

---

---

**Metallic coatings — Porosity tests —  
Porosity in gold or palladium coatings by  
sulfurous acid/sulfur dioxide vapour**

*Revêtements métalliques — Essais de porosité — Porosité des  
revêtements d'or ou de palladium par essai à l'acide sulfureux/vapeur de  
dioxyde de soufre*

iTeh STANDARD PREVIEW  
(standards.iteh.ai)

[ISO 15721:2001](https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-da75d985a585/iso-15721-2001)

[https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-  
da75d985a585/iso-15721-2001](https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-da75d985a585/iso-15721-2001)



**PDF disclaimer**

This PDF file may contain embedded typefaces. In accordance with Adobe's licensing policy, this file may be printed or viewed but shall not be edited unless the typefaces which are embedded are licensed to and installed on the computer performing the editing. In downloading this file, parties accept therein the responsibility of not infringing Adobe's licensing policy. The ISO Central Secretariat accepts no liability in this area.

Adobe is a trademark of Adobe Systems Incorporated.

Details of the software products used to create this PDF file can be found in the General Info relative to the file; the PDF-creation parameters were optimized for printing. Every care has been taken to ensure that the file is suitable for use by ISO member bodies. In the unlikely event that a problem relating to it is found, please inform the Central Secretariat at the address given below.

**iTeh STANDARD PREVIEW**  
**(standards.iteh.ai)**

ISO 15721:2001

<https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-da75d985a585/iso-15721-2001>

© ISO 2001

All rights reserved. Unless otherwise specified, no part of this publication may be reproduced or utilized in any form or by any means, electronic or mechanical, including photocopying and microfilm, without permission in writing from either ISO at the address below or ISO's member body in the country of the requester.

ISO copyright office  
Case postale 56 • CH-1211 Geneva 20  
Tel. + 41 22 749 01 11  
Fax + 41 22 749 09 47  
E-mail [copyright@iso.ch](mailto:copyright@iso.ch)  
Web [www.iso.ch](http://www.iso.ch)

Printed in Switzerland

## Contents

	Page
Foreword.....	iv
Introduction.....	v
1 Scope .....	1
2 Normative references .....	1
3 Terms and definitions .....	2
4 Apparatus .....	2
5 Reagents.....	2
6 Safety hazards.....	2
7 Procedure .....	3
8 Examination.....	4
9 Precision.....	5
10 Test report .....	5
Annex A (informative) Significance and use.....	6
Bibliography.....	7

iTeh STANDARD PREVIEW  
<https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-da75d985a585/iso-15721-2001>  
 (standards.iteh.ai)

## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 3.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this International Standard may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

International Standard ISO 15721 was prepared by Technical Committee ISO/TC 107, *Metallic and other inorganic coatings*, Subcommittee SC 7, *Corrosion tests*.

Annex A of this International Standard is for information only.

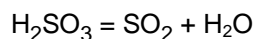
ITIH STANDARD PREVIEW  
(standards.iteh.ai)

[ISO 15721:2001](https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-da75d985a585/iso-15721-2001)

<https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-da75d985a585/iso-15721-2001>

## Introduction

The test method makes use of concentrated sulfurous acid ( $\text{H}_2\text{SO}_3$ ), which emits sulfur dioxide ( $\text{SO}_2$ ) gas according to the equilibrium reaction:



**WARNING — The sulfur dioxide gas that is released by the sulfurous acid is toxic, corrosive and irritating. The sulfurous acid shall only be handled and used in a fume cupboard. Prior to starting the test, suitable preparation shall be made to dispose of approximately 200 ml to 500 ml of sulfurous acid used in the test.**

The procedure is similar to one first proposed by Lee and Ternowski <sup>[4]</sup>. Exposure periods may vary, depending upon the degree of porosity to be revealed. Reaction of the gas with a corrodible base metal at pore sites produces reaction products that appear as discrete spots on the gold or palladium surface. Individual spots are counted with the aid of a lens or low-power stereomicroscope.

This test method is suitable for coatings containing 95 % or more of gold or palladium on substrates of copper, nickel and their alloys which are commonly used in electrical contacts.

This test is considered destructive because it reveals the presence of porosity by contaminating the surface with corrosion products and it undercuts the coating at pore sites or at the boundaries of the unplated areas. Any parts exposed to this test shall not be placed in service.

