

# SLOVENSKI STANDARD SIST EN 13194:2008

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### Kemikalije, ki se uporabljajo za pripravo pitne vode - Ocetna kislina

Chemicals used for treatment of water intended for human consumption - Acetic acid

Produkte zur Aufbereitung von Wasser für den menschlichen Gebrauch - Essigsäure

Produits chimiques pour le traitement de l'eau destinée à la consommation humaine -Acide acétique

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### <u>ICS:</u>

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Drinking water Chemicals for purification of water

SIST EN 13194:2008

en,fr,de



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#### SIST EN 13194:2008

# EUROPEAN STANDARD NORME EUROPÉENNE EUROPÄISCHE NORM

## EN 13194

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**English Version** 

# Chemicals used for treatment of water intended for human consumption - Acetic acid

Produits chimiques pour le traitement de l'eau destinée à la consommation humaine - Acide acétique

Produkte zur Aufbereitung von Wasser für den menschlichen Gebrauch - Essigsäure

This European Standard was approved by CEN on 28 June 2008.

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This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the CEN Management Centre has the same status as the official versions.

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

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### Foreword

This document (EN 13194:2008) has been prepared by Technical Committee CEN/TC 164 "Water supply", the secretariat of which is held by AFNOR.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by January 2009, and conflicting national standards shall be withdrawn at the latest by January 2009.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN [and/or CENELEC] shall not be held responsible for identifying any or all such patent rights.

This document supersedes EN 13194:2000.

Significant technical difference between this edition and EN 13194:2000 is as follows:

 Deletion of reference to EU Directive 80/778/EEC of July 15, 1980 in order to account the latest Directive in force (see [1]).

According to the CEN/CENELEC Internal Regulations, the national standards organizations of the following countries are bound to implement this European Standard: Austria, Belgium, Bulgaria, Cyprus, Czech Republic, Denmark, Estonia, Finland, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugat, Romania, Slovakia, Slovenia, Spain, Sweden, Switzerland and the United Kingdom.

### Introduction

In respect of potential adverse effects on the quality of water intended for human consumption, caused by the product covered by this European Standard:

- a) this European Standard provides no information as to whether the product may be used without restriction in any of the Member States of the EU or EFTA;
- b) it should be noted that, while awaiting the adoption of verifiable European criteria, existing national regulations concerning the use and/or the characteristics of this product remain in force.

NOTE Conformity with this European Standard does not confer or imply acceptance or approval of the product in any of the Member States of the EU or EFTA. The use of the product covered by this European Standard is subject to regulation or control by National Authorities.

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#### 1 Scope

This European Standard is applicable to acetic acid used for treatment of water intended for human consumption. It describes the characteristics of acetic acid and specifies the requirements and the corresponding test methods for acetic acid. It gives information on its use in water treatment.

#### 2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 1233, Water quality — Determination of chromium — Atomic absorption spectrometric methods

EN 1483, Water quality - Determination of mercury – Method using atomic absorption spectrometry

EN 26595, Water quality - Determination of total arsenic - Silver diethyldithiocarbamate spectrophometric method (ISO 6595:1982)

EN ISO 3696, Water for analytical laboratory use - Specification and test methods (ISO 3696:1987)

ISO 3165, Sampling of chemical products for industrial use — Safety in sampling

ISO 3856-2, Paints and varnishes — Determination of "soluble" metal content — Part 2: Determination of antimony content — Flame atomic absorption spectrometric method and Rhodamine B spectrophotometric method

ISO 6206, Chemical products for industrial use Sampling Vocabulary

ISO 8288:1986, Water quality — Determination of cobalt, nickel, copper, zinc, cadmium and lead — Flame atomic absorption spectrometric methods

ISO 9965, Water quality — Determination of selenium — Atomic absorption spectrometric method (hydride technique)

#### 3 Description

#### 3.1 Identification

#### 3.1.1 Chemical name

Acetic acid, ethanoic acid.

#### 3.1.2 Synonym or common name

Glacial acetic acid.

#### 3.1.3 Relative molecular mass

60,05

#### 3.1.4 Empirical formula

 $C_2H_4O_2$ 

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#### 3.1.5 Chemical formula

CH<sub>3</sub>COOH

#### 3.1.6 CAS Registry Number <sup>1)</sup>

64-19-7

#### 3.1.7 EINECS reference <sup>2)</sup>

200-580-7

#### 3.2 Commercial form

The product is available as colourless liquid.

#### 3.3 Physical properties

#### 3.3.1 Appearance

The product is colourless liquid at 20 °C.

#### 3.3.2 Density

# **iTeh STANDARD PREVIEW** The density at 20 °C is given in Table 1. (standards.iteh.ai)

### Table 1 — Density

htt <b>Concentration</b> ai/catalog/stan mass fraction %aa5bdb1e32/	lards/sist/667b62d4 <mark>DenSity</mark> 11-854f- sist-en-13194-2008 g/ml
80	1,068 to 1,072
99,85	1,049 to 1,050

#### 3.3.3 Solubility in water

Miscible.

#### 3.3.4 Vapour pressure (at 20 °C)

1,57 kPa (for pure acetic acid)

#### 3.3.5 Boiling point at 100 kPa <sup>3)</sup>

118 °C (for pure acetic acid)

#### 3.3.6 Melting point

16,2 °C (for pure acetic acid)

<sup>&</sup>lt;sup>1</sup>) Chemical Abstracts Service Registry Number.

<sup>2)</sup> European Inventory of Existing Commercial Chemical Substances.

<sup>3)</sup> 100 kPa = 1 bar.

