
**Iron ore pellets for shaft direct-reduction
feedstocks — Determination of the
clustering index**

*Boulettes de minerais de fer pour charges utilisées dans les procédés
par réduction directe — Détermination du pouvoir collant*

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 11256 was prepared by Technical Committee ISO/TC 102, *Iron ore and direct reduced iron*, Subcommittee SC 3, *Physical testing*.

This second edition cancels and replaces the first edition (ISO 11256:1998), which has been revised to homogenise with other physical test standards.

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Introduction

This International Standard concerns one of a number of physical test methods that have been developed to measure various physical parameters and to evaluate the behaviour of iron ores, including reducibility, disintegration, crushing strength, apparent density, etc. This method was developed to provide a uniform procedure, validated by collaborative testing, to facilitate comparisons of tests made in different laboratories.

The results of this test should be considered in conjunction with other tests used to evaluate the quality of iron ores as feedstocks for direct-reduction processes.

This International Standard may be used to provide test results as part of a production quality control system, as a basis of a contract, or as part of a research project.

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Iron ore pellets for shaft direct-reduction feedstocks — Determination of the clustering index

CAUTION — This International Standard may involve hazardous materials, operations and equipment. This standard does not purport to address all of the safety issues associated with its use. It is the responsibility of the user of this International Standard to establish appropriate safety and health practices and to determine the applicability of regulatory limitations prior to its use.

1 Scope

This International Standard specifies a method to provide a relative measure for evaluating the formation of clusters of iron ore pellets when reduced under conditions resembling those prevailing in shaft direct-reduction processes.

This International Standard is applicable to hot-bonded pellets.

2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 2597-1:2006, *Iron ores — Determination of total iron content — Part 1: Titrimetric method after tin(II) chloride reduction*

ISO 3082:2000¹⁾, *Iron ores — Sampling and sample preparation procedures*

ISO 9035:1989, *Iron ores — Determination of acid-soluble iron(II) content — Titrimetric method*

ISO 9507:1990, *Iron ores — Determination of total iron content — Titanium(III) chloride reduction methods*

ISO 11323:2002, *Iron ore and direct reduced iron — Vocabulary*

3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 11323 apply.

4 Principle

The test portion is isothermally reduced in a fixed bed, at 850 °C, under static load, using a reducing gas consisting of H₂, CO, CO₂ and N₂, until a degree of reduction of 95 %. The reduced test portion (cluster) is disaggregated by tumbling, using a specific tumble drum. The clustering index is calculated as the mass of clustered material accumulated after specified disaggregation operations.

1) Under revision to incorporate ISO 10836, *Iron ores — Method of sampling and sample preparation for physical testing*.

5 Sampling, sample preparation and preparation of test portions

5.1 Sampling and sample preparation

Sampling of a lot and preparation of a test sample shall be in accordance with ISO 3082.

The size range for pellets shall be 50 % – 16,0 mm + 12,5 mm, and 50 % – 12,5 mm + 10,0 mm.

A test sample of at least 10 kg, on a dry basis, of sized pellets shall be obtained.

Oven-dry the test sample to constant mass at $105\text{ °C} \pm 5\text{ °C}$ and cool it to room temperature before preparation of the test portions.

NOTE Constant mass is achieved when the difference in mass between two subsequent measurements becomes less than 0,05 % of the initial mass of the test sample.

5.2 Preparation of test portions

Collect each test portion by taking ore particles at random.

NOTE Manual methods of division recommended in ISO 3082, such as riffing, can be applied to obtain the test portions.

At least 5 test portions, each of approximately 2 000 g (\pm the mass of 1 particle) shall be prepared from the test sample: 4 test portions for testing and 1 for chemical analysis.

Weigh the test portions to the nearest 1 g and register the mass of each test portion on its recipient label.

6 Apparatus

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6.1 General

The test apparatus shall comprise:

- a) ordinary laboratory equipment, such as an oven, hand tools, a time-control device and safety equipment;
- b) a reduction-tube assembly, including a loading device;
- c) a furnace, equipped with a balance for permitting the mass loss of the test portion to be read at any time during the test;
- d) a system to supply the gases and regulate the flow rates;
- e) a tumble drum;
- f) a weighing device.

