Designation: E 458 – 72 (Reapproved 1996)$^{\epsilon 1}$

Standard Test Method for Heat of Ablation[1]

This standard is issued under the fixed designation E 458; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

ϵ^1 NOTE—Section 13 was added editorially in May 1996.

1. Scope

1.1 This test method covers determination of the heat of ablation of materials subjected to thermal environments requiring the use of ablation as an energy dissipation process. Three concepts of the property are described and defined: cold wall, effective, and thermochemical heat of ablation.

1.2 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*

E 285 Test Method for Oxyacetylene Ablation Testing of Thermal Insulation Materials[2]
E 341 Practice for Measuring Plasma Arc Gas Enthalpy by Energy Balance[2]
E 377 Practice for Internal Temperature Measurements in Ablative Materials[2]
E 422 Test Method for Measuring Heat Flux Using a Water-Cooled Calorimeter[2]
E 457 Test Method for Measuring Heat-Transfer Rate Using a Thermal Capacitance (Slug) Calorimeter[2]
E 459 Test Method for Measuring Heat-Transfer Rate Using a Thin-Skin Calorimeter[2]
E 470 Method for Measuring Gas Enthalpy Using Calorimetric Probes[3]
E 471 Test Method for Obtaining Char Density Profile of Ablative Materials by Machining and Weighing[2]
E 511 Test Method for Measuring Heat Flux Using a Copper-Constantan Circular Foil, Heat-Flux Gage[2]

3. Terminology

3.1 *Descriptions of Terms Specific to This Standard:*

3.1.1 *heat of ablation*—a property that indicates the ability of a material to provide heat protection when used as a sacrificial thermal protection device. The property is a function of both the material and the environment to which it is subjected. In general, it is defined as the incident heat dissipated by the ablative material per unit of mass removed, or

$$Q^* = q/m \tag{1}$$

where:
Q^* = heat of ablation, kJ/kg,
q = incident heat transfer rate, kW/m^2, and
m = total mass transfer rate, kg/m^2·s.

3.2 The heat of ablation may be represented in three different ways depending on the investigator's requirements:

3.3 *cold-wall heat of ablation*—The most commonly and easily determined value is the cold-wall heat of ablation, and is defined as the incident cold-wall heat dissipated per unit mass of material ablated, as follows:

$$Q^* = q_{cw}/m \tag{2}$$

where:
Q^*_{cw} = cold-wall heat of ablation, kJ/kg,
q_{hw} = heat transfer rate from the test environment to a cold wall, kW/m^2, and
m = total mass transfer rate, kg/m^2·s.

The temperature of the cold-wall reference for the cold-wall heat transfer rate is usually considered to be room temperature or close enough such that the hot-wall correction given in Eq 7 is less than 5 % of the cold-wall heat transfer rate.

3.4 *effective heat of ablation*—The effective heat of ablation is defined as the incident hot-wall dissipated per unit mass ablated, as follows:

$$Q^*_{eff} = q_{hw}/m \tag{3}$$

where:
Q^*_{eff} = effective heat of ablation, kJ/kg,
q_{hw} = heat transfer rate from the test environment to a nonablating wall at the surface temperature of the material under test, kW/m^2, and
m = total mass transfer rate, kg/m^2·s.

3.5 *thermochemical heat of ablation*—The thermochemical heat of ablation is defined as the incident hot-wall heat dissipated per unit mass ablated, corrected for reradiation heat rejection processes and material eroded by mechanical removal, as follows:

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[1] This test method is under the jurisdiction of ASTM Committee E-21 on Space Simulation and Applications of Space Technology and is the direct responsibility of Subcommittee E21.08 on Thermal Protection.

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[2] *Annual Book of ASTM Standards*, Vol 15.03.

[3] Discontinued, see 1982 *Annual Book of ASTM Standards*, Part 41.

$$Q^*_{tc} = (q_{hw} - q_{rr} - q_{mech})m_{tc} \quad (4)$$

$$m_{tc} = m - m_{mech} \quad (5)$$

$$q_{mech} = m_{mech}\ (h_m) \quad (6)$$

where:

Q^*_{tc} = thermochemical heat of ablation, kJ/kg,
q_{rr} = reradiative heat transfer rate, kW/m^2,
m_{tc} = mass transfer rate due to thermochemical processes, kg/m^2·s,
m_{mech} = mass-transfer rate due to mechanical processes, kg/m^2·s,
q_{mech} = heat-transfer rate due to energy carried away with mechanically removed material, kW/m^2, and
h_m = enthalpy of mechanically removed material, kJ/kg.

