

# SLOVENSKI STANDARD SIST EN 13176:2008

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# Kemikalije, ki se uporabljajo za pripravo pitne vode - Etanol

Chemicals used for treatment of water intended for human consumption - Ethanol

Produkte zur Aufbereitung von Wasser für den menschlichen Gebrauch - Ethanol

Produits chimiques pour le traitement de reau destinée à la consommation humaine -Ethanol

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Ta slovenski standard je istoveten z.ºdb15/EN°13176:2008

# <u>ICS:</u>

13.060.20Pitna voda71.100.80Kemikalije za čiščenje vode

Drinking water Chemicals for purification of water

SIST EN 13176:2008

en,fr,de



# iTeh STANDARD PREVIEW (standards.iteh.ai)

#### **SIST EN 13176:2008**

# EUROPEAN STANDARD NORME EUROPÉENNE EUROPÄISCHE NORM

# EN 13176

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**English Version** 

# Chemicals used for treatment of water intended for human consumption - Ethanol

Produits chimiques pour le traitement de l'eau destinée à la consommation humaine - Ethanol

Produkte zur Aufbereitung von Wasser für den menschlichen Gebrauch - Ethanol

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<u>SIST EN 13176:2008</u> https://standards.iteh.ai/catalog/standards/sist/2e69a54f-16d0-4a5e-856ae1752130db15/sist-en-13176-2008



EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

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# **SIST EN 13176:2008**

# EN 13176:2008 (E)

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# Foreword

This document (EN 13176:2008) has been prepared by Technical Committee CEN/TC 164 "Water supply", the secretariat of which is held by AFNOR.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by January 2009, and conflicting national standards shall be withdrawn at the latest by January 2009.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN [and/or CENELEC] shall not be held responsible for identifying any or all such patent rights.

This document supersedes EN13176:2000.

Significant technical difference between this edition and EN 13176:2000 is as follows:

 Deletion of reference to EU Directive 80/778/EEC of July 15, 1980 in order to take into account the latest Directive in force (see [1]).

According to the CEN/CENELEC Internal Regulations, the national standards organizations of the following countries are bound to implement this European Standard: Austria, Belgium, Bulgaria, Cyprus, Czech Republic, Denmark, Estonia, Finland, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugal, Romania, Slovakia, Slovenia, Spain, Sweden, Switzerland and the United Kingdom.

# Introduction

In respect of potential adverse effects on the quality of water intended for human consumption, caused by the product covered by this European Standard:

- a) this European Standard provides no information as to whether the product may be used without restriction in any of the Member States of the EU or EFTA;
- b) it should be noted that, while awaiting the adoption of verifiable European criteria, existing national regulations concerning the use and/or the characteristics of this product remain in force.

NOTE Conformity with this European Standard does not confer or imply acceptance or approval of the product in any of the Member States of the EU or EFTA. The use of the product covered by this European Standard is subject to regulation or control by National Authorities.

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# 1 Scope

This European Standard is applicable to synthetic ethanol used for treatment of water intended for human consumption. It describes the characteristics of synthetic ethanol and specifies the requirements and the corresponding test methods for synthetic ethanol. It gives information on its use in water treatment.

NOTE This European Standard does not apply to anhydrous ethanol which is not used for drinking water treatment.

# 2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 1233, Water quality - Determination of chromium - Atomic absorption spectrometric methods

EN 1483, Water quality - Determination of mercury – Method using atomic absorption spectrometry

EN 26595, Water quality - Determination of total arsenic - Silver diethyldithiocarbamate spectrophotometric method (ISO 6595:1982)

EN ISO 3696, Water for analytical laboratory use - Specification and test methods (ISO 3696:1987)

ISO 3165, Sampling of chemical products for industrial use - Safety in sampling

(standards.iteh.ai) ISO 3856-2, Paints and varnishes - Determination of "soluble" metal content - Part 2: Determination of antimony content - Flame atomic absorption spectrometric method and Rhodamine B spectrophotometric method

ISO 6206, Chemical products for industrial use - Sampling - Vocabulary e1752130db15/sist-en-13176-2008

ISO 8288:1986, Water quality - Determination of cobalt, nickel, copper, zinc, cadmium and lead - Flame atomic absorption spectrometric methods

ISO 9965, Water quality - Determination of selenium - Atomic absorption spectrometric method (hydride technique)

# 3 Description

# 3.1 Identification

#### 3.1.1 Chemical name

Ethanol.