The present test method is capable of detecting virtually all porosity or other defects that could participate in corrosion reactions with the substrate or underplate. The test is rapid, simple and inexpensive. In addition, it can be used on contacts with a complex geometry such as pin-sockets (it may be useful to open contact structures with deep recesses to permit reaction of the sulfur dioxide with the interior significant surfaces).

**iTeh STANDARD PREVIEW**  
**(standards.iteh.ai)**

ISO 15721:2001

<https://standards.iteh.ai/catalog/standards/sist/49c84486-3459-4e77-939b-da75d985a585/iso-15721-2001>

# Metallic coatings — Porosity tests — Porosity in gold or palladium coatings by sulfurous acid/sulfur dioxide vapour

**WARNING** — This International Standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this International Standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use. For specific hazard statements, see clause 6.

## 1 Scope

This test method covers equipment and methods for determining the porosity of gold or palladium coatings, particularly electrodeposits and clad metals used on electrical contacts.

It is designed to show whether the porosity level is less than or greater than a given value which by experience is considered by the user to be acceptable for the intended application.

This porosity test involves reactions in which the products delineate defect sites in coatings. Since the chemistry and properties of these products may not resemble those found in natural or service environments this test cannot be recommended for the prediction of the electrical performance of contacts unless correlation is first established with service experience.

This test is intended to be used for quantitative descriptions of porosity (such as number of pores per unit area or per contact) on coatings that have a sufficiently low pore density so that the corrosion sites are well separated and can be readily resolved. As a general guideline this can be achieved for pore densities up to about 100/cm<sup>2</sup>. Above this value the tests are useful for the qualitative detection and comparisons of porosity.

A variety of other porosity testing methods are described in the literature <sup>[5],[6]</sup>. Other ISO porosity test methods are described ISO 10308, ISO 12687, ISO 14647 and ISO 15720.

## 2 Normative references

The following normative documents contain provisions which, through reference in this text, constitute provisions of this International Standard. For dated references, subsequent amendments to, or revisions of, any of these publications do not apply. However, parties to agreements based on this International Standard are encouraged to investigate the possibility of applying the most recent editions of the normative documents indicated below. For undated references, the latest edition of the normative document referred to applies. Members of ISO and IEC maintain registers of currently valid International Standards.

ISO 2064, *Metallic and other inorganic coatings — Definitions and conventions concerning the measurement of thickness.*

ISO 2079, *Surface treatment and metallic coatings — General classification of terms.*

ISO 2080, *Electroplating and related processes — Vocabulary.*

ISO 10308, *Metallic coatings — Review of porosity tests.*

### 3 Terms and definitions

For the purposes of this International Standard, the terms and definitions given in ISO 2064, ISO 2079, ISO 2080 and ISO 10308 as well as the following apply.

#### 3.1

##### **corrosion products**

those reaction products emanating from the pores that protrude from, or are otherwise attached to, the coating surface after a vapour test exposure

#### 3.2

##### **metallic coatings**

includes platings, claddings or other metallic layers applied to the substrate

NOTE The coatings can comprise a single metallic layer or a combination of metallic layers.

#### 3.3

##### **porosity**

presence of any discontinuity, crack or hole in the coating that exposes a different underlying metal

### 4 Apparatus

**4.1 Test chamber**, of any convenient size, made of glass or acrylic resin with a gas-tight lid, such as a glass desiccator of 9 l to 10 l, capacity. The ratio of chamber volume in cubic centimeters to the generating solution (sulfurous acid) surface area in square centimeters shall not exceed  $25 \pm 1$ .

**4.2 Specimen holders, or supports**, of glass, polytetrafluoroethylene, or other inert materials. It is essential that the specimens be so arranged that they do not impede gas circulation. Specimens shall not be closer than 25 mm from the wall and 75 mm from the solution surface. The measurement areas of the specimens shall be at least 12 mm apart.

NOTE Do not use a porcelain plate or any other structure that would cover more than 30 % of the liquid surface cross-sectional area. This is to ensure that movement of air and vapour within the vessel will not be restricted during the test.

