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# Standard Test Method for Column Percolation Extraction of Mine Rock by the Meteoric Water Mobility Procedure<sup>1</sup>

This standard is issued under the fixed designation E2242; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\varepsilon$ ) indicates an editorial change since the last revision or reapproval.

#### 1. Scope

1.1 This test method provides a procedure for the column percolation extraction of mine rock in order to determine the potential for dissolution and mobility of certain constituents by meteoric water.

1.2 This test method is intended to describe the procedure for performing column percolation extractions only. It does not describe all types of sampling and analytical requirements that may be associated with its application.

1.3 The values stated in SI units are to be regarded as the standard. The values given in parentheses are provided for information only and are not considered standard.

1.4 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety safety, health, and health environmental practices and determine the applicability of regulatory limitations prior to use.

<u>1.5 This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.</u>

2. Referenced Documents

2.1 ASTM Standards:<sup>2</sup>

D1193 Specification for Reagent Water

D1293 Test Methods for pH of Water

D6234 Test Method for Shake Extraction of Mining Waste by the Synthetic Precipitation Leaching Procedure

D7237 Test Method for Free Cyanide and Aquatic Free Cyanide with Flow Injection Analysis (FIA) Utilizing Gas Diffusion Separation and Amperometric Detection

E29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications

E50 Practices for Apparatus, Reagents, and Safety Considerations for Chemical Analysis of Metals, Ores, and Related Materials E135 Terminology Relating to Analytical Chemistry for Metals, Ores, and Related Materials

E389 Test Method for Particle Size or Screen Analysis at 4.75 mm (No.4) Sieve and Coarser for Metal-Bearing Ores and Related Materials

E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method

<sup>&</sup>lt;sup>1</sup> This test method is under the jurisdiction of ASTM Committee E01 on Analytical Chemistry for Metals, Ores, and Related Materials and is the direct responsibility of Subcommittee E01.02 on Ores, Concentrates, and Related Metallurgical Materials.

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<sup>&</sup>lt;sup>2</sup> For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For Annual Book of ASTM Standards volume information, refer to the standard's Document Summary page on the ASTM website.



E877 Practice for Sampling and Sample Preparation of Iron Ores and Related Materials for Determination of Chemical Composition and Physical Properties

E882 Guide for Accountability and Quality Control in the Chemical Analysis Laboratory

E1601 Practice for Conducting an Interlaboratory Study to Evaluate the Performance of an Analytical Method

2.2 Other References:

Meteoric Water Mobility Procedure, Bureau of Mining Regulation and Reclamation, Nevada Division of Environmental Protection, 9/19/90

Standard Methods for the Examination of Water and Wastewater, 18th edition, APHA/AWWA/WEF, 1992

#### 3. Terminology

3.1 Definitions:

- 3.1.1 For definitions of terms used in this test method, see Terminology E135.
- 3.1.2 *meteoric <u>water-water, n-</u>*rainfall that has the potential to contact materials and mobilize soluble constituents from mine rock at a mining or mineral processing site.

3.1.3 *mine rock—<u>rock</u>, n*—ore, waste rock or overburden excavated in order to recover metals or minerals during mining operations or coarse processed ore such as heap leach spoils.

#### 4. Summary of Test Method

4.1 The test material is placed in a plastic column and contacted with water to produce an effluent containing an equal weight of solution to dry solids during a leaching period of up to 48 h. The extract is collected and preserved for analysis of inorganic constituents.

#### 5. Significance and Use

5.1 This test method is intended as a means for obtaining an extract from mine rock samples. The extract may be used to estimate the final pH and release of certain constituents of the test sample under the laboratory conditions described in this test method.

5.2 The pH of the extraction fluid used in this test method should reflect the pH of precipitation in the geographic region in which the mine rock is being evaluated. alog standards/sist/d5e5d134-c2b4-4103-bd35-be420eff792c/astm-e2242-21

5.3 This test method is designed to mobilize potential contaminants present in the solids, so that the resulting extract can be used to assess leachate that could potentially be produced from mine rock in the field.

