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An American National Standard

# Standard Specification for Denatured Fuel Ethanol for Blending with Gasolines for Use as Automotive Spark-Ignition Engine Fuel<sup>1</sup>

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# 1. Scope\*

- 1.1 This specification covers nominally anhydrous denatured fuel ethanol intended to be blended with unleaded or leaded gasolines at 1 to 10 volume % for use as a spark-ignition automotive engine fuel. The significance of this specification is shown in Appendix X1.
- 1.2 The values stated in SI units are to be regarded as the standard. The values given in parentheses are for information only.

#### 2. Referenced Documents

- 2.1 ASTM Standards: <sup>2</sup>
- D 86 Test Method for Distillation of Petroleum Products at Atmospheric Pressure
- D 381 Test Method for Gum Content in Fuels by Jet Evaporation
- D 512 Test Methods for Chloride Ion in Water
- D 891 Test Methods for Specific Gravity, Apparent, of Liquid Industrial Chemicals
- D 1152 Specification for Methanol (Methyl Alcohol)
- D 1193 Specification for Reagent Water
- D 1613 Test Method for Acidity in Volatile Solvents and Chemical Intermediates Used in Paint, Varnish, Lacquer, and Related Products
- D 1688 Test Methods for Copper in Water
- D 3505 Test Method for Density or Relative Density of Pure Liquid Chemicals
- D 4052 Test Method for Density and Relative Density of Liquids by Digital Density Meter
- D 4057 Practice for Manual Sampling of Petroleum and Petroleum Products
- D 4814 Specification for Automotive Spark-Ignition Engine Fuel
- <sup>1</sup> This specification is under the jurisdiction of ASTM Committee D02 on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee D02.A0 on Gasoline and Oxygenated Fuels.
- Current edition approved June 1, 2004. Published July 2004. Originally approved in 1988. Last previous edition approved in 2003 as D 4806–03.
- <sup>2</sup> For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

- D 5453 Test Method for Determination of Total Sulfur in Light Hydrocarbons, Motor Fuels and Oils by Ultraviolet Fluorescence
- D 5501 Test Method for Determination of Ethanol Content of Denatured Fuel Ethanol by Gas Chromatography
- D 5580 Test Method for Determination of Benzene, Toluene, Ethylbenzene, *p/m*-Xylene, *o*-Xylene, C<sub>9</sub> and Heavier Aromatics, and Total Aromatics in Finished Gasoline by Gas Chromatography
- D 6423 Test Method for Determination of pHe of Ethanol, Denatured Fuel Ethanol, and Fuel Ethanol (Ed75–Ed85)
- D 6550 Test Method for Determination of Olefin Content of Gasolines by Supercritical-Fluid Chromatography
- E 203 Test Method for Water Using Volumetric Karl Fischer Titration
- E 300 Practice for Sampling Industrial Chemicals
- E 1064 Test Method for Water in Organic Liquids by Coulometric Karl Fischer Titration
- 2.2 Other Standards:
- United States Code of Federal Regulations, Title 27, Parts 16 20 and 21<sup>3</sup>
- United States Federal Specification O-E-760b Ethyl Alcohol (Ethanol): Denatured Alcohol: and Proprietary Solvent<sup>4</sup>

#### 3. Terminology

- 3.1 Definitions:
- 3.1.1 *ethanol*, n—ethyl alcohol, the chemical compound  $C_2H_5OH$ .
- 3.1.2 *gasoline*, *n*—a volatile mixture of liquid hydrocarbons, generally containing small amounts of additives, suitable for use as a fuel in spark-ignition, internal combustion engines.

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3.1.3 *gasoline-ethanol blend*, *n*—a fuel consisting primarily of gasoline along with a substantial amount (more than 0.35 mass % oxygen) of denatured fuel ethanol. **D 4814** 

<sup>&</sup>lt;sup>3</sup> Order as Code of Federal Regulations Title 27 Parts 200-End: from U.S. Government Printing Office Superintendent of Documents, 732 N. Capitol St., NW, Mail Stop: SDE, Washington, DC 20401.

<sup>&</sup>lt;sup>4</sup> Order from U.S. Government Printing Office Superintendent of Documents, 732 N. Capitol St., NW, Mail Stop: SDE, Washington, DC 20401.



- 3.1.4 *oxygenate*, *n*—an oxygen-containing, ashless, organic compound, such as an alcohol or ether, which may be used as afuelorfuelsupplement. **D 4814** 
  - 3.2 Definitions of Terms Specific to This Standard:
- 3.2.1 *denaturants*—natural gasoline, gasoline components, unleaded gasoline, or toxic or noxious materials added to fuel ethanol to make it unsuitable for beverage use but not unsuitable for automotive use.
- 3.2.2 *denatured fuel ethanol*—fuel ethanol made unfit for beverage use by the addition of denaturants.
- 3.2.3 *fuel ethanol*—ethanol with impurities common to its production (including water but excluding denaturants).
- 3.2.4 *impurities*—in commercially produced fuel ethanol, compounds other than ethanol or denaturants present, such as methanol and fusel oil (for example, amyl and isoamyl alcohols).
  - 3.2.5 *pHe*—a measure of the acid strength of alcohol fuels.

