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Standard Guide for Depth Profiling in Auger Electron Spectroscopy¹

This standard is issued under the fixed designation E 1127; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This guide covers procedures used for depth profiling in Auger electron spectroscopy.

1.2 Guidelines are given for depth profiling by the following:

	Section
Ion Sputtering	6
Angle Lapping and Cross-Sectioning	7
Mechanical Cratering	8
Nondestructive Depth Profiling	9

1.3 This standard does not purport to address all of the safety problems, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

2. Referenced Documents

2.1 ASTM Standards:

E 673 Terminology Relating to Surface Analysis²

- E 684 Practice for Approximate Determination of Current Density of Large-Diameter Ion Beams for Sputter Depth Profiling of Solid Surfaces²
- E 827 Practice for Elemental Identification by Auger Electron Spectroscopy²
- E 1634 Guide for Performing Sputter Crater Depth Measurements²

3. Terminology

3.1 *Definitions*:

3.1.1 For definitions of terms used in this guide, refer to Terminology E 673.

4. Summary of Guide

4.1 In ion sputtering, the surface layers are removed by ion bombardment in conjunction with Auger analysis.

4.2 In angle lapping, the surface is lapped or polished at a small angle to improve the depth resolution as compared to a cross section.

4.3 In mechanical cratering, a spherical or cylindrical crater is created in the surface using a rotating ball or wheel. The sloping sides of the crater are used to improve the depth resolution as in angle lapping.

4.4 In nondestructive techniques, different methods of varying the electron information depth are involved.

5. Significance and Use

5.1 Auger electron spectroscopy yields information concerning the chemical and physical state of a solid surface in the near surface region. Nondestructive depth profiling is limited to this near surface region. Techniques for measuring the crater depths and film thicknesses are given in (**35**).

5.2 Ion sputtering is primarily used for depths of less than the order of 1 μ m.

5.3 Angle lapping or mechanical cratering is primarily used for depths greater than the order of 1 μ m.

5.4 The choice of depth profiling methods for investigating an interface depends on surface roughness, interface roughness, and film thickness (1).³

6. Ion Sputtering

6.1 First introduce the specimen into a vacuum chamber equipped with an Auger analyzer and an ion sputtering gun. Align the ion beam using a sputtering target or a Faraday cup, paying careful attention to the relative spot size of the electron beam, ion beam, and Faraday cup and their respective orientations to ensure accurate convergence of the two beams at the specimen surface.

6.1.1 Place the specimen in front of the Auger analyzer and direct the ion gun towards the analysis area. If the ion beam is not normal to the specimen surface then possible shadowing of the analysis area from the ion beam must be considered.

6.2 Choose the elements to be investigated from previous experience or from an initial Auger electron spectrum or an energy-dispersive X-ray spectrum since the latter spectrum can reveal additional elements present at depths greater than those that contribute to the Auger electron spectrum (2). Select a specific transition for each element. During the depth profiling, record the peak-to-peak heights for Auger derivative data, or peak heights or peak areas for N(E) data. The data may be

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² Annual Book of ASTM Standards, Vol 03.06.

³ The boldface numbers in parentheses refer to the list of references at the end of this guide.

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gathered during continuous sputtering or between timed sputter segments. Results may vary between the two techniques.

6.2.1 One source of their difference is due to the presence of ion-induced electrons during continuous sputter depth profiling, especially at low-electron kinetic energies, that can become comparable in intensity to the electrons induced by the probing incident electron beam. Unless one or the other of the excitation beams is modulated and detected synchronously these two types of emitted electrons are difficult to distinguish. These ion-induced electrons usually form a featureless background that rises steeply as their kinetic energy decreases, but sometimes ion-induced Auger peaks might be present whose lineshape may be different from those produced by the electron beam (3). As a result, care must be taken during continuous sputtering to ensure reliable results. Another source of difference is due to the buildup of adsorbed species during the data acquisition time in the discontinuous sputter depth profile mode (4). If portions of the ion-eroded surface expose very reactive phases, then Auger peaks due to adsorbed species, for example, oxygen or carbon, or both, will appear in the spectra and mask the actual depth distribution.

