



Designation: D 2193 – 97 (Reapproved 2001)

Standard Test Method for Hydroquinone in Vinyl Acetate¹

This standard is issued under the fixed designation D 2193; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope

1.1 This test method covers the determination of hydroquinone in the range from 1 to 20 ppm in refined, commercially available, vinyl acetate.

1.2 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

1.3 For hazard information and guidance, see the supplier's Material Safety Data Sheet.

2. Referenced Documents

2.1 *ASTM Standards:*

D 1193 Specification for Reagent Water²

3. Summary of Test Method

3.1 The vinyl acetate is evaporated at room temperature in a stream of inert gas or clean air to minimize the loss of hydroquinone by evaporation. The hydroquinone is dissolved in water and titrated with dilute standardized ceric acid sulfate using diphenylamine as indicator.

4. Significance and Use

4.1 This test method provides a measurement of inhibitor level in vinyl acetate. The results of these measurements can be used for specification acceptance.

5. Apparatus

5.1 *Buret*, 25-mL, graduated in 0.1-mL subdivisions.

5.2 *Beakers*, 50 and 600-mL capacity.

5.3 *Volumetric Flask*, 1000-mL capacity.

5.4 *Erlenmeyer Flasks*, 100 and 250-mL capacity.

5.5 *Nitrogen Cylinder*, or source of clean air.

5.6 *Pipets*, 10 and 50-mL capacity.

6. Reagents

6.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.³ Other grades may be used provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

6.2 *Purity of Water*—Unless otherwise indicated, references to water shall be understood to mean reagent water conforming to Type IV of Specification D 1193.

6.3 *Ceric Acid Sulfate, Standard Solution (0.002 N)*—Dissolve 1.096 g of ceric ammonium nitrate ((NH₄)₂Ce(NO₃)₆) in 28.0 mL of concentrated sulfuric acid (H₂SO₄, sp gr 1.84) contained in a 50-mL beaker. Slowly pour the ceric solution, while stirring, into 200 mL of water contained in a 600-mL beaker. When solution is complete, transfer this mixture to a 1000-mL volumetric flask and dilute to the mark with water.

6.4 *Diphenylamine Indicator Solution*—Dissolve 0.1 g of diphenylamine in 100 mL of H₂SO₄ (sp gr 1.84) and store this solution in a brown glass bottle.

6.5 *Hydroquinone Standard*—Dissolve 200.0 mg of hydroquinone, weighed to the nearest 0.1 mg, in water and dilute to 1000.0 mL in a volumetric flask. This solution is unstable and should be discarded after 1 week of normal use.

7. Standardization

7.1 Pipet 10-mL portions of the hydroquinone standard (see 6.5) into each of two 100-mL Erlenmeyer flasks. Add 3 drops of diphenylamine indicator solution to each flask. Using a 25-mL buret, titrate the contents of each flask with ceric acid sulfate solution to a faint blue end point that is permanent for

¹ This test method is under the jurisdiction of ASTM Committee D01 on Paint and Related Coatings, Materials, and Applications and is the direct responsibility of Subcommittee D01.35 on Solvents, Plasticizers, and Chemical Intermediates.

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² *Annual Book of ASTM Standards*, Vol 11.01.

³ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.