



Designation: E1064 – 23

# Standard Test Method for Water in Organic Liquids by Coulometric Karl Fischer Titration<sup>1</sup>

This standard is issued under the fixed designation E1064; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

## 1. Scope\*

1.1 This test method covers the determination of water from 0 % to 2.0 % mass in most liquid organic chemicals, with Karl Fischer reagent, using an automated coulometric titration procedure. Use of this test method is not applicable for liquefied gas products such as Liquid Petroleum Gas (LPG), Butane, Propane, Liquid Natural Gas (LNG), etc.

1.2 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.3 Review the current Safety Data Sheets (SDS) for detailed information concerning toxicity, first-aid procedures, handling, and safety precautions.

1.4 *This standard does not purport to address all of the safety problems, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.* Specific precautionary statements are given in Section 8.

1.5 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

## 2. Referenced Documents

2.1 *ASTM Standards:*<sup>2</sup>

[D1193 Specification for Reagent Water](#)

[D4672 Test Method for Polyurethane Raw Materials: Determination of Water Content of Polyols](#)

<sup>1</sup> This test method is under the jurisdiction of ASTM Committee D16 on Aromatic, Industrial, Specialty and Related Chemicals and is the direct responsibility of Subcommittee D16.04 on Instrumental Analysis.

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<sup>2</sup> For referenced ASTM standards, visit the ASTM website, [www.astm.org](http://www.astm.org), or contact ASTM Customer Service at [service@astm.org](mailto:service@astm.org). For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

[D6809 Guide for Quality Control and Quality Assurance Procedures for Aromatic Hydrocarbons and Related Materials](#)

[E180 Practice for Determining the Precision of ASTM Methods for Analysis and Testing of Industrial and Specialty Chemicals \(Withdrawn 2009\)](#)<sup>3</sup>

[E203 Test Method for Water Using Volumetric Karl Fischer Titration](#)

## 3. Summary of Test Method

3.1 This test method is based on the Karl Fischer reaction for determining water—the reduction of iodine by sulfur dioxide in the presence of water to form sulfur trioxide and hydriodic acid. The reaction becomes quantitative only when pyridine or other organic base and methanol or other alcohol are present. Unlike the volumetric Karl Fischer reagents that include iodine, the coulometric technique electrolytically generates iodine, with 10.71 C of generating current corresponding to 1 mg of water in accordance with Faraday's law.

## 4. Significance and Use

4.1 The coulometric technique is especially suited for determining low concentrations of water in organic liquids that would yield small titers by the Karl Fischer volumetric procedure. The precision and accuracy of the coulometric technique decreases for concentrations of water much greater than 2.0 % because of the difficulty in measuring the small size of sample required. The test method assumes 100 % efficiency of coulombs in iodine production. Provision is made for verifying this efficiency. (See [Table 1](#) and [Note 5](#).)

## 5. Interferences

5.1 Interfering substances are the same as are encountered in the volumetric Karl Fischer titration. A detailed discussion of interfering substances can be found in the treatise on "Aquametry."<sup>4</sup>

<sup>3</sup> The last approved version of this historical standard is referenced on [www.astm.org](http://www.astm.org).

<sup>4</sup> J. Mitchell, Jr. and D. M. Smith, "Aquametry"—*A Treatise on Methods for the Determination of Water, Part III*—The Karl Fischer Reagent, 2nd Ed., J. Wiley and Sons, Inc., New York, NY 1990.

\*A Summary of Changes section appears at the end of this standard

TABLE 1 Sample Size Estimation

Expected Water Content	Sample Size, mL
0 mg/kg to 500 mg/kg	5
500 mg/kg to 1000 mg/kg	2
1000 mg/kg to 2000 mg/kg	1
0.2 to 0.5 % mass	0.5
0.5 to 2.0 % mass	0.1

5.2 Test Method E203 discusses organic compounds in which water may be determined directly and compounds in which water cannot be determined directly, but in which interferences may be eliminated by suitable chemical reactions.

## 6. Apparatus

6.1 *Automatic Titrator*,<sup>5</sup> consisting of a control unit, titration vessel, dual platinum sensing electrode, generator assembly, and magnetic stirrer. The instrument is designed to coulometrically generate iodine that reacts stoichiometrically with the water present in the sample solution. The coulombs of electricity required to generate the reagent are converted to micrograms of water, which is obtained as a direct digital readout.

6.2 *Syringe*, 50 mL, fitted with an 115 mm hypodermic needle for removing excess solution from the titration chamber.

