the fixed designation D7649; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method describes a procedure primarily for the determination of carbon dioxide, argon, nitrogen, oxygen, and water in high pressure fuel cell grade hydrogen by gas chromatograph/mass spectrometer (GC/MS) with injection of sample at the same pressure as sample without pressure reduction, which is called “~~Jet Pulse Injection~~”-Injection.” The procedures described in this method were designed to measure carbon dioxide at ~~0.5 micromole~~ 0.5 micromole per mole (ppmv), Argon 1 ppmv, nitrogen 5 ppmv and ppmv, oxygen 2 ppmv, and water 4 ppmv.

1.2 Units—The values stated in SI units are to be regarded as standard. The values stated in inch-pound given in parentheses after SI units are provided for information only and are not considered standard.

1.3 The mention of trade names in standard does not constitute endorsement or recommendation for use. Other manufacturers of equipment or equipment models can be used.

~~1.4 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.~~

1.4 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.

1.5 This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.

2. Referenced Documents

2.1 ASTM Standards:²

[E177 Practice for Use of the Terms Precision and Bias in ASTM Test Methods](#)

[E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method](#)

2.2 Other Standards:³

[SAE TIR J2719 Information Report on the Development of a Hydrogen Quality Guideline for Fuel Cell Vehicles April 2008](#)

3. Terminology

3.1 Definitions of Terms Specific to This Standard:

3.1.1 ~~absolute pressure~~—pressure, n—pressure measured with reference to absolute zero pressure, usually expressed as kPa, mm Hg, bar or psi. ~~All the pressures mentioned in this method are absolute pressure.~~

3.1.1.1 Discussion—

¹ This test method is under the jurisdiction of ASTM Committee D03 on Gaseous Fuels and is the direct responsibility of Subcommittee D03.14 on Hydrogen and Fuel Cells.

Current edition approved April 1, 2017 Dec. 1, 2019, Published April 2017 February 2020. Originally approved in 2010. Last previous edition approved in 2010 as D7649D7649 – 10-10 (2017). DOI: 10.1520/D7649-10R17-10.1520/D7649-19.

² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from SAE International (SAE), 400 Commonwealth Dr., Warrendale, PA 15096-0001, <http://aerospace.sae.org>, 15096, <http://www.sae.org>.

All the pressures mentioned in this method are absolute pressure.

3.1.2 *constituent*—*constituent, n*—A component (or compound) found within a hydrogen fuel mixture.

3.1.3 *contaminant*—*contaminant, n*—impurity that adversely affects the components within the fuel cell system or the hydrogen storage system by reacting with its components. An adverse effect can be reversible or irreversible.

3.1.3.1 *Discussion*—

An adverse effect can be reversible or irreversible.

3.1.4 *dynamic calibration*—*calibration, n*—calibration of an analytical system using calibration gas standard generated by diluting known concentration compressed gas standards with hydrogen, as used in this method for carbon dioxide, argon, nitrogen, and oxygen (7.3 and 7.4).

3.1.5 *extracted ion chromatogram (EIC)*—*(EIC), n*—a GC/MS chromatogram where a selected ion is plotted to determine the compound(s) of interest.

3.1.6 *fuel cell grade hydrogen*—*hydrogen, n*—hydrogen satisfying the specifications in SAE TIR J2719.

3.1.7 *hydrogen fuel*—*fuel, n*—hydrogen to be tested without compositional change due to sample introduction, etc.

3.1.8 *jet pulse injection*—*injection, n*—high pressure hydrogen fuel sample is introduced instantaneously at the same pressure into GC/MS.

3.1.9 *relative humidity*—*humidity, n*—ratio of actual pressure of existing water vapor to maximum possible pressure of water vapor in the atmosphere at the same temperature, expressed as a percentage.

3.1.10 *response factor (RF)*—*(RF), n*—the amount in volume (μL) of an analyte divided by the EIC area of the analyte.

3.1.11 *static calibration*—*calibration, n*—calibration of an analytical system using standards in a matrix, state, or manner different than the samples to be analyzed, as used in this method for water concentration in hydrogen.

3.2 Acronyms:

3.2.1 *FCV*—fuel cell vehicle.

3.2.2 *PEMFC*—proton exchange membrane fuel cell.

