

INTERNATIONAL STANDARD

NORME INTERNATIONALE

Environmental testing –
Part 2-52: Tests – Test Kb: Salt mist, cyclic (sodium chloride solution)
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Essais d'environnement –
Partie 2-52: Essais – Essai Kb: Brouillard salin, essai cyclique (solution de
chlorure de sodium)
IEC 60068-2-52:2017
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IEC Central Office
3, rue de Varembe
CH-1211 Geneva 20
Switzerland

Tel.: +41 22 919 02 11
info@iec.ch
www.iec.ch

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Partie 2-52: Essais – Essai Kb: Brouillard salin, essai cyclique (solution de chlorure de sodium)

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ENVIRONMENTAL TESTING –**Part 2-52: Tests – Test Kb: Salt mist, cyclic
(sodium chloride solution)**

FOREWORD

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International Standard IEC 60068-2-52 has been prepared by IEC technical committee 104: Environmental conditions, classification and methods of test.

This bilingual version (2019-07) corresponds to the monolingual English version, published in 2017-11.

This third edition cancels and replaces the second edition published in 1996. This edition constitutes a technical revision.

This edition includes the following significant technical changes with respect to the previous edition:

- a) the entire content has been harmonized with ISO 9227 as far as possible;
- b) an introduction has been added;

- c) the scope has been simplified;
- d) normative references have been updated;
- e) the general description of the test has been changed;
- f) a dry chamber has been added to the test apparatus;
- g) severities have been changed to test methods;
- h) test methods 7 and 8 have been added;
- i) information on the test report has been added;
- j) Figure 1 has been changed to Table 1;
- k) a typical test apparatus example has been added in a new Annex A;
- l) a description of each test method has been added in a new Annex B;
- m) bibliographical references have been added.

The text of this International Standard is based on the following documents:

FDIS	Report on voting
104/751/FDIS	104/761/RVD

Full information on the voting for the approval of this International Standard can be found in the report on voting indicated in the above table.

The French version of this standard has not been voted upon.

This document has been drafted in accordance with the ISO/IEC Directives, Part 2.

A list of all parts in the IEC 60068 series, published under the general title *Environmental testing*, can be found on the IEC website.

The committee has decided that the contents of this document will remain unchanged until the stability date indicated on the IEC website under "<http://webstore.iec.ch>" in the data related to the specific document. At this date, the document will be

- reconfirmed,
- withdrawn,
- replaced by a revised edition, or
- amended.

INTRODUCTION

The mechanism of corrosion on metallic materials in a chloride-containing atmosphere is electrochemical, whereas the degradation effects experienced on non-metallic materials are caused by complex chemical reactions of the salts with the materials involved. The rate at which corrosive action takes place is dependent, to a large extent, on the supply of oxygenated salt solution to the surface of the test specimen(s), the temperature of the test specimen(s) and the temperature and humidity of the environment.

Apart from the corrosive effects, this cyclic salt mist test may be used to indicate deterioration of some non-metallic materials by assimilation of salts. In the various test methods described in this document, the period of spraying with the relevant salt solution is sufficient to wet the test specimen(s) thoroughly. Because this wetting is repeated after intervals of storage under humid conditions supplemented by storage under a standard atmosphere, it goes some way to reproducing the effects of natural environments.

Furthermore, considering natural environments for corrosion on metallic materials, neutral or acidified salt solution spray, humid, and dry conditions are also important factors as a cyclic corrosion test. Each condition is repeated after intervals of other conditions in different combinations to achieve corrosion on metallic materials and to get acceleration of corrosion.

The tests described in this document are accelerated compared with most expected conditions of use. As a result, it may be difficult to establish an overall acceleration factor for all kinds of test specimens. This also means that it is often not possible to use results gained from these tests as a comparative guide to the long-term behaviour of different coating systems since the corrosion stress during these tests differs significantly from the corrosion stresses encountered during use. Nevertheless, the method described gives a means of checking that the comparative quality of a metallic material is maintained.

