



Designation: ~~C114 – 22~~ C114 – 23

## Standard Test Methods for Chemical Analysis of Hydraulic Cement<sup>1</sup>

This standard is issued under the fixed designation C114; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

### 1. Scope\*

1.1 These test methods cover the chemical analyses of hydraulic cements. Any test methods of demonstrated acceptable precision and bias may be used for analysis of hydraulic cements, including analyses for referee and certification purposes, as explained in Section 4. Specific chemical test methods are provided for ease of reference for those desiring to use them. They are grouped as Reference Test Methods and Alternative Test Methods. The reference test methods are long accepted classical chemical test methods which provide a reasonably well-integrated basic scheme of analysis for hydraulic cements. The alternative test methods generally provide individual determination of specific analytes and may be used alone or as alternates and determinations within the basic scheme at the option of the analyst and as indicated in the individual method.

### 1.2 Contents:

Section	Subject
2	Referenced Documents
4	Description of Referee Analyses
4.1	Referee Analyses
5	Qualification for Different Analyses
5.1	Certified Reference Materials
5.2	Requirements for Qualification Testing
5.3	Alternative Analyses
5.4	Performance Requirements for Rapid Test Methods
6	General
6.1	Interferences and Limitations
6.2	Apparatus and Materials
6.3	Reagents
6.4	Sample Preparation
6.5	General Procedures
6.6	Recommended Order for Reporting Analyses
	Reference Test Methods
7	Insoluble Residue
8	Silicon Dioxide
8.2	Cements with Insoluble Residue Less Than 1 %
8.3	Cements with Insoluble Residue Greater Than 1 %
9	Ammonium Hydroxide Group
10	Ferric Oxide
11	Phosphorus Pentoxide
12	Titanium Dioxide
13	Zinc Oxide
14	Aluminum Oxide
15	Calcium Oxide
16	Magnesium Oxide
17	Sulfur

<sup>1</sup> These test methods are under the jurisdiction of ASTM Committee C01 on Cement and are the direct responsibility of Subcommittee C01.23 on Compositional Analysis. Current edition approved Dec. 15, 2022; Oct. 1, 2023. Published December 2022; October 2023. Originally approved in 1934. Last previous edition approved in 2018; 2022 as ~~C114 – 18~~; C114 – 22. DOI: ~~10.1520/C0114-22~~; 10.1520/C0114-23.

\*A Summary of Changes section appears at the end of this standard

17.1	Sulfur Trioxide
17.2	Sulfide
18	Loss On Ignition
18.1	Portland Cement
18.2	Portland Blast-Furnace Slag Cement and Slag Cement
19	Sodium and Potassium Oxides
19.1	Total Alkalis
19.2	Water-Soluble Alkalis
20	Manganic Oxide
21	Chloride
22	Chloroform-Soluble Organic Substances
	Alternative Test Methods
23	Calcium Oxide
24	Carbon Dioxide
25	Magnesium Oxide
26	Loss on Ignition
26.1	Portland Blast-Furnace Slag Cement and Slag Cement
27	Titanium Dioxide
28	Phosphorus Pentoxide
29	Manganic Oxide
30	Free Calcium Oxide
	Appendices
Appendix X1	Example of Determination of Equivalence Point for the Chloride Determination
Appendix X2	CO <sub>2</sub> Determinations in Hydraulic Cements

1.3 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety, health, and environmental practices and determine the applicability of regulatory limitations prior to use.*

1.5 *This international standard was developed in accordance with internationally recognized principles on standardization established in the Decision on Principles for the Development of International Standards, Guides and Recommendations issued by the World Trade Organization Technical Barriers to Trade (TBT) Committee.*

## 2. Referenced Documents

### 2.1 ASTM Standards:<sup>2</sup>

- [C25 Test Methods for Chemical Analysis of Limestone, Quicklime, and Hydrated Lime](#)
- [C219 Terminology Relating to Hydraulic and Other Inorganic Cements](#)
- [E29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications](#)
- [E275 Practice for Describing and Measuring Performance of Ultraviolet and Visible Spectrophotometers](#)
- [E350 Test Methods for Chemical Analysis of Carbon Steel, Low-Alloy Steel, Silicon Electrical Steel, Ingot Iron, and Wrought Iron](#)
- [E617 Specification for Laboratory Weights and Precision Mass Standards](#)
- [E832 Specification for Laboratory Filter Papers](#)

## 3. Terminology

3.1 *Definitions of Terms Specific to This Standard*—The terms used in this standard are defined in Terminology [C219](#).

### 3.2 Definitions:

3.2.1 *analyte, n*—a substance of interest when performing a quantitative analysis.

#### 3.2.1.1 Discussion—

*For the purposes of this test method*, analytes are considered to be those items listed in column 1 of [Table 1](#).

3.2.2 *reagent water, n*—water purified by the process of distillation, deionization, reverse osmosis, or any combination of the three processes.