5.4 This test method has not been demonstrated to simulate actual site leaching conditions.

5.5 This test method produces extracts that are amenable to the determination of both major and minor (trace) constituents. When minor constituents are being determined, it is especially important that precautions be taken in sample preservation, storage, and handling to avoid possible contamination of the extracts.

5.6 This test method is a comparative method intended for use as a routine method for monitoring mine rock. It is assumed that all who use this method will be trained analysts capable of performing skillfully and safely. It is expected that work will be performed in a properly equipped laboratory under appropriate quality control practices such as those described in Guide E882.

#### 6. Interferences

6.1 This test method may not be suitable for obtaining extracts from finely divided solids (such as: finely crushed drill cuttings, clays, sludges, mill tailings, etc.), due to difficulty in obtaining a representative extract solution. If it takes more than 48 h for the column to produce an extract which has a mass equivalent to 70 % of the dry test sample weight, use of an alternate extraction procedure may be required, such as Test Method D6234.

NOTE 1—See Appendix X1 for the alternative bottle roll extraction option for fine-grained solids.

## 7. Apparatus

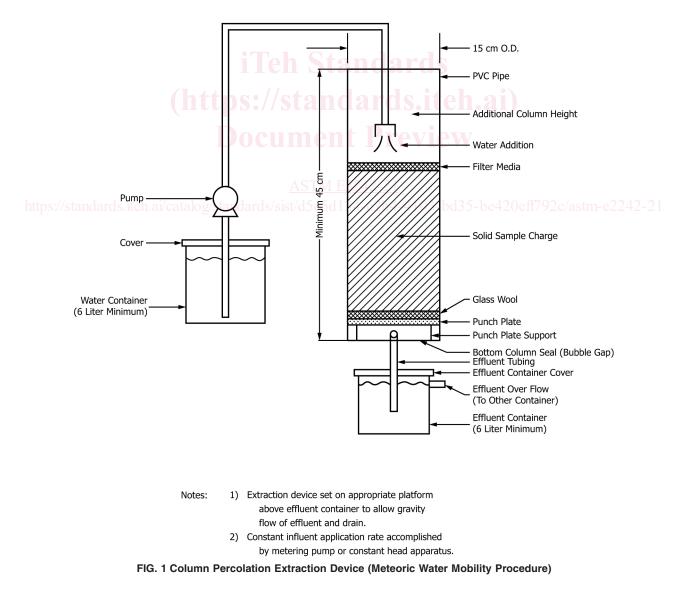
7.1 Drying Pans or Dishes, for moisture content determinations, 1 kg capacity.

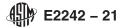
7.2 *Extraction Column*, PVC column of 15-cm(6-in.)15 cm(6 in.) O.D. of sufficient height to contain a minimum of 5 kg of test sample with a feed particle size that just passes a 5-cm(2-in.)5 cm(2 in.) sieve and sufficient additional height to contain applied water volume should poor percolation occur. For a 5-kg5 kg test sample, a 15-cm15 cm O.D. × 45-cm45 cm high column is recommended. Additional column height will be required for test sample quantities greater than 5 kg. The bottom of the column must be sealed and a solution discharge outlet situated above the sealed bottom of the column and below a perforated support plate. A drawing of a suitable extraction column is shown in Fig. 1.

7.3 *Filtration Device*, of a composition suitable to the nature of the analyses to be performed and equipped with a  $0.45 \mu m$  pore size filter. An assembly for pre-filtration or a centrifuge may be required if  $0.45 \mu m$  filtration is difficult.

7.4 *Filter Media*, glass wool (inert) or a plastic fiber pad is placed onto the support plate before loading the laboratory sample into the column, to minimize fines migration, and onto the top of the test sample after column loading, to aid in making an even water distribution.

