

## SLOVENSKI STANDARD SIST EN 15470:2008 01-februar-2008

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Liquefied petroleum gases - Determination of dissolved residues - Gas chromatographic method

Flüssiggase - Bestimmung des gelösten Rückstandes - Gaschromatographisches Verfahren

### iTeh STANDARD PREVIEW

Gaz de pétrole liquéfié - Détermination des résidus dissous - Méthode par chromatographie en phase gazeuse

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# EUROPEAN STANDARD NORME EUROPÉENNE EUROPÄISCHE NORM

## EN 15470

November 2007

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**English Version** 

### Liquefied petroleum gases - Determination of dissolved residues - High temperature Gas chromatographic method

Gaz de pétrole liquéfié - Détermination des résidus dissous - Méthode par chromatographie en phase gazeuse, à haute température Flüssiggase - Bestimmung der gelösten Rückstände -Hochtemperatur-Gaschromatographie-Verfahren

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

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### Foreword

This document (EN 15470:2007) has been prepared by Technical Committee CEN/TC 19 "Gaseous and liquid fuels, lubricants and related products of petroleum, synthetic and biological origin", the secretariat of which is held by NEN.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by May 2008, and conflicting national standards shall be withdrawn at the latest by May 2008.

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#### 1 Scope

This European Standard specifies a method for determining the dissolved residual matter in liquefied petroleum gases (LPG), in the range of 40 mg/kg to 100 mg/kg. Higher concentrations can be determined by adjusting the sample size.

The dissolved residue is the amount of organic compounds that are detectable by gas chromatography after evaporation of the sample at ambient temperature and then in an oven at 105 °C.

This method is not suitable for detecting solid materials or for possibly high molecular weight polymers (> 1 000 g/mol).

The advantages of this method are that a small quantity of LPG (50 g to 75 g) is required and the qualitative data available may indicate the origin of the residues (gas-oil, lubricants, plasticizers, etc.).

This method has been developed as a potential replacement of the commonly used method described in EN ISO 13757 [1], but is safer and more environmentally friendly. In addition, this new method is more accurate.

The precision data of the method have been determined from 20 mg/kg to 100 mg/kg. For a higher content of residue, the precision has not been tested, but remains of lesser interest as typical specifications are in the range of 20 mg/kg to 100 mg/kg.

NOTE An alternative European Standard, EN 15471 [2], specifies a gravimetric method, which has the same scope and slightly worst fidelity data.

WARNING — Use of this method involves hazardous materials and operations. It is the responsibility of the user to establish appropriate safety and health precautions. All handling must be performed in a fume hood.

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#### 2 Normative reference

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN ISO 4257, Liquefied petroleum gases - Method of sampling (ISO 4257:2001)

#### 3 Terms and definitions

For the purposes of this document, the following term and definition applies.

#### 3.1

#### liquefied petroleum gas (LPG)

petroleum gas that can be stored and/or handled in the liquid phase under moderate conditions of pressure and at ambient temperature, consisting predominantly of propane, butanes, with small proportions of propene, butenes and pentanes/pentenes

#### 4 Principle

A known mass of LPG, between 50 g and 75 g, is sampled and evaporated in a standard small glass flask. The residue is heated in an oven at 105 °C for 1 h. It is then diluted with a solvent and one internal standard is added. The mixture is then analyzed by a capillary gas chromatography and quantified by the internal standard method.

#### 5 Reagents

- 5.1 **Isopropylalcohol (IPA)**, analytical grade.
- 5.2 Carbon dioxide, solid for cooling.
- **5.3** Carbon disulfide, analytical grade (99,9 % minimum).
- **5.4** Normal Octane, analytical grade (99,9 % minimum).
- 5.5 **Pentane**, analytical grade (99,5 % minimum).

#### 6 Apparatus

**6.1 Sample cylinder**, of a total mass compatible with the balance used; preferably made of stainless steel fitted with two stainless steel valves free of oily materials, conforming to EN ISO 4257.

**6.2** Cooling bath, comprised of a (0,5 to 2) dl Dewar flask, filled with an appropriate liquid, e.g. IPA cooled with solid carbon dioxide, to achieve a temperature of about – 60 °C.

6.3 Sampling set, according to Annex A.

6.4 Balance for weighting, maximum weight depending of the sample cylinder weight (for instance 16 kg or 30 kg), with an accuracy of 1 g or better DARD PREVIEW

- 6.5 Analytical balance, suitable, accurate to within 0,1 mg or better.
- 6.6 Vent line, according to Annex A.
- SIST EN 15470:2008
- 6.7 Glass flasks, (100/ml/to 150 ml) with screw open caps7(for lexample2as in B.5 or other). 22014ff5e8dd/sist-en-15470-2008
- 6.8 Septum, adapted to the flask, PTFE/silicon (for example as in B.5 or other).
- 6.9 Laboratory gloves, of fine PVC, talc free.
- 6.10 Gas chromatograph, with its integrator and its automatic injector (see B.4 for an example).
- 6.11 Capillary column, (see B.6 for an example).