<sup>2</sup> For referenced ASTM standards, visit the ASTM website, [www.astm.org](http://www.astm.org), or contact ASTM Customer Service at [service@astm.org](mailto:service@astm.org). For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

**TABLE 1 Maximum Permissible Variations in Results<sup>A</sup>**

(Column 1) Analyte	(Column 2) Maximum Difference Between Duplicates <sup>B</sup>	(Column 3) Maximum Difference of the Average of Duplicates from CRM Certificate Values <sup>C,D,E</sup>
SiO <sub>2</sub> (silicon dioxide)	0.16	±0.2
Al <sub>2</sub> O <sub>3</sub> (aluminum oxide)	0.20	±0.2
Fe <sub>2</sub> O <sub>3</sub> (ferric oxide)	0.10	±0.10
CaO (calcium oxide)	0.20	±0.3
MgO (magnesium oxide)	0.16	±0.2
SO <sub>3</sub> (sulfur trioxide)	0.10	±0.1
LOI (loss on ignition)	0.10	±0.10
Na <sub>2</sub> O (sodium oxide)	0.03	±0.05
K <sub>2</sub> O (potassium oxide)	0.03	±0.05
TiO <sub>2</sub> (titanium dioxide)	0.02	±0.03
P <sub>2</sub> O <sub>5</sub> (phosphorus pentoxide)	0.03	±0.03
ZnO (zinc oxide)	0.03	±0.03
Mn <sub>2</sub> O <sub>3</sub> (manganic oxide)	0.03	±0.03
S (sulfide sulfur)	0.01	<i>E</i>
Cl (chloride)	0.003	±0.005
IR (insoluble residue)	0.10	<i>E</i>
Cx (free calcium oxide)	0.20	<i>E</i>
CO <sub>2</sub> (carbon dioxide)	0.12	<i>E,F</i>
Alk <sub>sol</sub> (water-soluble alkali) <sup>G</sup>	0.75/ <i>w</i>	<i>E</i>
Chl <sub>sol</sub> (chloroform-soluble organic substances)	0.004	<i>E</i>

<sup>A</sup> When seven CRM cements are required, as for demonstrating the performance of rapid test methods, at least six of the seven shall be within the prescribed limits and the seventh shall differ by no more than twice that value. When more than seven CRMs are used, as for demonstrating the performance of rapid test methods, at least 77 % shall be within the prescribed limits, and the remainder by no more than twice the value. When a lesser number of CRM cements are required, all of the values shall be within the prescribed limits.

<sup>B</sup> Where no value appears in column 3, CRM certificate values do not exist. In such cases, only the requirement for differences between duplicates shall apply.

<sup>C</sup> Inter-element corrections may be used for any oxide standardization provided improved accuracy can be demonstrated when the correction is applied to all seven CRM cements.

<sup>D</sup> Where an CRM certificate value includes a subscript number, that subscript number shall be treated as a valid significant figure.

<sup>E</sup> Not applicable. No certificate value given.

<sup>F</sup> Demonstrate performance by analysis, in duplicate, of at least one Portland cement. Prepare three standards, each in duplicate: Standard A shall be selected Portland cement; Standard B shall be Standard A containing 2.00 % Certified CaCO<sub>3</sub> (such as NIST 915a); Standard C shall be Standard A containing 5.00 % Certified CaCO<sub>3</sub>. Weigh and prepare two separate specimens of each standard. Assign the CO<sub>2</sub> content of Standard A as the average of the two values determined, provided they agree within the required limit of column 2. Assign CO<sub>2</sub> values to Standards B and C as follows: Multiply the Certified CaCO<sub>3</sub> value (Y) for CO<sub>2</sub> (from the certificate value) by the mass fraction of Certified CaCO<sub>3</sub> added to that standard (percentage added divided by 100); multiply the value determined for Standard A by the mass fraction of Standard A in each of the other standards (that is, 0.98 and 0.95 for Standards B and C, respectively); add the two values for Standard A and for Standard B, respectively; call these values B and C.

Example:

$$B = 0.98A + 0.02Y.$$

$$C = 0.95A + 0.05Y.$$

Where for Certified CaCO<sub>3</sub>, if Y = 39.9 %

$$B = 0.98A + 0.80 \% \text{ by mass.}$$

$$C = 0.95A + 2.00 \% \text{ by mass.}$$

Maximum difference between the duplicate CO<sub>2</sub> values for Standards B and C, respectively, shall be 0.170, 17 % and 0.24 % by mass. Averages of the duplicate values for Standards B and C shall differ from their assigned values (B and C) by no more than 10 % of those respective assigned values.

<sup>G</sup> *w* = weight, in grams, of samples used for the test.

3.2.2.1 *distillation, n*—the process of purification by the evaporation and vaporization of water and its subsequent condensation and collection.

3.2.2.2 *deionization, n*—the process of purification using the two-step process of converting soluble salts into acids by passing them through a hydrogen exchanger after which they are removed by an acid absorbent or synthetic resin.

3.2.2.3 *reverse osmosis, n*—water purification technology that uses a semipermeable membrane to remove ions, molecules, and larger particles from drinking water.

3.2.3 *water (potable), n*—water that is suitable for drinking.

#### 4. Description of Referee Analyses

4.1 *Referee Analyses*—When conformance to chemical specification requirements is questioned, perform referee analyses as described in 4.1.1. The reference test methods that follow in Sections 7 – 22, or other test methods qualified according to 5.4, the *Performance Requirements for Rapid Test Methods* section, are required for referee analysis. A cement shall not be rejected for failure to conform to chemical requirements unless all determinations of constituents involved and all necessary separations prior to the determination of any one constituent are made entirely by these methods. When reporting the results of referee analyses, specify which test methods were used.

4.1.1 Referee analyses shall be made in duplicate and the analyses shall be made on different days. If the two results do not agree within the permissible variation given in Table 1, the determination shall be repeated until two or three results agree within the permissible variation. When two or three results do agree within the permissible variation, their average shall be accepted as the correct value. When an average of either two or three results can be calculated, the calculation shall be based on the three results. For the purpose of comparing analyses and calculating the average of acceptable results, the percentages shall be calculated to the nearest 0.01 (or 0.001 in the case of chloroform-soluble organic substances), although some of the average values are reported to 0.1 as indicated in the test methods. When a blank determination (see Note 1) is specified, one shall be made with each individual analysis or with each group of two or more samples analyzed on the same day for a given analyte.

