



Standard Test Method for Solvents Analysis in Hazardous Waste Using Gas Chromatography¹

This standard is issued under the fixed designation D 5830; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method is used to determine qualitatively and quantitatively the presence of the following compounds in waste samples using gas chromatography. This test method is designed for use as a screening method with a typical reporting level of 0.1 %.

Dichodifluoromethane	Tetrahydrofuran
Trichlorofluoromethane	Acetone
1,1,2-Trichloro-1,2,2-trifluoroethane	Methyl Ethyl Ketone
Methanol	MIBK
Ethanol	Cyclohexanone
Isopropanol	Ethyl Acetate
<i>n</i> -Propanol	Propyl Acetate
Isobutanol	Butyl Acetate
<i>n</i> -Butanol	Benzene
<i>tert</i> -Butanol	Toluene
Methylene Chloride	Ethylbenzene
Chloroform	Xylenes
Carbon Tetrachloride	Styrene
1,1-Dichloroethane	Chlorobenzene
1,2-Dichloroethane	Dichlorobenzenes
1,2-Dichloropropane	Nitrobenzene
1,1-Dichloroethylene	Fluorobenzene
1,2-Dichloroethene	<i>n</i> -Propyl Benzene
1,1,1-Trichloroethane	Isopropyl Benzene
Tetrachloroethylene	Isobutyl Benzene
Trichloroethylene	<i>n</i> -Butyl Benzene
Tetrachloroethane	2-Ethoxyethanol
Cyclopentane	2-Butoxyethanol
Pentane	2-Ethoxyethanol Acetate
Hexane	2-Methoxyethanol
Heptane	Bromoform
Cyclohexane	Carbitol
Isooctane	Ethyl Ether
Nitropropane	1,4-Dioxane
Ethanolamine	Diacetone Alcohol
Nitromethane	Acetonitrile
Ethylene Chloride	Pyridine
Benzyl Chloride	Toluidine
	Ethylene Glycol
	Propylene Glycol

1.1.1 This compound list is a compilation of hazardous solvents and other constituents that are routinely seen in hazardous waste samples.

1.2 The scope of this test method may be expanded to include other volatile and semivolatile organic constituents.

1.2.1 Hydrocarbon mixtures such as kerosene and mineral spirits.

1.2.2 High-boiling organics, defined here as compounds which boil above *n*-Hexadecane.

1.2.3 Other organics that the analyst is able to identify, either through retention time data or gas chromatography/mass spectrometric (GC/MS) analysis.

1.3 Gas chromatographic methods are recommended for use only by, or under close supervision of, an experienced analyst.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 ASTM Standards:

D 1193 Specification for Reagent Water²

2.2 EPA Document:

Gas Chromatography/Mass Spectrometry Method 8260, Test Methods for Evaluating Solid Waste Physical/Chemical Methods, SW-846, Third Edition, Final Update 1, July 1992³

3. Summary of Test Method

3.1 Waste samples are analyzed by direct injection, or by carbon disulfide, *M*-Pyrol, or other suitable solvent extraction and injection of the extract into a gas chromatograph. Detection is achieved using a detector which is specific for the needed application, for example, flame ionization detector (FID), electron capture detector (ECD), thermal conductivity detector (TCD), photoionization detector (PID), or mass selective detector (MSD). This test method may be expanded to utilize other detector types not previously mentioned.

4. Significance and Use

4.1 This test method is useful in identifying the major solvent constituents in hazardous waste samples. This test method is designed to support field or site assessments,

¹ This test method is under the jurisdiction of ASTM Committee D-34 on Waste Management and is the direct responsibility of Subcommittee D34.02 on Physical and Chemical Characterization.

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² *Annual Book of ASTM Standards*, Vol 11.01.

³ Available from the Superintendent of Documents, U.S. Government Printing Office, Washington, DC 20402.

recycling operations, plant operations, or pollution control programs.