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ENVIRONMENTAL TESTING –

Part 2-52: Tests – Test Kb: Salt mist, cyclic (sodium chloride solution)

1 Scope

This part of IEC 60068-2 specifies the application of the cyclic salt mist test to components or equipment designed to withstand a salt-laden atmosphere as salt can degrade the performance of parts manufactured using metallic and/or non-metallic materials.

2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

IEC 60068-1, *Environmental testing – Part 1: General and guidance*

IEC 60068-2-78, *Environmental testing – Part 2-78: Tests – Test Cab: Damp heat, steady state*

ISO 9227, *Corrosion tests in artificial atmospheres – Salt spray tests*

<https://standards.iteh.ai/catalog/standards/sist/b7f80722-2c2d-43a8-861c-60402923c64f/iec-60068-2-52-2017>

3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- IEC Electropedia: available at <http://www.electropedia.org/>
- ISO Online browsing platform: available at <http://www.iso.org/obp>

4 General description of the test

4.1 Description of each test condition

4.1.1 General

This document consists of the cyclic test conditions of salt mist, dry condition, humid condition and standard atmosphere. The effect of each test condition is as follows.

NOTE Salt mist is also called salt spray.

The test specimen(s) is typically not energized during the test.

4.1.2 Salt mist

The test specimen is corroded by an electrochemical or complex chemical reaction with neutral or acidified salt solution. Salt solution forms a thin electrolytic film on the test specimen surface. This can initiate corrosion and can allow it to proceed.

4.1.3 Dry condition

Heated air in the chamber may cause evaporation of water from the test specimen surface from decreasing relative humidity and increased specimen temperature. This causes the concentration of the solution to increase and speeds up chemical reactions, accelerating the corrosion process. Salt precipitates out of the solution onto the test specimen surface during this condition. Evaporation and heating rates may vary based on the test chamber and test specimen.

4.1.4 Humid condition

4.1.4.1 Dry condition to humid condition

As the relative humidity increases, precipitated salt crystals on the specimen surface absorb water vapour until a liquid electrolytic solution forms, reinitializing the corrosion process.

4.1.4.2 Salt mist to humid condition

The humid condition maintains the existing wetness on the test specimen surface at the end of the salt mist period, without excessive dilution of the solution that may result from condensing humidity.

4.1.5 Standard atmosphere

The test specimen is allowed to equilibrate with standard laboratory conditions, which results in gradual drying and relaxation from most corrosion reactions. The period of dry atmosphere may occur, in practice, during breaks of operation, for example during the weekend. The inclusion of such a dry period may lead to corrosion mechanisms which can be quite different from those under constant wet conditions. The test schedule shall be chosen in order to respect the timelines given in 9.4. Additional dry periods (e.g. during a weekend) shall be avoided.

5 Test apparatus

5.1 General

Each test method includes two or more environmental conditions, which may be achieved by the use of multiple chambers or a single chamber that automatically transitions from one condition to the next. In either case, transitions are not assumed to be instantaneous. Care shall be taken to minimize any adverse effects associated with handling of specimens during transfer between chambers.

5.2 Salt mist chamber

The chamber shall conform to the requirements of ISO 9227. It shall maintain a temperature of $35\text{ °C} \pm 2\text{ K}$.

NOTE A schematic diagram of one possible design of spray cabinet is shown in Annex A.

5.3 Humidity chamber

The chamber shall conform to the requirements of IEC 60068-2-78. It shall maintain a relative humidity of $93\% \pm 3\%$ at a temperature of $40\text{ °C} \pm 2\text{ K}$ or a relative humidity of over 95% at a temperature of $50\text{ °C} \pm 2\text{ K}$.

5.4 Standard atmosphere chamber

The chamber shall conform to the requirements of IEC 60068-1. It shall maintain a relative humidity of 50 % \pm 5 % at a temperature of 23 °C \pm 2 K.

5.5 Dry chamber

The chamber shall maintain a relative humidity of less than 30 % at a temperature of 60 °C \pm 2 K.

6 Salt solution

6.1 Preparation of the sodium chloride solution

The solution shall conform to the requirements of ISO 9227.

NOTE The sodium chloride concentration of the sprayed solution collected is 50 g/l \pm 5 g/l.

