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Eksplzivne atmosfere - 20-1. del: Lastnosti materiala in razvrstitev za pline in hlapne - Preskusne metode in podatki (ISO/IEC/DIS 80079-20-1:2015)

Explosive atmospheres - Part 20-1: Material characteristics for gas and vapour classification - Test methods and data (ISO/IEC/DIS 80079-20-1:2015)

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Atmosphères explosives - Partie 20-1: Caractéristiques des produits pour le classement des gaz et des vapeurs - Méthodes et données d'essai (ISO/IEC/DIS 80079-20-1:2015)

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ICS:

29.260.20	Električni aparati za eksplozivna ozračja	Electrical apparatus for explosive atmospheres
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Explosive atmospheres —

Part 20-1:

Material characteristics for gas and vapour classification — Test methods and data

Atmosphères explosives —

Partie 20-1: Caractéristiques des produits pour le classement des gaz et des vapeurs — Méthodes et données d'essai

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ISO/CEN PARALLEL PROCESSING

This draft has been developed within the International Organization for Standardization (ISO), and processed under the **ISO lead** mode of collaboration as defined in the Vienna Agreement.

This draft is hereby submitted to the ISO member bodies and to the CEN member bodies for a parallel five month enquiry.

Should this draft be accepted, a final draft, established on the basis of comments received, will be submitted to a parallel two-month approval vote in ISO and formal vote in CEN.

To expedite distribution, this document is circulated as received from the committee secretariat. ISO Central Secretariat work of editing and text composition will be undertaken at publication stage.



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ISO copyright office
Ch. de Blandonnet 8 • CP 401
CH-1214 Vernier, Geneva, Switzerland
Tel. +41 22 749 01 11
Fax +41 22 749 09 47
copyright@iso.org
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155 It is published as a double logo standard.

156 The text of this standard is based on the following documents:

FDIS	Report on voting
31M/xxx/xxx	31M/xxx/xxx

157

158 Full information on the voting for the approval of this standard can be found in the report on
159 voting indicated in the above table. In ISO, the standard has been approved by xx P members
160 out of xx having cast a vote.

161 This publication has been drafted in accordance with the ISO/IEC Directives, Part 2.

162 A list of all parts of the IEC 60079 respectively ISO/IEC 80079 series, under the general title:
163 *Explosives atmospheres* can be found on the IEC website.

164 The committee has decided that the contents of this publication will remain unchanged until
165 the stability date indicated on the IEC website under "<http://webstore.iec.ch>" in the data
166 related to the specific publication. At this date, the publication will be

- 167 • reconfirmed,
- 168 • withdrawn,
- 169 • replaced by a revised edition, or
- 170 • amended.

171

172 The National Committees are requested to note that for this publication the stability date
173 is 2020

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174 THIS TEXT IS INCLUDED FOR THE INFORMATION OF THE NATIONAL COMMITTEES AND WILL BE
175 DELETED AT THE PUBLICATION STAGE. [2/sist-en-iso-iec-80079-20-1-2020](#)

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EXPLOSIVE ATMOSPHERES –

Part 20-1: Material characteristics for gas and vapour classification – Test methods and data

186 **1 Scope**

187 This part of ISO/IEC 80079 provides guidance on classification of gases and vapours. It
188 describes a test method intended for the measurement of the maximum experimental safe
189 gaps (MESG) for gas-air mixtures or vapour-air mixtures under normal conditions of
190 temperature¹ and pressure (20 °C, 100 kPa) so as to permit the selection of an appropriate
191 group of equipment. The standard describes also a test method intended for use in the
192 determination of the auto-ignition temperature (AIT) of a vapour-air mixture or gas-air mixture
193 at atmospheric pressure, so as to permit the selection of an appropriate temperature class of
194 equipment.

195 Values of chemical properties of materials are provided to assist in the selection of equipment
196 to be used in hazardous areas. Further data may be added as the results of validated tests
197 become available.

198 The materials and the characteristics included in a table (see Annex B) have been selected
199 with particular reference to the use of equipment in hazardous areas. The data in this
200 standard have been taken from a number of references which are given in the bibliography.

