

SLOVENSKI STANDARD oSIST prEN 14429:2013

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Karakterizacija odpadkov - Izluževalni preskus - Vpliv pH na izluževanje z začetnim dodatkom kisline/baze

Characterisation of waste - Leaching behaviour test - Influence of pH on leaching with initial acid/base addition

Charakterisierung von Abfällen - Untersuchung des Auslaugungsverhaltens - Einfluss des pH-Wertes unter vorheriger Säure/Base Zugabe

Caractérisation des déchets - Essais de comportement à la lixiviation - Influence du pH sur la lixiviation avec ajout initial d'acide/base

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Characterisation of waste - Leaching behaviour test - Influence of pH on leaching with initial acid/base addition

Caractérisation des déchets - Essais de comportement à la lixiviation - Influence du pH sur la lixiviation avec ajout initial d'acide/base

Charakterisierung von Abfällen - Untersuchung des Auslaugungsverhaltens - Einfluss des pH-Wertes unter vorheriger Säure/Base-Zugabe

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EUROPEAN COMMITTEE FOR STANDARDIZATION COMITÉ EUROPÉEN DE NORMALISATION EUROPÄISCHES KOMITEE FÜR NORMUNG

CEN-CENELEC Management Centre: Avenue Marnix 17, B-1000 Brussels

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Foreword

This document (prEN 14429:2013) has been prepared by Technical Committee CEN/TC 292 "Characterization of waste", the secretariat of which is held by NEN.

This document is currently submitted to the CEN Enquiry.

This document will supersede CEN/TS 14429:2005.

This document has been prepared under a mandate given to CEN by the European Commission and the European Free Trade Association.

This document has been developed primarily to support the requirements for leaching behaviour testing within EU and EFTA countries.

This document specifies test to determine the influence of pH on the leachability of inorganic constituents from waste materials.

For the complete characterization of the leaching behaviour of waste under specified conditions the application of other test methods is required, (see EN 12920).

Anyone dealing with waste and sludge analysis should be aware of the typical risks of that kind of material irrespective of the parameter to be determined. Waste and sludge samples can contain hazardous (e.g. toxic, reactive, flammable, infectious) substances, which can be liable to biological and/or chemical reaction.

Consequently these samples should be handled with special care. Gases which can be produced by microbiological or chemical activity are potentially flammable and will pressurise sealed bottles. Bursting bottles are likely to result in hazardous shrapnel, dust and/or aerosol. National regulations should be followed with respect to all hazards associated with this method.

Introduction

In the different European countries, tests have been developed to characterize and assess the constituents which can be leached from waste materials. The release of soluble constituents upon contact with water is regarded as one of the main mechanism of release which results in a potential risk to the environment during life-cycle of waste materials (disposal or re-use scenario). The intent of these tests is to identify the leaching properties of waste materials. The complexity of the leaching process makes simplifications necessary. Not all of the relevant aspects of leaching behaviour can be addressed in one single standard.

Procedures to characterize the behaviour of waste materials can generally be divided into three steps, using different tests in relation to the objective. The following test hierarchy is taken from the Landfill Directive ¹) and the Decision on Annex II of this Directive ²) for disposal of waste.

Basic characterization constitutes a full characterization of the waste by gathering all the necessary information for a safe management of the waste in the short and long term. Basic characterization may provide information on the waste (type and origin, composition, consistency, leachability, etc.), information for understanding the behaviour of waste in the considered management scenario, comparison of waste properties against limit values, and detection of key variables (critical parameters as liquid/solid (L/S) ratios, leachant composition, factors controlling leachability such as pH, redox potential, complexing capacity and physical parameters) for compliance testing and options for simplification of compliance testing. Characterization may deliver ratios between test results from basic characterization and results from simplified test procedures as well as information on a suitable frequency for compliance testing. In addition to the leaching behaviour, the composition of the waste should be known or determined by testing. The tests used for basic characterization should always include those to be used for compliance testing.

Compliance testing is used to demonstrate that the sample of today fits the population of samples tested before by basic characterization and through that, is used to carry out compliance with regulatory limit values. The compliance test should therefore always be part of the basic characterization program. The compliance test focuses on key variables and leaching behaviour identified by basic characterization tests. Parts of basic characterization tests can also be used for compliance purposes.

On-site verification tests are used as a rapid check to confirm that the waste is the same as that which has been subjected to characterization or compliance tests. On-site verification tests are not necessarily leaching tests.

The test procedure described in this document is a basic characterization test and falls in category a).

In this European Standard leaching is carried out at different pH values as a result of the reaction between pre-selected amounts of acid or base and test portions of the waste material. Size reduction is performed to facilitate approaching equilibrium. This test is different from the "pH dependence test with continuous pH control" (pH static test, see EN 14997) in which the pH is controlled at pre-selected values over the entire testing period by continuous measurement and automatic addition of acid or base. The test is aiming at approaching equilibrium at the end of the procedure.