5. Interferences

5.1 Interferences may be encountered from any number of organic compounds that respond in the detector. Also, closely eluting components may complicate identification based solely on retention time. When these types of interferences are encountered, the analyst must rely on other sources of information for positive identification, such as:

5.1.1 Gas chromatography/mass spectrometric (GC/MS) confirmation, see EPA Method 8260, direct injection technique;

5.1.2 Use of confirmation column, or confirmatory detector;

5.1.3 Use of varying temperature programs or standard comparison, or both;

5.1.4 Sample history, for example, any information available from the waste generator; and,

5.1.5 Physical characteristics, for example, flammability, specific gravity, or miscibility with water.

5.2 Interferences may also be encountered from syringe carryover. Immediately following each injection, the syringe

should be thoroughly rinsed with carbon disulfide, or *M*-Pyrol. Other solvents such as methanol may be used as rinse solvents if sample types necessitate their use, but be aware that carryover and possible interferences may occur if the rinse solvent is not completely cleaned from the syringe before reuse. Before each injection the syringe must be thoroughly rinsed with the sample to be injected, where the first two pumps are flushed into a separate waste receptacle.

5.3 When carbon disulfide (CS₂) is used to extract solids or sludges that contain significant amounts of water, low recovery of the water miscible solvents may result.

5.4 Some grades of CS₂ may contain trace amounts of benzene.

5.5 *M*-Pyrol seems to degrade slowly with time. The low-level degradation products interfere with some late eluting compounds on some columns (approximately five small peaks).