4. Summary of Test Method

4.1 The simultaneous analysis of carbon dioxide, argon, nitrogen, oxygen, and water at 0.5 – 5 ppmv (micromole per mole) in hydrogen fuel samples from fueling stations is challenging due to high hydrogen fuel sample pressure and possible contaminations from ambient air.

4.2 In this method, a small stainless steel loop is initially pressurized with high pressure hydrogen standard or sample without any pressure regulation or restriction (“Sample Loop Pressurization”, Fig. 1). The hydrogen in the loop is then released entirely

Figure 1 Sample Loop Pressurization

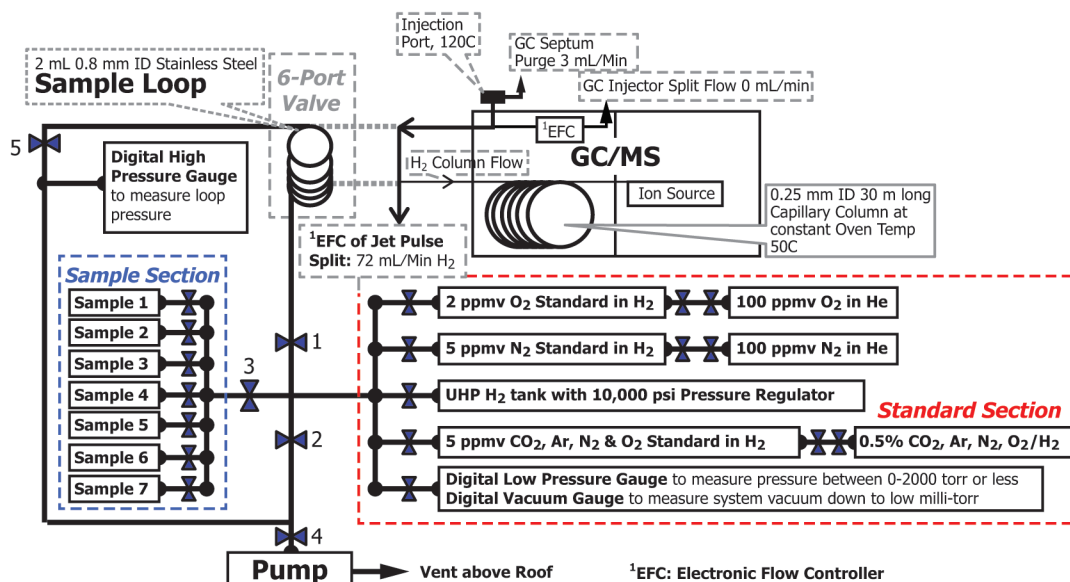


FIG. 1 Sample Loop Pressurization

as a “jet pulse” into a T-union which splits sample into a 0.25 μm ID 30 m long capillary column and an electronic flow controller (EFC) used to vent excess hydrogen to the atmosphere (“Jet Pulse Injection”, Fig. 2). Less than 1% of hydrogen enters the capillary column with the remaining sample venting to atmosphere through EFC. As demonstrated in Appendix X1, the hydrogen volume “jet pulse injected” into the capillary column is a constant volume and independent of the sample loop pressure when the sample loop pressure is over 90 psi. Therefore, the constant hydrogen volume from standards or samples is GC/MS analyzed regardless of standard or sample pressures.

4.3 Jet pulse injected volume into the capillary column is approximately 100 μL (In Appendix X1, this volume is calculated to be $\frac{200(200 \text{ psi} - 14.7 \text{ psi})}{14.7 \text{ psi}} \times 2 \text{ mL}$ or 27 mL. Hence, 99.5% of the hydrogen sample vents to atmosphere. This type of “Jet Pulse Injection” has been found acceptable for the analysis of high pressure hydrogen fuel sample since the hydrogen volume injected is independent of the pressures of hydrogen standards or samples. Consequently, it is unnecessary to regulate standards and hydrogen samples to the same pressure. In addition to possible trace leaks or air trapped inside, regulators are not recommended as moisture on the regulator surface can be released into the sample resulting in a high moisture determination.

4.4 A mass spectrometer provides sensitive and selective detection towards carbon dioxide, argon, nitrogen, oxygen, and water.