NOTE 1—Rinse all glass syringes and needles with anhydrous acetone after cleaning, then dry in an oven at 100 °C for at least 1 h and store in a desiccator. Plastic syringes shall be disposed of following use.

6.3 *Syringe*, 20 mL, fitted with an 115 mm hypodermic needle for introduction of neutralizing solution into the titration chamber (see Note 1).

6.4 *Syringes*, 1 mL and 5 mL, fitted with 115 mm hypodermic needles for introduction of samples into titration chamber (see Note 1).

6.5 *Syringe*, 5 µL, fitted with 115 mm hypodermic needle for standardization of instrument (see Note 1).

6.6 *Fluorocarbon Sealing Grease or TFE-Fluorocarbon*, to seal the titration chamber against atmospheric moisture.

6.7 *Septa*, to seal sample port but allow introduction of samples by a needle with a minimum of moisture contamination. Replace serum caps and septa as required to prevent air leakage as indicated by instrument drift.

6.8 *Serum Bottles*.

6.9 *Oven*, temperature 100 °C ± 5 °C.

6.10 *Dessicator*, standard laboratory type with color change indicator.

6.11 *Analytical Balance*, capable of weighing to ±0.0001 g.

## 7. Reagents

7.1 *Purity of Reagents*—Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemi-

<sup>5</sup> Basic references to the automatic coulometric titrator: M. T. Kelley, R. W. Stelzner, W. R. Laing, and D. J. Fisher, *Analytical Chemistry* 31, No. 2, 220 (1959) and A. W. Meyer, Jr. and C. M. Boyd, *Analytical Chemistry* 31, No. 2, 215 (1959).

cal Society, where such specifications are available.<sup>6</sup> Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Purity of Water*—Unless otherwise indicated, reference to water shall be understood to mean Type I or Type II reagent water, conforming to Specification D1193.

7.3 *Karl Fischer Reagents*—Commercial coulometric KF reagents and reagent systems of various types are available for use with autotitrators for water determination. Traditionally, pyridine was the organic base used in KF reagents. Pyridine-free formulations are available and are preferred by most KF instrument manufacturers for use with their equipment. The pyridine-free reagents are less toxic, less odorous, and more stable than those containing pyridine. The use of pyridine-free reagents is recommended whenever possible. Coulometric titrations normally require two reagent solutions. An anolyte or solvent titration solution and a catholyte or generator titrant solution. However, now reagents can be purchased in one or two component reagent systems. A one component reagent system contains all the components required for a Karl Fischer titration in a single solution. A two component system incorporates separate solutions for the solvent and titrant.

NOTE 2—Two good references on pyridine-free reagents are the Hydranal<sup>®7</sup> Manual-Eugen Schotz Reagents for Karl Fischer Titration, from Riedel-deHaen (www.rdhlab.de) or Sigma Aldrich (www.sigmaaldrich.com) and Moisture Measurement by Karl Fischer Titrimetry, 2nd ed., by GFS Chemicals, Inc., January 2004.

7.3.1 *Generator Titrant Solution (catholyte)*, containing iodine, sulfur dioxide, pyridine or other organic base and methanol or other alcohol to provide iodine in the reaction mixture.

7.3.2 *Solvent Titration Solution (anolyte)*, prepared as per instrument specifications.

7.3.3 *Neutralizing Solution*, methanol containing approximately 20 mg H<sub>2</sub>O/mL.

## 8. Safety Precautions

8.1 The reagents contain one or more of the following: iodine, pyridine or other organic base, sulfur dioxide, and methanol or other alcohol. Wear chemical resistant gloves when mixing the reagents and removing solution from the titration chamber. Care must be exercised to avoid inhalation of reagent vapors, or direct contact of the reagent with the skin.

## 9. Sampling

9.1 Because of the low concentration of water to be measured, maximum care must be exercised at all times to avoid contaminating the sample with moisture from the sample container, the atmosphere, or transfer equipment.

<sup>6</sup> ACS Reagent Chemicals, *Specifications and Procedures for Reagents and Standard-Grade Reference Materials*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.

<sup>7</sup> Hydranal<sup>®</sup> is a registered trademark of Sigma-Aldrich, Inc, St. Louis, MO.

9.1.1 Dry the sample bottles and caps overnight in an oven at 100 °C before using. Allow to cool in a desiccator before filling and sealing.

9.1.2 Fill the sample bottle as rapidly as possible to within 25 mm of the top and immediately seal.

9.2 Remove the test specimens for analysis from the sample bottle with a dry hypodermic syringe. Inject dry nitrogen into the sample bottle with the syringe to displace the sample that is removed.