#### 3.1.2 Synonym or common name

Ethyl alcohol.

3.1.3 Relative molecular mass

46,07

# 3.1.4 Empirical formula

 $C_{2}H_{6}0$ 

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# 3.1.5 Chemical formula

 $C_2H_5OH$ 

# 3.1.6 CAS Registry Number <sup>1)</sup>

64-17-5

# 3.1.7 EINECS reference <sup>2)</sup>

200-57-86

# 3.2 Commercial form

The product is available as colourless liquid.

# 3.3 Physical properties

# 3.3.1 Appearance

The product is colourless liquid at 20 °C.

# 3.3.2 Density

# Tehle 1 (acc [2] and [4])

# The density at 20 °C is given in Table 1 (see [3] and [4])

Table 1 — Density			
https://concentration/catalog/star Mass fraction %	dards/sist/2e69a <b>5e</b> nsity-4a5e-856a- sist-en-13176-2008 g/ml		
95	0,8114		
95,1	0,8110		
95,2	0,8106		
95,3	0,8104		
95,4	0,8100		
95,5	0,8096		
95,6	0,8092		
95,7	0,8088		
95,8	0,8084		
95,9	0,8080		
96	0,8076		

# 3.3.3 Solubility in water

Miscible.

<sup>1)</sup> Chemical Abstracts Service Registry Number.

<sup>2)</sup> European Inventory of Existing Commercial Chemical Substances.

# 3.3.4 Vapour pressure (at 20 °C)

5,81 kPa (for pure ethanol)

# 3.3.5 Boiling point at 100 kPa <sup>3)</sup>

78,2 °C (for pure ethanol)

# 3.3.6 Melting point

-112,3 °C (for pure ethanol)

#### 3.3.7 Specific heat

2,399 kJ/(kg K) at 20 °C (for pure ethanol)

#### 3.3.8 Viscosity, dynamic

1,2 mPa.s at 20 °C (for pure ethanol)

# 3.3.9 Critical temperature (for gas)

240,77 °C

# 3.3.10 Critical pressure (for gas)

6 400 kPa

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SIST EN 13176:2008 3.3.11 Physical hardness https://standards.iteh.ai/catalog/standards/sist/2e69a54f-16d0-4a5e-856ae1752130db15/sist-en-13176-2008

Not applicable.

# 3.4 Chemical properties

Ethanol is a polar and protic organic solvent.

# 4 Purity criteria

# 4.1 General

This European Standard specifies the minimum purity requirements for Ethanol used for the treatment of water intended for human consumption. Limits are given for impurities commonly present in the product. Depending on the raw material and the manufacturing process other impurities may be present and, if so, this shall be notified to the user and when necessary to relevant authorities.

NOTE Users of this product should check the national regulations in order to clarify whether it is of appropriate purity for treatment of water intended for human consumption, taking into account raw water quality, required dosage, contents of other impurities and additives used in the product not stated in the product standard.

Limits have been given for impurities and chemicals parameters where these are likely to be present in significant quantities from the current production process and raw materials. If the production process or raw materials lead(s) to significant quantities of impurities, by-products or additives being present, this shall be notified to the user.

<sup>3) 100</sup> kPa = 1 bar

# 4.2 Composition of commercial product

The product shall contain a mass fraction of at least 95 % synthetic ethanol.

NOTE The commercial product may contain up to a mass fraction of 5% water.

# 4.3 Impurities and main by-products

The acetaldehyde content shall not exceed 100 mg/kg of pure ethanol.

# 4.4 Chemical parameters

NOTE For the purpose of this European Standard, "chemical parameters" are those defined in the EU Directive 98/83/EC of 3 November 1998 (see [1]).