5. Significance and Use

5.1 Low operating temperature fuel cells such as proton exchange membrane fuel cells (PEMFCs) require high purity hydrogen for maximum performance. The following are the reported effects (SAE TIR J2719) of the compounds determined by this test method.

5.2 Carbon Dioxide (CO₂), acts largely as a diluent, however, in the fuel cell environment, CO₂ can be transformed into CO.

5.3 Water (H₂O), is an inert impurity, as it does not affect the function of a fuel cell stack; however, it provides a transport mechanism for water-soluble contaminants, such as Na⁺ or K⁺. In addition, it may form ice on valve internal surface at cold weather or react exothermally with metal hydride used as hydrogen fuel storage.

5.4 Inert Gases (N₂ and Ar), do not normally react with a fuel cell components or fuel cell system and are considered diluents. Diluents can decrease fuel cell stack performance.

5.5 Oxygen (O₂), in low concentrations is considered an inert impurity, as it does not adversely affect the function of a fuel cell stack; however, it is a safety concern for vehicle on board fuel storage as it can react violently with hydrogen to generate water and heat.

6. Apparatus

6.1 Mass Spectrometer (MS)—The MS can perform mass calibration with a scanning range from m/e 15 to 650. The background peak intensities of water, nitrogen, argon, oxygen, and carbon dioxide in the mass spectrum of FC-43 (perfluorotributylamine),

Figure 2 Jet Pulse Injection

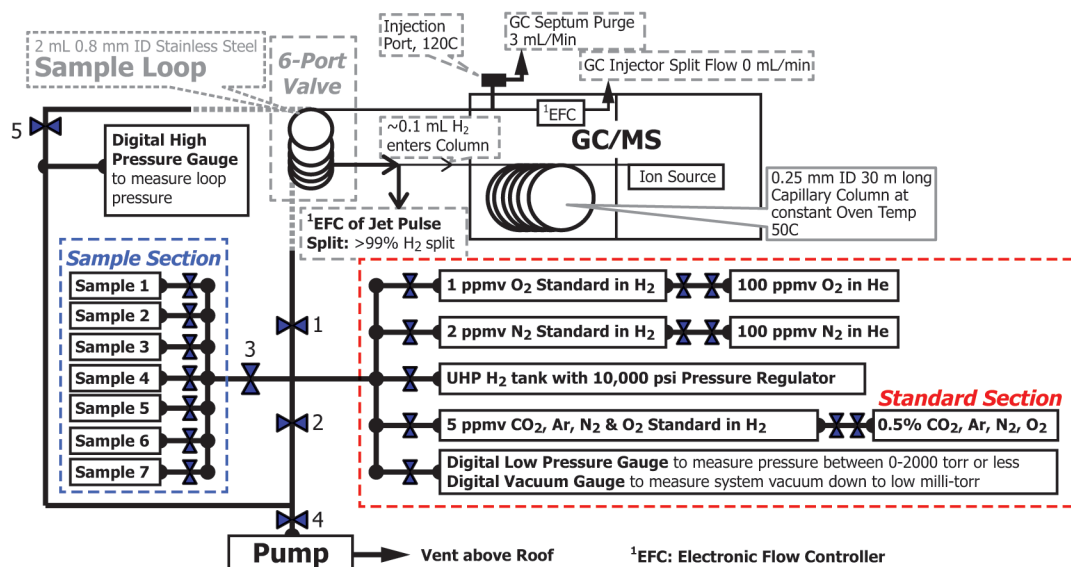


FIG. 2 Jet Pulsed Pulse Injection

used for mass calibration, should be less than 10% of m/e 69 to demonstrate a background acceptable for the determination of these analytes before beginning sample analysis. All analytes determined according to this method have a molecular mass less than 44 amu; therefore, the mass scanning range of m/e 15 to 50 is typically used.

6.2 *Data System*—A computer or other data recorder loaded with appropriate software for data acquisition, data reduction, and data reporting and possessing the following capabilities is required:

- 6.2.1 Graphic presentation of the total ion chromatogram (TIC) and extracted ion chromatogram (EIC).
- 6.2.2 Digital display of chromatographic peak areas.
- 6.2.3 Identification of peaks by retention time and mass spectra.
- 6.2.4 Calculation and use of response factors.
- 6.2.5 External standard calculation and data presentation.

