



Designation: D 5135 – 02^{ε1}

Standard Test Method for Analysis of Styrene by Capillary Gas Chromatography¹

This standard is issued under the fixed designation D 5135; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last approval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

^{ε1} NOTE—Research report was added editorially in September 2004.

1. Scope*

1.1 This test method covers the determination of the impurities typically found in styrene and uses those values to determine the purity of styrene by difference by gas chromatography. Typical impurities in commercial styrene are found in Table 1.

1.2 This test method is applicable to impurity concentrations in the range of 0.001 to 1.00 % and for styrene purities of 99 % or higher when using the internal standard procedure.

1.3 The following applies to all specified limits in this standard: for purposes of determining conformance with this standard, an observed value or a calculated value shall be rounded off “to the nearest unit” in the last right-hand digit used in expressing the specification limit, in accordance with the rounding-off method of Practice E 29.

1.4 *This standard does not purport to address all the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.* For a specific hazard statement, see Section 9.

2. Referenced Documents

2.1 ASTM Standards:²

- D 3437 Practice for Sampling and Handling Liquid Cyclic Products
- D 4307 Practice for Preparation of Liquid Blends for Use as Analytical Standards
- D 4790 Terminology of Aromatic Hydrocarbons and Related Chemicals

¹ This test method is under the jurisdiction of ASTM Committee D16 on Aromatic Hydrocarbons and Related Chemicals and is the direct responsibility of Subcommittee D16.07 on Styrene, Ethylbenzene, and C₉ and C₁₀ Aromatic Hydrocarbons.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard’s Document Summary page on the ASTM website.

TABLE 1 Impurities Known or Suggested to be Present in Commercial Styrene

Ethylbenzene
o-xylene
m-xylene
p-xylene
Isopropylbenzene
n-propylbenzene
m-ethyltoluene
p-ethyltoluene
α-methylstyrene
m-vinyltoluene
p-vinyltoluene
Phenylacetylene

E 29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications

E 355 Practice for Gas Chromatography Terms and Relationships

E 691 Practice for Conducting an Interlaboratory Study to Determine the Precision of Test Methods

E 1510 Practice for Installing Fused Silica Open Tubular Capillary Columns in Gas Chromatographs

2.2 Other Document

OSHA Regulations—29CFR paragraphs 1910.1000 and 1910.1200³

3. Terminology

3.1 See Terminology D 4790 for definitions of terms used in this test method.

4. Summary of Test Method

4.1 In this test method, the chromatogram peak area for each impurity is compared to the peak area of the internal standard (*n*-heptane or other suitable known) added to the sample. From the response factors of these impurities relative to that of the internal standard and the amount of internal standard added, the concentration of the impurities are calculated. The styrene

³ Available from U.S. Government Printing Office, Superintendent of Documents, 732 North Capitol St., NW, Mail Stop SDE, Washington, DC 20401.

*A Summary of Changes section appears at the end of this standard.

content is obtained by subtracting the total amount of all impurities from 100.00.

5. Significance and Use

5.1 This test method is designed to obtain styrene purity on the basis of impurities normally present in styrene and may be used for final product inspections and process control.

5.2 This test method will detect the impurities shown in Table 1, non-aromatic hydrocarbons containing ten carbons or less, and others where specific impurity standards are available. Absolute purity cannot be determined if unknown impurities are present.

6. Interferences

6.1 The internal standard chosen must be satisfactorily resolved from any impurity as well as the product peak. A peak will be satisfactorily resolved from a neighboring peak if the distance from the valley to the baseline between the two peaks is not greater than 50 % of the peak height of the smaller of the two peaks.

7. Apparatus

7.1 *Gas Chromatograph*—Any instrument having a flame ionization detector that can be operated at the conditions given in Table 2. The system shall have sufficient sensitivity to obtain a minimum peak height response for 0.001 weight % impurity of twice the height of the background noise.

7.2 *Column*—A capillary column containing a stationary phase of cross-linked polyethylene glycol has been found to be satisfactory. Any column can be used that is capable of resolving all significant impurities from styrene and from the internal standard.

7.3 *Recorder*—Electronic integration, with tangent capabilities is recommended.

7.4 *100-mL Volumetric Flask*.

7.5 *Microsyringes*, assorted volumes.

8. Reagents and Materials

8.1 *Purity of Reagent*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Commit-

tee on Analytical Reagents of the American Chemical Society,⁴ where such specifications are available.

8.2 *Carrier Gas*—Chromatographic grade helium is recommended (minimum purity of 99.95 mole %).

8.3 *Hydrogen and Air*, for FID detector.

8.4 *n-Heptane*, 99.0 % minimum purity, or other internal standard, such as *n*-octane, previously analyzed to be free of compounds coeluting with impurities in the sample.

8.5 *Styrene*, the highest purity available, but not less than 99.7 %.

8.6 *Pure Compounds*, for calibration, shall be those compounds that are typically present in commercial styrene. These should be at least 99 % pure as they are to be used for determining response factors.

9. Hazards

9.1 Consult the latest OSHA regulations, suppliers' Material Safety Data Sheets, and local regulations for all materials used in this procedure.

10. Sampling

10.1 Sample the material in accordance with Practice D 3437.

11. Preparation of Apparatus

11.1 Follow manufacturer's instructions for mounting and conditioning the column into the chromatograph and adjusting the instrument conditions described in Table 2, allowing sufficient time for the equipment to reach equilibrium. See Practices E 355 and E 1510 for additional information on gas chromatographic practices and terminology.

12. Calibration

12.1 Prepare a calibration mixture containing approximately 99.5 weight % styrene and the expected significant impurities at their expected concentration (see Practice D 4307). Weigh all components to the accuracy required to calculate the concentration of each to the nearest 0.001 %.

12.2 With a microsyringe, add 50 μ L of internal standard to a 100-mL volumetric flask about three-fourths full of calibration mixture. Mix well. Add calibration mixture to mark and again mix well. If *n*-heptane is used as the internal standard, using a density of 0.684 for *n*-heptane and 0.906 for styrene, this solution will contain 0.0377 weight % *n*-heptane.

12.3 Also prepare a sample of the styrene used for the calibration blend with and without *n*-heptane to determine the concentration of existing impurities and interfering compounds with internal standard. If impurities in the styrene emerge with the chosen internal standard, an alternate internal standard must be used.

12.4 Inject an appropriate amount of sample into the chromatograph and obtain a chromatogram.

TABLE 2 Typical Instrument Parameters

Carrier gas	helium
Carrier gas flow rate at 110°C, mL/min	1.2
Detector	flame ionization
Detector temperature, °C	240
Injection port temperature, °C	230
Hydrogen flow rate, mL/min	30
Air flow rate, mL/min	275
Make up gas	nitrogen
Make up gas flow rate, mL/min	23
Split flow, mL/min	150
Column	60 m \times 0.32 mm ID \times 0.5 μ m bonded polyethylene glycol-fused silica capillary
Column temperature, °C	110
Chart speed, cm/min	1
Sample size, μ L	0.6

⁴ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.