INTERNATIONAL STANDARD



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Dense shaped refractory products — Determination of resistance to sulfuric acid

Produits réfractaires façonnés denses — Détermination de la résistance à l'acide sulfurique (standards.iteh.ai)

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Foreword

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Dense shaped refractory products — Determination of resistance to sulfuric acid

1 Scope and field of application

This International Standard specifies a method for determining the resistance of dense shaped refractory products to attack by sulfuric acid.

NOTE — Sulfuric acid is used since it gives results which are typical of the results of exposing refractory materials to many acids other than hydrofluoric acid.

2 References

ISO 383, Laboratory glassware Conterchangeable conical PREVIEW ground joints. (standards.ifturns in the coil (see ISO 4799).

list).

0,001 g.

thermometer.

ISO 565, Test sieves — Woven metal wire cloth and perforated

plate — Nominal sizes of apertures.

ISO 8890:1988

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ISO 1770, Solid-stem general purpose thermometers. (length approximately 4:10 thm): (see ISO 1770).c-4:102fb487c/iso-8890-1988

ISO 1773, Laboratory glassware — Boiling flasks (narrownecked). **4.8** Sand bath or oil bath.

ISO 4799, Laboratory glassware - Condensers.

ISO 5022, Shaped refractory products — Sampling and acceptance testing.

3 Principle

The test sample, crushed in a specified manner, is subjected for 6 h to attack by 70 % (m/m) boiling sulfuric acid, and the resultant mass loss is determined and expressed as a percentage of the initial mass of the dry material.

4 Apparatus

Ordinary laboratory apparatus and

4.1 Suitable mechanical crushing device, preferably not steel.

4.2 Woven metal wire cloth sieve, 0,80 mm aperture, conforming to the requirements of ISO 565 (supplementary size list).

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4.9 Porcelain filter crucible.

4.10 Thermostatically controlled oven, capable of maintaining a temperature of 110 \pm 5 °C.

4.3 Woven metal wire cloth sieve, 0.63 mm aperture, con-

forming to the requirements of ISO 565 (supplementary size

4.4 Balance, capable of weighing 25 g to the nearest

4.5 Round-bottomed flasks, of capacity 500 ml (see ISO 1773), each with a short, narrow neck, equipped with a ground glass stopper (see ISO 383) for the insertion of a

5 Reagents

During the test, use only reagents of recognized analytical grade and only distilled water or water of equivalent purity.

5.1 Sulfuric acid, 70 % (m/m), $\rho_{15} = 1,615$ g/cm³.

5.2 Barium chloride, 50 g/l solution.

6 Test samples

6.1 The number of samples to be taken shall be in accordance with ISO 5022 or with another standard sampling plan.

6.2 From each sample, take two pieces of a total mass of approximately 250 g, one from the centre and one from an edge.

6.3 Grind the two pieces together using the crushing device (4.1), sieving the material several times and re-grinding the residues until all the test material has passed through the 0,80 mm aperture sieve (4.2).

 ${\sf NOTE}$ — The material of the mortar should be harder than the test material. If a steel mortar has to be used, treat the crushed material carefully with a magnet.

6.4 Sieve the material through the 0,63 mm aperture sieve (4.3). Clear the residue on the sieve of all dust particles by washing with distilled water. This residue forms the test material.

 $\mathsf{NOTE}-\mathsf{A}$ systematic error will be introduced if grain sizes are used other than between the limits set in 6.3 and 6.4.

6.5 Dry the residue on the 0,63 mm aperture sieve in the oven (4.10), controlled at 110 \pm 5 °C, until constant mass is reached. Before each weighing, allow the sieve with contents to cool to ambient temperature in a desiccator. Weigh to the nearest \pm 0,001 g.

7 Procedure

where

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7.1 It is advisable to carry out at least two tests in parallel.

(standardsmittis the mass, in grams, of the residue.

 $\frac{-m_2}{-}$ × 100

7.2 For each test, weigh, to the nearest 0,001 g, about 20 g of the dried material (mass m_1).

ISO 8890:1488 9 Https://standards.iteh.ai/catalog/standards/sist/12e419f2-b91f-4947-a6ac-

7.3 Place the weighed test material in a round-bottomed flask b487c/The test report shall include the following information:

(4.5) and cover it with 200 ml of the sulfuric acid (5.1). Attach a coil condenser (4.6) and insert a thermometer (4.7) so that it is immersed at least 15 mm in the liquid.

7.4 Over a period of about 30 min, bring the contents of the flask to the boiling point in the sand or oil bath (4.8) [the boiling point of 70 % (m/m) sulfuric acid is approximately 170 °C]. Note the temperature of the liquid after boiling has begun.

7.5 Keep the liquid boiling lightly for a period of 6 h. Note the temperature of the liquid at the end of this time.

7.6 After the liquid has boiled for 6 h, take the flask out of the bath and allow it to cool for 1 h. Decant the clear acid floating on top of the sample. Cautiously pour in approximately 500 ml of distilled water, and wash the whole contents of the flask into the porcelain filter crucible (4.9), previously dried and

- a) the testing establishment;
- b) the date of the test;

c) a reference to this International Standard, i.e. "Determination of resistance to acids in accordance with ISO 8890";

weighed to the nearest 0,001 g, using an aspirator to aid filter-

ing. Wash the residue in the filter crucible with distilled water

until the filtrate remains unclouded when a few drops of the

7.7 Dry the crucible containing the residue in the oven, con-

trolled at 110 \pm 5 °C, until constant mass is reached. Before each weighing, allow the crucible and its contents to cool to

Weigh the crucible and its contents to the nearest 0,001 g (the

Calculate the loss of mass of the test material, as a percentage

barium chloride solution (5.2) are added.

ambient temperature in a desiccator.

Expression of results

of its initial mass, using the formula

net mass is m_2).

 m_1

d) the material tested (manufacturer, type, batch number, etc.);

e) the temperature of the acid shortly after boiling began and shortly before the end of the test;

f) the individual values and the calculated mean value of the proportionate loss of mass of each sample tested.

NOTE — The individual values are used in determining the mean value. The mean value is used in further statistical analysis, for example in accordance with ISO 5022.

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