Mechanical removal of material takes place in the more severe test environments where relatively high aerodynamic shear or particle impingement is present. The effects of mechanical removal and theories relating to the mechanism of this process are presented in Refs (**1-5**).[4] If the effects of mechanical removal of material cannot be determined or are deemed unimportant, the term q_{mech} in Eq 4 goes to zero. The investigator should, however, be aware of the existence of this phenomenon and also note whether this effect was considered when reporting data.

3.6 The three heat of ablation values described in 3.2 require two basic determinations: the heat-transfer rate and the mass-transfer rate. These two quantities then assume various forms depending on the particular heat of ablation value being determined.

4. Significance and Use

4.1 The heat of ablation provides a measure of the ability of a material to serve as a heat protection element in a severe thermal environment. The property is a function of both the material and the environment to which it is subjected. It is therefore required that laboratory measurements of heat of ablation simulate the service environment as closely as possible. Some of the parameters affecting the property are pressure, gas composition, heat transfer rate, mode of heat transfer, and gas enthalpy. As laboratory duplication of all parameters is usually difficult, the user of the data should consider the differences between the service and the test environments. Screening tests of various materials under simulated use conditions may be quite valuable even if all the service environmental parameters are not available. These tests are useful in material selection studies, materials development work, and many other areas.

5. Determination of Heat Transfer Rate

5.1 *Cold-Wall Heat Transfer Rate*:

5.1.1 Determine the cold-wall heat-transfer rate to a specimen by using a calorimeter. These instruments are available commercially in several different types, some of which can be readily fabricated by the investigator. Selection of a specific type is based on the test configuration and the methods used, and should take into consideration such parameters as instrument response time, test duration, and heat transfer rate (**6**).

5.1.1.1 The calorimeters discussed in 5.1.1 measure a "cold-wall" heat-transfer rate because the calorimeter surface temperature is much less than the ablation temperature. The value thus obtained is used directly in computing the cold-wall heat of ablation.

5.1.2 Install the calorimeter in a calorimeter body that duplicates the test model in size and configuration. This is done in order to eliminate geometric parameters from the heat-transfer rate measurement and to ensure that the quantity measured is representative of the heat-transfer rate to the test model. If the particular test run does not allow an independent heat-transfer rate measurement, as in some nozzle liner and pipe flow tests, mount the calorimeter as near as possible to the location of the mass-loss measurements. Take care to ensure that the nonablating calorimeter does not affect the flow over the area under test. In axisymmetric flow fields, measurements of mass loss and heat-transfer rate in the same plane, yet diametrically opposed, should be valid.

5.2 *Computation of Effective and Thermochemical Heats of Ablation*:

5.2.1 In order to compute the effective and thermochemical heats of ablation, correct the cold-wall heat-transfer rate for the effect of the temperature difference on the heat transfer. This correction factor is a function of the ratio of the enthalpy potentials across the boundary layer for the hot and cold wall as follows:

$$q_{hw} = q_{cw}\ [(h_e - h_{hw})/(h_e - h_{cw})] \quad (7)$$

where:

h_e = gas enthalpy at the boundary layer edge, kJ/kg,
h_{hw} = gas enthalpy at the surface temperature of the test model, kJ/kg, and
h_{cw} = gas enthalpy at a cold wall, kJ/kg.

5.2.2 This correction is based upon laminar flow in air and subject to the restrictions imposed in Ref (**7**). Additional corrections may be required regarding the effect of temperature on the transport properties of the test gas. The form and use of these corrections should be determined by the investigator for each individual situation.

5.3 *Gas Enthalpy Determination*:

5.3.1 The enthalpy at the boundary layer edge may be determined in several ways: energy balance, enthalpy probe, spectroscopy, etc. Details of the methods may be found elsewhere (**8-11**). Take care to evaluate the radial variation of enthalpy in the nozzle. Also, in low-density flows, consider the effect of nonequilibrium on the evaluation. Determination of the gas enthalpy at the ablator surface and the calorimeter surface requires pressure and surface temperature measurements. The hot-wall temperatures are generally measured by optical methods such as pyrometers, radiometers, etc. Other methods such as infrared spectrometers and monochromators have been used (**12,13**). Effects of the optical properties of the boundary layer of an ablating surface make accurate determinations of surface temperature difficult.

5.3.2 Determine the wall enthalpy from the assumed state of the gas flow (equilibrium, frozen, or nonequilibrium), if the

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[4] The boldface numbers in parentheses refer to the references listed at the end of the standard.