NOTE 1—A blank determination is a procedure which follows all steps of analysis but in the absence of a sample. It is used for detection and compensation of systematic bias.

#### 5. Qualification for Different Analyses

5.1 *Certified Reference Materials*—A Certified Reference Material (CRM) must be used in the qualification of test methods and analysts. Acceptable reference cements are NIST CRMs, or other reference cements traceable to the NIST CRMs. The reference cement must have an assigned value for the analyte being determined. Traceability consists of documentary evidence that the assigned values of the reference cement are compatible with the certified values of NIST CRMs. To demonstrate traceability for a given analyte, perform a referee analysis (as defined in 4.1) on the proposed reference cement, using a NIST CRM for demonstration of precision and accuracy. The reference cement is acceptable if its assigned value agrees with the average referee value within the limits given in column 3 of Table 1. If the reference cement, as supplied, has no documented guarantee of homogeneity, establish its homogeneity by analyzing at least six randomly selected samples. No result shall deviate from the assigned value by more than the limits given in column 2 of Table 1. An acceptable reference cement must be accompanied by a document showing the data produced in demonstrating traceability and homogeneity.

5.2 *Requirements for Qualification Testing*—Qualified test methods are required whenever testing is performed for the following reasons: (1) for Referee analyses; (2) for analyses intended for use as a basis for acceptance or rejection of a cement; or, (3) for manufacturer's certification. When Reference Methods are used, qualification testing of the analyst is required as described in 5.2.1. When Rapid Methods are used, qualification testing of both the analyst and the test method are required as described in 5.2.1 and 5.4. Such demonstration may be made concurrently with analysis of the cement being tested. The requirements for qualification of a test method and analyst are summarized in Table 2.

5.2.1 Qualification of the analyst shall be demonstrated by analysis of each analyte of concern using at least one CRM cement in duplicate, no matter what test method is used (Note 2). Duplicate samples shall be tested on different days. The analyst is considered qualified when the difference between the duplicate results does not vary by more than the value listed in column 2 of Table 1 and the average of the two samples agrees with the certificate value of the CRM within the limits listed in column 3 of Table 1 after correction for minor components when needed. The same test methods to be used for analysis of cement being tested shall be used for analysis of the CRM cement. If either of the two requirements listed above are not met, identify and correct any problems or errors found in the procedure. Repeat the determinations until a set of duplicate results agree within the permissible variations. Requalification of the analyst is required every two years.

**TABLE 2 Minimum Number of CRMs Required for Qualification of Chemical Testing**

	Method Type	
	Reference <sup>A</sup>	Other <sup>B</sup>
Equipment Qualification	None	7
Analyst Qualification <sup>C</sup>	1	1

<sup>A</sup> Reference Methods are those outlined in Sections 7 – 22.

<sup>B</sup> These may be any test method as described in 5.3, the *Alternative Analyses* section, or any instrumental or rapid test method, which must be qualified in accordance with 5.4, the *Performance Requirements for Rapid Test Methods* section.

<sup>C</sup> Each analyst performing acceptance or reference analyses must be qualified in accordance with 5.2.1, the *Performance Requirements for Rapid Test Methods* section, at a frequency of two years. If qualification of the instrument is completed by a single analyst, the analyst has demonstrated individual qualifications per 5.2.1.

NOTE 2—When qualifying a Rapid Method with seven CRMs in accordance with 5.4.2, the analyst performing the qualification of the test method may simultaneously qualify for the requirement of 5.2.1.

5.2.2 Qualification data demonstrating that the same operator or analyst making the acceptance determination obtained precise and accurate results with CRM cements in accordance with 5.2.1 shall be made available on request to all parties concerned when there is a question of acceptance of a cement. If the CRM used is not a NIST cement, the traceability documentation of the CRM used shall also be made available on request.

5.3 *Alternative Analyses*—The alternative test methods provide, in some cases, procedures that are shorter or more convenient to use for routine determination of certain constituents than are the reference test methods (Note 3). Longer, more complex procedures, in some instances, have been retained as alternative test methods to permit comparison of results by different procedures or for use when unusual materials are being examined, where unusual interferences may be suspected, or when unusual preparation for analysis is required. Test results from alternative test methods may be used as a basis for acceptance or rejection when it is clear that a cement does or does not meet the specification requirement. Any change in test method procedures from those procedures listed in Sections 7 – 30 requires method qualification in accordance with 5.4, the *Performance Requirements for Rapid Test Methods* section.

### ASTM C114-23

NOTE 3—It is not intended that the use of reference test methods be confined to referee analysis. A reference test method may be used in preference to an alternative test method when so desired. A reference test method must be used where an alternative test method is not provided.

5.3.1 Duplicate analyses and blank determinations are not required when using the alternative test methods. If, however, a blank determination is desired for an alternative test method, one may be used and it need not have been obtained concurrently with the analysis. The final results, when corrected for blank values, should, in either case, be so designated.

#### 5.4 *Performance Requirements for Rapid Test Methods*:<sup>3,4</sup>

5.4.1 *Definition and Scope*—Where analytical data obtained in accordance with this test method are required, any test method may be used that meets the requirements of 5.4.2, the *Qualification of a Test Method* section. A test method is considered to consist of the specific procedures, reagents, supplies, equipment, instrument, and so forth, selected and used in a consistent manner by a specific laboratory. See Note 4 for examples of procedures.

