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# Standard Test Method for Foam In Aqueous Media (Bottle Test)<sup>1</sup>

This standard is issued under the fixed designation D 3601; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

## 1. Scope

1.1 This test method covers the measurement of the increase in volume of a low-viscosity aqueous liquid (less than 3 cSt at  $40^{\circ}$ C) due to its tendency to foam under low shear conditions.

Note 1—Foam under high shear is covered by Test Method D 3519 which uses a commercial blender.

1.2 The values stated in SI units are to be regarded as the standard. The values given in parentheses are provided for information only.

1.3 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.Figs. 1-3

### 2. Referenced Documents

2.1 ASTM Standards:

- D 1126 Test Method for Hardness in Water<sup>2</sup>
- D 3519 Test Method for Foam In Aqueous Media (Blender Test)<sup>3</sup>

### 3. Summary of Test Method

3.1 The increase in volume is determined by the increase in total height of the test fluid, including foam, after vigorous shaking of the emulsion at  $25 \pm 1^{\circ}$ C ( $77 \pm 1.8^{\circ}$ F).

NOTE 2—Water that is normally used to reduce the concentrate to a working consistency can be used. However, if this type of water is not desirable or available, two other types of water may be used to prepare the test liquid: (1) distilled water or (2) distilled water with a subsequent seeding of the test liquid with synthetic hard water.

#### 4. Significance and Use

4.1 The results obtained by this test method are useful as guides in determining the tendency of a water-based metalworking coolant to produce foam under low shear conditions. No correlation with changes in heat transfer, pumpability, or other factors affected by foam is intended. The foam generated by any given industrial process depends on the method by



FIG. 1 Boston Round Bottle.

which the foam is generated and may not be directly proportional to that produced by this controlled laboratory test method. Further, the foam generated at the specified test temperature will not necessarily predict the foaming tendency of the liquid (that is, metalworking coolant) at some other use temperature.

#### 5. Apparatus

5.1 *Water Bath*, constant-temperature, suitable to hold several bottled emulsions at  $25 \pm 1^{\circ}$ C ( $77 \pm 1.8^{\circ}$ F) for 1 to 2 h.

Note 3—A common household dishpan is satisfactory when the test temperature is close to room temperature.

5.2 Stop Watch or Timer, capable of measuring 5 min  $\pm$  0.2 s.

5.3 *Bottles*, clean or new, clear glass, 16-oz (narrow mouth), with screw neck.<sup>4</sup> The 16-oz bottle is  $6^{41}/_{64}$  in. (169 mm) tall and has a maximum diameter of 2  $^{31}/_{32}$  in. (75 mm). The outside neck is  $^{9}/_{32}$  in. (7.1 mm) and the shoulder radius is 1 in. (25.4 mm).

- 5.4 Syringe or Transfer Pipet.<sup>5</sup>
- 5.5 Rule, millimetre, at least 150 mm in length.

#### 6. Materials

6.1 Distilled Water.

<sup>&</sup>lt;sup>1</sup> This test method is under the jurisdiction of ASTM Committee D-2 on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee D02.LO.01 Metal Removal Fluids and Lubricants.

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<sup>&</sup>lt;sup>2</sup> Annual Book of ASTM Standards, Vol 11.01.

<sup>&</sup>lt;sup>3</sup> Annual Book of ASTM Standards, Vol 05.02.

<sup>&</sup>lt;sup>4</sup> 16-oz Round Packer, Brockway Glass Co., Mold No. 1094 or 16-oz Boston Round, as shown in Fig. 1 is suitable for this purpose. Bottle takes Cap No. 28. Standard Glass Container Series No. 80 specifies approximate weight of glass, etc., on drawing No. C-8010 obtainable from Glass Container Manufacturers' Institute Inc., 330 Madison Ave., New York, NY 10017.

<sup>&</sup>lt;sup>5</sup> Disposable pipets, Thomas Scientific Co. Saedsboro, NJ 08085-0099, No. 8937-R62, Corning No. 7077 or Fisher No. 13-671-108E with attached 25 to 50-cm<sup>3</sup> (1 to 2-oz) capacity rubber bulb have been found suitable for this purpose as have Becton-Dickson disposable syringes No. 21G, without needle.

#### 1. Preparation of Emulsion

1.1 (7.2)	Sample description	
1.2 (7.2)	Concentration, %	
1.3 (7.3)	Source of water used	
1.4 (7.3)	Water hardness, ppm	
• • •	Method of preparing emulsion	
2. Test Data	ta	
2.1 (7.5)	) Temperature at start of test	0°
	) Initial height (/)	mm
	) Maximum total height at zero time (M)	
		mm

2.4 (7.11) Residual total height after 5 min (R)	mm
2.5 (7.10) Time to defoam to 10 mm (to nearest 1/2 min)	min
2.6 (7.12) Temperature at end of test	°C
2.7 (8.1) Maximum foam, $F_m = M - I$	mm
2.8 (8.2) Residual foam, Fr = R - I	

Caution—The round robin on this test used distilled water and a controlled synthetic hard water to make data comparative to the products under test at different places and at different times. Care must be exercised when natural waters are used that comparative samples are used in exactly the same water, taken at the same time from the same source. (For instance, well waters can change in hardness rapidly depending on the change in demand within the hour.)

NOTE—Numbers in parentheses indicate the section within the body of the method where the observations to be recorded are made.

FIG. 2 Suggested Test Form for Recording Data.

