

SLOVENSKI STANDARD SIST ISO 5351-2:1996

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Razredčena raztopina celuloze - Določanje mejne viskoznosti - 2. del: Metoda v raztopini železo(III) natrijev tartrat kompleks (EWNN mod NaCI)

Cellulose in dilute solutions -- Determination of limiting viscosity number -- Part 2: Method in iron(III) sodium tartrate complex (EWNN mod NaCl) solution

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Cellulose en solutions diluées - Détermination de l'indice de viscosité limite -- Partie 2: Méthode utilisant une solution du complexe fer(III)-tartrate de sodium (EWNN mod NaCI)

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International Standard



INTERNATIONAL ORGANIZATION FOR STANDARDIZATION® MEXTON PRANDALAR OPPAHUSALUM NO CTAHDAPTUSALUM® ORGANISATION INTERNATIONALE DE NORMALISATION

Cellulose in dilute solutions — Determination of limiting viscosity number — Part 2 : Method in iron(III) sodium tartrate complex (EWNN_{mod NaCl}) solution

Cellulose en solutions diluées – Détermination de l'indice de viscosité limite – Partie 2 – Méthode utilisant une solution du complexe fer(III)-tartrate de sodium (EWNN méd Nachdards.iteh.ai)

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards institutes (ISO member bodies). The work of developing International Standards is carried out through ISO technical committees. Every member body interested in a subject for which a technical committee has been set up has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work.

Draft International Standards adopted by the technical committees are circulated to the member bodies for approval before their acceptance as International Standards by the ISO Council.

International Standard ISO 5351/2 was developed by Technical Committee ISO/TC 6, Paper, board and pulps, and was circulated to the member bodies in January 1980.

It has been approved by the member bodies of the following countries :

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Austria	France aa8353b7	35 Romania 5351-2-1996		
Belgium	Germany, F.R.	3525/581-150-5351-2-1996 South Africa, Rep. of		
Brazil	Hungary	Spain		
Chile	India	Sweden		
China	Kenya	Turkey		
Czechoslovakia	Korea, Rep. of	United Kingdom		
Egypt. Arab Rep. of	Norway	USA		
Finland	Philippines			

No member body expressed disapproval of the document.

[😳] International Organization for Standardization, 1981 🔹

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Cellulose in dilute solutions – Determination of limiting viscosity number -Part 2 : Method in iron(III) sodium tartrate complex (EWNN_{mod NaCl}) solution

0 Introduction

The viscosity (or dynamic viscosity), symbol η , of a fluid is defined by the Newtonian equation

$$\tau = \eta \cdot \dot{\gamma}$$

where

- τ is the shear stress;
- iTeh STANDARI η is the viscosity;

ISO 5351/1 specifies a method for the determination of the limiting viscosity number of cellulose in dilute cupri-ethylenediamine (CED) solution.

References 2

ISO 638, Pulps – Determination of dry matter content.

ISO 1833, Textiles – Binary fibre mixtures – Quantitative chemical analysis.

ISO 5089, Textiles Preparation of laboratory test samples and test specimens for chemical testing. $\dot{\gamma} = \frac{dv}{dz}$ is the velocity gradient (v is the velocity of doies.)

ISO/TR 5090, Textiles - Methods for the removal of nonplane relative to the other, and z the co-ordinate perpenfibrous matter prior to quantitative analysis of fibre mixtures. dicular to the two planes). https://standards.iteh.ai/catalog/standards/sist/6fc804d8-79b6-4174-979b-

In non-Newtonian behaviour, normally the 2835 With highst-iso-5351-2-1996 polymer solutions such as cellulose, the ratio of the shear stress to the velocity gradient varies with the shear stress.

The data required for evaluation of the limiting viscosity number of cellulose in dilute solutions (for definitions and symbols, see clause 4) are derived by means of a capillary-tube viscometer. The results of these measurements are seriously affected by the shear rate. The solution to the problem that thereby arises can be sought either by determination of the viscometric properties at so low a concentration of cellulose that the effect of shear rate is small, or by determination at a shear rate closely reproducible in different laboratories. In this document, the first condition, low concentration of cellulose, is applied.

Scope and field of application 1

This part of ISO 5351 specifies a method for the determination of the limiting viscosity number of cellulose in dilute iron(III) sodium tartrate complex (EWNN_{mod NaCl}) solution.