**4.3 Stereomicroscope**, having a  $\times 10$  magnification shall be used for pore counting. In addition a movable source of illumination capable of giving oblique lighting on the specimen surface is required. An eyepiece is recommended for the convenience in locating the contact area or other significant measurement areas.

### 5 Reagents

Sulfurous acid ( $\text{H}_2\text{SO}_3$ ), "analytical grade reagent", or better, in individual well-sealed bottles.

### 6 Safety hazards

**6.1** Carry out these test procedures in a clean, working fume cupboard. The  $\text{SO}_2$  gas that is emitted is toxic, corrosive, and irritating.

**6.2** Use caution however, when actually performing the tests that the drafts often found in fume cupboards do not cause significant cooling of the chamber walls which may lead to condensation of water and acceleration of the test. It is often convenient to enclose the reaction vessel in a box with a loose-fitting cover and to keep the box in a fume cupboard during the test.

**6.3** Observe normal precautions in handling corrosive acids. In particular, wear eye protection completely enclosing the eyes, and make eye wash facilities readily available.



## 7 Procedure

### 7.1 Cleaning

Handle specimens as little as possible and only with tweezers, microscope-lens tissue, or clean, soft cotton gloves. Prior to the test, inspect the samples under  $\times 10$  magnification using the stereomicroscope (4.3) for evidence of particulate matter. If present, such particles shall be removed by blowing them off with clean, oil-free air. If this is not successful, discard the sample and draw a replacement sample from the batch. Then clean the samples with solvents or solutions that do not contain chlorinated hydrocarbons, CFCs or other known ozone-destroying compounds. The procedure outlined in note 1 has given satisfactory results for platings with mild to moderate surface contamination.

NOTE 1 Suggested cleaning procedure.

- a) Keep individual contacts separated if there is a possibility of damaging the measurement areas during the various cleaning steps.
- b) Clean samples for 5 min in an ultrasonic cleaner which contains a hot ( $65\text{ }^{\circ}\text{C}$  to  $85\text{ }^{\circ}\text{C}$ ) 2 % aqueous solution of a mildly alkaline (pH 7,5 to pH 10) detergent.
- c) After ultrasonic cleaning, rinse samples under warm running tap water for at least 5 s.
- d) Rinse samples ultrasonically for 2 min in fresh deionized water to remove the last detergent residues.
- e) Immerse in fresh methanol or isopropanol, and ultrasonically "agitate" for at least 30 s in order to remove the water from the samples.
- f) Remove and dry samples until the alcohol has completely evaporated. If an air blast is used as an aid for drying, the air shall be oil-free, clean and dry.
- g) Do not touch measurement area of the samples with bare fingers after cleaning.
- h) Re-inspect samples (under  $\times 10$  magnification ) for particulate matter on the surface. If particulates are found, repeat the cleaning steps. Surface cleanliness is extremely important: contaminants, such as plating salts, organic films and metal flakes may give erroneous indications of defects and are unacceptable.

NOTE 2 Omit the cleaning steps for samples with corrosion-inhibiting, lubricant coatings, or both, if it is desired to determine the efficacy of these coatings in the  $\text{SO}_2$  atmosphere.

### 7.2 Test temperature

The test temperature shall be  $23\text{ }^{\circ}\text{C} \pm 3\text{ }^{\circ}\text{C}$ , unless otherwise specified and the relative humidity in the immediate vicinity of the test chamber shall be no greater than 60 %. If the relative humidity is greater than 60 % do not run the test.

### 7.3 Sequence of operations

- a) Add the sulfurous acid ( $\text{H}_2\text{SO}_3$ ) carefully to the bottom of the clean and dry test chamber in a fume cupboard, insert the samples, using suitable fixtures (4.2) and replace the lid. The ambient relative humidity shall be no greater than 60 % while  $\text{H}_2\text{SO}_3$  is added and the samples are inserted.
- b) Expose the samples to the test environment for 2 h for gold coating thicknesses of  $1,25\text{ }\mu\text{m}$  or greater, 90 min for gold thicknesses less than  $1,25\text{ }\mu\text{m}$  and 60 min for palladium coatings.
- c) Remove the samples at the end of the test period, and allow them to stabilize at low humidity in the fume cupboard for  $10\text{ min} \pm 5\text{ min}$ .