The content of chemical parameters shall conform to the requirements specified in Table 2.

Parame	ter	Limit in mg/kg of pure ethanol	
Arsenic (As)	max.	1	
Cadmium (Cd)	max.	2	
Chromium (Cr)	STAN Max R	D PRÆVIE	W
Mercury (Hg)	(stand <sup>max</sup> ds	iteh <u>3</u> i)	
Nickel (Ni)	max.	1	
Lead (Pb)	SIST MAX 1317	<u>6:2008</u> <b>2</b>	
Antimony (Sb)	s.iteh.ai/catalog/ <b>max</b> lards/	sist/2e69a54f-16d0-4a5	e-85
Selenium (Se)	e1752130db15/sist-er	-13176-2008	
NOTE Cyanid Pesticides and by-products of the	e does not exist in etha polycyclic aromatic hy e manufacturing proces	nol solvent medium. drocarbons are not s.	

#### Table 2 — Chemical parameters

# 5 Test methods

# 5.1 Sampling

# 5.1.1 Relevant Standards

Observe the general recommendations of ISO 3165 and take account of ISO 6206.

#### 5.1.2 Sampling from drums and bottles

#### 5.1.2.1 General

**5.1.2.1.1** Mix the contents of the container to be sampled by shaking the container, by rolling it or by rocking it from side to side, taking care not to damage the container or spill any of the liquid.

**5.1.2.1.2** If the design of the container is such (for example, a narrow-necked bottle) that it is impracticable to use a sampling implement, take a sample by pouring after the contents have been thoroughly mixed. Otherwise, proceed as described in 5.1.2.1.3.

**5.1.2.1.3** Examine the surface of the liquid. If there are signs of surface contamination, take samples from the surface as described in 5.1.2.2; otherwise, take samples as described in 5.1.2.3.

#### 5.1.2.2 Surface sampling

Take a sample using a suitable ladle. Lower the ladle into the liquid until the rim is just below the surface, so that the surface layer runs into it. Withdraw the ladle just before it fills completely and allow any liquid adhering to the ladle to drain off. If necessary, repeat this operation so that, when the other selected containers have been sampled in a similar manner, the total volume of sample required for subsequent analysis is obtained.

#### 5.1.2.3 Bottom sampling

Take a sample using an open sampling tube, or a bottom-valve sampling tube, suited to the size of container and the viscosity of the liquid.

When using an open sampling tube, close it at the top and then lower the bottom end to the bottom of the container. Open the tube and move it rapidly so that the bottom of the tube traverses the bottom of the container before the tube is filled. Close the tube, withdraw it from the container and allow any liquid adhering to the outside of the tube to drain off.

When using a bottom-valve sampling tube, close the valve before lowering the tube into the container and then proceed in a similar manner to that when using an open sampling tube.

#### 5.1.3 Sampling from tanks and tankers

From each access point, take samples as follows:

- a) from the surface of the liquid using a ladle as described in 5.1.2.2; VIEW
- b) from the bottom of the tank or tanker, using a sampling tube as described in 5.1.2.3 or using a specially designed bottom-sampling apparatus;
- c) from one or more positions, depending on the overall depth, between the bottom and the surface using a weighted sampling cantps://standards.iteh.a/catalog/standards/sist/2e69a54f-16d0-4a5e-856ae1752130db15/sist-en-13176-2008

# 5.2 Analysis

#### 5.2.1 Ethanol (main product)

#### 5.2.1.1 Principle

The ethanol content is determined by measuring the density using a digital density meter.

The measuring principle of the digital density meter is based on the change of the frequency of a hollow oscillator when filled with different liquids. The mass, and thus the density of the liquid, changes this frequency due to a gross mass change of the oscillator caused by the introduction of the liquid.

The oscillator consists of a hollow elastic glass tube which is electronically excited in an undamped harmonic fashion. The density meter gives a direct read-out of the density result.

#### 5.2.1.2 Apparatus

- **5.2.1.2.1** Digital density meter capable of measuring at  $(20 \pm 0,1)$  °C.
- **5.2.1.2.2** Glass syringe, capacity 2 ml.