This method is applicable to EWNN-soluble samples of cellulose, for instance in pulps and textiles.

 NOTE - The viscosity test is a means for determining the extent of cellulose degradation produced by cooking and bleaching. This degradation greatly affects the suitability of pulp for dissolving purposes and papermaking.

Measurement of the times of efflux of the diluted solvent and solution of cellulose through a capillary-tube viscometer at a specified concentation at 20 °C. Calculation by Schulz-Blaschke formula of the limiting viscosity number from these measurements, and from the known concentration of the solution.

4 Definitions

For the purposes of this International Standard, the following definitions apply.

4.1 shear rate, G: The velocity gradient of a fluid layer, parallel to the direction of flow, at the periphery of the capillary, defined by the equation

$$G = \frac{4V}{\pi r^3 t_{\rm f}}$$

where

V is the volume between two arbitrary calibration marks of the viscometer, in millilitres;

- r is the radius of the capillary tube, in centimetres;
- is the efflux time of the fluid, in seconds. t_f

4.2 viscosity ratio : The ratio of the viscosities η and η_0 of the polymer solution of stated concentration and of the solvent respectively at the same temperature :

η

 η_0

This ratio, as a number, is dimensionless.

4.3 viscosity relative increment : The viscosity ratio (4.2) minus one :

 $\frac{\eta}{\eta_0} - 1 = \frac{\eta - \eta_0}{\eta_0}$

As a number, it is dimensionless.

4.4 viscosity number : The ratio of the viscosity relative increment (4.3) to the polymer concentration c in the solution :

$$\frac{\eta - \eta_0}{\eta_0 \cdot c}$$

Its unit is the millilitre per gram.

4.5 limiting viscosity number (L.V.N.P. n) The limiting value of the viscosity number (4.4) at infinite dilution :

$$[\eta] = \lim_{c \to 0} \frac{\eta - \eta_0}{\eta_0 \cdot c}$$

Reagent 5

Iron(III) sodium tartrate complex solution (EWNN_{mod NaCl}).

Iron(III) sodium tartrate complex dissolved in dilute sodium hydroxide solution, for convenience referred to as iron sodium tartrate complex solution, and prepared as described in annex A.

Apparatus 6

Ordinary laboratory apparatus and

6.1 Constant-temperature bath, capable of being controlled at 20 \pm 0,1 °C, accommodating the dissolving flasks (6.2) and the viscometer (6.5).

6.2 Conical flasks, with plastics stoppers, capacity 50 ml.

6.3 Balance, accurate to \pm 0,1 mg.

6.4 Timing device, capable of being read to the nearest 0.1 s.

6.5 Capillary-tube viscometer, length of capillary tube at least 90 mm, for example an Ubbelohde viscometer of type 1 a.

6.6 Shaking machine, accommodating the dissolving flasks (6.2).

Pretreatment of test samples 7

7.1 Pulp samples

Take a sample corresponding to approximately 10 g of oven-dry mass. Split and tear the pulp into pieces of about $5 \text{ mm} \times 5 \text{ mm}$ in size.

7.2 Textile samples

7.2.1 Generally

KEVIEV Extract a sample of at least 3 g of air-dry mass in a Soxhlet standar apparatus with light petroleum for 1 h at a minimum rate of 6 cycles per hour. Allow the light petroleum to evaporate from the sample; soak the specimen in cold distilled or deionized SIST ISO water for 1 h, and then in a fresh portion of water at 65 \pm 5 °C Its unit is the millilitre per gram.¹⁾ https://standards.itch.ai/catalog/standards/statlog/statlog/statlog/statlog/statlog/statlog/statlog/statlog/statlog/statlog/statlog/statlog/statlog/st aa8353b73525/sistoo: 53 Algitate the liquor from time to time. Remove the excess water from the sample by squeezing, suction or by means of a centrifuge and then allow the sample to become air-dry. For additional information, see ISO 1833 and ISO/TR 5090.

Prepare the sample as specified in ISO 5089.

7.2.2 For bast fibres

Prepare the sample as described in annex B.

Procedure 8

8.1 Weighing of sample

8.1.1 If the approximate limiting viscosity number of the sample is known, weigh, to an accuracy of \pm 0,5 mg, into the dissolving flask (6.2), an amount of the sample so chosen that the value of the viscosity ratio becomes 1,1 to 1,5 (see the note).