NOTE 4—Examples of test methods used successfully by their authors for analysis of hydraulic cement are given in the list of references. Included are test methods using atomic absorption X-ray spectrometry and spectrophotometry-EDTA.

5.4.1.1 If more than one instrument, even though substantially identical, is used in a specific laboratory for the same analyses, use of each instrument shall constitute a separate test method and each must be qualified separately.

<sup>3</sup> Gebhardt, R. F., “Rapid Methods for Chemical Analysis of Hydraulic Cement,” ASTM STP 985, 1988.

<sup>4</sup> Barger, G. S., “A Fusion Method for the X-Ray Fluorescence Analysis of Portland Cements, Clinker and Raw Materials Utilizing Cerium (IV) Oxide in Lithium Borate Fluxes,” Proceedings of the Thirty Fourth Annual Conference on Applications of X-Ray Analysis, Denver Conference, Volume 29 pp. 581–585, August 5, 1985.

5.4.2 *Qualification of a Test Method*—Prior to use for analysis of hydraulic cement, each test method (see 5.4.1) must be qualified individually for such analysis. Qualification data, or if applicable, requalification data, shall be made available pursuant to the Manufacturer’s Certification section of the appropriate hydraulic cement specification.

5.4.2.1 Using the test method chosen, make single determinations for each analyte under consideration on at least seven CRM samples. Requirements for a CRM are listed in 5.1, the *Certified Reference Material* section. Complete two rounds of tests on different days repeating all steps of sample preparations. Calculate the differences between values and averages of the values from the two rounds of tests.

5.4.2.2 When seven CRMs are used in the qualification procedure, at least six of the seven differences between duplicates obtained of any single analyte shall not exceed the limits shown in column 2 of Table 1 and the remaining differences by no more than twice that value. When more than seven CRMs are used, the values for at least 77 % of the samples shall be within the prescribed limits, while the values for the remainder shall differ by no more than twice that value.

5.4.2.3 For each analyte and each CRM, the average obtained shall be compared to the certified concentrations. Where a certificate value includes a subscript number, that subscript shall be assumed to be a significant number. When seven CRMs are used in the qualification procedure, at least six of the seven averages for each analyte shall not differ from the certified concentrations by more than the value shown in column 3 of Table 1, and the remaining average by more than twice that value. When more than seven CRMs are used in the qualification procedure, at least 77 % of the averages for each analyte shall not differ from the certified concentrations by more than the value shown in column 3 of Table 1, and the remaining average(s) by more than twice that value.

5.4.2.4 The standardization, if needed, used for qualification and for analysis of each constituent shall be determined by valid curve-fitting procedures. A point-to-point, saw-tooth curve that is artificially made to fit a set of data points does not constitute a valid curve-fitting procedure. A complex polynomial drawn through the points is similarly not valid. For the same reason, empirical inter-element corrections may be used, only if  $\leq (N-3)(N-3)/2$  are employed, where N is the number of different standards used. The qualification testing shall be conducted with specimens newly prepared from scratch, including all the preparation stages applicable for analysis of an unknown sample, and employing the reagents currently in use for unknown analyses.

5.4.3 *Partial Results*—Test Methods that provide acceptable results for some analytes but not for others may be used only for those analytes for which acceptable results are obtained.

ASTM C114-23

5.4.4 *Report of Results*—When performing chemical analysis and reporting results for Manufacturer’s Certification, the type of method (Reference or Rapid) and the test method used along with any supporting qualification testing shall be available on request.

5.4.5 *Rejection of Material*—See 4.1, the *Referee Analyses* section, and 5.3, the *Alternative Analyses* section.

5.4.6 *Requalification of a Test Method:*

5.4.6.1 Requalification of a test method shall be required upon receipt of substantial evidence that the test method may not be providing data in accordance with Table 1 for one or more constituents. Such requalification may be limited to those constituents indicated to be in error and shall be carried out prior to further use of the method for analysis of those constituents.

5.4.6.2 Substantial evidence that a test method may not be providing data in accordance with Table 1 shall be considered to have been received when a laboratory is informed that analysis of the same material by Reference Test Methods run in accordance with 4.1.1, the final average of a CCRL sample, a certificate value of an NIST CRM, the assigned value of an alternate CRM, or an accepted value of a known secondary standard differs from the value obtained by the test method in question by more than twice the value shown in column 2 of Table 1 for one or more constituents. When indirect test methods are involved, as when a value is obtained by difference, corrections shall be made for minor constituents in order to put analyses on a comparable basis prior to determining the differences. For any constituents affected, a test method also shall be requalified after any substantial repair or replacement of one or more critical components of an instrument essential to the test method.

5.4.6.3 If an instrument or piece of equipment is replaced, even if by one of identical make or model, or is significantly modified, a previously qualified test method using such new or modified instrument or equipment shall be considered a new method and must be qualified in accordance with 5.4.2.

5.4.7 *Precision and Bias*—Different analytical test methods are subject to individual limits of precision and bias. It is the responsibility of the user to demonstrate that the test methods used at least meet the limits of precision and bias shown in **Table 1**.

## 6. General

### 6.1 *Interferences and Limitations:*

6.1.1 These test methods were developed primarily for the analysis of portland cements. However, except for limitations noted in the procedure for specific constituents, the reference test methods provide for accurate analyses of other hydraulic cements that are completely decomposed by hydrochloric acid, or where a preliminary sodium carbonate fusion is made to ensure complete solubility. Some of the alternative test methods may not always provide accurate results because of interferences from elements which are not removed during the procedure.