201 These methods for determining the MESG or the AIT may also be used for gas-air-inert
202 mixtures or vapour-air-inert mixtures. However, data on air-inert mixtures are not tabulated.

203 **2 Normative references**

204 The following referenced documents are indispensable for the application of this document.
205 For dated references, only the edition cited applies. For undated references, the latest edition
206 of the referenced document (including any amendments) applies.

207 IEC 60079-11, *Explosive atmospheres – Part 11: Equipment protection by intrinsic safety "i"*

208 IEC 60079-14, *Explosive atmospheres – Part 14: Electrical installations design, selection and*
209 *erection*

210 **3 Terms and definitions**

211 For the purposes of this document, the following terms and definitions apply.

212 Note 1 to entry: For the definitions of any other terms, particularly those of a more general nature, reference should
213 be made to IEC 60050-426 or other appropriate parts of the IECV (International Electrotechnical Vocabulary).

214 **3.1**

215 **ignition by hot surface (auto-ignition)**

216 a reaction which is evidenced by a clearly perceptible flame and/or explosion, and for which
217 the ignition delay time does not exceed 5 min

218
219

Note 1 to entry: See 7.2.2 for a test method

1 An exception is made for substances with vapour pressures which are too low to permit mixtures of the required concentrations to be prepared at normal ambient temperatures. For these substances, a temperature 5 K above that needed to give the necessary vapour pressure or 50 K above the flash point is used.

- 220 **3.2**
221 **ignition delay time**
222 time between the completed injection of the flammable material and the ignition
- 223 **3.3**
224 **auto-ignition temperature**
225 **AIT**
226 lowest temperature (of a hot surface) at which under specified test conditions an ignition of a
227 flammable gas or vapour in mixture with air or air-inert gas occurs
- 228 Note 1 to entry: See Clause 7 for a test method.
- 229 **3.4**
230 **maximum experimental safe gap**
231 **MESG**
232 maximum gap of a joint of 25 mm in width which prevents any transmission of an explosion
233 during the tests made under specified conditions
- 234 Note 1 to entry: See Clause 6 for a test method.
- 235 **3.5**
236 **minimum igniting current**
237 **MIC**
238 minimum current in a specified test circuit that causes the ignition of the explosive test mixture
239 in the spark-test apparatus according to IEC 60079-11
- 240 Note 1 to entry: See 5.1.6 for the test circuit.
- 241 **3.6**
242 **flammable limits**
243 the flammable limits of a gas or vapor are the lower (LFL) and upper (UFL) flammable limit, stated in
244 percent by volume or in g/m³ of gas in a gas-air mixture, between which a flammable mixture is formed
- 245 Note 1 to entry: In the past, flammable and explosive have been used interchangeably in many texts, but the trend
246 is to avoid the confusion that this causes. The term flammable relates to the properties of the material that
247 determine its ability to produce self-sustaining flame propagation in any direction (upwards, sideways or
248 downwards). The term explosive relates to flame propagation that is accompanied by pressure and temperature
249 rise and noise (usually higher-speed propagation) and is significantly affected by (non-material related) test-
250 chamber conditions (geometry, degree of confinement...). LFL concentrations are typically lower than LEL
251 concentrations for the same material and UFL concentrations are typically higher than UEL concentrations for the
252 same material.
- 253 **3.7**
254 **equipment grouping**
255 classification system of equipment related to the explosive atmosphere for which they are intended to
256 be used
- 257 Note 1 to entry: IEC 60079-0 identifies three equipment groups:
- 258 Group I, equipment for mines susceptible to fire damp;
- 259 Group II, which is divided into sub-groups, equipment for all places with an explosive gas atmosphere
260 other than mines susceptible to fire damp
- 261 Group III, which is divided into sub-groups, equipment for all places with an explosive dust atmosphere
262 other than mines susceptible to fire damp
- 263 **3.8**
264 **flash point**
265 **FP**
266 lowest liquid temperature at which, under specified test conditions, a liquid gives off vapours in
267 quantity such as to be capable of forming an ignitable vapour-air mixture
- 268 **3.9**
269 **gas**
270 gaseous phase of a substance that cannot reach equilibrium with its liquid or solid state in the
271 temperature and pressure range of interest
- 272 Note 1 to entry: This is a simplification of the scientific definition, and merely requires that the substance is above
273 its boiling point or sublimation point at the ambient temperature and pressure.