5.6 Interference from the CS₂ solvent peak may occur if using a TCD.

5.7 When using a TCD, be aware that water, as well as oxygenated compounds, for example, MEK, MIBK, may

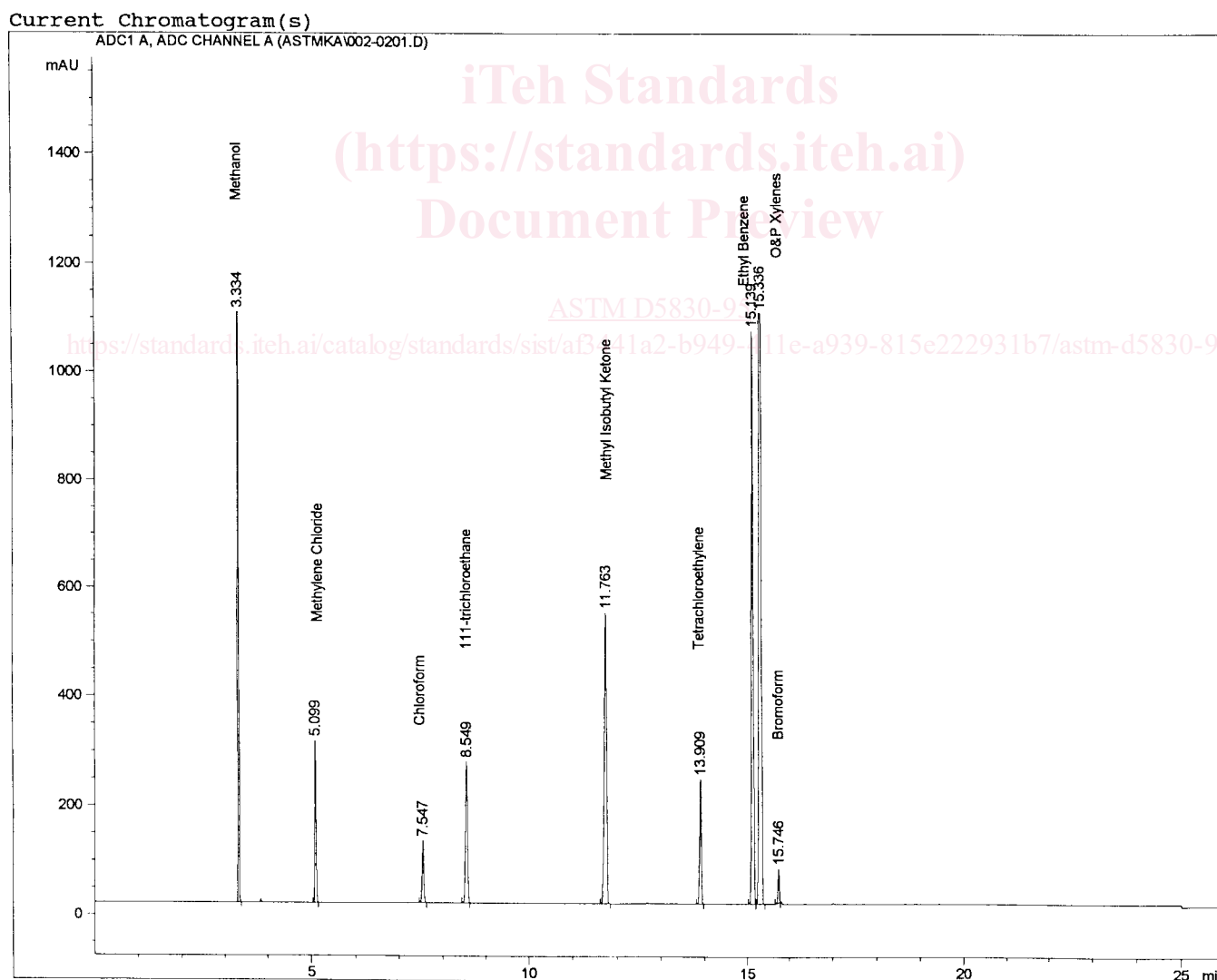


FIG. 1 Daily QC Standard FID/DB-1701

suppress detector response.

5.8 If an electrolytic conductivity detector (ECD) must be used, be aware that CS₂, *M*-Pyrol, and high concentrations of halogenated compounds may overload and possibly damage the detector. It is recommended that the ECD be used only when very low detection levels of halogenated compounds are expected and direct injection of the sample is possible.

6. Apparatus

6.1 *Gas Chromatograph System*—Equipped with capillary or packed column injection ports, or both, detector, and data system.

6.2 *Recommended Chromatographic Columns:*

6.2.1 *Capillary; Microbore or Megabore.*

6.2.1.1 DB-1701, 30M × 0.25-mm inside diameter, 0.25-μm film thickness.

6.2.1.2 DB-624, 30M × 0.3-mm inside diameter, 1.8-μm film thickness.

6.2.2 *Packed: Stainless Steel or Glass.*

6.2.2.1 1% SP-1000, 60/80 Carbowax B, 8-ft by 1/8-in. inside diameter.

6.2.2.2 10% SP-2100, 100/120 Chromosorb WHP, 2M × 2 mm ID.

NOTE 1—These columns are recommended and have shown to give good results. Operating conditions for each is listed in Section 10. Equivalent or alternative columns, or both, may be used depending on application.

6.3 *Glass Screw-Cap Vials or Equivalent*—To collect samples and store standards. Polytetrafluoroethylene or other inert material should be used for the cap liner.

6.4 *Microsyringes*, 1.0, 10, and 100 μL.

6.5 *Analytical Balance*, accurate to 0.0001 g.

6.6 *Pipettes*, glass, disposable, or volumetric micropipettor or equivalent.

6.7 *Microdisk Filters*, 0.45, 1.0, or 5.0 μm, optional.

6.8 *Centrifuge*, optional.

6.9 *Vortex-Type Mixer*.

7. Reagents and Materials

7.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society where such specifications are available.⁴ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Purity of Water*— Unless otherwise indicated, references to water shall be understood to mean reagent water as defined by Type II of Specification D 1193.

7.3 *Nitrogen or Helium (High Purity)*—For carrier and makeup gases. Air and hydrogen (high purity) for fuel gases.

⁴ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmaceutical Convention, Inc. (USPC), Rockville, MD.

Gases may be obtained from a gas generator if available, through purification of a lower grade, or from a high-purity tank supply.

7.4 *Carbon Disulfide, CS₂*—Chromatography grade.

7.5 *M-Pyrol, C₆H₆NO*—Available through several chemical suppliers and sources as 1-methyl-2-pyrrolidone.

7.6 *Individual Standards for Each Component of Interest*—99% purity available from many vendors.