NOTE 5—Instrumental analyses can usually detect only the element sought. Therefore, to avoid controversy, the actual procedure used for the elemental analyses should be noted when actual differences with reference procedures can exist. For example,  $P_2O_5$  and  $TiO_2$  are included with  $Al_2O_3$  in the usual wet test method and sulfide sulfur is included in most instrumental procedures with  $SO_3$ .

6.1.2 When using a test method that determines total sulfur, such as most instrumental test methods, sulfide sulfur will be determined with sulfate and included as such. In most hydraulic cements, the difference resulting from such inclusion will be insignificant, less than 0.05 weight %. In some cases, notably slags and slag-containing cements but sometimes other cements as well, significant levels of sulfide may be present. In such cases, especially if there is a question of meeting or not meeting a specification limit or when the most accurate results are desired, analytical test methods shall be chosen so that sulfate and sulfide can be reported separately.

6.1.2.1 Where desired, when using instrumental test methods for sulfate determination, if sulfide has been determined separately, correct the total sulfur results (expressed as an oxide) in accordance with the following calculation:

$$SO_3 = S_{total} - (2.5 \cdot S^-) \quad (1)$$

where:

$SO_3$  = sulfur trioxide excluding sulfide sulfur,

$S_{total}$  = total sulfur in the sample, expressed as the oxide, from instrumental results,

2.5 = molecular ratio of  $SO_3/S^-$  to express sulfur as  $SO_3$ , and

2.5 = molecular ratio of  $SO_3/S^-$  to express sulfur as  $SO_3$ , and

$S^-$  = sulfide sulfur present.

### 6.2 *Apparatus and Materials:*

6.2.1 *Balance*—The analytical balance used in the chemical determinations shall conform to the following requirements:

6.2.1.1 The balance shall be capable of reproducing results within 0.0002 g with an accuracy of  $\pm 0.0002$  g. Direct-reading balances shall have a sensitivity not exceeding 0.0001 g (**Note 6**). Conventional two-pan balances shall have a maximum sensibility reciprocal of 0.0003 g. Any rapid weighing device that may be provided, such as a chain, damped motion, or heavy riders, shall not increase the basic inaccuracy by more than 0.0001 g at any reading and with any load within the rated capacity of the balance.

NOTE 6—The sensitivity of a direct-reading balance is the weight required to change the reading one graduation. The sensibility reciprocal for a conventional balance is defined as the change in weight required on either pan to change the position of equilibrium one division on the pointer scale at capacity or at any lesser load.

6.2.2 *Weights*—Weights used for analysis shall conform to Types I or II, Grades S or O, Classes 1, 2, or 3 as described in Specification **E617**. They shall be checked at least once a year, or when questioned, and adjusted at least to within allowable tolerances for Class 3 weights (**Note 7**). For this purpose each laboratory shall also maintain, or have available for use, a reference set of standard weights from 50 g to 10 mg, which shall conform at least to Class 3 requirements and be calibrated at intervals not exceeding five years by the National Institute of Standards and Technology (NIST). After initial calibration, recalibration by the NIST may be waived provided it can be shown by documented data obtained within the time interval specified that a weight comparison between summations of smaller weights and a single larger weight nominally equal to that summation, establishes that

the allowable tolerances have not been exceeded. All new sets of weights purchased shall have the weights of 1 g and larger made of stainless steel or other corrosion-resisting alloy not requiring protective coating, and shall meet the density requirements for Grades S or O.

NOTE 7—The scientific supply houses do not presently list weights as meeting Specification E617. They list weights as meeting NIST or OIML standards. The situation with regard to weights is in a state of flux because of the trend toward internationalization. Hopefully this will soon be resolved.

NIST Classes S and S-1 and OIML Class F<sub>1</sub> weights meet the requirements of this standard.

6.2.3 *Glassware and Laboratory Containers*—Standard volumetric flasks, burets, and pipets should be of precision grade or better. Standard-taper, interchangeable, ground-glass joints are recommended for all volumetric glassware and distilling apparatus, when available. Wherever applicable, the use of special types of glassware, such as colored glass for the protection of solutions against light, alkali-resistant glass, and high-silica glass having exceptional resistance to thermal shock is recommended. Polyethylene containers are recommended for all aqueous solutions of alkalis and for standard solutions where the presence of dissolved silica or alkali from the glass would be objectionable. Such containers shall be made of high-density polyethylene having a wall thickness of at least 1 mm.

6.2.4 *Desiccators*—Desiccators shall be provided with a good desiccant, such as magnesium perchlorate, activated alumina, or sulfuric acid. Anhydrous calcium sulfate may also be used provided it has been treated with a color-change indicator to show when it has lost its effectiveness. Calcium chloride is not a satisfactory desiccant for this type of analysis.

6.2.5 *Filter Paper*—Filter paper shall conform to the requirements of Specification E832, Type II, Quantitative. When coarse-textured paper is required, Class E paper shall be used, when medium-textured paper is required, Class F paper shall be used, and when retentive paper is required, Class G shall be used.

6.2.6 *Crucibles*:

6.2.6.1 *Platinum Crucibles* for ordinary chemical analysis should preferably be made of pure unalloyed platinum and be of 15 mL to 30 mL capacity. Where alloyed platinum is used for greater stiffness or to obviate sticking of crucible and lid, the alloyed platinum should not decrease in weight by more than 0.2 mg when heated at 1200 °C for 1 h.

6.2.6.2 *Porcelain Crucibles*, glazed inside and out, except outside bottom and rim of 5 mL to 10 mL capacity.