6.3 *Gas chromatography*—Chromatographic system capable of obtaining retention time repeatability of 0.05 min throughout the analysis.

6.3.1 *Interface with MS*—A heated interface connecting the GC column to the MS ion source.

6.3.2 *GC Column*—A 0.25 mm ID 30 m 0.25 μm film thickness DB-5 column has been successfully used to perform this analysis. Other capillary columns may be used provided chromatographic peaks do not significantly tail. One end of the GC column is connected to the Jet Pulse Splitjet pulse split (6.4.5) and the other end is connected to the ion source inlet of a mass spectrometer.

6.3.3 *Carrier Gas*—Ultra high purity hydrogen is used as carrier gas. Use of helium carrier gas results in unacceptable broadening of the water chromatographic peak. An example of water peaks is shown in Fig. 3.

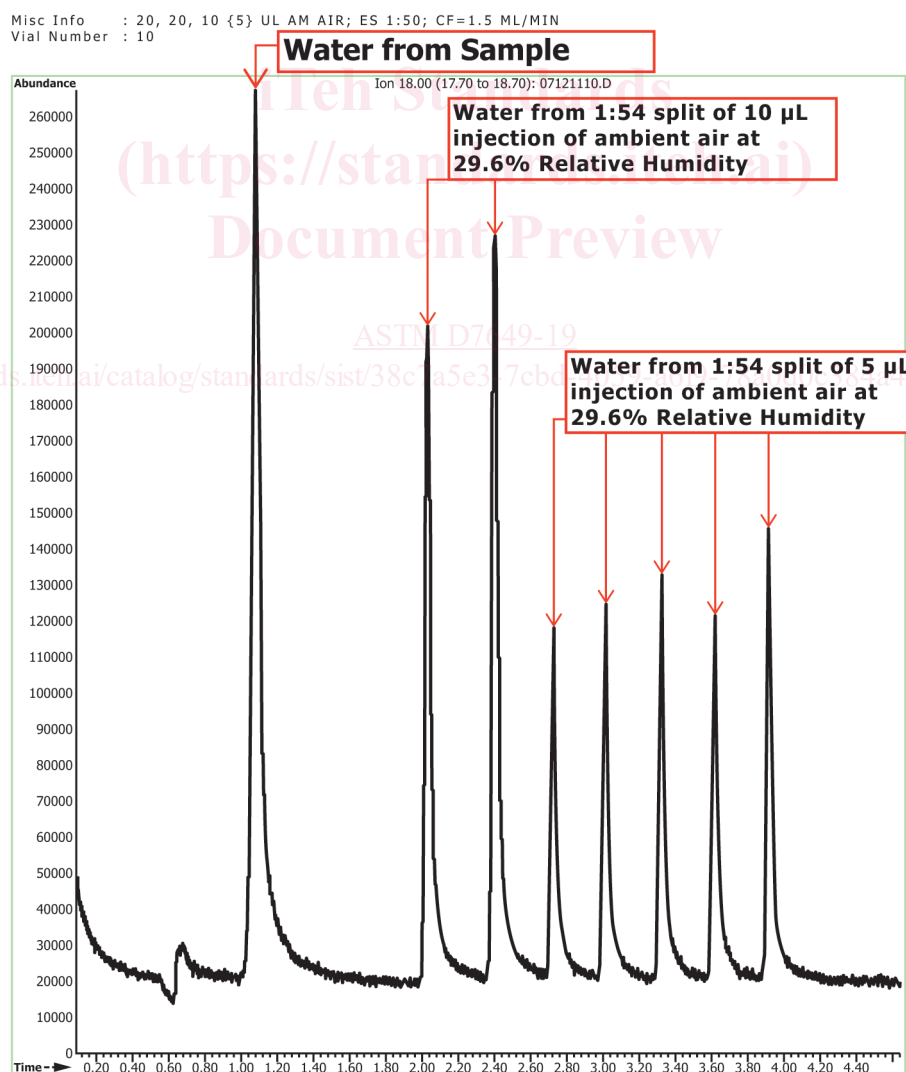


FIG. 3 m/e 18 Extracted Ion Chromatogram of Sample Analysis with Co-Injection of Ambient Air

6.3.4 *GC Injector*—An injector port with a glass insert and a septum is connected through a 1/16 in. OD stainless steel tubing to a jet pulse split (6.4.5) in the inlet system (6.4). The injector temperature is set to at ~~220°C~~ 220 °C to ensure that all water vapor in injected ambient air ~~are is~~ not condensed in the injector. The GC column and total split flow rate are electronically set at 1.5 and 75 mL/min, respectively. The GC total split flow includes a GC septum purge flow of ~~3 mL/min~~ 3 mL/min (Fig. 1 and Fig. 2) and GC injector split flow of ~~72 mL/min~~ 72 mL/min.