#### 4. Performance Requirements

4.1 *Denatured Fuel Ethanol*—When fuel ethanol is denatured as specified in Section 5, it shall conform to the following requirements at the time of blending with a gasoline.

Ethanol, volume %, min Methanol, volume %, max 0.5 Solvent-washed gum, 5.0 mg/100 mL, max Water content, volume %, max 1 (Note 1) Denaturant content, volume %, min 1.96 volume %, max 4.76 Inorganic Chloride content, mass 40 (32) ppm (mg/L), max Copper content, mg/kg, max 0.1 0.007 (56) (Note 3) Acidity (as acetic acid CH<sub>3</sub>COOH), mass % (mg/L), max 6.5 to 9.0 Appearance Visibly free of suspended or precipitated contaminants (clear and bright)

Note 1—In some cases, a lower water content may be necessary to avoid phase separation of a gasoline-ethanol blend at very low temperatures. This reduced water content, measured at the time of delivery, shall be agreed upon between the supplier and purchaser.

Note 2—If denatured fuel ethanol is prepared by the addition of denaturants to undenatured fuel ethanol after it has been produced rather than during the dehydration process, the 15.56/15.56°C (60/60°F) specific gravity in air of the undenatured fuel ethanol shall be in the range from 0.7937–0.7977.

Note 3—Denatured fuel ethanol may contain additives, such as corrosion inhibitors and detergents, that may affect the titratable acidity (acidity as acetic acid) of the finished fuel ethanol. Although the base fuel ethanol may meet the acidity specification, the effect of these additives may produce an apparent high titratable acidity of the finished product. Contact the ethanol supplier if there is a question regarding the titratable acidity of your denatured fuel ethanol to verify that the base ethanol meets the acidity requirements of 4.1.

4.2 Other Properties—Limits more restrictive than those specified above, or the specification of additional properties such as color, may be agreed upon between the supplier and the purchaser.

#### 5. Denaturants

5.1 The only denaturants used for fuel ethanol shall be natural gasoline, gasoline components, or unleaded gasoline at

a minimum concentration of two parts by volume per 100 parts by volume of fuel ethanol. One denatured formula specifically designed for fuel use by the Bureau of Alcohol, Tobacco, and Firearms (BATF) of the U.S. Treasury Department is Formula C.D.A. 20. It requires that for every 100 gal of ethanol of not less than 195 proof, a total of 2.0 gal of denaturant be added. Another fuel alcohol rendered unfit for beverage use and manufactured at an alcohol fuel plant requires the addition of 2 gal or more of materials listed by the director to each 100 gal of ethanol. The fuel ethanol formulas approved by the U.S. Treasury Department include materials, which are not allowed by this ASTM specification. This specification prohibits the use of hydrocarbons with an end boiling point higher than 225°C (437°F) as determined by Test Method D 86, although they may be permitted by BATF regulations. Some kerosines, for instance, promote piston scuff in automotive engines. The denaturants permitted by this specification may be included as part of the 10 volume % denatured fuel ethanol blended with a gasoline if they do not exceed five parts by volume per 100 parts by volume of fuel ethanol. This is permitted in the United States by law. Any part of these denaturants that are present at concentrations higher than five parts by volume per 100 parts by volume of fuel ethanol are considered as part of the base gasoline.

Note 4—BATF regulations concerning the preparation, use, and handling of denatured ethanols are published in the United States Code of Federal Regulations, Title 27, Parts 19, 20, and 21. 27 CFR 19.1005 contains regulations for rendering fuel alcohol unfit for beverage use. 27 CFR 21.24 contains the formula for manufacturing completely denatured alcohol, C.D.A. 20.

5.2 Prohibited Denaturants—Although this specification permits only hydrocarbons in the gasoline boiling range to be used as denaturants, specific mention must be made of some materials that have extremely adverse effects on fuel stability, automotive engines, and fuel systems. These materials shall not be used as denaturants for fuel ethanol under any circumstances. They are as follows: methanol which does not meet Specification D 1152, pyrroles, turpentine, ketones, and tars (high-molecular weight pyrolysis products of fossil or nonfossil vegetable matter). While any significant amount of methanol will lower the water tolerance and increase the vapor pressure of a gasoline-ethanol blend, these effects become more serious when methanol is present at more than 2.5 parts by volume per 100 parts by volume of fuel ethanol. Also, methanol, which does not meet Specification D 1152, frequently contains impurities, such as turpentine and tars. Similarly, ketone denaturants tend to degrade fuel stability or increase the tendency of a gasoline-ethanol blend to corrode metals and attack elastomers. These effects become more serious if the concentration of a ketone such as 4-methyl pentanone (methyl isobutyl ketone) exceeds one part by volume per 100 parts by volume of fuel ethanol. There is no information available on the effects of denaturants other than those mentioned above; but unless a denaturant, such as a higher aliphatic alcohol or ether, is known to have no adverse effect on a gasoline-ethanol blend or on automotive engines or fuel systems, it shall not be used.

