



**SLOVENSKI STANDARD**  
**SIST ISO 4689-2:2005**

**01-november-2005**

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Železne rude -- Določitev vsebnosti žvepla -- Del 2: Metoda porušitve/titracijska metoda

Iron ores -- Determination of sulfur content -- Part 2: Combustion/titration method

**iTeh STANDARD PREVIEW**

Minerais de fer -- Dosage du soufre -- Partie 2: Méthode par combustion et titration

**Ta slovenski standard je istoveten z: ISO 4689-2:2004**

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**ICS:**

73.060.10      Železove rude      Iron ores

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INTERNATIONAL  
STANDARD

ISO  
4689-2

First edition  
2004-04-01

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**Iron ores — Determination of sulfur  
content —**

**Part 2:  
Combustion/titration method**

*Minerais de fer — Dosage du soufre —*

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*Partie 2: Méthode par combustion et titration*  
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Reference number  
ISO 4689-2:2004(E)

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Published in Switzerland

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## ISO 4689-2:2004(E)

## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 4689-2 was prepared by Technical Committee ISO/TC 102, *Iron ore and direct reduced iron*, Subcommittee SC 2, *Chemical analysis*.

This first edition, in part cancels and replaces ISO 4690:1986, which has been technically revised.

ISO 4689 consists of the following parts, under the general title *Iron ores — Determination of sulfur content*:

- *Part 2: Combustion/titration method* [SIST ISO 4689-2:2005](https://standards.iteh.ai/catalog/standards/sist/647963e9-e9bd-49cb-932d-00f94676e041/sist-iso-4689-2-2005)
- *Part 3: Combustion/infrared method* <https://standards.iteh.ai/catalog/standards/sist/647963e9-e9bd-49cb-932d-00f94676e041/sist-iso-4689-2-2005>

The following part is under preparation:

- *Part 1: Barium sulfate gravimetric method*

## Introduction

This part of ISO 4689 was originally published as, ISO 4690:1986, *Iron ores — Determination of sulfur content — Combustion method*. Under a policy of rationalization of the numbering system used in ISO/TC 102, it has been decided to re-designate this document as ISO 4689-2. It was further decided to introduce a combustion/infrared method, numbered ISO 4689-3.

When next revised, ISO 4689:1986, *Iron ores — Determination of sulfur content — Barium sulfate gravimetric method*, will be re-designated ISO 4689-1.

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# Iron ores — Determination of sulfur content —

## Part 2: Combustion/titration method

**WARNING** — This part of ISO 4689 may involve hazardous materials, operations and equipment. This part of ISO 4689 does not purport to address all of the safety problems associated with its use. It is the responsibility of the user of this part of ISO 4689 to establish appropriate health and safety practices and determine the applicability of regulatory limitations prior to use.

### 1 Scope

This part of ISO 4689 specifies a combustion/titration method for the determination of the sulfur content of iron ores.

This method is applicable to sulfur contents between 0,002 % (mass fraction) and 0,25 % (mass fraction) in natural iron ores, iron ore concentrates and agglomerates, including sinter products. The results are not affected by the presence of fluoride.

### 2 Normative references

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The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 385-1:1984, *Laboratory glassware — Burettes — Part 1: General requirements*

ISO 648:1977, *Laboratory glassware — One-mark pipettes*

ISO 1042:1998, *Laboratory glassware — One-mark volumetric flasks*

ISO 3082:2000, *Iron ores — Sampling and sample preparation procedures*

ISO 7764:1985, *Iron ores — Preparation of predried test samples for chemical analysis*

### 3 Principle

The sample is mixed with tungsten(VI) oxide and heated at 1 200 °C in a resistance furnace, using nitrogen as a carrier gas.

The evolved sulfur dioxide is absorbed in dilute hydrochloric acid solution containing starch and potassium iodide, and the solution is titrated continuously during evolution with a standard volumetric solution of potassium iodate.

### 4 Reagents

During analysis, use only reagents of recognized analytical grade, and only distilled water or water of equivalent purity.

## ISO 4689-2:2004(E)

**4.1 Magnesium perchlorate**,  $\text{Mg}(\text{ClO}_4)_2$ , anhydrous, free flowing, particle size 0,5 mm to 2 mm.

**4.2 Tungsten(VI) oxide**,  $(\text{WO}_3)$ .

**4.3 Tin(II) chloride**,  $\text{SnCl}_2 \cdot 2\text{H}_2\text{O}$ , particle size 0,5 mm to 2 mm.

**4.4 Soda-asbestos**, particle size 0,5 mm to 2 mm.

**4.5 Hydrochloric acid**,  $\rho$  1,16 g/ml to 1,19 g/ml, diluted 1 + 66.

**4.6 Potassium iodide**, KI, 3 % (mass fraction) solution.

**4.7 Starch**, 2 % (mass fraction) solution.

Make a suspension of 2,0 g of starch in 10 ml of water, add to 50 ml of boiling water and stir. Cool, dilute to 100 ml and mix.

Prepare this solution immediately prior to use.

**4.8 Potassium iodate**,  $\text{KIO}_3$ , standard volumetric solution, 0,001 042 mol/l.

NOTE The potassium iodate should be previously dried for 2 h at 130 °C.

