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Rubbers and latices — Nomenclature

Caoutchouc et latex — Nomenclature iTeh STANDARD PREVIEW (standards.iteh.ai)

<u>ISO 1629:1995</u> https://standards.iteh.ai/catalog/standards/sist/fd2d1de5-2f1b-44da-b371-1c1b81040628/iso-1629-1995



Reference number ISO 1629:1995(E)

Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 1629 was prepared by Technical Committee VIEW ISO/TC 45, *Rubber and rubber products*.

This third edition cancels and replaces the second edition (ISO 96299987), i) which has been technically revised.

Annex A of this International Standard is for information only. https://standards.iten.av/catalog/standards/sist/fd2d1de5-2f1b-44da-b371-1c1b81040628/iso-1629-1995

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International Organization for Standardization

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Rubbers and latices — Nomenclature

1 Scope

1.1 This International Standard establishes a system of symbols for the basic rubbers in both dry and latex forms, based on the chemical composition of the polymer chain.

1.2 The purpose of this International Standard is to standardize the terms used in industry, commerce and government, and it is not intended to conflict with, but rather to act as a supplement to existing trade names and trade marks.

- **R** Rubbers having an unsaturated carbon chain, e.g. natural rubber and synthetic rubbers derived at least partly from conjugated dienes
- **T** Rubbers having carbon, oxygen and sulfur in the polymer chain
- **U** Rubbers having carbon, oxygen and nitrogen in the polymer chain

RIZ Rubbers having phosphorus and nitrogen in the polymer chain

(standards.iteh.ai) NOTE 1 In technical papers or presentations, the name of the rubber should be used if possible. The symbols should follow the chemical name for use in later references. https://standards.iteh.ai/catalog/standards/sist/fd2d1de5-2f1b-44da-b371-1c1b81040628/iso-13.1-1The "M" group

2 Rubbers

Rubbers, in both dry and latex forms, are grouped and symbolized on the basis of the chemical composition of the polymer chain in the following manner:

- **M** Rubbers having a saturated carbon chain of the polymethylene type
- **N** Rubbers having carbon and nitrogen in the polymer chain

NOTE 2 No rubber has so far been symbolized in the "N" group.

- **O** Rubbers having carbon and oxygen in the polymer chain
- **Q** Rubbers having silicon and oxygen in the polymer chain

The "M" group comprises rubbers having a saturated chain of the polymethylene type. The following symbols are used:

- ACM Copolymer of ethyl acrylate (or other acrylates) and a small amount of a monomer which facilitates vulcanization. (Usually known as acrylic rubber)
- AEM Copolymer of ethyl acrylate (or other acrylates) and ethylene
- **ANM** Copolymer of ethyl acrylate (or other acrylates) and acrylonitrile
- **CM** Chloropolyethylene¹⁾
- CSM Chlorosulfonylpolyethylene

¹⁾ In ISO 1043-1^[1], the abbreviation given for chloropolyethylene is PE-C.

- **EPDM** Terpolymer of ethylene, propylene and a diene with the residual unsaturated portion of the polymerized diene in the side chain
- **EPM** Ethylene-propylene copolymer
- **EVM** Ethylene-vinyl acetate copolymer²⁾
- FEPM Copolymer of tetrafluoroethylene and propylene
- **FFKM** Perfluoro rubber in which all substituent groups on the polymer chain are fluoro, perfluoroalkyl or perfluoroalkoxy groups
- **FKM** Fluoro rubber having substituent fluoro, perfluoroalkyl or perfluoroalkoxy groups on the polymer chain
- IM Polyisobutene³⁾
- **NBM** Fully hydrogenated acrylonitrile-butadiene copolymer (see 3.4.2)

- **FMQ** Silicone rubber having both methyl and fluorine substituent groups on the polymer chain
- **FVMQ** Silicone rubber having methyl, vinyl and fluorine substituent groups on the polymer chain
- **MQ** Silicone rubber having only methyl substituent groups on the polymer chain, such as dimethyl polysiloxane
- **PMO** Silicone rubber having both methyl and phenyl substituent groups on the polymer chain
- **PVMQ** Silicone rubber having methyl, vinyl and phenyl substituent groups on the polymer chain
- **VMQ** Silicone rubber having both methyl and vinyl substituent groups on the polymer chain

The letter for substituent group(s) on the polymer chain is inserted to the left of the code letter for rubber with silicon and oxygen in the backbone (Q) in descending order of per cent present, i.e. largest nearest the Q.