6.4 *Inlet System*—A system introduces high pressure samples or standards into GC/MS for analysis. The sample or standard enter the inlet system through “Sample Loop Pressurization” (Fig. 1) and then leave the inlet system to GC/MS through “Jet Pulse Injection” (Fig. 2). While the inlet system is in “Sample Loop Pressurization”, ~~Pressurization,~~ the sample loop (6.4.4) is pressurized directly with hydrogen samples or calibration standards without pressure regulation or flow restriction. Afterwards, a six-port sample valve (6.4.1) switches the inlet system to “Jet Pulse Injection”, ~~Injection,~~ in which pressurized hydrogen in the sample loop is released instantaneously onto the GC column (6.3.2) and jet pulse split (6.4.5). Since the sample pressure is high, all parts of the inlet system must be capable of working at pressures of 1500 psi or higher.

6.4.1 *Six Ports Valve*—This valve is used to switch from “Sample Loop Pressurization” (Fig. 1) to “Jet Pulse Injection” (Fig. 2).

6.4.2 *Samples and Calibration Standards*—All calibration standards and samples are prepared or collected in 1800 psi pressure rated containers with a DOT 3A1800 label (United States Department of Transportation mandated label) affixed to the outside surface. All calibration standards and samples are connected to the inlet system before beginning an analytic sequence to minimize the potential for air or moisture contamination due to addition or replacement of standard or sample containers.

6.4.3 *Vacuum Pump*—~~Pump,~~ an oil vacuum pump that can pump down to 50 mtorr or less.

6.4.4 *Sample Loop*—~~Loop,~~ stainless steel tubing with ~~1/46~~ 1/16 in. OD and 2 mL inside volume. Both ends of the sample loop are connected to a six port valve (6.4.1).

6.4.5 *Jet Pulse Split*—~~a~~ T-union connects the following three portions.

6.4.5.1 Six port valve (6.4.1),

6.4.5.2 Inlet of GC column (6.4.2), and

6.4.5.3 Inlet of an electronic flow controller (EFC) with its outlet to ambient air. The flow rate of this EFC is always electronically set at ~~150 mL/min~~ 150 mL/min to vent most of the GC injector split flow (~~72 mL/min~~) (72 mL/min) during “Sample Loop Pressurization” (Fig. 1) and released hydrogen from pressurized sample loop in “Jet Pulse Injection” (Fig. 2).

6.4.6 *Digital Vacuum Gauge*—~~Gauge,~~ capable of measuring absolute pressure at vacuum range 0 to ~~12,000~~ 12 000 milli-torr (mtorr or 10^{-3} torr). For the vacuum range from 0 to 1000 mtorr, the accuracy is ~~± 10%~~ ± 10 % or ± 10 % or ± 10 mtorr, whichever is larger.

6.4.7 *Digital Pressure Gauges*—Two types of digital pressure gauges are required. A pressure gauge 0 to 1000 psig is used to measure sample and standard final pressure. Another digital pressure gauge in the low and narrow pressure range, such as 0 to 2000 torr, is used to measure the pressure of pure gases in initial standard preparation.

6.4.8 *Pressure Regulator*—A ~~10,000~~ 10 000 psi pressure regulator is used to reduce UHP hydrogen pressure to ~~approximately 400 psi~~ approximately 400 psi for calibration standard preparation. It is also used to pressurize the inlet system during method blank analysis, and during inlet system flushing.

7. Reference Standards

7.1 Typical reference standards are listed in Fig. 1. Two standards prepared in helium containing 100 ppmv O₂ and ~~100 ppmv~~ 100 ppmv N₂, are ~~commercial~~ commercially available. Remaining standards listed in Fig. 1 are prepared as per below.

