



Designation: D 1274 – 95 (Reapproved 2000)

## Standard Test Methods for Chemical Analysis of Pentachlorophenol<sup>1</sup>

This standard is issued under the fixed designation D 1274; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

### 1. Scope

1.1 These test methods cover the determination of the chemical analysis of pentachlorophenol for use in the preservative treatment of wood.

1.1.1 Test Method D 38 covers the sampling of wood preservatives prior to testing.

1.2 The analytical procedures appear in the following order:

	Sections
Total Acidity	6 and 7
Alkali-Insoluble Material	8 and 9
Freezing Point	10

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

### 2. Referenced Documents

#### 2.1 ASTM Standards:

D 38 Test Methods for Sampling Wood Preservatives Prior to Testing<sup>2</sup>

D 1193 Specification for Reagent Water<sup>3</sup>

D 1272 Specification for Pentachlorophenol<sup>2</sup>

#### 2.2 Other Standard:

AWPA A5 Standard Methods for Analysis of Oil-Borne Preservatives<sup>4</sup>

### 3. Summary of Test Methods

3.1 *Pentachlorophenol*—A measured sample is dissolved in methanol. The phenol is titrated with a standard solution of Sodium Hydroxide.

<sup>1</sup> These test methods are under the jurisdiction of ASTM Committee D-7 on Wood and are the direct responsibility of Subcommittee D07.06 on Treatments for Wood Products.

Current edition approved July 15, 1995. Published September 1995. Originally published as D 1274 – 53. Last previous edition D 1274 – 94.

These test methods are identical in substance with appropriate sections of the American Wood-Preservers' Association Standard Methods for Analysis of Oil-Borne Preservatives (A5-93). Acknowledgment is made to the American Wood-Preservers' Association for its development of the subject matter covered in this standard.

<sup>2</sup> *Annual Book of ASTM Standards*, Vol 04.10.

<sup>3</sup> *Annual Book of ASTM Standards*, Vol 11.01.

<sup>4</sup> Available from American Wood Preservers' Association, P.O. Box 286, Woodstock, MD 21163-0286.

3.2 *Sodium Hydroxide Insoluble Matter*—A weighed sample is digested in 0.1 N NaOH and filtered. The residue is dried and weighed.

3.3 *Freezing Point*—A sample is heated in an enameled cup or nickel tube to 10°C above the expected melting point, then gradually cooled, and the melting point measured by a standardized thermometer.

### 4. Significance and Use

4.1 Pentachlorophenol for use in the preservative treatment of wood must conform with Specification D 1272.

### 5. Reagents

5.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.<sup>5</sup> Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

5.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean reagent water conforming to Specification D 1193.

### TOTAL ACIDITY

### 6. Reagents

6.1 *Sodium Hydroxide, Standard Solution (0.1 N)*—Dissolve 4.0 g of sodium hydroxide (NaOH) in 1000 mL of water.

6.1.1 *Standardization of Sodium Hydroxide Solution*—Transfer four 0.1 N potassium acid phthalate replicates (from 6.2), each of 25.00-mL volume, into 125-mL Erlenmeyer flasks. Add 3 to 5 drops of phenolphthalein indicator solution to each replicate. Titrate to a permanent faint pink end point with 0.1 N NaOH solution (from 6.1). Read the buret to the nearest estimated 0.01 mL. Record the volume of NaOH solution used for each replicate.

<sup>5</sup> *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Anal. Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmaceutical Convention, Inc. (USPC), Rockville, MD.