8. Standard Preparation

8.1 *Stock Standard Solutions*—Stock standards are prepared from pure standard materials. It is recommended that the standards be prepared so that each component is 5 to 10% by weight. The stock standards must be prepared by directly weighing each component. For extremely volatile components, such as ether and freons, it is recommended that a new stock standard be prepared daily or as needed. If a dilution solvent is needed when preparing the stock standards, use the same solvent used for sample extraction or dilution in Section 7.

NOTE 2—Due to the incompatibility of some standard compounds, that is, some compounds are not miscible with each other, and also because of the number of compounds typically looked for in a single chromatographic run, it is advisable to prepare 3 or 4 standard solutions each composed of 10 to 15 compounds. A set of standard chromatograms and a retention timetable should be available for reference.

8.2 *Secondary Working Standards*—These are prepared from stock standard solutions using the appropriate solvent. Secondary standards should encompass the linear range of the GC system.

NOTE 3—Linear response and range must be established with all detectors and chromatography systems used for quantitation. All calibration and sample analysis must be done within the established linear range.

8.3 *Calibration Check Standard*—A calibration check standard should be prepared. The standard mixture should provide a good overall check of the GC/detector system. The compounds should cover the major compound types, for example, alcohols, aromatics, aliphatics, ketones, and halogenates. A typical calibration check standard flame ionization detector (FID) chromatogram is shown in Fig. 1.

9. Sample Collection, Preservation, and Handling

9.1 Sample collection should be in accordance with appropriate sampling protocols.

9.2 Samples should be collected in glass containers, that have tightly sealing caps. If very volatile organics are of particular interest, the headspace in the container should be kept to a minimum.

9.3 *Sample Transfer Implements*—Implements are required to transfer portions of waste samples from the sample containers to the laboratory containers. Liquid samples may be transferred using disposable pipets. Solids and semisolids may be transferred using a conventional laboratory spatula.

9.4 Samples shall be handled maintaining safe laboratory practices. Any samples with special hazards must be appropriately labeled.

9.5 Unused sample material, laboratory dilutions, and waste from the samples may be regulated. Consult your specialist or the regulations, or both, for guidance in the proper handling and disposal of laboratory wastes.

10. Procedure

10.1 Sample Preparation:

10.1.1 Analyze liquid matrices with relatively low viscosity using direct injection into the GC, either as received or after dilution with CS₂, M-Pyrol, or other suitable solvent.

10.1.2 Analyze solid or semisolid samples as follows:

10.1.2.1 For carbon disulfide or M-Pyrol preparation, weigh 3 g of the waste sample in a 15-mL glass vial. Add 3 g of carbon disulfide or M-Pyrol to the vial and the mixture is vortexed vigorously. After allowing the solids to settle, inject the CS₂ or M-Pyrol extract into the GC.

10.1.2.2 Use alternate sample sizes and extraction solvent weights if necessary. Actual sample size and solvent weight must be recorded in the appropriate sample preparation log book. It is essential for accurate waste sample analysis that sample size be sufficient to ensure a representative sample. If alternate sample size or extraction solvent volumes, or both, are used, this must be reflected in the calculations under the dilution factor in Section 11.

10.1.3 Multiple phases or layers are typically present in

hazardous waste samples. Depending on treatment or process requirements, it may be necessary to analyze each phase or layer individually.

10.2 Recommended GC Operating Conditions:

10.2.1 For Capillary DB-1701 with FID

Column flow rate	1–1.5 mL/min
Make-up gas flow rate	30–60 mL/min
Split flow	60 cm ³ /min
Injector temperature	250°C
Detector temperature	250°C
Airflow (FID)	Approximately 300 mL/min
Hydrogen flow (FID)	Approximately 30 mL/min
Initial oven temperature	35°C
Initial time	6 min
Level 1 rate	6°C/min
Level 1 final value	180°C
Level 2 rate	10°C/min
Level 2 final value	230°C
Run time	40 min
Threshold	4 units
Peak width	0.04 min

NOTE 4—Typical chromatograms are shown in Figs. 2-5.