274 **3.10**275 **vapour**

276 gaseous phase of a substance that can reach equilibrium with its liquid or solid state in the
277 temperature and pressure range of interest

278 Note 1 to entry: This is a simplification of the scientific definition, and merely requires that the substance is below
279 its boiling point or sublimation point at the ambient temperature and pressure.

280 **4 Classification of gases and vapours**281 **4.1 General**

282 Group I addresses mines susceptible to firedamp.

283 Note Firedamp consists mainly of methane, but always contains small quantities of other gases, such as nitrogen,
284 carbon dioxide, and hydrogen, and sometimes ethane and carbon monoxide. The terms firedamp and methane are
285 used frequently in mining practice as synonyms.

286 Group II addresses flammable gases and vapours other than in mines susceptible to
287 firedamp. Group II gases and vapours are classified according to their MESG and/or MIC ratio
288 into subgroups.

289 All flammable materials are classified according to their AIT into temperature classes.

290 **4.2 Classification according to the maximum experimental safe gaps (MESG)**

291 Gases and vapours may be classified according to their MESG into Groups IIA, IIB or IIC,
292 based on the determination method described in this standard. This method does not take into
293 account the possible effects of obstacles on the safe gaps².

294 NOTE 1 The standard method for determining MESG is described in 6.2, but where determinations have been
295 undertaken only in an 8 l spherical vessel with ignition close to the flange gap these can be accepted provisionally.

296 For the purpose of classification the MESG limits are:

297 Group IIA: $MESG \geq 0,90 \text{ mm}$.

298 Group IIB: $0,50 \text{ mm} < MESG < 0,90 \text{ mm}$.

299 Group IIC: $MESG \leq 0,50 \text{ mm}$.

300 Determination of both the MESG and MIC ratio is required when $0,50 < MESG < 0,55$. Then
301 the Group is determined by MIC ratio,

302 NOTE 2 For gases and highly volatile liquids, the MESG is determined at 20 °C.

303 NOTE 3 If it was necessary to do the MESG determination at temperatures higher than ambient temperature a
304 temperature 5 K above that needed to give the necessary vapour pressure or 50 K above the flash point is used
305 and this value of MESG is given in the table and the classification of the equipment group is based on this result.

306 **4.3 Classification according to the minimum igniting current ratio (MIC ratio)**

307 Gases and vapours may be classified according to the ratio of their minimum igniting currents
308 (MIC) to the ignition current of laboratory methane into Groups IIA, IIB or IIC. The purity of
309 laboratory methane shall be not less than 99,9 %.

310 NOTE The standard method of determining MIC ratios is with the apparatus described in IEC 60079-11, but
311 where determinations have been undertaken in other apparatus these can be accepted provisionally.

312 For the purpose of classification the MIC ratios are:

313 Group IIA: $MIC > 0,80$.

314 Group IIB: $0,45 \leq MIC \leq 0,80$.

315 Group IIC: $MIC < 0,45$.

2 The design of the test apparatus for safe gap determination, other than that used for selecting the appropriate group of enclosure for a particular gas, may need to be different to the one described in this standard. For example, the volume of the enclosure, flange width, gas concentrations and the distance between the flanges and any external wall or obstruction may have to be varied. As the design depends on the particular investigation which is to be undertaken, it is impracticable to recommend specific design requirements, but for most applications the general principles and precautions indicated in the clauses of this standard will still apply.

316 Determination of both the MESH and MIC ratio is required when $0,80 < MIC < 0,90$ or
317 $0,45 < MIC < 0,50$. Then the Group is determined by MESH.

318 4.4 Classification according to a similarity of chemical structure

319 When a gas or vapour is a member of a homologous series of compounds, the classification
320 of the gas or vapour can provisionally be inferred from the data of the neighbouring members
321 of the series.

322 If the classification of the neighbouring members is based on both MESH and MIC ratio
323 classification according to similarity of chemical structure is not allowed.