3.2 The "O" group in the neuron of the following earliest and ards.iteh.ai)

oxygen in the polymer chain. The following symbols are used:

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- CO Polychloromethyloxirafies:/(dsually/s.khow/natalsg/standards/sist/fd2d1de5-2f1b-44da-b371epichlorohydrin rubber) 1c1b81040628/3c41129 Description
- **ECO** Copolymer of ethylene oxide (oxirane) and chloromethyloxirane (also known as epi-chlorohydrin copolymer or rubber)
- **GECO** Terpolymer of epichlorohydrin-ethylene oxide-allyl glycidyl ether
- **GPO** Copolymer of propylene oxide and allyl glycidyl ether (also known as polypropylene oxide rubber)

3.3 The "Q" group

The "Q" group is defined by inserting the name of the substituent group on the polymer chain prior to the silicone designation. The following symbols are used:

The "R" group, in both dry and latex forms, is defined by inserting, before the word "rubber", the name of the monomer or monomers from which the rubber was prepared (except for natural rubber). The letter preceding the letter "R" signifies the conjugated diene from which the rubber was prepared (except for natural rubber). Any letter or letters preceding the diene letter signifies the comonomer or comonomers, substituent groups or chemical modifications. The designation may be prefixed by the letter E and a hyphen to signify an emulsion-polymerized rubber or the letter S and a hyphen to signify a solution-polymerized rubber.

For latices, the designated symbol is followed by the word latex, e.g. "SBR latex".

The symbols given in 3.4.2 to 3.4.4 are used.

²⁾ In ISO 1043-1^[1], the abbreviation given for ethylene-vinyl acetate copolymer is E/VAC.

³⁾ In ISO 1043-1^[1], the abbreviation given for polyisobutene is PIB.

3.4.2 (General	3.4.4 chain	Rubbers containing halogen on the polymer
ABR	Acrylate-butadiene rubber	BIIR	Bromo-isobutene-isoprene rubber (usually known as bromobutyl rubber)
BR	Butadiene rubber	CIIR	Chloro-isobutene-isoprene rubber (usually
CR	Chloroprene rubber		known as chlorobutyl rubber)
ENR	Epoxidized natural rubber	3.5 T	he "T" group
HNBR	Hydrogenated NBR (some unsaturation re- mains, see 3.1)	The "T gen an known are use	group comprises rubbers having carbon, oxy- d sulfur in the polymer chain. They are usually as polysulfide rubbers. The following symbols ed:
liR	lsobutene-isoprene rubber (usually known as butyl rubber)	от	A rubber having either a
ю	leeprope rubber, synthetic		—CH ₂ —CH ₂ —O—CH ₂ —O—CH ₂ —CH ₂ —
MSBR	α-methylstyrene-butadiene rubber		group or occasionally an R group, where R is an aliphatic hydrocarbon, not usually —CH ₂ —CH ₂ —, between the polysulfide linkages in the polymer chain.
NBR	Acrylonitrile-butadiene rubber (usually known	ΕΟΤ	A rubber having a
	as nitrile rubber)		CH ₂ CH ₂ OCH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ CH ₂
NIR	Acrylonitrile-isoprene rubber STANDAR	D PR	group and R groups which are usually —CH ₂ —CH ₂ — but occasionally other
NR	Natural rubber (standards	liteh.	2ali phatic groups between the polysulfide linkages in the polymer chain.
PBR	Vinylpyridine-butadiene rubber https://standards.iteh.ai/catalog/standards/si3.462d The-21H0-49tro-49tro-49tro-1-		
PSBR	Ic1b81040628/iso- Vinylpyridine-styrene-butadiene rubber	gen an svmbol	⁵ " group comprises rubbers having carbon, oxy- d nitrogen in the polymer chain. The following Is are used:
SBR	Styrene-butadiene rubber	AFMU	Terpolymer of tetrafluoroethylene trifluoro-
	E-SBR Emulsion-polymerized SBR	,	nitrosomethane and nitrosoperfluorobutyric acid
	S-SBR Solution-polymerized SBR	AU	Polyester urethane
SIBR	Styrene-isoprene-butadiene rubber	EU	Polyether urethane
		3.7 T	he "Z" group
3.4.3 F groups	Rubbers having substituent carboxylic acid (COOH) on the polymer chain	The "Z and nit bols are	" group comprises rubbers having phosphorus rogen in the polymer chain. The following sym- e used:
XBR	Carboxylic-butadiene rubber	FZ	A rubber having a —P—N— chain and having fluoroalkoxy groups attached to the phos- phorus atoms in the chain
XCR	Carboxylic-chloroprene rubber		
XNBR	Carboxylic-acrylonitrile-butadiene rubber	PZ	A rubber having aP
XSBR	Carboxylic-styrene-butadiene rubber		the chain

Annex A

(informative)

Bibliography

[1] ISO 1043-1:1987, Plastics — Symbols — Part 1: Basic polymers and their special characteristics.

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