

# SLOVENSKI STANDARD oSIST prEN 15562:2008

01-september-2008

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Fertilizers - Determination of cyanamide nitrogen

Düngemittel - Bestimmung von Cyanamidstickstoff

Engrais - Détermination de l'azote cyanamidé

Ta slovenski standard je istoveten z: prEN 15562

<u> 8181 EN 15562:2009</u>

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ICS:

65.080 Gnojila Fertilizers

oSIST prEN 15562:2008 en,fr,de

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# EUROPEAN STANDARD NORME EUROPÉENNE EUROPÄISCHE NORM

## FINAL DRAFT prEN 15562

June 2008

ICS 65.080

Will supersede CEN/TS 15562:2007

#### **English Version**

### Fertilizers - Determination of cyanamide nitrogen

Engrais - Détermination de l'azote cyanamidé

Düngemittel - Bestimmung von Cyanamidstickstoff

This draft European Standard is submitted to CEN members for unique acceptance procedure. It has been drawn up by the Technical Committee CEN/TC 260.

If this draft becomes a European Standard, CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration.

This draft European Standard was established by CEN in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the CEN Management Centre has the same status as the official versions.

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#### SIST EN 15562:2009

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

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### prEN 15562:2008 (E)

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prEN 15562:2008 (E)

#### **Foreword**

This document (prEN 15562:2008) has been prepared by Technical Committee CEN/TC 260 "Fertilizers and liming materials", the secretariat of which is held by DIN.

This document is currently submitted to the Unique Acceptance Procedure.

This document will supersede CEN/TS 15562:2007.

This document has been prepared under a mandate given to CEN by the European Commission and the European Free Trade Association.

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#### prEN 15562:2008 (E)

#### 1 Scope

This European Standard specifies a method for the determination of cyanamide nitrogen in fertilizers. The method is applicable to calcium cyanamide and calcium cyanamide/nitrate mixtures.

#### 2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 1482-2, Fertilizers and liming materials — Sampling and sample preparation — Part 2: Sample preparation

EN 12944-1:1999, Fertilizers and liming materials and soil improvers — Vocabulary — Part 1: General terms

EN 12944-2:1999, Fertilizers and liming materials and soil improvers — Vocabulary — Part 2: Terms relating to fertilizers

EN ISO 3696:1995, Water for analytical laboratory use — Specification and test methods (ISO 3696:1987)

#### 3 Terms and definitions

For the purposes of this document, the terms and definitions given in EN 12944-1:1999 and EN 12944-2:1999 apply.

### **Document Preview**

#### 4 Principle

Precipitation of cyanamide nitrogen as a silver complex. Digestion in sulfuric acid with the aid of a catalyst. Distillation of the ammonia from an alkaline solution, absorption in an excess of standard sulfuric acid solution and back-titration with standard sodium or potassium hydroxide solution.

#### 5 Reagents

#### 5.1 General

Use only reagents of recognized analytical grade and distilled or demineralized water, free from carbon dioxide and all nitrogenous compounds (grade 3 according to EN ISO 3696:1995).

#### 5.2 Glacial acetic acid

- **5.3 Ammonia solution**, containing 10 % of ammonia gas by mass ( $\rho_{20}$  = 0,96 g/ml).
- 5.4 Ammoniacal silver solution, according to Tollens.

Mix 500 ml of 10 % silver nitrate (AgNO<sub>3</sub>) solution in water with 500 ml of 10 % ammonia solution (5.3).

Do not expose unnecessarily to light, heat or air. The solution normally keeps for years. As long as the solution remains clear, the reagent is of good quality.