6.2.7 *Muffle Furnace*—The muffle furnace shall be capable of operation at the temperatures required and shall have an indicating pyrometer accurate within  $\pm 25$  °C, as corrected, if necessary, by calibration. More than one furnace may be used provided each is used within its proper operating temperature range.

6.3 *Reagents*:

6.3.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.<sup>5</sup> Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

6.3.2 Use reagent water as defined in 3.2.2 for all tests.

6.3.3 *Concentration of Reagents*:

6.3.3.1 *Prepackaged Reagents*—Commercial prepackaged standard solutions or diluted prepackaged concentrations of a reagent may be used whenever that reagent is called for in the procedures provided that the purity and concentrations are as specified. Verify purity and concentration of such reagents by suitable tests.

6.3.3.2 *Concentrated Acids and Ammonium Hydroxide*—When acids and ammonium hydroxide are specified by name or chemical

<sup>5</sup> *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeia Convention, Inc. (USPC), Rockville, MD; *ACS Reagent Chemicals, Specifications and Procedures for Reagents and Standard-Grade Reference Materials*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.



formula, it shall be understood that concentrated reagents of the following approximate specific gravities or concentrations by weight are intended:

Acetic acid (HC <sub>2</sub> H <sub>3</sub> O <sub>2</sub> )	99.5 %
Hydrochloric acid (HCl)	sp gr 1.19
Hydrofluoric acid (HF)	48 %
Nitric acid (HNO <sub>3</sub> )	sp gr 1.42
Phosphoric acid (H <sub>3</sub> PO <sub>4</sub> )	85 %
Sulfuric acid (H <sub>2</sub> SO <sub>4</sub> )	sp gr 1.84
Ammonium hydroxide (NH <sub>4</sub> OH)	sp gr 0.90

6.3.3.3 The desired specific gravities or concentrations of all other concentrated acids shall be stated whenever they are specified.

6.3.4 *Diluted Acids and Ammonium Hydroxide*—Concentrations of diluted acids and ammonium hydroxide, except when standardized, are specified as a ratio stating the number of volumes of the concentrated reagent to be added to a given number of volumes of water, for example: HCl (1+99) means 1 volume of concentrated HCl (sp gr 1.19) added to 99 volumes of water.

6.3.5 *Standard Solutions*—Concentrations of standard solutions shall be expressed as normalities (*N*) or as equivalents in grams per millilitre of the analyte to be determined, for example: 0.1 *N* Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> solution or K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> (1 mL = 0.004 g Fe<sub>2</sub>O<sub>3</sub>). The average of at least three determinations shall be used for all standardizations. When a material is used as a primary standard, reference has generally been made to the standard furnished by NIST. However, when primary standard grade materials are otherwise available they may be used or the purity of a salt may be determined by suitable tests.

6.3.6 *Nonstandardized Solutions*—Concentrations of nonstandardized solutions prepared by dissolving a given weight of the solid reagent in a solvent shall be specified in grams of the reagent per litre of solution, and it shall be understood that water is the solvent unless otherwise specified, for example: NaOH solution (10 g/L) means 10 g of NaOH dissolved in water and diluted with water to 1 L. Other nonstandardized solutions may be specified by name only, and the concentration of such solutions will be governed by the instructions for their preparation.

6.3.7 *Indicator Solutions*:

6.3.7.1 *Methyl Red*—Prepare the solution on the basis of 2 g of methyl red/L of 95 % ethyl alcohol.

6.3.7.2 *Phenolphthalein*—Prepare the solution on the basis of 1 g of phenolphthalein/L of 95 % ethyl alcohol.

6.4 *Sample Preparation*:

**TABLE 3 Rounding of Reported Results**

Analyte	Decimal Places
SiO <sub>2</sub> (silicon dioxide)	1
Al <sub>2</sub> O <sub>3</sub> (aluminum oxide)	1
Fe <sub>2</sub> O <sub>3</sub> (ferric oxide)	2
CaO (calcium oxide)	1
MgO (magnesium oxide)	1
SO <sub>3</sub> (sulfur trioxide)	2
LoI (loss on ignition)	1
Na <sub>2</sub> O (sodium oxide)	2
K <sub>2</sub> O (potassium oxide)	2
SrO (strontium oxide)	2
TiO <sub>2</sub> (titanium dioxide)	2
P <sub>2</sub> O <sub>5</sub> (phosphorous pentoxide)	2
ZnO (zinc oxide)	2
Mn <sub>2</sub> O <sub>3</sub> (manganic oxide)	3
S (sulfide sulfur)	2
Cl (chloride)	3
IR (insoluble residue)	2
FL (free calcium oxide)	1
CO <sub>2</sub> (carbon dioxide)	1
Water-soluble Alkali	2
Chloroform-soluble Organic Substances	3

6.4.1 Before testing, pass representative portions of each sample through a No. 20 (850  $\mu\text{m}$ ) sieve, or any other sieve having approximately 20 openings/1 in., in order to mix the sample, break up lumps, and remove foreign materials. Discard the foreign materials and hardened lumps that do not break up on sieving or brushing.

6.4.2 By means of a sample splitter or by quartering, the representative sample shall be reduced to a laboratory sample of at least 50 g. Where larger quantities are required for additional determinations such as water-soluble alkali, chloride, duplicate testing, and so forth, prepare a sample of at least 100 g.

6.4.3 Pass the laboratory sample through a U.S. No. 100 sieve (sieve opening of 150  $\mu\text{m}$ ). Further grind the sieve residue so that it also passes the No. 100 sieve. Homogenize the entire sample by again passing it through the sieve.