7.2 *0.5% 0.5 % CO₂, Ar, N₂, and O₂ in hydrogen*—*Hydrogen*—An evacuated ~~1-1 L~~ 1 L cylinder is connected to four pressure-regulated compressed gas cylinders containing reagent or UHP grade CO₂, Ar, N₂, and O₂. The system is evacuated to less than 500 mtorr with all the regulators opened and the main cylinder valves closed. With the system isolated from vacuum pump, the ~~1-1 L~~ 1 L cylinder valve is opened, and 100 torr of each target compound from the compressed gas cylinders is expanded into the system and ~~1-1 L~~ 1 L cylinder. The ~~1-1 L~~ 1 L cylinder is then pressured using UHP hydrogen to 390 psi, or $390/14.7 \times 760 = 2.02 \times 10^4$ torr. The concentration of each target compound is $100 \text{ torr} / (2.02 \times 10^4 \text{ torr}) = 0.5 \%$. This standard can be used as a co-injection standard (9.3.3) and further diluted to prepare a 5 ppmv standard (7.3). The UHP hydrogen used for preparation of both ~~0.5% 0.5 %~~ 0.5 % (7.2) and 5 ppmv standards (7.3) are free from CO₂, Ar, N₂, and O₂ by this test method.

7.3 *5 ppmv 5 ppmv CO₂, Ar, N₂, and O₂ in hydrogen*—*Hydrogen*:

7.3.1 Close all the valves in Fig. 1, except leave Valves 2, 4, and both valves of the cylinder labeled with “5 ppmv CO₂, Ar, N₂ & O₂ Standard in H₂” open. Evacuate the “Standard Section” in Fig. 1 to less than 100 mtorr.

7.3.2 Close Valve 2 and pressurize the “Standard Section” in Fig. 1 to above 100 psi by UHP hydrogen.

7.3.3 Open Valve 2 and pump the “Standard Section” in Fig. 1 to less than 100 mtorr and then close it.

7.3.4 Close the valve of the standard cylinder close to the ~~0.5% 0.5 %~~ 0.5 % standard and open the valve of the cylinder containing the ~~0.5% 0.5 %~~ 0.5 % CO₂, Ar, N₂, and O₂ in hydrogen for about 10 s and then close it.

7.3.5 Open the valve of the standard cylinder close to the ~~0.5% 0.5 %~~ 0.5 % standard for 5 s and close it. Measure the pressure by the digital pressure gauge in Fig. 1.

7.3.6 Pressurize the “Standard Section” in Fig. 1 to less than 400 psi with UHP H₂ with ~~10,000~~10 000 psi pressure regulator and close the other valve of the standard cylinder.

7.3.7 The concentration of the standard is calculated as following. If the pressure in 7.3.5 is 20 torr and final pressure 385 psi (7.3.6), the concentration is $(20 \text{ torr} \times (0.50/100))/(385/14.7 \times 760) = 5.0 \text{ ppmv}$.

7.3.8 Low concentration standards other than 5 ppmv can be prepared and used as calibration standards.

7.4 *Detection Limit Standards of Oxygen and Nitrogen*—~2 ppmv O₂ and ~5 ppmv N₂ in hydrogen. The preparation of these two standards from commercially available 100 ppmv O₂ and N₂ in helium, respectively, is in the same as that for ~~5 ppmv~~5 ppmv CO₂, Ar, N₂, and O₂ in hydrogen (7.3). The detection limit standards are analyzed in each analytical sequence to validate acceptable detection of oxygen and nitrogen at the detection limit.

8. Preparation of Apparatus

8.1 *GC/MS*—Place in service in accordance to the manufacturer’s instructions. Perform daily mass calibration using FC-43. As stated in 6.1, each of the peak intensities of m/e 18, 28, 29, and 32 should be less than ~~10%~~10 % of m/e 69 in the mass spectrum of FC-43 used for mass calibration. In order to achieve this condition, the GC column flow rate of GC/MS system should be set at a high flow rate, such as, ~~2 mL/min~~2 mL/min, while the system is in standby mode to remove any air in the carrier gas line. In addition, when any air may be introduced into the carrier gas system, such as when changing the hydrogen carrier gas tank, the GC total split flow rate is set at ~~100 mL/min~~100 mL/min for an hour to rapidly remove air in the carrier gas line.