Figure 1 shows an example of the test apparatus.

6.2 Reduction tube, with a double wall made of non-scaling, heat-resistant metal to withstand temperatures higher than 850 °C and resistant to deformation. The internal diameter of the inner reduction tube shall be $125\text{ mm} \pm 1\text{ mm}$. A removable perforated plate, made of non-scaling, heat-resistant metal to withstand temperatures higher than 850 °C , shall be mounted in the reduction tube to support the test portion and to ensure uniform gas flow through it. The perforated plate shall be 10 mm thick, with diameter 1 mm less than the internal diameter of the tube. The holes in the plate shall be 3 mm to 4 mm in diameter at a pitch center distance of 5 mm to 6 mm. The internal diameter of the outer reduction tube shall be large enough to allow gas-flow preheating before entering the inner reduction tube.

Figure 2 shows an example of a reduction tube.

6.3 Loading device, capable of supplying a total static load of $147 \text{ kPa} \pm 2 \text{ kPa}$ evenly to the test portion. The load shall be transferred by means of a ram with a rigid perforated footplate, so as to distribute it evenly to the surface of the porcelain balls placed on top of the test portion. The footplate shall be 10 mm thick and its diameter shall be 1 mm less than the internal diameter of the tube. The holes in the plate shall be 3 mm to 4 mm in diameter at a pitch center distance of 5 mm to 6 mm.

6.4 Porcelain balls, having a size range between 10,0 mm and 12,5 mm, and of sufficient quantity to form two double-layer beds on the perforated plate.

6.5 Furnace, having a heating capacity and temperature control able to maintain the entire test portion, as well as the gas entering the test portion, at $850 \text{ }^\circ\text{C} \pm 5 \text{ }^\circ\text{C}$.

6.6 Balance, capable of weighing the reduction-tube assembly, including the test portion, to an accuracy of 1 g. The balance shall have an appropriate device to suspend the reduction-tube assembly.

6.7 Gas-supply system, capable of supplying the gases and regulating gas flow rates. It shall be ensured that a frictionless connection between the gas-supply system and the reduction tube does not affect the weight-loss determination during reduction.

6.8 Tumble drum, made of steel plate at least 5 mm in thickness, having an internal diameter of 1 000 mm and an internal length of 500 mm. Two equally spaced L-shaped steel lifters, 50 mm flat \times 50 mm high \times 5 mm thick and 500 mm long, shall be solidly attached longitudinally inside the drum by welding, so as to prevent accumulation of material between the lifter and drum. Each lifter shall be fastened so that it points towards the axis of the drum, with its attached leg pointing away from the direction of rotation, thus providing a clear unobstructed shelf for lifting the sample. The door shall be so constructed as to fit into the drum to form a smooth inner surface. During the test, the door shall be rigidly fastened and sealed to prevent loss of the sample. The drum shall be rotated on stub axles attached to its ends by flanges welded so as to provide smooth inner surfaces. The drum shall be replaced, in any case, when the thickness of the plate is reduced to 3 mm in any area. The lifters shall be replaced when the height of the shelf is reduced to less than 47 mm.

Figure 3 shows an example of a tumble drum.

6.9 Drum-rotation equipment, capable of ensuring that the drum attains full speed in one revolution, rotates at a constant speed of $25 \text{ r/min} \pm 1 \text{ r/min}$ and stops within one revolution. The equipment shall be fitted with a revolution counter and with an automatic device for stopping the drum after a predetermined number of revolutions.

6.10 Weighing device, capable of weighing the test sample and test portions to an accuracy of 1 g.

7 Test conditions

7.1 General

Volumes and flow rates of gases are as measured at a reference temperature of $0 \text{ }^\circ\text{C}$ and at a reference atmospheric pressure of $101,325 \text{ kPa}$ ($1,013 25 \text{ bar}$).

7.2 Reducing gas

7.2.1 Composition

The reducing gas shall consist of:

CO	$30,0 \text{ } \% \pm 1,0 \text{ } \%$ (volume fraction)
CO ₂	$15,0 \text{ } \% \pm 1,0 \text{ } \%$ (volume fraction)

H₂ 45,0 % ± 1,0 % (volume fraction)

N₂ 10,0 % ± 1,0 % (volume fraction)

7.2.2 Purity

Impurities in the reducing gas shall not exceed:

O₂ 0,1 % (volume fraction)

H₂O 0,2 % (volume fraction)

7.2.3 Flow rate

The flow rate of the reducing gas, during the entire reducing period, shall be maintained at 40 L/min ± 0,5 L/min.