### 6. Workmanship

- 6.1 The fuel ethanol shall be visually free of sediment and suspended matter. It shall be clear and bright at the ambient temperature or 21°C (70°F), whichever is higher.
- 6.2 The specification defines only a basic purity for this product. The product shall be free of any adulterant or contaminant that may render the material unacceptable for its commonly used applications.

## 7. Sampling

- 7.1 Samples may be obtained by an appropriate procedure of Practice D 4057 or E 300, except that water displacement (in section on Sampling for Specific Tests in D 4057) must not be used. Where practical, fuel ethanol should be sampled in glass containers. If samples must be collected in metal containers, do not use soldered metal containers, although they are specified in the Sampling Equipment section in Practice E 300, because the soldering flux in the containers may contaminate the sample.
- 7.2 Sample Size—A minimum of about 1 L or 1 U.S. qt is recommended. If specific gravity is to be determined by a hydrometer method, additional volume may be required. This depends on the size of the hydrometer.
- 7.3 Lot Size—A lot shall normally consist of the amount contained in a tanker compartment or other bulk container in which it is delivered. If this definition does not apply, the definition of a lot must be agreed upon between the supplier and purchaser.

Note 5—See Sections 5, 6, and 7 on Significance, Safety, and Statistical Considerations, respectively, of Practice E 300 for a detailed discussion of the statistics of sampling.

#### 8. Test Methods

- 8.1 The scope of some of the test methods specified in 8.2-8.10 do not include denatured fuel ethanol. The precisions of those test methods may differ from the reported precisions when testing denatured fuel ethanol.
  - 8.2 Water Content—Test Methods E 203 or E 1064.
- 8.3 Solvent-Washed Gum Content—Test Method D 381, air jet apparatus.
  - 8.4 Acidity—Test Method D 1613.
  - 8.5 pHe—Test Method D 6423.
- 8.6 Appearance—The product shall be visibly free of suspended or precipitated contaminants (clear and bright). This shall be determined at indoor ambient temperature unless otherwise agreed upon between the supplier and the purchaser.

- 8.7 Specific Gravity—Test Methods D 891, Procedure B or Test Method D 4052. For Test Methods D 891, Procedure B (hydrometer), no formal precision statement is available, but practical experience indicates that precision is no better than 0.0005. Test Methods D 891 Procedure C (pycnometer), with an interlaboratory precision (reproducibility) of 0.0002, should be used as a referee method.
- 8.8 *Inorganic Chloride Content*—Modification of Test Methods D 512–81(1985) $^{\epsilon 1}$ , Method C.
- 8.8.1 The modification of Test Methods D 512–81(1985)<sup>€1</sup>, Method C consists of using 5 mL of sample diluted with 20 mL of water in place of the 25-mL sample specified in the standard procedure. The water shall meet Specification D 1193, Type II. The volume of the sample prepared by this modification will be slightly larger than 25 mL. To allow for the dilution factor, report the chloride ion present in the fuel ethanol sample as the chloride ion present in the diluted sample multiplied by five.
- 8.8.2 The precision of this modified method has not been determined, but for the actual amount of chloride ion found in the diluted sample, it is expected to be similar to the precision of Test Methods D  $512-81(1985)^{\epsilon 1}$ , Method C.
- 8.9 Copper Content—Modification of Test Methods D 1688, Test Method A.
- 8.9.1 The modifications of Test Methods D 1688, Test Method A (atomic absorption, direct) consists of mixing reagent-grade ethanol (which may be denatured according to BATF of the U.S. Treasury Department Formula 3A or 30) in place of water as the solvent or diluent for the preparation of reagents and standard solutions. However, this must not be done to prepare the stock copper solution described in the section on Copper Solution, Stock in Test Method D 1688. Because a violent reaction may occur between the acid and the ethanol, use water, as specified, in the acid solution part of the procedure to prepare the stock copper solution. Use ethanol for the rinse and final dilution only.
- 8.9.2 The precision of this modified method has not been determined, but it is expected to be similar to the precision of Test Method D 1688, Test Method A.
  - 8.10 Ethanol Content—Test Method D 5501.

#### 9. Keywords

9.1 acidity; automotive spark-ignition engine fuel; base gasoline; chloride ion content; copper content; corrosion inhibitors; denaturants; denatured fuel ethanol; ethanol content; ethanol purity; fuel; fuel ethanol; gasoline; gasoline-ethanol blend; impurities; oxygenate; solvent-washed gum; water content