324 4.5 Classification of mixtures of gases

325 Mixtures of gases should generally be allocated to a group only after a special determination
326 of MESH or MIC ratio. One method to estimate the group is to determine the MESH of the
327 mixture by applying a form of Le Châtelier's principle:

$$328 \quad MESH_{mix} = \frac{1}{\sum_i \left(\frac{X_i}{MESH_i} \right)}$$

329 Where X_i is the percentage by volume of material i and $MESH_i$ is the MESH of material i .

330 This method should not be applied in case of exceptions to the Le Châtelier's principle and to
331 mixtures and/or streams that have:

- 332 a) acetylene or its equivalent hazard (e.g. self decomposition properties);
- 333 b) oxygen or other strong oxidizer as one of the components;
- 334 c) large concentrations (over 5 %) of carbon monoxide. Because unrealistically high MESH
335 values may result, caution should be exercised with two component mixtures where one of
336 the components is an inert, such as nitrogen.

337 For mixtures containing an inert such as nitrogen in concentrations less than 5 % by volume,
338 use an MESH of infinity. For mixtures containing an inert such as nitrogen in concentrations
339 5 % and greater by volume, use an MESH of 2.

340 NOTE An alternate method that includes stoichiometric ratios is presented in the essay "Maximum experimental
341 safe gap of binary and ternary mixtures," by Brandes and Redeker [1].

342 5 Data for flammable gases and vapours, relating to the use of equipment

343 5.1 Determination of the properties

344 5.1.1 General

345 The compounds listed in this standard are in accordance with Clause 4, or have physical
346 properties similar to those of other compounds in that list.

347 5.1.2 Equipment group

348 The groups are the result of MESH or MIC ratio determination except where there is no value
349 listed for MESH or MIC ratio. For these, the group is based on chemical similarity (see
350 Clause 4).

351 NOTE If it was necessary to do the MESH determination at temperatures higher than ambient temperature a
352 temperature 5 K above that needed to give the necessary vapour pressure or 50 K above the Flash Point is used
353 and this value of MESH is given in the table of Annex B and the classification of the equipment group is based on
354 this result.

355 5.1.3 Flammable limits

356 Determinations have been made by a number of different methods, but the preferred method
357 is with a low energy ignition at the bottom of a vertical tube. The values (in percentage by
358 volume and mass per volume) are listed in the table of Annex B.

359 If the flash point is high, the compound does not form a flammable vapour air/mixture at
 360 normal condition of temperature (20 °C). Where flammability data are presented for such
 361 compounds the determinations have been made at a temperature sufficiently elevated to allow
 362 the vapour to form a flammable mixture with air.

363 5.1.4 Flash point FP

364 The value given in the table of Annex B is the “closed cup” measurement. When this data was
 365 not available the “open cup” value is quoted and indicated by (oc). The symbol < (less than),
 366 indicates that the flash point is below the value (in degree Celsius) stated, this probably being
 367 the limit of the apparatus used.

368 5.1.5 Temperature class

369 The temperature class of a gas or vapour is given according IEC 60079-14:

370 **Table 1 – Classification of temperature class and range of auto-ignition temperatures**

Temperature class	Range of auto-ignition temperature (AIT) °C
T1	> 450
T2	300 < AIT ≤ 450
T3	200 < AIT ≤ 300
T4	135 < AIT ≤ 200
T5	100 < AIT ≤ 135
T6	85 < AIT ≤ 100

371
 372 **5.1.6 Minimum igniting current (MIC)**
 373 The apparatus for the determination of minimum igniting current is defined in IEC 60079-11.
 374 The test apparatus shall be operated in a 24 V d.c. circuit containing a (95 ± 5) mH air-cored
 375 coil. The current in this circuit is varied to a minimum value until ignition of the most easily
 376 ignited concentration of the specific gas or vapour in air is obtained.

377 5.1.7 Auto-ignition temperature (AIT)

378 The value of auto-ignition temperature depends on the method of testing. The preferred
 379 method and data obtained is given in Clause 7 and in Annex A.

380 If the compound is not included in these data, the data obtained in similar apparatus, such as
 381 the apparatus described by ASTM International standard (ASTM E659), is listed ³.