7.3 Heating and cooling gas

Nitrogen (N₂) shall be used as the heating and cooling gas. Impurities shall not exceed 0,1 % (volume fraction).

The flow rate of N₂ shall be maintained at 20 L/min until the test portion reaches 850 °C and at 40 L/min during the temperature-equilibration period. During cooling, it shall be maintained at 20 L/min.

7.4 Temperature of the test portion

The temperature of the entire test portion shall be maintained at 850 °C ± 5 °C during the entire reducing period and, as such, the reducing gas shall be preheated before entering the test portion.

7.5 Loading of the test portion

After 60 min of reduction, the test portion shall be subjected to a constant load of 147 kPa ± 2 kPa applied over the surface of the bed.

8 Procedure

8.1 Number of determinations for the test

Carry out the test as many times as required by the procedure in Annex A.

8.2 Chemical analysis

Take, at random, one of the test portions prepared in 5.2 and use it for the determination of the iron(II) oxide content (w_1) in accordance with ISO 9035 and total iron content (w_2) in accordance with ISO 2597-1 or ISO 9507.

8.3 Reduction

In order to achieve a more uniform gas flow, place a double-layer bed of porcelain balls (6.4) in the reduction tube (6.2) on the perforated plate, and level its surface. Place a ceramic-fibre wool along the inner wall of the reduction tube to avoid the test portion sticking to the wall.

Take, at random, another test portion prepared in 5.2 and record its mass (m_0). Place it in the reduction tube (6.2) and level its surface.

Place a further double layer of the porcelain balls upon the test portion.

Close the top of the reduction tube with the loading device (6.3). Insert the reduction-tube assembly into the furnace (6.5) and suspend it centrally from the balance (6.6), ensuring that there is no contact with the furnace wall or heating elements.

Connect the thermocouple, ensuring that its tip is in the centre of the test portion.

Connect the gas-supply system (6.7) and the compressed air to the loading device.

Pass a flow of N_2 through the test portion at a rate of at least 20 L/min and commence heating. When the temperature of the test portion approaches 850 °C, increase the flow rate to 40 L/min. Continue heating while maintaining the flow of N_2 until the mass of the test portion is constant and the temperature is constant at 850 °C for 10 min.

DANGER — Carbon monoxide and the reducing gas, which contains carbon monoxide, are toxic and therefore hazardous. Testing shall be carried out in a well ventilated area or under a hood. Precautions should be taken for the safety of the operator, in accordance with the safety codes of each country.

Record the mass of the test portion (m_1) and the time. Immediately introduce the reducing gas at a flow rate of 40 L/min to replace the N_2 . Record the mass of the test portion (m_t) continuously or at least every 3 min for the first 15 min and thereafter at 10 min intervals.

After 60 min of reduction, apply a load of $147 \text{ kPa} \pm 2 \text{ kPa}$, evenly to the test portion.

Calculate the degree of reduction, R_t , relative to the iron(III) state, after t min, as follows:

$$R_t = \left(\frac{0,111 w_1}{0,430 w_2} + \frac{m_1 - m_t}{m_0 \times 0,430 w_2} \times 100 \right) \times 100$$

where

m_0 is the mass, in grams, of the test portion;

m_1 is the mass, in grams, of the test portion immediately before starting the reduction;

m_t is the mass, in grams, of the test portion, after reduction time t ;

w_1 is the iron(II) oxide content, as a percentage by mass, of the test sample prior to the test, determined in accordance with ISO 9035, calculated from the iron(II) content by multiplying it by the oxide conversion factor $\text{FeO/Fe(II)} = 1,286$;

w_2 is the total iron content, as a percentage by mass, of the test portion prior to the test, determined in accordance with ISO 2597-1 or ISO 9507.

When the degree of reduction reaches 95 %, stop the flow of the reducing gas, remove the load and record the time.

Cool the reduced test portion to below 50 °C, under N_2 , at a flow rate of 20 L/min.