



Designation: **E2313–21a** **E2313 – 22**

Standard Test Method for Aldehydes in Mono-, Di-, and Triethylene Glycol (using Spectrophotometry)¹

This standard is issued under the fixed designation E2313; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope*

1.1 This test method describes the spectrophotometric determination of total aldehyde content in mono-, di-, and triethylene glycol (MEG, DEG, and TEG). Alkoxyalcohols (hemiacetals), if present, are co-determined, whereas dialkoxyalkanes (acetals), if present, are not. The results provide a measure of the purity of the sample with respect to total aldehyde content.

1.2 This test method is applicable to samples with total aldehyde concentration (as acetaldehyde) to 50 mg/kg or 35 mg/kg (as formaldehyde). The limit of detection (LOD) is 0.4 mg/kg and the limit of quantitation is 1.2 mg/kg.

NOTE 1—LOD and LOQ were calculated using the lowest level sample in the ILS.

1.3 The following applies for the purposes of determining the conformance of the test results using this test method to applicable specifications, results shall be rounded off in accordance with the rounding-off method of Practice E29.

1.4 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.5 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.* For specific hazard statements, see Section 8 and 7.1.1 – 7.1.6.

1.6 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

2. Referenced Documents

2.1 ASTM Standards:²

D1193 Specification for Reagent Water

D2194 Test Method for Concentration of Formaldehyde Solutions

D6809 Guide for Quality Control and Quality Assurance Procedures for Aromatic Hydrocarbons and Related Materials

E29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications

¹ This test method is under the jurisdiction of ASTM Committee D16 on Aromatic, Industrial, Specialty and Related Chemicals and is the direct responsibility of Subcommittee D16.14 on Alcohols & Glycols.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For Annual Book of ASTM Standards volume information, refer to the standard's Document Summary page on the ASTM website.

*A Summary of Changes section appears at the end of this standard

E180 Practice for Determining the Precision of ASTM Methods for Analysis and Testing of Industrial and Specialty Chemicals
(Withdrawn 2009)³

E275 Practice for Describing and Measuring Performance of Ultraviolet and Visible Spectrophotometers

E288 Specification for Laboratory Glass Volumetric Flasks

E300 Practice for Sampling Industrial Chemicals

E969 Specification for Glass Volumetric (Transfer) Pipets

E1547 Terminology Relating to Industrial and Specialty Chemicals

2.2 *Other Document:*

Manufacturer's Instruction Manual of Spectrophotometer

3. Terminology

3.1 See Terminology **E1547** for definitions of terms used in this test method.

4. Summary of Test Method

4.1 The total aldehyde content of the sample is determined spectrophotometrically. The aldehydes present are reacted with excess 3-methyl-2-benzothiazolinone hydrazone (MBTH) to form azines. An aqueous solution of iron (III) chloride is added, causing the oxidation of the excess MBTH to a reactive cation, which reacts with the previously formed azines producing bluish-green cations. Acetone or methanol is added to stop the oxidation reaction, to mask the excess MBTH and to obtain a clear solution. The absorbance of the test solution is measured spectrophotometrically at 635 nm. The amount of aldehyde present in the test solution is derived from a previously prepared calibration graph. The aldehyde content of the sample, expressed as either acetaldehyde or formaldehyde, is calculated from the data obtained.

5. Significance and Use

5.1 Knowledge of the total aldehyde content of MEG is required to establish whether the product meets the requirements of its quality specifications.

6. Apparatus

6.1 *Spectrophotometer*, suitable for measurement at a wavelength of 635 nm, provided with cells with a pathlength of 10 mm.

NOTE 2—Further information on spectrophotometers can be obtained from Practice **E275**.

6.2 *Analytical Balance*, readable to 0.1 mg, calibrated. Recalibrate or verify at regular intervals.

6.3 *Stopwatch*, measuring accuracy ± 1 s or better. Verify at regular intervals.

6.4 *Buret*, calibrated, 100 mL, with a 50- or 75-mL reservoir on top of a lower portion calibrated in 0.1-mL divisions. A PTFE-fluorocarbon resin stopcock is suitable for this purpose. Recalibrate or verify at regular intervals.

6.5 *Adjustable Volume Dispensing Pipette(s)*, capable of dispensing volumes in the range 1.00 to 5.00 mL, with suitable disposable pipette tips. It is recommended to verify the capacity of the pipette(s) at regular intervals.

6.6 *Glassware:*

NOTE 3—In cases where volumetric glassware is to be used, it shall conform to Grade (or Class) A specifications as described in, for example, Specification **E288** for volumetric flasks, or Specification **E969** for one-mark pipettes.

7. Reagents

7.1 *Purity of Reagents*—Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the

³ The last approved version of this historical standard is referenced on www.astm.org.

committee on Analytical Reagents of the American Chemical Society where such specifications are available.⁴ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.1.1 *Acetone (2-propanone)*, A.R. minimum purity 99.99 % (V/V). (**Warning**—Acetone is highly flammable. Keep container in a well-ventilated place. Keep away from sources of ignition. Do not breathe vapor. Take precautionary measures against static discharges.)