NOTE In Annex B specific uses of both the pH dependence test with initial acid/ base addition and the pH dependence test with continuous pH control are indicated.

¹⁾ Council Directive 1999/31/EC of 26 April 1999 on the landfill of waste.

²⁾ Council Decision 2003/33/EC of 19 December 2002.

1 Scope

This European Standard is applicable for determining the influence of pH on the leachability of inorganic constituents from a waste material. Equilibrium condition as defined in the standard is established by addition of pre-determined amounts of acid or base to reach desired end pH values. This test method produces eluates, which are subsequently characterized physically and chemically.

This European Standard is a parameter specific test as specified in EN 12920. The application of this test method alone is not sufficient for the determination of the detailed leaching behaviour of a waste under specified conditions.

NOTE This generally requires the application of several test methods, behavioural modelling and model validation as specified in EN 12920.

2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN 14346, Characterization of waste – Calculation of dry matter by determination of dry residue and water content.

EN 14899, Characterization of waste – Sampling of waste materials – Framework for the preparation and application of a sampling plan

EN 15002, Characterization of waste – Preparation of test portions from the laboratory sample

EN 16192, Characterization of waste – Analysis of eluates

EN ISO 3696, Water for analytical laboratory use - Specification and test methods 9677-166a917684 [c/sist-

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EN ISO 5667-3, Water quality – Sampling – Part 3: Preservation and handling of samples

3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

3.1 dry residue *W*_{dr} remaining mass fraction of a sample after a drying process at 105 °C

[SOURCE: EN 14346:2007]

3.2

eluate solution obtained by a laboratory leaching test

[SOURCE: EN 12457-1:2002]

3.3

equilibrium

condition achieved when the pH deviation during a checking period at the last 4 h of the test is below 0,3 pH unit

3.4 laboratory sample

sample or subsample(s) sent to or received by the laboratory

[SOURCE: IUPAC, 1997]

When the laboratory sample is further prepared (reduced) by subdividing, cutting, sawing, coring, or Note 1 to entry: by combinations of these operations, the result is the test sample. When no preparation of the laboratory sample is required, the laboratory sample is the test sample. A test portion is removed from the test sample for the performance of the test or for analysis. The laboratory sample is the final sample from the point of view of sampling but it is the initial sample from the point of view of the laboratory.

Several laboratory samples may be prepared and sent to different laboratories or to the same Note 2 to entry: laboratory for different purposes. When sent to the same laboratory, the set is generally considered as a single laboratory sample and is documented as a single sample.

3.5

leachant

liquid that is brought into contact with the test portion in the leaching procedure

3.6 liquid to solid-ratio L/S

ratio between the amount of liquid (L) and of solid (S) in the test

Note 1 to entry: L/S is expressed in I/kg dry matter.

3.7

suspension

standards.iteh.ai) mixture of leachant and test portion

3.8

test portion 9f7684fc/sistamount or volume of the test sample taken for analysis, usually of known weight or volume

[SOURCE: IUPAC, 1997]

3.9

test sample

sample, prepared from the laboratory sample, from which test portions are removed for testing or analysis

[SOURCE: IUPAC, 1997]

Symbols and abbreviations 4

For the purposes of this document, the following symbols and abbreviations apply.

- ANC acid neutralization capacity
- BNC base neutralization capacity

DM dry matter

- DOC dissolved organic carbon
- L/S liquid to solid-ratio
- $M_{\rm d}$ dried mass of the test portion

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- *m*_d mass after drying
- *M*_w un-dried mass of the test portion
- *m*_r mass before drying
- *t*₀ time at the start of the leaching test
- V_{A/B} volume of acid or base used in leachant
- V_{demin} volume of demineralized water used in leachant
- V_L volume of added leachant
- $w_{\rm dr}$ dry residue of the waste

5 Principle

This European Standard describes a method to determine the influence of pH on the leachability of inorganic constituents from a waste material.

Separate test portions are leached at a fixed L/S ratio with leachants containing pre-selected amounts of acid or base in order to reach stationary pH values at the end of the extraction period. Each leachant is added in three steps in the beginning of the test. At least 8 final pH-values are required, covering at the minimum the range pH 4 to pH 12 (both included i.e. the lowest value \leq 4 and the highest value \geq 12). The amounts of acid or base needed to cover the pH range can be derived from the results of a preliminary titration, from available experimental data on the material to be tested or from an arbitrary division of the predetermined maximum consumption of acid and base. The tests are carried out at a fixed contact time at the end of which equilibrium can be assumed to be approached for most constituents in most waste materials to be characterized. The approaching of equilibrium as defined in the standard is verified at the end of the extraction period.

