INTERNATIONAL STANDARD



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Steel and iron — Determination of nickel, copper and cobalt contents — Inductively coupled plasma atomic emission iTeh Sspectrometric method \\

(standards.iteh.ai)

Determination of copper content

https://standards.iteh.ai/catalog/standards/sist/59296a6f-ea9a-47d9-b407-02010bc74e30/iso-13898-3-1997

> Aciers et fontes — Dosage du nickel, du cuivre et du cobalt — Méthode par spectrométrie d'émission atomique avec plasma induit par haute fréquence —

Partie 3: Dosage du cuivre



Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75% of the member bodies casting a vote.

International Standard ISO 13898-3 was prepared by Technical Committee ISO/TC 17, *Steel*, Subcommittee SC 1, *Methods of determination of chemical composition*. ISO 13898-3:1997

https://standards.iteh.ai/catalog/standards/sist/59296a6f-ea9a-47d9-b407-ISO 13898 consists of the following parts, under the general title Steel and iron — Determination of nickel, copper and cobalt contents — Inductively coupled plasma atomic emission spectrometric method.

- Part 1: General requirements and sample dissolution
- Part 2: Determination of nickel content
- Part 3: Determination of copper content
- Part 4: Determination of cobalt content

Annexes A and B of this part of ISO 13898 are for information only.

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Steel and iron — Determination of nickel, copper and cobalt contents — Inductively coupled plasma atomic emission spectrometric method —

Part 3:

Determination of copper content

1 Scope

This part of ISO 13898 specifies an inductively coupled plasma atomic emission spectrometric method for the determination of copper content in unalloyed steel and unalloyed iron.

The method is applicable to copper contents between 0,001 % (m/m) and 0,40 % (m/m).

2 Normative references (standards.iteh.ai)

The following standards contain provisions which, through reference in this text, constitute provisions of this part of ISO 13898. At the time of publication, the editions indicated were valid. All standards are subject to revision, and parties to agreements based on this part of ISO 13898 are encouraged to investigate the possibility of applying the most recent editions of the standards indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 3696:1987, Water for analytical laboratory use - Specification and test methods.

ISO 5725-2:1994, Accuracy (trueness and precision) of measurement methods and results - Part 2: Determination of repeatability and reproducibility of a standard measurement method.

ISO 5725-3:1994, Accuracy (trueness and precision) of measurement methods and results - Part 3: Intermediate measures of the precision of a standard measurement method.

ISO 13898-1:1997, Steel and iron - Determination of nickel, cobalt and copper contents - Inductively coupled plasma atomic emission spectrometric method - Part 1: General requirements and sample dissolution.

ISO 14284:1996, Steel and iron - Sampling and preparation of samples for the determination of chemical composition.

3 Principle

The principle is described in clause 3 of ISO 13898-1:1997.

4 Reagents

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade and only grade 2 water as specified in ISO 3696.

In addition to the reagents given in clause 4 of ISO 13898-1:1997, the following are required.

4.1 Copper, standard solution.

4.1.1 Stock solution, corresponding to 0,80 g of copper per litre.

Weigh, to the nearest 0,1 mg, 0,800 g of copper metal [purity >99,99 % (m/m)] and transfer to a 200 ml beaker. Add 50 ml of nitric acid (4.3 of ISO 13898-1:1997) cover with a watch-glass and heat to dissolve gently. Cool to ambient temperature, transfer into a 1 000 ml one-mark volumetric flask quantitatively and dilute to the mark with water and mix.

1 ml of this stock solution contains 0,80 mg Cu.

4.1.2 Standard solution A, corresponding to 0,08 g of copper per litre.

Transfer 20,0 ml of the copper stock solution (4.1.1) to a 200 ml one-mark volumetric flask, dilute to the mark with water and mix.

Prepare this standard solution immediately before use.

1 ml of this standard solution contains 0,08 mg CuNDARD PREVIEW

4.1.3 Standard solution B, corresponding to 0,016 g of copper peclitre a1)

Transfer 10,0 ml of the copper stock solution (4.1.1) to a 500 ml one-mark volumetric flask, dilute to the mark with water and mix. https://standards.iteh.ai/catalog/standards/sist/59296a6f-ea9a-47d9-b407-

If the calibration graph is found to be non-linear, an additional calibration series may be used.

Prepare this standard solution immediately before use.