382 5.2 Properties of particular gases and vapours

383 5.2.1 Coke oven gas

384 Coke oven gas is a mixture of hydrogen, carbon monoxide and methane. If the sum of the
 385 concentrations (vol %) of hydrogen and carbon monoxide is less than 75 % of the total,
 386 flameproof equipment of Group IIB is recommended, otherwise equipment of Group IIC is
 387 recommended.

388 5.2.2 Ethyl nitrite

389 The auto-ignition temperature of ethyl nitrite is 95 °C, above which the gas suffers explosive
 390 decomposition.

391 NOTE Ethyl nitrite is not be confused with its isomer, nitroethane.

³ Results from using the apparatus described in ASTM D2155 (now replaced by ASTM E659) were reported by C.J. Hilado and S.W. Clark. The apparatus is similar to the one used by Zabetakis. If there is no determination by either the IEC apparatus, nor similar apparatus, the lowest value obtained in other apparatus is listed. A more comprehensive list of data for auto ignition temperature, with the reference to sources, is given by Hilado and Clark.

392 5.2.3 MESG of carbon monoxide

393 The MESG for carbon monoxide relates to a mixture with air saturated with moisture at normal
394 ambient temperature. This determination indicates the use of Group IIB equipment in the
395 presence of carbon monoxide. A larger MESG may be observed with less moisture. The
396 lowest MESG (0,65 mm) is observed for a mixture of CO/H₂O near 7: molar ratio. Small
397 quantities of hydrocarbon in the carbon monoxide-air mixture have a similar effect in reducing
398 the MESG so that Group IIB equipment is required.

399 5.2.4 Methane, Group IIA

400 Industrial methane, such as natural gas, is classified as Group IIA, provided it does not
401 contain more than 25 % (V/V) of hydrogen. A mixture of methane with other compounds from
402 Group IIA, in any proportion is classified as Group IIA.

403 6 Method of test for the maximum experimental safe gap (MESG)

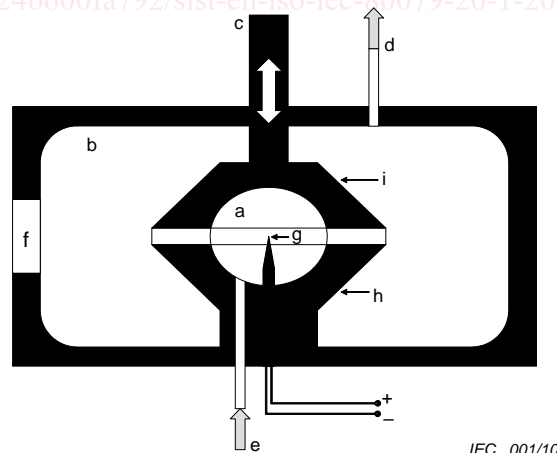
404 6.1 Outline of method

405 The interior and exterior chambers of the test apparatus are filled with a known mixture of the
406 gas or vapour in air, under normal conditions of temperature⁴ and pressure (20 °C, 100 kPa)
407 and with the circumferential gap between the two chambers accurately adjusted to the
408 desired value. The internal mixture is ignited and the flame propagation, if any, is observed
409 through the windows in the external chamber. The maximum experimental safe gap for the
410 gas or vapour is determined by adjusting the gap in small steps to find the maximum value
411 of gap which prevents ignition of the external mixture, for any concentration of the gas or
412 vapour in air.

413 6.2 Test apparatus

414 6.2.1 General

415 The apparatus is described in the following subclauses and is shown schematically in
416 Figure 1. It is also possible to use an automatic set-up when it is proven that the same results
417 are obtained as with a manual apparatus.



418

Key

a	interior spherical chamber	e	inlet of mixture
b	exterior cylindrical enclosure	f	observation windows
c	adjustable part	g	spark electrode
d	outlet of mixture	h	lower gap plate, fixed
		i	upper gap plate, adjustable

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Figure 1 – Test apparatus

⁴ An exception is made for substances with vapour pressures which are too low to permit mixtures of the required concentrations to be prepared at normal ambient temperatures. For these substances, a temperature 5 K above that needed to give the necessary vapour pressure or 50 K above the flash point is used.