9. Procedure

9.1 The detailed procedures used to perform jet pulse injection followed by GC/MS analysis are listed below. The procedures are split into two sections – sample loop pressurization (Fig. 1) and jet pulse injection/GC/MS analysis (Fig. 2).

9.2 *Sample Loop Pressurization* (Fig. 1):

9.2.1 With all the valves closed in Fig. 1, Valves 2, 3, and 4 are opened and pumped down including both standard and sample sections to less than 100 mtorr. If this pressure cannot be lowered to at least 100 mtorr, perform a leak check.

9.2.2 Open Valves 1 and 5 and close Valves 2 and 4, followed with pressurization of the sample loop with UHP hydrogen to 300 – 400 psi.

9.2.3 Open Valve 4 only to pump down the entire system to less than 10 torr and then open Valve 2 to less than 1 torr. Simultaneously, measure the flow rate from EFC of the jet pulse split (6.4.5); this should be close to 72 mL/min.

9.2.4 Close both Valves 2 and 4 and pressurize the entire system with standard or sample and measure the loop pressure using a digital high pressure gauge as depicted in Fig. 1. For safety ~~reasons~~reasons, it is recommended that the loop pressure not be over 500 psi. For the method blank analysis using UHP hydrogen, the loop should be pressurized to ~~approximate 400 psi~~approximately 400 psi.

9.3 *Jet Pulse Injection/GC/MS Analysis* (Fig. 2):

9.3.1 Switch the six port valve to “Jet Pulse Injection” (Fig. 2) and simultaneously start GC/MS acquisition.

9.3.2 Measure the GC injector split flow rate, which should be 0 mL/min since most injector split flow vents out from the jet pulse split (6.4.5). Measure the GC septum purge flow, which should be approximate 3 mL/min.

9.3.3 After 1.5 ~~minutes~~min from the start of GC/MS acquisition, the ~~0.5%~~0.5 % CO₂, Ar, N₂, and ~~0.2%~~0.2 % O₂ in hydrogen (6.2), ~~1%~~1 % CO₂ in nitrogen or ambient air is co-injected three times and 18 ~~seconds~~s apart. The volume and time interval of co-injections can be varied, for example, those in Table X1.1 of Appendix X1. The reasons for co-injection are listed under “Co-Injection” of Appendix X1. Each GC/MS analysis is completed in less than 5 ~~minutes~~min.

9.3.4 A typical analytic sequence is shown in Table 1. As shown in this table, except for the method blank, each standard or sample is analyzed consecutively three times. The three analyses of each standard or sample should generate the same extracted ion chromatogram (EIC) areas of target compounds within analytical error, as described in Appendix X1; this demonstrates that the jet pulse injected the same volume of each hydrogen standard or sample. The EIC of target compounds are ~~m/e 44~~m/e 44 for CO₂, 40 Ar, 29 N₂, 32 O₂, and 18 H₂O. In general, the variation of EIC areas of 5 ppmv CO₂, Ar, N₂, and O₂ in hydrogen (7.3) in three consecutive analyses in the analytical sequence (Table 1) ~~are~~is less than ~~10%~~10 % RSD for CO₂ and Ar. For N₂ and O₂, the %RSD should be less than ~~30%~~30 % due to the baseline noise of GC/MS analysis.

10. Calculation

10.1 The concentrations of carbon dioxide, argon, nitrogen, and oxygen in hydrogen sample can be calculated from the EIC areas of carbon dioxide at ~~m/e 44~~m/e 44, argon ~~m/e 40~~m/e 40, nitrogen ~~m/e 29~~m/e 29, and oxygen ~~m/e 32~~m/e 32 using Eq. (1) from Fig. 4.

10.2 The jet pulse injected volume described in 9.3.4 can be calculated from the RF of co-injection (~~(2)Eq from Fig. 4~~) and the EIC area of jet pulse injected 5 ppmv standard (7.2), as shown in Eq (3) from Fig. 4. The split ratio is the ratio of sum of the flow rates of the jet pulse split, GC injector split, and injector septum purge over the column flow rate. In general, it is close to 50. Since air contains ~~0.934%~~0.934 % of argon, the RF of argon can be also calculated from ambient air co-injection.