7.5 Laboratory Balance, capable of weighing to 1.0 g.





7.6 *pH Meter*, with a readability of 0.01 units and an accuracy of at least  $\pm$  0.05 units at 25 °C.

7.7 *Tubing*, surgical or synthetic tubing sufficient in diameter and length for the extraction column assembly (pump, column outlet).

7.8 *Water and Extract Containers*, sufficient in size to contain the water added during extraction. Containers must be covered to avoid contamination.

7.9 *Water Metering Device*, a metering pump or constant head device to insure constant rate extraction fluid application during column percolation.

### 8. Reagents and Materials

8.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.<sup>3</sup> Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

8.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean Type II reagent water at 18 °C to 27 °C conforming to Specification D1193. Type III or IV may be used if they effect no measurable change in the blank or sample.

#### 9. Hazards

9.1 For precautions to be observed in the use of analytical methods associated with this standard, refer to Practices E50.

# 10. Sampling and Sample Preparation

10.1 The amount of gross sample to be sent to the laboratory should he sufficient to perform the moisture content determination as specified in 10.3, and to provide at least 5 kg of test sample on a dry weight basis for extraction. Gross sample weights in the range of 7 kg to 25 kg are appropriate.

#### ASTM E2242-21

NOTE 2—Information on obtaining representative samples can also be found in Pierre Gy's Sampling Theory and Sampling Practice<sup>4</sup> and in Practice E877.

10.2 In order to prevent sample contamination or constituent loss prior to extraction, keep the samples in closed containers appropriate to sample type and desired analysis.

10.3 *Moisture Determination*—Remove the gross sample from the container and blend by coning or rolling to obtain a sample for feed moisture content with a minimum of 500-g500 g dry weight.

10.3.1 Weigh the moisture test sample and dry to constant weightmass ( $\pm$  0.05 %) at 105 °C  $\pm$  2 °C and record the weightmass.

10.3.2 Calculate the moisture content of the test sample as follows:

$$M = \frac{100(B-A)}{B} \tag{1}$$

where:

- A = mass of sample after drying, g,
- B = wet mass of sample, g, and

M =moisture content, %.

<sup>&</sup>lt;sup>3</sup> Reagent Chemicals, American Chemical Society Specifications, American Chemical Society, Washington, DC.-DC, www.chemistry.org. For suggestions on the testing of reagents not listed by the American Chemical Society, see the United States Pharmacopeia and National Formulary, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD:MD, http://www.usp.

<sup>&</sup>lt;sup>4</sup> Pitard, F., *Pierre Gy's Sampling Theory and Sampling Practice*, Vols. I and II, CRC Press, 1989.

🕼 E2242 – 21

10.4 Separate the remaining laboratory sample on a 5-cm (2-in.)5 cm (2 in.) sieve. Save the minus 5-cm5 cm fraction for recombination with the crushed plus 5-cm5 cm fraction. The screening time or end point is when additional periods of shaking fail to change the results by more than 0.3 % (Test Method E389). For highly friable material an acceptable end point shall be determined experimentally.

10.5 Weigh the plus and minus 5-cm5 cm sieve fractions, calculate and record the weightmass percent retained on the 5-cm5 cm sieve as follows:

$$R = \frac{100C}{(C+D)} \tag{2}$$

where:

- R = weight retained on the 5-cm sieve, %,
- C = mass retained on the 5-cm sieve, g, and
- D = mass passing the 5-cm sieve, g.
- $\underline{R} \equiv \underline{\text{mass retained on the 5 cm sieve, }}\%$
- $\overline{C} \equiv \overline{\text{mass retained on the 5 cm sieve, g, and}}$
- $\overline{D}$  = mass passing the 5 cm sieve, g.