10.2.2 For Capillary DB-624 with FID

Current Chromatogram(s)

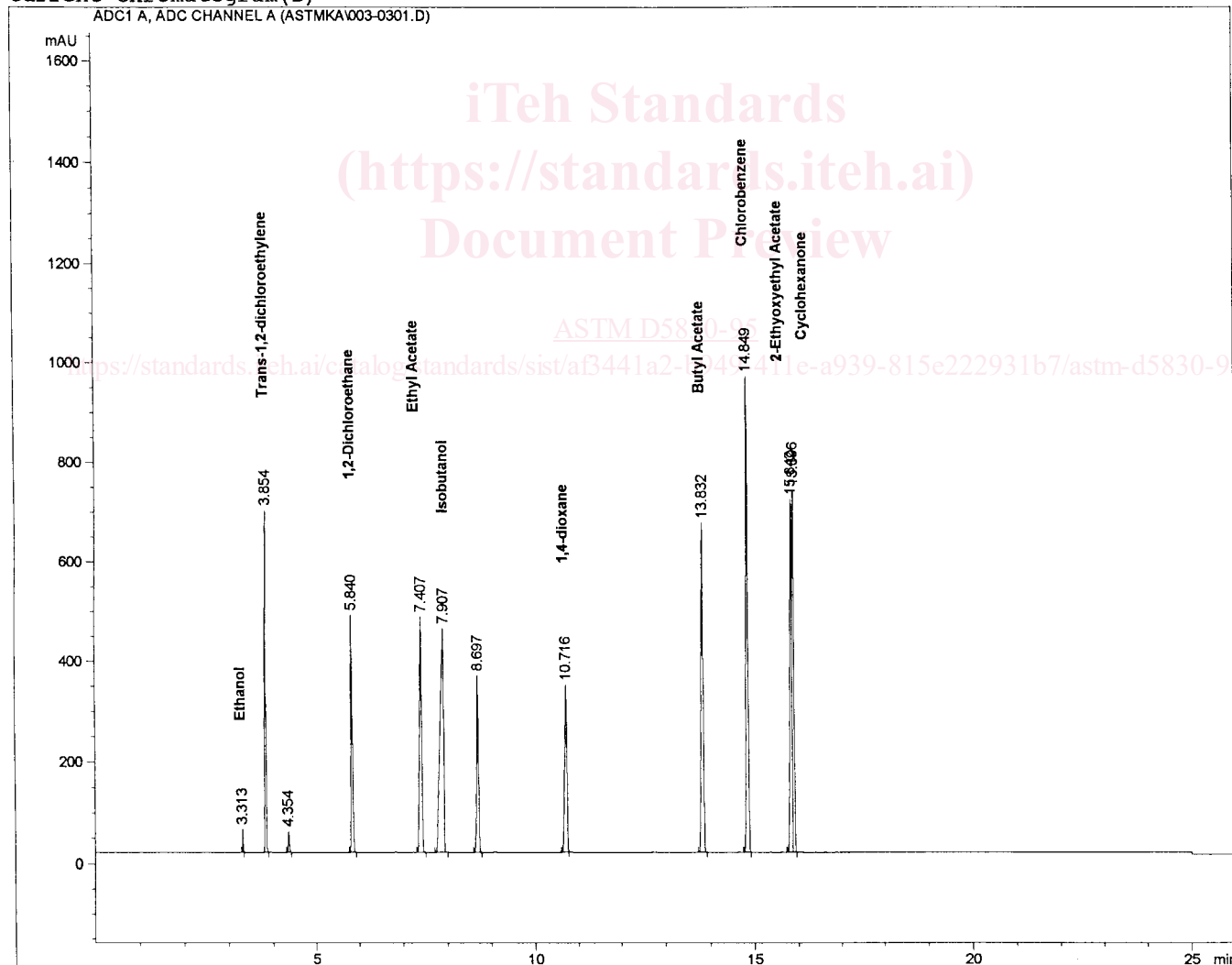


FIG. 2 Solvent Standard #1 FID/DB-1701