6.4.4 Transfer the sample to a clean, dry, glass container with an airtight lid and further mix the sample thoroughly.

6.4.5 Expedite the above procedure so that the sample is exposed to the atmosphere for a minimum time.

### 6.5 General Procedures:

6.5.1 *Weighing*—The calculations included in the individual test methods assume that the exact weight specified has been used. Accurately weighed samples, that are approximately but not exactly equal to the weight specified, may be used provided appropriate corrections are made in the calculations. Unless otherwise stated, weights of all samples and residues should be recorded to the nearest 0.0001 g.

6.5.2 *Tared or Weighed Crucibles*—The tare weight of crucibles shall be determined by preheating the empty crucible to constant weight at the same temperature and under the same conditions as shall be used for the final ignition of a residue and cooling in a desiccator for the same period of time used for the crucible containing the residue.

6.5.3 *Constancy of Weight of Ignited Residues*—To definitely establish the constancy of weight of an ignited residue for referee purposes, the residue shall be ignited at the specified temperature and for the specified time, cooled to room temperature in a desiccator, and weighed. The residue shall then be reheated for at least 30 min, cooled to room temperature in a desiccator, and reweighed. If the two weights do not differ by more than 0.2 mg, constant weight is considered to have been attained. If the difference in weights is greater than 0.2 mg, additional ignition periods are required until two consecutive weights agree within the specified limits. For ignition loss, each reheating period shall be 5 min.

6.5.4 *Volatilization of Platinum*—The possibility of volatilization of platinum or alloying constituents from the crucibles must be considered. On reheating, if the crucible and residue lose the same weight (within 0.2 mg) as the crucible containing the blank, constant weight can be assumed. Crucibles of the same size, composition, and history shall be used for both the sample and the blank.

6.5.5 *Calculation*—In all operations on a set of observed values such as manual multiplication or division, retain the equivalent of at least two more places of figures than in the single observed values. For example, if observed values are read or determined to the nearest 0.1 mg, carry numbers to the nearest 0.001 mg in calculation. When using electronic calculators or computers for calculations, perform no rounding, except in the final reported value.

6.5.6 *Rounding Figures*—Rounding of figures to the number of significant places required in the report should be done after calculations are completed, in order to keep the final results substantially free of calculation errors. The rounding procedure should follow the principle outlined in Practice E29.<sup>6</sup> In assessing analyst- and method-qualification in accordance with Section 4, the individual duplicate results, the difference between them, the average of duplicates on CRMs, and the difference of this average from the certificate value shall be left un-rounded for comparison with the required limits. Round results for reporting as shown in Table 3.

NOTE 8—The rounding procedure referred to in 6.5.6, in effect, drops all digits beyond the number of places to be retained if the next figure is less than 5. If it is more than 5, or equal to 5 and subsequent places contain a digit other than 0, then the last retained digit is increased by one. When the next digit is equal to 5 and all other subsequent digits are 0, the last digit to be retained is unchanged when it is even and increased by one when it is odd. For example 3.96 (50) remains 3.96 but 3.95 (50) becomes 3.96.

<sup>6</sup> See also the ASTM Manual on Presentation of Data and Control Chart Analysis, STP 15D, 1976.

6.6 *Recommended Order for Reporting Analyses*—The following order is recommended for reporting the results of chemical analysis of hydraulic cement:

SiO<sub>2</sub> (silicon dioxide)  
 Al<sub>2</sub>O<sub>3</sub> (aluminum oxide)  
 Fe<sub>2</sub>O<sub>3</sub> (ferric oxide)  
 CaO (calcium oxide)  
 MgO (magnesium oxide)  
 SO<sub>3</sub> (sulfur trioxide)  
 Loss on ignition  
 Na<sub>2</sub>O (sodium oxide)  
 K<sub>2</sub>O (potassium oxide)  
 TiO<sub>2</sub> (titanium dioxide)  
 P<sub>2</sub>O<sub>5</sub> (phosphorus pentoxide)  
 ZnO (zinc oxide)  
 Mn<sub>2</sub>O<sub>3</sub> (manganic oxide)  
 Sulfide sulfur  
 Insoluble residue  
 Free calcium oxide  
 CO<sub>2</sub> (Carbon Dioxide)  
 Water-soluble alkali  
 Chloroform—soluble organic substances

## REFERENCE TEST METHODS

### 7. Insoluble Residue (*Reference Test Method*)

#### 7.1 *Summary of Test Method:*

7.1.1 In this test method, insoluble residue of a cement is determined by digestion of the sample in hydrochloric acid followed, after filtration, by further digestion in sodium hydroxide. The resulting residue is ignited and weighed (**Note 9**).

**NOTE 9**—This test method, or any other test method designed for the estimation of an acid-insoluble substance in any type of cement, is empirical because the amount obtained depends on the reagents and the time and temperature of digestion. If the amount is large, there may be a little variation in duplicate determinations. The procedure should be followed closely in order to reduce the variation to a minimum.

7.1.2 When this test method is used on blended cement, the decomposition in acid is considered to be complete when the portland-cement clinker is decomposed completely. An ammonium nitrate solution is used in the final washing to prevent finely-ground insoluble material from passing through the filter paper.

#### 7.2 *Reagents:*

7.2.1 *Ammonium Nitrate Solution* (20 g NH<sub>4</sub>NO<sub>3</sub>/L).

7.2.2 *Sodium Hydroxide Solution* (10 g NaOH/L).

