
Določevanje paladija v zlitinah za nakit iz paladija - Gravimetrična metoda z dimetil glioksimom (ISO 11490:1995)

Determination of palladium in palladium jewellery alloys - Gravimetric method with dimethyl glyoxime (ISO 11490:1995)

Bestimmung von Palladium in Palladium-Schmucklegierungen - Gravimetrische Bestimmung mit Dimethylglyoxim (ISO 11490:1995)

Dosage du palladium dans les alliages de palladium pour la bijouterie-joaillerie - Méthode gravimétrique à la diméthylglyoxime (ISO 11490:1995)

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Ta slovenski standard je istoveten z: EN ISO 11490:1995

ICS:

39.060

Nakit

Jewellery

SIST EN ISO 11490:1998

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EUROPEAN STANDARD

EN ISO 11490

NORME EUROPÉENNE

EUROPÄISCHE NORM

May 1995

ICS 39.060

Descriptors: chemical analysis, determination of content, palladium, gravimetric analysis

English version

**Determination of palladium in palladium jewellery
alloys - Gravimetric method with dimethyl
glyoxime (ISO 11490:1995)**

Dosage du palladium dans les alliages de
palladium pour la bijouterie-joaillerie
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Palladium-Schmucklegierungen - Gravimetrische
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CEN

European Committee for Standardization
Comité Européen de Normalisation
Europäisches Komitee für Normung

Central Secretariat: rue de Stassart, 36 B-1050 Brussels

Foreword

The text of the International Standard ISO 11490:1995 has been prepared by Technical Committee ISO/TC 174 "Jewellery" in collaboration with CEN/TC 283 "Precious metals - Applications in jewellery and associated products". It has been submitted to Parallel Vote and has been approved by CEN on 1995-04-27 as a European Standard.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by November 1995, and conflicting national standards shall be withdrawn at the latest by November 1995.

According to the CEN/CENELEC Internal Regulations, the following countries are bound to implement this European Standard: Austria, Belgium, Denmark, Finland, France, Germany, Greece, Iceland, Ireland, Italy, Luxembourg, Netherlands, Norway, Portugal, Spain, Sweden, Switzerland, and the United Kingdom.

Endorsement notice

The text of the International Standard ISO 11490:1995 was approved by CEN as a European Standard without any modification.

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Annex ZA (normative)**Normative references to international publications
with their relevant European publications**

This European Standard incorporates by dated or undated reference, provisions from other publications. These normative references are cited at the appropriate places in the text and the publications are listed hereafter. For dated references, subsequent amendments to or revisions of any of these publications apply to this European Standard only when incorporated in it by amendment or revision. For undated references the latest edition of the publication referred to applies.

<u>Publication</u>	<u>Year</u>	<u>Title</u>	<u>EN</u>	<u>Year</u>
ISO 9202	1991	Jewellery - Fineness of precious metal Alloys	EN 29202	1992

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INTERNATIONAL STANDARD

ISO
11490

First edition
1995-05-01

Determination of palladium in palladium jewellery alloys — Gravimetric determination with dimethylglyoxime

iTeh STANDARD PREVIEW

(standard from iTeh.ai)
*Dosage du palladium dans les alliages de palladium pour la
bijouterie-joaillerie — Dosage gravimétrique par la diméthylglyoxime*

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Reference number
ISO 11490:1995(E)

ISO 11490:1995(E)**Foreword**

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 11490 was prepared by Technical Committee ISO/TC 174, *Jewellery*.

Annex A of this International Standard is for information only.

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International Organization for Standardization
Case Postale 56 • CH-1211 Genève 20 • Switzerland

Printed in Switzerland

Determination of palladium in palladium jewellery alloys — Gravimetric determination with dimethylglyoxime

1 Scope

This International Standard specifies a gravimetric method for the determination of palladium in palladium jewellery alloys, preferably within the range of fineness stated in ISO 9202.

These alloys which may contain silver, indium, gallium, copper, cobalt, nickel, tin and ruthenium. Coprecipitated elements have to be determined by a suitable method and a correction applied.

2 Normative reference

The following standard contains provisions which, through reference in this text, constitute provisions of this International Standard. At the time of publication, the edition indicated was valid. All standards are subject to revision, and parties to agreements based on this International Standard are encouraged to investigate the possibility of applying the most recent edition of the standard indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 9202:1991, *Jewellery — Fineness of precious metal alloys*.

3 Principle

The sample is dissolved in *aqua regia*. Palladium is precipitated with dimethylglyoxime. If present, silver is separated as silver chloride. The palladium dimethylglyoxime compound is converted to metallic palladium by ignition and the latter is then determined gravimetrically.

4 Reagents

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade and only distilled water or water of equivalent purity.

4.1 Nitric acid, 69 % (m/m), $\rho_{20} = 1,41 \text{ g/cm}^3$.

4.2 Hydrochloric acid, 38 % (m/m), $\rho_{20} = 1,19 \text{ g/cm}^3$.

4.3 Dilute hydrochloric acid, 8,5 % (m/m), $\rho_{20} = 1,04 \text{ g/cm}^3$.

4.4 Dimethylglyoxime solution.

Dissolve 10 g of dimethylglyoxime in 1 litre of ethanol.

4.5 Ammonium chloride.

4.6 Dilute nitric acid, 1,39 % (m/m), $\rho_{20} = 1,00 \text{ g/cm}^3$.

Cautiously add 10 ml of nitric acid (4.1) to 1 000 ml of water and mix.

4.7 Hydrofluoric acid, 40 % (m/m), $\rho_{20} = 1,13 \text{ g/cm}^3$.

4.8 Dilute sulfuric acid, 49 % (m/m), $\rho_{20} = 1,4 \text{ g/cm}^3$.

Cautiously add 100 ml of sulfuric acid [98 % (m/m), $\rho_{20} = 1,86 \text{ g/cm}^3$] to 100 ml of water and mix.

4.9 Reducing gas, such as hydrogen or a hydrogen/nitrogen mixture.