6.2 pH adjustment

6.2.1 Neutral salt solution

The pH adjustment shall conform to the requirements of ISO 9227 for the neutral salt solution test.

6.2.2 Acidified salt solution

Add the following reagents to 10 litres of the prepared neutral 5 % sodium chloride solution as follows: 12 ml of nitric acid (HNO₃, ρ = 1,42 g/ml), 17,3 ml of sulfuric acid (H₂SO₄, ρ = 1,84 g/ml) and sufficient quantity of 10 % mass fraction of sodium hydroxide (NaOH) solution to adjust the pH of the solution to 3,5 \pm 0,1 (about 300 ml will be required). The pH of the sprayed solution collected within the chamber is 3,4 to 3,6 at 25 °C \pm 2 K.

6.3 Filtration

If necessary, filter the solution before placing it in the reservoir of the apparatus to remove any solid matter which might block the apertures of the spraying device.

7 Initial measurements

The test specimen(s) shall be visually inspected and, if necessary, electrically and mechanically checked as required by the relevant specification.

8 Preconditioning

The relevant specification shall specify the cleaning procedure to be applied immediately before the test; it shall also state whether temporary protective coatings shall be removed.

The cleaning method used should not interfere with the effect of the salt mist on the test specimen(s), nor introduce any secondary corrosion. Touching of the test surfaces by hand should be avoided as far as possible before the test.

9 Testing

9.1 Test chamber

For test methods 1 and 2, a salt mist chamber and humidity chamber are used.

For test methods 3 to 6, a salt mist chamber, humidity chamber and standard atmosphere chamber are used.

For test methods 7 and 8, a salt mist chamber, dry chamber and humidity chamber are used.

All test methods may be performed in a single chamber provided it is capable of maintaining the required conditions. If not, care should be taken to avoid loss of salt solution deposits on the test specimen(s) and to avoid any damage to the test specimen(s) due to manual handling. For test methods 7 and 8, each condition should be carried out in the same chamber due to the difficulty in achieving the rapid drying and re-wetting intended if specimens are manually moved from one chamber to another.

It is recommended to use a specific chamber for testing test method 8 exclusively. Cross effects of remaining acidified salt solution when consequently using test method 1 to 7 cannot be ignored.

9.2 Arrangement of the test specimen(s)

The arrangement shall conform to the requirements of ISO 9227.

9.3 Conditions during salt mist

The test conditions shall conform to the requirements of ISO 9227.

9.4 Test methods

9.4.1 General

The relevant specification shall indicate which of the eight following test methods shall be used. A description of each test method is given in Annex B. When not specified, the test method shall be agreed by the interested parties.

The user should be aware of mass loss for each test condition.

9.4.2 Test method 1

One cycle is seven days. One cycle shall consist of spraying the specimen with a salt solution at $35\text{ °C} \pm 2\text{ K}$ for 2 h, followed by the humid condition at $40\text{ °C} \pm 2\text{ K}$, $93\% \pm 3\% \text{ RH}$ for six days and 22 h. The required number of cycles is four (28 days).

In the case of manual handling, the transition time (maximum 2 h) should be included in the humid condition period of six days and 22 h.

9.4.3 Test method 2

One cycle is one day. One cycle shall consist of spraying the specimen with a salt solution at $35\text{ °C} \pm 2\text{ K}$ for 2 h, followed by the humid condition at $40\text{ °C} \pm 2\text{ K}$, $93\% \pm 3\% \text{ RH}$ for 22 h. The required number of cycles is three (three days).

In the case of manual handling, the transition time (maximum 2 h) should be included in the humid condition period of 22 h.

9.4.4 Test method 3

One cycle is seven days. One cycle shall consist of spraying the specimen with a salt solution at $35\text{ °C} \pm 2\text{ K}$ for 2 h, followed by the humid condition at $40\text{ °C} \pm 2\text{ K}$, $93\% \pm 3\% \text{ RH}$ for 22 h. This shall be repeated four times. The test specimens shall then be stored under standard atmosphere at $23\text{ °C} \pm 2\text{ K}$ and $50\% \pm 5\% \text{ RH}$ for three days. The required number of cycle is one (seven days).