7.1.2 *Formaldehyde*, A.R., 36 to 40 % mass formaldehyde solution (Formalin). (**Warning**—Formaldehyde is classified as a human carcinogen. Formaldehyde is toxic by inhalation, in contact with skin and if swallowed. Causes burns. Possible risks of irreversible effects. May cause sensitization by skin contact. In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. Wear suitable protective clothing, gloves and eye/face protection. In case of accident or if you feel unwell, seek medical advice immediately. Use only in well ventilated areas.)

7.1.3 *Hydrochloric Acid*, A.R., concentrated [36 % mass]. (**Warning**—Hydrochloric Acid is toxic by inhalation. Causes severe burns. Keep container in a well-ventilated place. In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. Wear suitable protective clothing, gloves and eye/face protection. In case of accident or if you feel unwell, seek medical advice immediately.)

7.1.4 *Iron (III) Chloride*, (FeCl₃·6H₂O), A.R. (**Warning**—Iron (III) Chloride causes burns. Avoid contact with skin and eyes. In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. Take off immediately all contaminated clothing. Wear eye/face protection. If swallowed, seek medical advice immediately and show the container or label.)

7.1.5 *Methanol*, A.R. minimum purity 99.99 % (V/V). (**Warning**—Methanol is highly flammable. Toxic by inhalation and ingestion. Keep container in a well-ventilated place. Keep container tightly closed when not in use. Keep away from sources of ignition. Avoid contact with skin. In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible.)

7.1.6 *3-Methyl-2-Benzothiazolinone Hydrazone Hydrochloride (MBTH)*. (**Warning**—MBTH is toxic if swallowed. Irritating to eyes. Keep locked up. Avoid contact with skin and eyes. In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. This material and its container must be disposed of in a safe way. Wear suitable protective clothing. In case of accident or if you feel unwell, seek medical advice immediately. If swallowed, seek medical advice immediately and show the container or label.)

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7.2 *Water*; reagent water conforming to Type I or II of Specification **D1193**.

7.3 Solutions:

7.3.1 *Standard Formaldehyde Solution*, 8.0 µg/mL, aqueous. Determine the exact concentration (c_1) of an A.R. 36 to 40 % mass formaldehyde solution (**Warning**—See 7.1.2.) according to Test Method **D2194**, expressing the result in g/100 mL. Weigh approximately 2.0 to 2.2 g (m_1) of the formaldehyde solution to the nearest 0.1 mg, into a 100-mL volumetric flask. Make up to the mark with water and mix thoroughly. Pipette 1.0 mL of this solution into a 1-L volumetric flask. Make up to the mark with water and mix thoroughly. Transfer this solution to a brown bottle with screw cap. Calculate the exact formaldehyde content of this solution (c_2), in µg/mL, by means of the following equation:

$$c_2, \mu\text{g/mL} = \frac{m_1 \times c_1}{10} \quad (1)$$

where:

c_1 = concentration of concentrated formaldehyde solution used, g/100 mL, and

m_1 = mass of concentrated formaldehyde solution taken, g.

The concentrated and diluted formaldehyde solutions are stable in excess of six months if stored at room temperature. Do not store in a refrigerator. Prepare fresh solutions every six months.

⁴ ACS Reagent Chemicals, Specifications and Procedures for Reagents and Standard-Grade Reference Materials, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

7.3.2 *Hydrochloric Acid, 50 % (V/V)*. Dilute one volume of A.R. concentrated hydrochloric acid (**Warning**—See 7.1.3.) with one volume of water.

7.3.3 *Iron (III) Chloride Solution, 0.3 % mass, aqueous*. Dissolve 5 g of A.R. iron (III) chloride, 6H₂O (**Warning**—See 7.1.4.) in water, adding 1 or 2 drops of A.R. concentrated hydrochloric acid to stabilize solution (**Warning**—See 7.1.3.) and dilute to 1 L with water. Prepare a fresh solution daily.

7.3.4 *Methyl-2-benzothiazolone Hydrazone Solution (MBTH solution), 0.4 % mass, aqueous*. Dissolve 400 ± 1 mg of MBTH hydrochloride (**Warning**—See 7.1.6.) in water in a 100-mL volumetric flask. Make up to the mark and mix thoroughly. Prepare a fresh solution daily.

7.4 *Reagents*, as described in Test Method D2194, for the standardization of formaldehyde solution (see 7.3.1).

7.4.1 *Sodium Sulfite Solution, 12.5 % mass, aqueous*. Dissolve 125 g of anhydrous sodium sulfite (Na₂SO₃) in water and dilute to 1 L.

NOTE 4—Sodium sulfite gradually oxidizes to sodium sulfate on exposure to air and therefore should be kept in a tightly closed container. For best results freshly prepared reagent should be used.

7.4.2 *Sulfuric Acid, 0.5 mol/L (1.0 N)*. Prepare and standardize sulfuric acid (H₂SO₄) against 0.5 mol/L (0.5 N) sodium hydroxide (NaOH) solution which has been standardized against potassium acid phthalate to a thymolphthalein end point.