#### 7 Test procedure

#### 7.1 Safety

CAUTION — It is essential to work under an explosion-proof hood. In order to remove both the LPG and carbon disulfide vapours take all the necessary safety measures and, in particular, earth the equipment in order to eliminate the risks associated with static electricity.

#### 7.2 Summary of the method

A small quantity of LPG (typically around 75 g) is transferred in liquid phase from the sample cylinder to a closed glass flask cooled in an IPA /  $CO_2$  ice mixture. The transfer line shall not be cooled to reduce the possibility that a frozen substance is retained. The dual-phase mixture will condense when transferred in the cooled flask. The sampling flow is controlled by the needle valve inserted in the sampling line. The sample weight is determined by weighing the sample cylinder.

After transfer, the LPG is left to evaporate at ambient temperature. The glass container is then opened and placed for 1 h in an oven at 105 °C.

The resulting residue is dissolved in CS<sub>2</sub> and analyzed by gas chromatography using an internal standard.

#### 7.3 Sampling procedure

- a) Put the sample cylinder (6.1) on the balance (6.4);
- b) Assemble the apparatus, using the set shown in Annex A and following Figure A.1, by connecting the liquid phase valve of the sample cylinder to the sampling line comprising:
  - PTFE line;
  - spindle-needle valve;
  - 1/16" stainless steel injection tubing.
- c) Fix the PTFE line to a support in order that the sampling operations do not disturb the balance;
- d) Prepare a cooling bath (6.2) by filling a beaker or small Dewar with solid carbon dioxide and IPA;
- e) Clean a glass flask (6.7) with analytical pentane;
- f) Prepare a cap and a septum (6.8) with two small holes permitting two passages for 1/16 inch tubings;
- g) Place a clean glass flask (with its cap, its pierced PTFE septum and its venting line in position) in the cooling bath for at least 1 min before starting the sampling;
- NOTE This enables a better introduction of the sample into the glass container.

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- h) Close the needle valve; 22014ff5e8dd/sist-en-15470-2008
- i) Open the sample cylinder cautiously and fill/purge the sampling line by opening the needle valve slightly and progressively;
- j) Close the needle valve when the LPG flows regularly;
- k) Introduce the injection end in the flask;
- I) Record the weight of the sample cylinder,  $m_1$ ;
- m) Sample slowly the LPG by opening the needle valve;
- n) When about 100 ml has been sampled, close the needle valve;
- o) Record the weight of the sample cylinder,  $m_2$ ;
- p) The weight *m* of the LPG sample in grams is given by the calculation:

$$m = m_2 - m_1$$

(1)

q) Repeat the procedure for the required number of samples.

#### 7.4 Evaporation procedure

Disconnect the sampling line and leave the vaporized gas to exit through the venting line.

NOTE The lower end of the venting line should not be placed in the boiling liquid phase.

The evaporation of the sample starts as soon as the sampling begins; the flask is left at ambient temperature or gently warmed by any appropriate device to a temperature not exceeding 25 °C.

The gas is vented through the hood.

When the natural evaporation is completed, the flask is left for the duration of 1 h at 105 °C, with the <u>cap</u> <u>opened</u>. The flask is then cooled down for 30 min in a desiccator and closed with its cap and septum. The test flask now contains the dissolved evaporation residue, the amount of which will be determined by gas chromatography in the second phase of the analysis.

# CAUTION — Close the flask with the septum with the PTFE side downwards. Store the flask in the upright position.

#### 7.5 Gas chromatography analysis of the evaporation residue

#### 7.5.1 Summary of the method

The evaporation residue obtained in 7.4 is diluted and washed with a solution of  $CS_2$  (solution A) containing an internal standard (octane). The solution is then injected onto the GC capillary column. The residue amount is determined by the internal standard method. The result should be corrected by a blank determination.

#### 7.5.2 Preparation of the solution A

The solution A contains about 40 mg of octane weighted at 0,1 mg diluted in 100 g of CS<sub>2</sub>.

# 7.5.3 Operating conditions

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A recommended scheme of operating conditions is reported in Table 1:

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Carrier gas	22014ff5e8dd/sist on 1547 nature	helium
	flow	4 ml/min
Injector	type	splitless
	temperature	380 °C
Injection	type	automatic
	volume	2 μΙ
Detector	type	FID
	temperature	400 °C
	hydrogen flow	30 ml/min
	air flow	350 ml/min
	make-up gas flow	30 ml/min
Signal	mode	compensation
Oven	initial temperature	10 °C during 1 min
	final temperature	400 °C
	rate	15 °C/min