8.1.2 If the approximate value of the limiting viscosity number of the sample is not known, test a sample of 15 to 20 mg.

¹⁾ The SI unit of [η] is the cubic metre per kilogram. Information from ISO/TC 61, *Plastics*, indicates plans in TC 61 to adopt this unit in practice.

8.1.3 If the viscosity ratio so obtained is not within the range specified, make the test by choice of the correct amount according to the value of the viscosity ratio so derived. At the same time, weigh out a separate sample, for determination of the dry matter in accordance with ISO 638 or ISO 1833 sub-clause 1.7.

NOTE - For routine analytical purposes, a constant amount of 15 to 20 mg of the sample may be used.

8.2 Preparation of test solution

With a pipette, add 50,0 ml of the EWNN solution (clause 5) to the sample. Close the flask, and shake it by hand for about 5 s. Transfer the flask to the shaking machine (6.6) and continue the shaking for about 16 h (see note 1). Take care to preclude raising of the temperature of the solution above 18 °C (see note 2).

NOTES

1 For the routine analysis of a large number of samples, the flasks may conveniently be put in the shaker and shaken overnight. For $[\eta]$ values below 1 000 ml/g, the shaking time is reducible to 2 h.

2 Dependent upon the room temperature, this may be done by refrigeration of the sample to 0 to 4 °C and isolation of the dissolving flasks with suitable covers, or more safely by direct cooling with water NDAR during the shaking. II eh SIA

where

$$\frac{\eta - \eta_0}{\eta_0 \cdot c} = \frac{t - t_0}{t_0 \cdot c}$$
 is the viscosity number, in millilitres per gram;

 $-t_{0}$ is the viscosity relative increment (dimensionless):

k is an empirical constant (for the cellulose-EWNN system, k = 0.3;

c is the concentration (oven-dry basis), in grams per millilitre, of the cellulose in the solvent.

Example of calculation

For a spruce sulphite pulp, the following data were obtained :

 $t = 74.2 \, s$

 $t_0 = 56,2 \text{ s}$

Accordingly, the viscosity ratio, $\frac{\eta}{\eta_0} = 1,320$, and the viscosity Determination of efflux times (standards.itelative increment 8.3

Keep the dissolving flask containing the solution in the Keep the dissolving flask containing the solution in the $351-2:199\eta - \eta_0 \eta$ constant-temperature bath (6.1), until a temperature of $351-2:199\eta - \eta_0 \eta$ 20 ± 0,1 °C is reached. Immerse the viscometer 36.55 manuards/sist/61c80448-756,417419790,320 constant-temperature bath, fix it in a vertical position, and iso-5351-2-1 996 allow it to stand until equilibrium temperature is reached. Rinse the viscometer twice with the test solution prior to each determination. For temperature adjustment, apply suction to draw the solution once or twice to the maximum liquid level of the viscometer.

Draw a portion of the test solution into the viscometer by suction. Allow the fluid to drain. When the meniscus is at the upper mark on the upper bulb, start the timing device (6.4), and measure the efflux time (t) for the fluid to drain to the lower mark on the bulb. Measure the efflux time several times to an accuracy of \pm 0,1 s. Measure, as described above, the efflux time (t_0) of the solvent.

Make at least two determinations the results of which shall agree within \pm 2,5 %. If the deviation exceeds this, the test shall be repeated.

9 Calculation

The limiting viscosity number is given, in millilitres per gram, by the Schulz-Blaschke formula :

$$(L.V.N.) = [\eta] = \frac{\begin{pmatrix} \eta - \eta_0 \\ \eta_0 \cdot c \end{pmatrix}}{1 + k_\eta \cdot \left(\frac{\eta - \eta_0}{\eta_0}\right)}$$

For a cellulose concentration of $c = 0.304 \times 10^{-3}$ g/ml, the Schulz-Blaschke formula yields a limiting viscosity number L.N.V. = $[\eta] = 950 \text{ ml/g}.$

10 Test report

The test report shall include the following particulars :

- a) reference to this International Standard;
- b) all the information necessary for complete identification of the sample;
- preliminary treatment, such as purification; c)
- type and constants of the viscometer; d)
- the result, expressed in millilitres per gram; e)
- f) any unusual features observed in the course of the test;

g) any operations not specified in this International Standard, or in the International Standards to which reference is made, or regarded as optional, which might have affected the results.