#### $F_{m_i}$ Millimetres Maximum Foam = M - I

						Sam	ole Number					
	1	2	3	4	5	6	7	8	9	10	11	12
X <sub>1</sub>	30	35	20	10	65 <sup>A</sup>	65 <sup>A</sup>	65 <sup>A</sup>	5	60	20	15	15
X <sub>2</sub>	30	40	20	10	65 <sup>A</sup>	65 <sup>A</sup>	65 <sup>A</sup>	10	65	20	10	10
<i>X</i> <sub>1</sub>	39	49	20	20	101	101	101	21	101	11	10	10
X <sub>2</sub>	48	48	19	19	100	100	101	12	101	14	20	10
					F <sub>r</sub> , Millimetr	es Residual	Foam = R					
<i>X</i> <sub>1</sub>	0	0	0	0	0	00	45	0	55	5	0	0
$X_2$	0	0	0	0	0	0	45	0	66	5	5	0
$x_1$	0	0	0	0	0	0	101	0	101	7	4	10
X <sub>2</sub>	0	0	0	0	0	0	101		101	8	3	4
				T, Tim	e to Defoan	n to 10-mm	Net Foam L	evel, min				
<i>X</i> <sub>1</sub>	0.5	0.5	0.25 <sup>B</sup>	0	2.0	2.5	5+	evie	5+	1.0	0	0
X <sub>2</sub>	0.25 <sup>B</sup>	0.5	0.25 <sup>B</sup>	0	1.5	3.0	5+	0	5+	1.25	0	0
$x_1$	0.25 <sup>B</sup>	0.25 <sup>B</sup>	0.25 <sup>B</sup>	0.25 <sup>B</sup>	1.5	2.0	5+	0.25 <sup>B</sup>	5+	0.25 <sup>B</sup>	0.25 <sup>B</sup>	1.0
<i>X</i> <sub>2</sub>	0.25 <sup><i>B</i></sup>	0.25 <sup>B</sup>	0.25 <sup>B</sup>	0.25 <sup>B</sup>	1.5	2.0	5+	0.25 <sup><i>B</i></sup>	5+	0.25 <sup>B</sup>	0.25 <sup>B</sup>	0.25 <sup>B</sup>
	x <sub>2</sub> x <sub>1</sub> x <sub>2</sub> x <sub>1</sub> x <sub>2</sub> x <sub>1</sub> x <sub>2</sub> x <sub>1</sub> x <sub>2</sub> x <sub>1</sub> x <sub>2</sub>	$\begin{array}{cccc} x_2 & 30 \\ x_1 & 39 \\ x_2 & 48 \\ x_1 & 0 \\ x_2 & 0 \\ x_1 & 0 \\ x_2 & 0 \\ x_1 & 0 \\ x_2 & 0 \\ \end{array}$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

<sup>A</sup> Maximum foam possible in bottle.

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<sup>B</sup>Request was for nearest 0.5 min. Where <0.5 min is shown, it was recorded as 0.25 min.  $x_1$  and  $x_2$  are duplicate test results.

# Sample Descriptions

Sample 1 L-1-3A in distilled water - 10 % butyl Cellosolve® (trademark of Union Carbide)

3 L-1-3B in distilled water - 10 % butyl Carbitol® (trademark of Union Carbide)

5 L-1-3C in distilled water - Sample 3 plus nonionic ether

7 L-1-3D in distilled water - 5 % commercial long oil soluble coolant

9 L-1-3E in distilled water – 3 % commercial synthetic coolant

11 L-1-3F in distilled water - 3 % commercial synthetic plus defoamer

2, 4, 6, 8, 10, 12 - Same in hard water, respectively

#### FIG. 3 Sample Data (Bottle Method).

6.2 *Hard Water*, 20 000 ppm prepared by dissolving 29.4 g of reagent grade (ACS standard)  $CaCl_2 \cdot 2H_2O$  in 1 litre of freshly boiled distilled water (used only where distilled water is used as in Note 5).

#### 7. Procedure

7.1 Fix a label or other means of marking total heights to the outside of the bottle.

NOTE 4—Waterproof medical adhesive tape makes satisfactory waterproof labels, which readily take pencil marks.<sup>6</sup>

7.2 Using the manufacturer's recommended procedure, prepare 200 ml of emulsion at the recommended concentration and pour the emulsion into the 16-oz round bottle (or equivalent 500-cm<sup>3</sup> bottle).

7.3 When natural water is used, record water hardness (using Test Method D 1126), source, and date obtained.

NOTE 5—In the absence of manufacturers' recommendations, place 190 ml of distilled water in the 16-oz round bottle. Using a 10-ml syringe or pipet, accurately measure 10 ml of the coolant concentrate into this water. Immediately cap and shake to thoroughly mix this 5 % emulsion or solution.

7.4 Store this test liquid in the constant-temperature bath for a minimum of 1 h and a maximum of 2 h at  $25 \pm 1^{\circ}$ C (77  $\pm 1.8^{\circ}$ F).

NOTE 6—Storage in water to level above air-liquid interface of emulsion will suffice to stabilize emulsion (or solution) temperature.

7.5 Using any suitable thermometer, measure and record the

<sup>&</sup>lt;sup>6</sup> One suitable transparent tape is sold under the brand "Scalefix Scales" by Bel-Art Products Inc., Pequannock, NJ, Catalog No. H-2075 (1974).