1 ml of this standard solution contains 0,016 mg Cu.

5 Apparatus

The apparatus required is given in clause 5 of ISO 13898-1:1997.

6 Sampling

Carry out sampling in accordance with ISO 14284.

7 Procedure

7.1 Test portion

Weigh, to the nearest 1 mg, approximately 1,00 g of the test sample.

7.2 Blank test (corresponding to the zero member)

Proceed as directed in 7.2 of ISO 13898-1:1997.

7.3 Determination

7.3.1 Preparation of the test solution

Proceed as directed in 7.3.1 of ISO 13898-1:1997.

7.3.2 Preparation of the calibration solutions

Introduce into a series of six 200 ml beakers 1,00 g \pm 0,001 g of the pure iron (**4.1** of ISO 13898-1:1997). Add to each beaker 10 ml of nitric acid (**4.3** of ISO 13898-1:1997), cover the beaker with a watch-glass and heat gently until the end of effervescence. Add 10 ml of hydrochloric acid (**4.2** of ISO 13898-1:1997) and continue the heating until complete dissolution occurs.

Cool to ambient temperature and transfer the solution to six 200 ml one-mark volumetric flasks, rinsing the beakers with the minimum quantity of water.

Using a pipette or burette, add to the volumetric flasks the volume of copper standard solution A (4.1.2) indicated in table 1. If the calibration graph is found to be non-linear, additional calibration series may be used (e.g. tables 2 and 3). If the internal standard technique is used, using a pipette, add 2 ml of the scandium internal standard solution (4.4 of ISO 13898-1:1997) or 10 ml of the yttrium internal standard solution (4.5 of ISO 13898-1:1997). Dilute to the mark with water and mix.

7.4 Spectrometric measurements

7.4.1 Optimization of the instrument

iTeh STANDARD PREVIEW Proceed as directed in 7.4.1 of ISO 13898-1:1997 (standards.iteh.ai)

7.4.2 Measurements of the emitted intensities

ISO 13898-3:1997 Proceed as directed in 7.412 of 180 at 3898-414997g/standards/sist/59296a6f-ea9a-47d9-b407-02010bc74e30/iso-13898-3-1997

7.4.3 Preparation of the calibration graph

Proceed as directed in 7.4.3 of ISO 13898-1:1997.

Table 1 — Copper contents between 0,001 % (m/m) and 0,40 % (m/m)

Volume of copper standard solution A	Copper concentration	Corresponding copper content in the test portion
mi	μg/mi	% (<i>m/m</i>)
O ¹⁾	0	0
10,0	4,00	0,080
15,0	6,00	0,120
20,0	8,00	0,160
30,0	12,00	0,240
50.0	20,00	0.400

Volume of copper standard solution B	Copper concentration	Corresponding copper content in the test portion	
ml	μg/ml	% (m/m)	
O ¹⁾	0	0	
0,5	0,040	0,000 8	
1,0	0,080	0,001 6	
2,0	0,160	0,003 2	
3,0	0,240	0,004 8	
5,0	0,400	0,008 0	

Table 2 — Example for copper contents up to 0,008 % (m/m)

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Table 3 — Example for copper contents between 0,008 % (m/m) and 0,080 % (m/m)

Volume of copper standard solution B	Copper concentration	Corresponding copper content in the test portion
ml	μg/ml	% (m/m)
0 1)	0	0
5,0	0,40	0,008
10,0	0,80	0,016
20,0	1,60	0,032
30,0	2,40	0,048
50,0	4,00	0,080
1) Zero member.		

8 Expression of results

8.1 Method of calculation

Proceed as directed in 8.1 of ISO 13898-1:1997.

8.2 Precision

A planned trial of this method was carried out by 26 laboratories in 12 countries at 10 levels of copper, each laboratory making three determinations (see notes) of copper content at each level.

The test samples used are listed in table A.1.

The results obtained were treated statistically in accordance with ISO 5725-1, ISO 5725-2 and ISO 5725-3, using the data obtained from the samples containing 10 levels of copper within the application range.

The data obtained showed a logarithmic relationship between copper content and repeatability limit (r) and reproducibility limits (R and Rw) of the test results (see notes), as summarized in table 4. The graphical representation of the data is given in annex B.