6.2.2 It is advisable when analyzing an unknown specimen to periodically examine survey scans to detect any new elements that were not present in the initial survey scan and to determine if any of the Auger peaks have been displaced outside of their analysis windows (5).

6.3 Crater-edge profiling of the sputter-formed crater by using Auger line scans is a technique similar to the analysis of the mechanically formed craters in Section 8 (6). Forming the crater by sputtering may introduce the additional complications of ion-induced damage and asymmetric crater dimensions.

6.4 If specimen rotation is used to reduce ion-induced roughness, then the rotational speed, rotation axis runout relative to ion beam sputtered area or wobble and data acquisition rate should be reported (7 and 8).

6.5 Identify the elements in the survey scans using Practice E 827.

6.6 The Auger data and the sputtering conditions should be reported as described in Practice E 996.

6.7 There is extensive information available in the literature on the effects of ion bombardment on solid surfaces (9-14).

6.8 Special care must be exercised whenever specimen temperature changes are present because effects due to surface diffusion, surface segregation or diffusion limited bulk processes such as point defect migration can occur and dramatically alter the specimen composition, even over depths larger than the ion beam penetration depth which is typically a few nanometers (15 and 16). The concept of preferential sputtering in multielement, single-phase specimens has altered significantly so that chemical effects such as surface segregation are considered to be at least as important as physical effects such as mass differences in the evolution of the near surface composition during sputter depth profiling (17-20). Since the probing depths in Auger electron spectroscopy are usually smaller than the ion-penetration depth these effects are very important in any interpretation of Auger signal intensity in terms of composition during ion-beam profiling. Computer modelling of these and other ion-induced phenomena has been extensively studied and has provided new insights into this field (**21 and 22**).

6.8.1 It should be determined for each specimen if compositional changes or other sputter effects are likely to occur. It may be possible to minimize these effects in some instances by adjusting the sputtering parameters.

6.9 Ion guns used in Auger analysis are normally selfcontained units capable of producing a focused beam of ions. The specimen is not used as an anode for the gun. Many ion guns are able to raster the ion beam. A rastered ion beam will produce a more uniform ion current distribution on the specimen surface in the region of analysis.

6.10 If the ion gun is differentially pumped, the vacuum pumps may be left on during sputtering, removing most of the sputtered gases. If not, then the chamber must be back filled with gas and provisions for removing the sputtered active gases must be considered. Titanium sublimation is effective in removing these gases.

6.11 Noble gas ions are normally used in sputtering and the most commonly used gas is argon. Xenon is occasionally used with high beam energies when rapid sputtering is needed. Active gases such as oxygen and metal ions are used in special circumstances.

6.11.1 Ion energies commonly used for depth profiling using noble gases are in the range from 1 to 5 keV where lower ion energies are usually preferred for improved depth resolution. Higher ion energies usually can be obtained with higher ion currents and less preferential sputtering.

6.11.2 Ion beam current density can be measured by a Faraday cup or by following Practice E 684.

6.11.3 The sputter rate is needed to calibrate the depth scale (33, 34, Guide E 1634) when depth profiling using ion sputtering. Several reference standards are available for this purpose. One reference material consists of 30 and 100-nm thick tantalum pentoxide films (23).⁴ Another reference material is an alternating nickel and chromium thin film structure; each layer is nominally 50-nm thick.⁵

7. Angle Lapping and Cross-Sectioning

7.1 In cross-sectioning, polish the specimen perpendicular to the interface, while in angle lapping, polish the specimen at an angle to increase the depth resolution as shown in Fig. 1 (24). Polishing usually includes the use of silicon carbide papers, diamond paste, and alumina. Use progressively finer polishing particles to obtain the desired surface finish. Possible limitations of the techniques include smearing of material across the interface, surface roughness, and the electron probe diameter limiting the spatial resolution.

7.2 In angle lapping mount the specimen on a flat gage block and measure the angle with a collimator. The accuracy depends on the flatness of the specimen. In practice an angle of 0.1° can be accurately measured.

⁴ Available from the National Physical Laboratory, Teddington, Middlesex, England. Listed as Certified Reference Material NPL No. S7B83, BCR No. 261.

⁵ Available from the National Institute of Standards and Technology, U.S. Department of Commerce, Gaithersburg, MD 20899. Listed as NIST Standard Reference Material 2135.