10.6 Crush or hand break the materials retained on the  $\frac{5-\text{cm}5 \text{ cm}}{5-\text{cm}5 \text{ cm}}$  sieve so that they just pass through the  $\frac{5-\text{cm}(2-\text{in.})5 \text{ cm}(2-\text{in.})5}{5-\text{cm}5 \text{ cm}}$  sieve and mix the prepared sample.

10.7 Thoroughly blend the prepared sample and calculate the minimum test sample weight<u>mass</u> containing  $\frac{5-\text{kg5} \text{kg}}{\text{weight}}$  dry weight, mass, based on the feed moisture content from 10.3, as follows:



where:

S = minimum sample mass for testing, g and

M =moisture content, %.

10.8 Divide the prepared sample, if necessary, to obtain a test sample suitable for processing in the column apparatus but not less than the minimum sample weightmass from 10.7. and weigh to  $\pm 1$  g. Label and reserve any excess prepared sample for any additional testing which may be required.

10.9 Calculate the dry weightmass of the test sample as follows:

$$L = \frac{E(100 - M)}{100}$$

(4)

where:

L = dry mass of the test sample, g,

E = gross mass of the test sample, g, and

M = moisture content of the gross sample, %.

#### 11. Procedure

11.1 Place filter media (7.4) in the bottom of the extraction column and load the laboratory sample incrementally (-1 kg/increment)(-1 kg/increment) into the column. To minimize particle segregation and compaction during column loading, the sample portions shall be dropped from no more than 0.6 m (24 in.) when introduced from the top of the column, and no tamping, shaking, or other methods to compact the sample will be employed. Place filter media over the test sample in the column.

11.2 Use a water addition rate of 3.5 mL/min for minimum weightmass samples in the range of 5-kg5 kg to 5.5-kg5.5 kg dry weight.mass.

11.2.1 For samples greater than  $\frac{5.5 \text{ kg}}{5.5 \text{ kg}}$  dry weight, mass, calculate the rate of water addition to the column to add a weight mass of water equal to the dry weight mass of the test sample during 24 h as follows:

🖗 E2242 – 21

$$F = \frac{L}{1440} \tag{5}$$

where:

F = water addition rate, mL/min, and

L = dry mass of the laboratory sample, g.

11.3 Measure and record the initial temperature and pH of the leaching water in accordance with Test Methods D1293. Record the time and date that water addition is started.

11.4 Add water to the column at the rate specified in 11.2 until the weightmass of the effluent is equal to the dry mass of the laboratory sample. If the water will not freely percolate through the column, abort the test and repeat the evaluation using an alternate testing procedure, such as described in X1.2 or Test Method D6234.

11.5 Discontinue water addition and effluent collection when the effluent  $\frac{\text{weight} \text{mass}}{\text{weight} \text{mass}}$  is equal to the dry  $\frac{\text{weight} \text{mass}}{\text{weight} \text{mass}}$  of the test sample or 48 h has elapsed, whichever occurs first. If, after 48 h, the effluent  $\frac{\text{weight} \text{mass}}{\text{weight} \text{mass}}$  is not at least 70 % of the dry  $\frac{\text{weight} \text{mass}}{\text{weight} \text{mass}}$  of the test sample, abort the test and select an alternate testing method for the sample, such as described in X1.3 or Test Method D6234.

11.6 Thoroughly mix the effluent and let the column effluent continue to drain into a waste container.

11.7 Immediately measure and record the final pH of the effluent in accordance with Test Methods D1293. Record the date, time, and final effluent weight.mass.

11.8 *Extract*—Quantitatively transfer a 2-L2L portion of the effluent through a large glass funnel to a filtration device equipped with a  $0.45 + \mu m 0.45 \mu m$  filter for analyses of dissolved constituents. Pre-filtration using a coarse porosity glass filter and centrifuge separation procedures may be necessary prior to filtration for some difficult to filter effluents. Immediately measure and record the final pH of the extract in accordance with Test Methods D1293. Record the final effluent and extract weights.masses.