## 10. Preparation of Apparatus

10.1 Clean, dry, and assemble the titration chamber as directed in the manufacturer's instructions. Use fluorocarbon grease or TFE-fluorocarbon to seal the upper and lower sections of the chamber from atmospheric moisture. Connect the leads from the sensing and generator electrodes to the titrator.

10.2 Prepare the titration solution (7.3.2) as directed by the instrument manufacturer and fill the instrument reservoir as directed by the manufacturer.

10.3 Add the generator solution (7.3.1) to the generator assembly and replace the cover cap. The surface of the generator solution must be below the level of the titration solution to prevent backflow contamination of the titration solution. The generator solution may have to be removed periodically to maintain the lower level.

10.4 Follow the manufacturer's procedure to dry the titration cell.

## 11. Verification of Calibration

11.1 Different autotitrators may vary in calibration procedures. Consult the operating manual for the autotitrator in use. Stable, prepackaged Quality Control (QC) water standards are commercially available with 10 mg/kg, 100 mg/kg and 1 % (m/m) water content for this purpose. It is desirable to verify calibration with a standard solution that approximates the same range of water expected to be in the samples. If the measured value exceeds  $\pm 5$  % of the known amount, take appropriate action before proceeding with the sample test.

NOTE 3—This may require replacing or regenerating the reagent solutions.

## 12. Procedure

12.1 Assemble a dry syringe and needle and attach a small piece of rubber septum to the needle tip. Withdraw 1 mL to 2 mL of the sample into the syringe and discard the contents into a waste container. Using the following table as a guide, withdraw the proper amount of test sample into the syringe and seal with the rubber septum. Obtain a tare weight to  $\pm 0.1$  mg. See Table 1, Note 4 and Note 5.

NOTE 4—If multiple analyses are performed on the same test sample or if an appreciable volume of test sample is withdrawn, a slight pressure should be maintained on the sample bottle by means of a hypodermic needle attached to a dry nitrogen source.

NOTE 5—Alternatively, if syringes of suitable accuracy are available, a measured volume of sample can be injected and the mass calculated from the volume and density.

12.2 With the analyzer stabilized, carefully insert the needle of the sample syringe through the septum and below the level of solution in the titration chamber. Inject the sample slowly into the titration solution and begin titration. Withdraw the syringe needle, seal and again weigh to the nearest  $\pm 0.1$  mg to determine the exact sample mass. Allow the titration to proceed until the end-point is indicated. Record the micrograms of water determined.

NOTE 6—After numerous analyses, the level of solvent accumulated in the titration chamber may have to be reduced. This can be accomplished with a 50 mL syringe or by partially draining the solution if the titration chamber is provided with a stopcock. Discard the solution and replace with fresh titration solution if a stable reading cannot be obtained.

NOTE 7—Replace the generator solution when it becomes yellow and a stable reading cannot be obtained.

## 13. Calculation

13.1 Calculate the water content of the sample to the nearest 0.001 % mass as follows:

$$\text{water content, \% mass} = \frac{\mu\text{g water found}}{\text{g sample} \times 10000} \quad (1)$$

## 14. Report

14.1 Report the percentage of water to the nearest 0.001 % mass.

## 15. Precision and Bias

15.1 *Precision*—The following criteria should be used to judge the acceptability of results when using pyridine-based reagents (see Note 8).

15.1.1 *Repeatability (Single Analyst)*—The coefficient of variation for a single determination has been estimated to be 1.39 % relative at 61 DF. The 95 % limit for the difference between two such runs is 3.9 % relative.

15.1.2 *Laboratory Precision (Within-Laboratory, Between Days Variability)*—The coefficient of variation of results (each the average of duplicate determinations), obtained by the same analyst on different days, was estimated to be 2.00 % relative at 31 DF. The 95 % limit for the difference between two such averages is 5.6 % relative.

15.1.3 *Reproducibility (Multilaboratory)*—The coefficient of variation of results (each the average of duplicate determinations), obtained by analysts in different laboratories, has been estimated to be 6.12 % relative at 7 DF. The 95 % limit for the difference between two such averages is 17.1 % relative.

15.2 *Bias*—The bias of this test method has not been determined since no acceptable reference material has been identified.

NOTE 8—The above precision estimates are based on an inter-laboratory study on samples of toluene, ethyl acetate, ethanol, and acetonitrile, containing approximately 0.02, 0.16, 0.52, and 2.04 % mass water, respectively. One analyst in each of 8 laboratories performed duplicate determinations on two different days, for a total of 128 determinations. Five values for toluene were judged to be outliers. Practice E180 and a special computer program for the analysis of variance of unbalanced nested experiment designs were used to develop these precision estimates.