7.4.3 *Thymolphthalein Indicator, 0.1 % mass, alcohol solution*. Dissolve 1.0 g of thymolphthalein in 100 mL of methanol, ethanol, or isopropanol and dilute to 1 L with additional alcohol.

7.5 *Ethylene Glycol Quality Control Sample*—It is recommended to select an Ethylene Glycol sample (fiber grade MEG, DEG or TEG), similar to the product being analyzed and to use it as a quality control sample (**Warning**—See 8.28.3.). To this end, ensure to obtain a sufficient amount and store it in such a way that it is stable for a known period of time and use it as such during this period of time only. For more details, see 16.1.2.

8. Hazards

8.1 Consult current OSHA regulations and supplier's Safety Data Sheets and local regulations for all materials used in this test method.

8.2 Note health and safety warnings in Section 7.

8.3 *Ethylene Glycol*—Although (mono-, di-, and tri-) ethylene glycol in general is not classified as dangerous or flammable and is not expected to impose a health hazard when used under normal conditions, it is recommended to avoid inhalation and contact with skin and eyes. Wear suitable protective clothing and gloves. Do not breathe gas, fumes, vapor, or spray. Use only in well ventilated areas. In case of contact with eyes rinse with plenty of water and seek medical advice.

8.4 Formaldehyde is classified as a human carcinogen.

9. Sampling, Test Specimens, and Test Units

9.1 Follow the relevant instructions for sampling as given in Practice E300.

10. Preparation of Apparatus

10.1 *Spectrophotometer*—Check the performance of the spectrophotometer at regular intervals according to the guidelines given in Practice E275 and the manufacturer's instruction manual.

10.2 *Glassware*—Thoroughly clean the cells and other glassware using 50 % (V/V) hydrochloric acid (**Warning**—See 7.1.3.)

according to the guidelines in Practice E275. If the glassware is persistently stained by iron (III) chloride, clean the affected glassware in a concentrated hydrochloric acid bath. Do not use acetone or methanol for cleaning, since this reacts with MBTH.

11. Calibration

11.1 Introduce the amounts of reagents as listed in Table 1 into separate 100-mL volumetric flasks in the following way: First introduce the standard formaldehyde solution and water by means of a volumetric pipette, followed by MBTH solution.

11.2 Proceed as described in 12.5 through 12.7, measuring the absorbances of the calibration solutions, including the blank solutions, against water at 635 nm in a 10-mm cell.

11.3 If the mean blank absorbance exceeds its upper control limit, clean the glassware and repeat the calibration.

11.4 Calculate the exact amount, in μg , of formaldehyde present in each of the calibration solutions, which will be in the order of 0.0, 8.0, 16.0, 24.0, 32.0, and 40.0 μg .

11.5 Construct a calibration graph by plotting the absorbances against the corresponding amount of formaldehyde in each calibration solution. Apply simple linear regression to obtain the best straight line (calibration function).

11.6 Evaluate the linearity of the calibration function obtained by calculating the correlation coefficient or by plotting the y-residuals versus the known amounts, or both, and visually evaluate the results.

NOTE 5—Many spectrophotometers have the ability to calculate a calibration graph automatically after measuring the calibration solutions and subsequently to show the amount/concentration of the component being measured directly on a display. In such cases, no calibration graphs need to be constructed and the calculation can be started with 13.2. It is, however, recommended to verify the calculation procedure of the instrument and to establish the characteristics of the calibration graph according to suitable regression analysis software.

11.7 *Verification*—The entire calibration graph is based on only one starting solution. In such a case, a weighing error, for example, could introduce significant errors when reading concentrations in test solutions. It is, therefore, recommended to either:

11.7.1 Prepare a separately weighed control solution with an accurately known amount of approximately 20 μg of formaldehyde and perform 11.2, measuring the absorbance of the control solution. Obtain the amount of formaldehyde of the control solution from the calibration graph/function. If this value and the calculated value of the control solution differ by more than the confidence limits for the control solution, as derived from the confidence limits for intercept and slope of the calibration function, repeat the calibration.

11.7.2 Maintain a record or control chart of the slope of the calibration graph. If a significant difference with earlier values is obtained, repeat the calibration.

12. Procedure

12.1 Place a 100-mL volumetric flask onto the balance and tare or record the tare weight to the nearest 0.1 mg. Pipet 1 mL of sample into the flask, reweigh the flask and record the increase in weight as the mass of the test portion (m_2).

TABLE 1 Amounts of Reagents to be Added to the Calibration Solutions

Calibration Solution, No.	Standard Formaldehyde Solution, mL	Water, mL	MBTH Solution, mL
Blank 1	0.0	5.0	5.0
Blank 2	0.0	5.0	5.0
Blank 3	0.0	5.0	5.0
1	1.0	4.0	5.0
2	2.0	3.0	5.0
3	3.0	2.0	5.0
4	4.0	1.0	5.0
5	5.0	0.0	5.0