11.8.1 Retain a 250-mL250 mL portion of the extract for metals analysis with its pH adjusted to less than 2 with nitric acid.

11.8.2 Retain a 250-mL250 mL portion of the extract for nutrients analysis with its pH adjusted to less than 2 with sulfuric acid.

11.8.2.1 If the test material is a cyanidation residue or contains cyanate, perform nutrient analysis on the refrigerated extract from 11.8.4, since acidification of any cyanate may result in conversion to ammonia-nitrogen.

11.8.3 Retain a  $\frac{1-L}{L}$  portion of the extract for cyanide analysis with its pH adjusted to between 12 and 12.5 with sodium hydroxide. Store in a dark bottle at 4 °C ± 2 °C, if the cyanide analysis is not performed immediately.

11.8.3.1 Preserve a free cyanide analysis portion to pH 11 in accordance with Test Method D7237 if free cyanide is to be determined on the extract and analysis is not to be performed immediately.

11.8.4 Refrigerate the remaining extract at 4 °C  $\pm$  2 °C for mineral analyses.

11.9 Analyze the unfiltered effluent for specific constituents or properties, or use for biological testing procedures as necessary.

11.10 Analyses should be performed using appropriate ASTM test methods. Where no appropriate ASTM test methods exist, other test methods may be used, such as Standard Methods for the Examination of Water and Wastewater, and record the methods used in the report. Whether visible phase separation during storage of the extract or effluent occurs or not, appropriate mixing should be used to ensure the homogeneity of the extract and effluent prior to their use in such analyses or testing.



#### **TABLE 1 Statistical Information—Test Variables**

Material	Number of Labs	Variable	Mean	Measurement Units
Mine Waste Rock	8	Time	27.5	Hours
Heap Leach Spoils	8	Time	26.7	Hours
Mine Waste Rock	8	Temperature	21.7	Degrees C
Heap Leach Spoils	8	Temperature	21.8	Degrees C
Mine Waste Rock	8	Volume	4.99	Litres
Heap Leach Spoils	8	Volume	5.05	Litres

11.11 Allow the remaining column effluent to drain until at least two minutes elapses between drops of effluent from the test column. There should be no apparent free water phase in the column at the end of the draining period. Record any visual changes in the column residue, extract or effluent which occurred during the test period.

11.12 Remove the test residue from the column and blend by coning or rolling and obtain a sample of at least 500 g for residual moisture content. Determine the residual moisture as in 10.3 and record the results. Reserve the residue for any additional testing required.

#### 12. Report

12.1 Rounding of test results obtained using this test method shall be performed as directed in Practice E29, Rounding Method, unless an alternative rounding method is specified by the customer or applicable material specification.



12.2.1 Source of the gross sample, date of sampling, methods of sampling and sample preservation, storage conditions, handling procedures, and length of time between sample collection and extraction;

12.2.2 Moisture content (10.3), time and temperature used in the determination of the moisture content and results from any analyses performed on other splits of the laboratory sample;

12.2.3 Description of the gross sample, including its physical characteristics and sieve analysis results (10.5);

- 12.2.4 WeightMass of the test sample on a wet and dry basis;
  - 12.2.5 The pH of the water used for the extraction;
- 12.2.6 Water application rate, temperature, extraction time, effluent pH<sub>2</sub> and the final effluent weight; mass;
  - 12.2.7 Filter pore size used and filter composition;
  - 12.2.8 Use of a centrifuge or pre-filter, the pre-filter pore size and composition;
  - 12.2.9 Observations of changes in the test material or leaching solution (11.11);
- 12.2.10 The final weightmass and pH of the extract, and the results of specific analyses;

12.2.11 Moisture content of the residue, time and temperature used in the determination of the moisture content, and results of any analyses performed on the residue; and

12.2.12 Dates on which extraction was started and completed, preservation used for extract portions, and dates of analyses. A detailed laboratory worksheet that may be helpful in performing the test method is shown in Fig. 2.