



Designation: D 1757 – 03

Standard Test Method for Sulfate Sulfur in Ash from Coal and Coke¹

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1. Scope

1.1 This test method pertains to the determination of sulfate sulfur in coal or coke ash.

1.2 The values stated in SI units (Practice E 380) shall be regarded as the standard.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 ASTM Standards:

D 1193 Specification for Reagent Water²

D 2795 Test Methods for Analysis of Coal and Coke Ash³

D 3174 Test Method for Ash in the Analysis Sample of Coal and Coke from Coal³

D 3177 Test Methods for Total Sulfur in the Analysis Sample of Coal and Coke³

D 3682 Test Method for Major and Minor Elements in Combustion Residues from Coal Utilization Processes³

D 4326 Test Method for Major and Minor Elements in Coal and Coke Ash by X-Ray Fluorescence³

E 380 Practice for Use of the International System of Units (SI) (the Modernized Metric System)⁴

2.2 British Standard:

BS 1016, Part 14 Analysis of Coal Ash and Coke Ash⁵

3. Summary of Test Methods

3.1 *Test Method A—Modified British Method*—A specified quantity of ash is digested in boiling dilute hydrochloric acid

¹ This test method is under the jurisdiction of ASTM Committee D05 on Coal and Coke and is the direct responsibility of Subcommittee D05.29 on Major Elements in Ash and Trace Elements of Coal.

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² *Annual Book of ASTM Standards*, Vol 11.01.

³ *Annual Book of ASTM Standards*, Vol 05.06.

⁴ *Annual Book of ASTM Standards*, Vol 14.04.

⁵ Available from American National Standards Institute (ANSI), 25 W. 43rd St., 4th Floor, New York, NY 10036..

solution to which bromine water is added to convert sulfite that may be present to the sulfate form. After neutralization and precipitation of iron with ammonium hydroxide, the mixture is filtered and sulfate in the filtrate is determined gravimetrically as barium sulfate (BaSO_4).

3.2 *Test Method B—Eschka Mixture*—A specified quantity of ash and Eschka mixture are ignited together in air. The sulfates are dissolved in hot water and separated from undissolved ash residue and magnesium oxide by filtration. Sulfate in the filtrate is determined gravimetrically as barium sulfate (BaSO_4).

4. Significance and Use

4.1 Laboratory furnace temperature and furnace ventilation have an influence on SO_3 retention in laboratory ash. Consequently, sulfur in ash as determined in the laboratory cannot be assumed to be equivalent to sulfur present in the mineral matter in coal or to the retention of sulfur in ash produced under the conditions of commercial utilization.

4.2 The sulfate sulfur determined by these test methods can be used to calculate the sulfur trioxide portion of ash so that the ash content or ash composition can be reported on a sulfur trioxide free basis.

5. Interferences

5.1 Barium in coal ashes can result in incomplete recovery of sulfate sulfur.

5.2 High iron content can be encountered in coal and coke ash and can introduce error if, during the sulfate precipitation, the iron is either partially adsorbed by the BaSO_4 precipitate or coprecipitated as iron sulfate.

6. Apparatus

6.1 *Muffle Furnace*, electric, capable of maintaining a temperature of $800 \pm 25^\circ\text{C}$ for igniting barium sulfate.

6.2 *Crucibles or Capsules*, porcelain, platinum, alundum, or silica of 10 to 15-mL capacity, for igniting BaSO_4 .