Weigh, to the nearest 0,000 2 g, 0,223 g of dried potassium iodate and dissolve in water. Cool, transfer to a 1 litre volumetric flask, dilute to volume and mix.

1 ml of this standard volumetric solution is equivalent to 0,10 mg of sulfur.

## 5 Apparatus

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Ordinary laboratory apparatus, including one-mark pipettes and one-mark volumetric flasks complying with the specifications of ISO 648 and ISO 1042 respectively as well as the following. See Figure 1.

**5.1 Nitrogen supply.**

**5.2 Drying tower**, containing soda-asbestos (4.4) and magnesium perchlorate (4.1).

**5.3 Flowmeter**, graduated from 0 to 1 l/min.

**5.4 Resistance furnace**, capable of maintaining a temperature of  $1\ 200\ ^\circ\text{C} \pm 25\ ^\circ\text{C}$ .

**5.5 Refractory combustion tube**, capable of withstanding a temperature of  $1\ 200\ ^\circ\text{C} \pm 25\ ^\circ\text{C}$ .

**5.6 Refractory combustion boat**, with loose-fitting lid, or a suitable refractory combustion boat capable of accommodating a porous cartridge, as shown in Figures 2a) or 2b) respectively. The following dimensions can be used as guidelines.

Combustion boat		Cartridge	
Length	80 mm	Length	50 mm
Width	13 mm	Internal diameter	15 mm
Depth	9 mm		

**5.7 Tin(II) chloride absorption tube**, of narrow glass tubing containing approximately 10 g of solid tin(II) chloride (4.3), suitably plugged with inert wool to prevent the reagent being blown into the absorption vessel. A U-tube is suitable. Where necessary this is inserted in the train at the exit of the combustion tube at point X in Figure 1.

NOTE If the sample is known or thought to contain chloride, e.g., in the form of sodium chloride, chlorapatite or scapolite, the chlorine formed during combustion of the sample should be removed by passing the gas stream through solid tin(II) chloride in a suitable tube or vessel (5.7) prior to absorption and titration. If numerous samples having chloride contents  $> 1\%$  (mass fraction) are to be analysed, it is advisable to determine the absorption capacity of the tin(II) chloride in order to be able to judge the most suitable interval for reagent replacement. In such cases, a larger absorption vessel may be preferred.

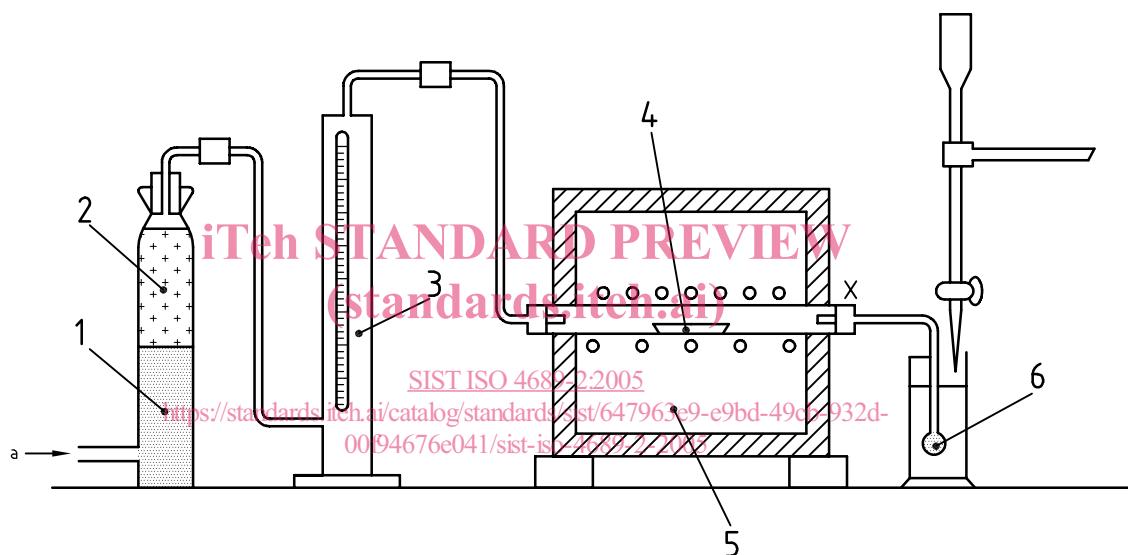
The tin(II) chloride absorption tube should also be fitted if the chloride content of the test sample is unknown.

**5.8 Cylinder**, 100 ml, tall form, to hold the absorbing solution.

**5.9 Bubbler**, multi-hole, to reach to the bottom of the cylinder (5.8).

**5.10 Burette**, conforming to ISO 385-1, graduated in divisions of 0,05 ml.

NOTE For determination of low sulfur contents ( $< 0,005\%$  [mass fraction]), alternative titration devices such as a piston burette or a suitable micro-syringe capable of delivering small, precise aliquots of titrant may be used.



#### Key

- 1 soda-asbestos
- 2 anhydrous magnesium perchlorate
- 3 flowmeter
- 4 combustion boat
- 5 resistance furnace
- 6 multihole bubbler

<sup>a</sup> Nitrogen.

**Figure 1 — Combustion apparatus**