#### 3.3.7 Specific heat

2,047 kJ/(kg K) at 20 °C (for pure acetic acid)

#### 3.3.8 Viscosity, dynamic

1,222 mPa.s at 20 °C (for pure acetic acid)

#### 3.3.9 Critical temperature (for gas)

Not applicable.

#### 3.3.10 Critical pressure (for gas)

Not applicable.

#### 3.3.11 Physical hardness

Not applicable.

#### 3.4 Chemical properties

Acetic acid is a weak acid.

**Purity criteria** 

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#### 4.1 General

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This European Standard specifies the minimum purity requirements for Acetic acid used for the treatment of water intended for human consumption. Limits are given for impurities commonly present in the product. Depending on the raw material and the manufacturing process other impurities may be present and, if so, this shall be notified to the user and when necessary to relevant authorities.

NOTE Users of this product should check the national regulations in order to clarify whether it is of appropriate purity for treatment of water intended for human consumption, taking into account raw water quality, required dosage, contents of other impurities and additives used in the product not stated in the product standard.

Limits have been given for impurities and chemicals parameters where these are likely to be present in significant quantities from the current production process and raw materials. If the production process or raw materials leads to significant quantities of impurities, by-products or additives being present, this shall be notified to the user.

#### 4.2 Composition of commercial product

The product shall contain a minimum mass fraction of 80 percent acetic acid.

NOTE The commercial product may contain up to a mass fraction of 20% water.

#### 4.3 Impurities and main by-products

The product shall conform to the requirements specified in Table 2.

Impurity		Limit in mg/kg of pure acetic acid
Formic acid	max.	500
Acetaldehyde	max.	50

#### Table 2 — Impurities

#### 4.4 Chemical parameters

NOTE For the purpose of this European Standard, "chemical parameters" are those defined in the EU Directive 98/83/EC of 3 November 1998 (see [1]).

The content of chemical parameters shall conform to the requirements specified in Table 3.

Parameter		Limit in mg/kg of pure acetic acid	
Arsenic (As)	max.	0,5	
Cadmium (Cd) Chromium (Cr) AND	max.	0,5 PRE 0,5 IEW	
Mercury (Hg)standa	rdmaxte	<b>h.ai</b> <sup>0,5</sup>	
Nickel (Ni)	max.	0,5	
Lead (Pb) https://standards.iteh.ai/catalog/sta	EN 13194:200 max. andards/sist/66	8 0,5 7b62d4-4a52-4b11-854	
Antimony (Sb) <sub>9aa5bdb1e3</sub>	2/sistmax1.319	94-2008 0,5	
Selenium (Se)	max.	0,5	
NOTE Cyanide does not exist in the acetic acid medium. Pesticides and polycyclic aromatic hydrocarbons are not by-products of the manufacturing process.			

#### Table 3 — Chemical parameters

#### 5 Test methods

#### 5.1 Sampling

#### 5.1.1 Relevant Standards

Observe the general recommendations of ISO 3165 and take account of ISO 6206.

#### 5.1.2 Sampling from drums and bottles

#### 5.1.2.1 General

**5.1.2.1.1** Mix the contents of the container to be sampled by shaking the container, by rolling it or by rocking it from side to side, taking care not to damage the container or spill any of the liquid.

**5.1.2.1.2** If the design of the container is such (for example, a narrow-necked bottle) that it is impracticable to use a sampling implement, take a sample by pouring after the contents have been thoroughly mixed. Otherwise, proceed as described in 5.1.2.1.3.

**5.1.2.1.3** Examine the surface of the liquid. If there are signs of surface contamination, take samples from the surface as described in 5.1.2.2; otherwise, take samples as described in 5.1.2.3.

#### 5.1.2.2 Surface sampling

Take a sample using a suitable ladle. Lower the ladle into the liquid until the rim is just below the surface, so that the surface layer runs into it. Withdraw the ladle just before it fills completely and allow any liquid adhering to the ladle to drain off. If necessary, repeat this operation so that, when the other selected containers have been sampled in a similar manner, the total volume of sample required for subsequent analysis is obtained.

#### 5.1.2.3 Bottom sampling

Take a sample using an open sampling tube, or a bottom-valve sampling tube, suited to the size of container and the viscosity of the liquid.

When using an open sampling tube, close it at the top and then lower the bottom end to the bottom of the container. Open the tube and move it rapidly so that the bottom of the tube traverses the bottom of the container before the tube is filled. Close the tube, withdraw it from the container and allow any liquid adhering to the outside of the tube to drain off.

When using a bottom-valve sampling tube, close the valve before lowering the tube into the container and then proceed in a similar manner to that when using an open sampling tube.

#### 5.1.3 Sampling from tanks and tankers

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From each access point, take samples as follows:

- a) from the surface of the liquid, using a ladle as described in 5.1.2.2;
- b) from the bottom of the tank or tanker, using a sampling tube as described in 5.1.2.3 or using a specially designed bottom-sampling apparatus; <sub>19aa5bdb1e32/sist-en-13194-2008</sub>
- c) from one or more positions, depending on the overall depth, between the bottom and the surface using a weighted sampling can.

#### 5.2 Analysis

#### 5.2.1 Acetic acid (main product)

#### 5.2.1.1 Principle

An accurately weighed quantity of the sample is diluted with water and then titrated with a standard volumetric sodium hydroxide solution using phenolphthalein as an indicator.

#### 5.2.1.2 Reagents

All reagents shall be of a recognized analytical grade and the water used shall conform to grade 3 in accordance with EN ISO 3696.

- **5.2.1.2.1** Sodium hydroxide solution c (NaOH) = 1,0 mol/l.
- **5.2.1.2.2** Phenolphthalein indicator solution, 5 g/l.