#### 7.3 *Procedure:*

7.3.1 To 1 g of the sample (**Note 10**) add 25 mL of cold water. Disperse the cement in the water and while swirling the mixture, quickly add 5 mL of HCl. If necessary, warm the solution gently, and grind the material with the flattened end of a glass rod for a few minutes until it is evident that decomposition of the cement is complete (**Note 11**). ~~Dilute~~ Next, dilute the solution to 50 mL with hot water (nearly boiling) and heat the covered mixture rapidly to near boiling by means of a high-temperature hot plate. ~~Then~~ Then, using the hot plate, digest the covered mixture for 15 min at a temperature just below boiling (**Note 12**). Filter the solution through a medium-textured paper into a 400 mL beaker, beaker and wash the beaker, paper, and residue thoroughly with hot water, and water. If desired, reserve the filtrate for the sulfur trioxide determination, if desired determination (**Note 13**). Transfer the filter paper and contents to the original beaker, safely add up to ~~400 mL~~ 100 mL of hot (near boiling) NaOH solution

(10 g/L), and macerate the filter paper using a glass rod. After maceration, add any remaining NaOH solution to total 100 mL and proceed to digest covered mixture at a temperature just below boiling for 15 min. During the digestion, occasionally stir the mixture. Acidify the solution with HCl using methyl red or bromocresol purple as the indicator and add an excess of 4 or 5 drops of HCl. Filter through medium-textured paper and wash the residue at least 14 times with hot  $\text{NH}_4\text{NO}_3$  solution (20 g/L) making certain to wash the entire filter paper and contents during each washing. Slowly char and ignite or cover and ignite residue in a weighed platinum or porcelain crucible at 900 °C to 1000 °C, cool in a desiccator, and weigh.

NOTE 10—If sulfur trioxide is to be determined by turbidimetry it is permissible to determine the insoluble residue on a 0.5 g sample. In this event, the percentage of insoluble residue should be calculated to the nearest 0.01 by multiplying the weight of residue obtained by 200. However, the cement should not be rejected for failure to meet the insoluble residue requirement unless a 1 g sample has been used.

NOTE 11—If a sample of portland cement contains an appreciable amount of manganic oxide, there may be brown compounds of manganese which dissolve slowly in cold diluted HCl but rapidly in hot HCl in the specified strength. ~~In all cases, dilute the solution as soon as decomposition is complete.~~

NOTE 12—~~In order to keep the solutions closer to the boiling temperature, it is recommended that these digestions be carried out on an electric hot plate rather than in a steam bath.~~

NOTE 12—~~Continue~~ If continuing with the sulfur trioxide determination (described ~~17.1.2.1—17.1.3~~ by diluting in Section 17 to 250 mL or 200 mL as required, continue at, “Dilute the filtrate to 250 mL and heat to boiling” (17.1.2.1 by the appropriate section).

7.3.2 *Blank*—Make a blank determination, following the same procedure and using the same amounts of reagents, and correct the results obtained in the analysis accordingly.

7.4 *Calculation*—Calculate the percentage of the insoluble residue to the nearest 0.01 by multiplying the weight in grams of the residue (corrected for the blank) by 100.

## 8. Silicon Dioxide (Reference Test Method)

8.1 *Selection of Test Method*—For cements other than portland and for which the insoluble residue is unknown, determine the insoluble residue in accordance with Section 7 of these test methods. For portland cements and other cements having an insoluble residue less than 1 %, proceed in accordance with 8.2. For cements having an insoluble residue greater than 1 % proceed in accordance with 8.3.

### 8.2 Silicon Dioxide in Portland Cements and Cements with Low Insoluble Residue:

8.2.1 *Summary of Test Method*—In this test method silicon dioxide ( $\text{SiO}_2$ ) is determined gravimetrically. Ammonium chloride is added and the solution is not evaporated to dryness. This test method was developed primarily for hydraulic cements that are almost completely decomposed by hydrochloric acid and should not be used for hydraulic cements that contain large amounts of acid-insoluble material and require a preliminary sodium carbonate fusion. For such cements, or if prescribed in the standard specification for the cement being analyzed, the more lengthy procedure in 8.3 shall be used.

8.2.2 *Reagent*—Ammonium chloride ( $\text{NH}_4\text{Cl}$ ).

#### 8.2.3 Procedure:

8.2.3.1 Mix thoroughly 0.5 g of the sample and about 0.5 g of  $\text{NH}_4\text{Cl}$  in a 50 mL beaker, cover the beaker with a watch glass, and add cautiously 5 mL of HCl, allowing the acid to run down the lip of the covered beaker. After the chemical action has subsided, lift the cover, add 1 or 2 drops of  $\text{HNO}_3$ , stir the mixture with a glass rod, replace the cover, and set the beaker on a steam bath for 30 min (Note 14). During this time of digestion, stir the contents occasionally and break up any remaining lumps to facilitate the complete decomposition of the cement. Fit a medium-textured filter paper to a funnel, transfer the jelly-like mass of silicic acid to the filter as completely as possible without dilution, and allow the solution to drain through. Scrub the beaker with a policeman and rinse the beaker and policeman with hot HCl (1+99). Wash the filter two or three times with hot HCl (1+99) and then with ten or twelve small portions of hot water, allowing each portion to drain through completely. Reserve the filtrate and washings for the determination of the ammonium hydroxide group (Note 14).

NOTE 13—A hot plate may be used instead of a steam bath if the heat is so regulated as to approximate that of a steam bath.

Under conditions where water boils at a lower temperature than at sea level: such as at higher elevations, 30 min may not be sufficient to recover all of the silica. In such cases, increase the time of digestion as necessary to get complete recovery of the silica. In no case should this time exceed 60 min.