Copper content	Repeatability limit	Reproducibility limits	
% (m/m)			R _w
0,001	0,000 16	0,000 23	0,000 24
0,002	0,000 23	0,000 37	0,000 35
0,005	0,000 37	0,000 70	0,000 58
0,010	0,000 54	0,001 1	0,000 84
0,020	0,000 77 <u>ISO</u>	<u>3898-3:1900</u> 01 8	0,001 2
0,050	https://standa0c00iteBai/catalog/st	andards/sis 0;003)4 a6f-ea9a-47	d9-b407- 0,002 0
0,100	0,001 82010bc74e	30/iso-138 98095 1 4 97	0,002 9
0,200	0,002 6	0,008 4	0,004 2
0,400	0,003 8	0,014	0,006 1

Tal	ble 4	
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NOTES

1 Two of the three determinations were carried out under repeatability conditions as defined in ISO 5725-1, i.e. one operator, same apparatus, identical operating conditions, same calibration, and a minimum period of time.

2 The third determination was carried out at a different time (on a different day) by the same operator as in note 1 using the same apparatus with a new calibration.

3 From the results obtained on day 1 the repeatability limit (r) and reproducibility limit (R) were calculated using the procedure specified in ISO 5725-2. From the first result obtained on day 1 and the result obtained on day 2, the within-laboratory reproducibility limit (R) was calculated using the procedure given in ISO 5725-3.

9 Test report

Proceed as directed in clause 9 of ISO 13898-1:1997.

Annex A

(informative)

Additional information on the international cooperative tests

The repeatability and reproducibility data in table 4 were derived from the results of international analytical trials carried out in 1993 on nine steel samples and one iron sample in 12 countries involving 26 laboratories.

The results of the trials were reported in document ISO/TC 17/SC 1 N 1024, March 1994. The graphical presentation of the precision data is given in annex B.

The test samples used are listed in table A.1.

Sample	Copper content % (m/m)		Precision data			
	Certified	TANDARD F		Repeat- Rability III W limits		ucibility iits
		(standa ^w Cu,1	rds.ite ^w Cu,2	h.aij)	R	R _W
JSS 003-3 (unalloyed steel) h	0,001 4 ttps://standards.	0,001 450 iteh.ai/catalog/s	1380,801:1397 andards/sist/59	0,000 17 296a6f-ea9a-47	0,000 43 1d9-b407-	0,000 33
NR 1C (Unalloyed steel)	0,001 4	0,001 45 ⁷⁴⁶	³⁰ 0,001 42	-1097 0,000 30	0,000 36	0,000 59
NR 21 (Unalloyed steel)	0,045	0,042 6	0,042 6	0,001 2	0,002 3	0,001 4
NBS 15h (Unalloyed steel)	0,013	0,012 6	0,012 6	0,000 37	0,000 82	0,000 55
NBS 16f (Unalloyed steel)	0,006	0,005 1	0,005 1	0,000 57	0,000 87	0,000 84
BAS 087-1 (Unalloyed steel)	0,171	0,172	0,172	0,003 6	0,008 0	0,007 1
BCS 452 (Unalloyed steel)	0,22	0,217	0,217	0,003 3	0,016 0	0,006 3
IRSID 081-1 (Unalloyed steel)	0,026	0,025 4	0,025 5	0,000 91	0,001 6	0,000 73
IRSID 010-1 (Unalloyed steel)	0,279	0,281	0,281	0,003 2	0,013 0	0,000 70
EURO 487-1 (Pig iron)	0,021 6	0,022 1	0,022 1	0,000 80	0,001 7	0,000 83
$\overline{w}_{Cu,1}$: general mean within a day $\overline{w}_{Cu,2}$: general mean between days						

Table A.1



Graphical representation of precision data



 $\lg r = 0,530 \ 9 \ \lg \overline{w}_{Cu,1} - 2,209 \ 4$

 $\lg R = 0,681 \ 6 \ \lg \overline{w}_{Cu,1} - 1,588 \ 8$

 $\lg R_W = 0.537 \ 4 \ \lg \overline{w}_{Cu,2} - 2.000 \ 3$

where

 $\overline{w}_{Cu,1}$ is the average copper content, expressed as a percentage by mass, obtained within a day;

 $\overline{w}_{Cu,2}$ is the average copper content, expressed as a percentage by mass, obtained between days.

Figure B.1 — Logarithmic relationship between copper content (\overline{w}_{Cu}) and repeatability limit (r) or reproducibility limits (R and R_W)