The results are expressed in mg/l of constituents for each final pH value. For each final pH value also the quantity of acid that is added is expressed in mol H^+/kg DM and the quantity of base that is added is expressed as negative mol OH⁻/kg DM (-mol OH⁻/kg DM).

NOTE 1 Other expression of results is possible (including mg leached /kg dry matter). Since this test is aiming at approaching equilibrium i.e. solubility controlled, the results alone cannot be used to quantify the soluble mass fraction.

The acid or base neutralization capacity (ANC, BNC) of the waste is also determined.

NOTE 2 The pH range covered by the test may be restricted to a pH range relevant for the specific material and the considered problem (see 9.2).

6 Reagents

Use only reagents of recognized analytical grade, unless otherwise specified.

6.1 Distilled water, demineralised water, de-ionised water or water of equivalent purity (5 < pH < 7,5) with a conductivity < 0,1 mS/m according to grade 2 specified in EN ISO 3696

6.2 Nitric acid, c(HNO₃) = 0,1 mol/l to 5 mol/l

6.3 Sodium hydroxide, *c*(NaOH) = 0,1 mol/l to 5 mol/l

NOTE Sodium hydroxide is unstable due to possible uptake of CO₂. Therefore it is recommended to prepare a fresh solution.

7 Equipment

7.1 General

Check the materials and equipment specified in 7.2.1, 7.2.4 and 7.2.9 before use for proper operation and absence of interfering elements that may affect the result of the test.

Calibrate the equipment specified in 7.2.2, 7.2.6, 7.2.7 and 7.2.8.

7.2 Laboratory equipment

Usual laboratory apparatus, and in particular the following:

7.2.1 Agitation device, end-over-end tumbler (5 r/min to 10 r/min) or roller table rotating at about 10 r/min

7.2.2 Analytical balance, with an accuracy of at least 0,1 g

7.2.3 Glass or plastic bottles, e.g. high density polyethylene (HDPE)/polypropylene (PP)/polytetrafluoroethylene (PTFE)/polyethyleneterephtalate (PET)

Use bottles with an appropriate volume (250 ml for the test portions of 15 g of dry mass, 500 ml for test portions of 30 g dry mass and 1 l for test portions of 60 g dry mass), and with screw cap, for eluate collection and preservation of eluate samples (rinsed in accordance with EN ISO 5667-3).

7.2.4 Crushing equipment, jaw crusher or a cutting device

NOTE Crushing is prescribed to avoid unnecessary grinding to very fine particle sizes, such as takes place in a rotary swing mill, ball mill or similar device.

7.2.5 Membrane filters, with a pore size of $0,45 \,\mu m_{0.0015}$

Membrane filters for the filtration device, fabricated from inert material, which is compatible with the waste. Filter shall be pre-rinsed with demineralised water or similarly clean in order to remove DOC.

- 7.2.6 **pH meter**, with an accuracy of at least 0,05 pH units
- 7.2.7 Conductivity meter, with an accuracy of at least 0,1 mS/m
- 7.2.8 Redox potential meter (optional)
- 7.2.9 Vacuum filtration device or pressure filtration device

8 Sample preparation

8.1 Laboratory sample

The laboratory sample shall consist of a mass equivalent of at least 1 kg of dry mass. In case less material is available, a justification shall be provided in the test report.

Perform sampling in accordance with EN 14899 or a standard derived from EN 14899 in order to obtain a representative laboratory sample.

8.2 Preparation of test sample

The tests shall be made on material with a grain size of 95 % less than 1 mm. In order to ensure that the test sample consists in 95 % mass of particles less than 1 mm in diameter, it shall be sieved, using the sieving equipment (see 7.2.4), to separate the oversized particles. If oversized material exceeds 5 % (mass), the entire oversized fraction shall be crushed. Any non-crushable material (e.g. metallic parts such as nuts, bolts, scrap) shall be separated from the oversized fraction and the weight and nature of the non-crushable material shall be recorded. Crushed and uncrushed material shall be mixed to constitute the test sample.

Moist material that is not possible to sieve needs to be dried prior to sieving and/or crushing. The drying temperature shall not exceed 40 °C.

Perform size reduction, drying, if needed, and sub-sampling according to specifications provided in EN 15002.

NOTE 1 There is no obligation to sieve if it is obvious that the material will fully pass the sieve on 1 mm.

NOTE 2 The crushed material can change upon storage due to ageing of fresh surfaces. It is therefore preferable to test the material as soon as possible after crushing.

NOTE 3 It is recommended that materials with a high natural pH be crushed under nitrogen in order to avoid contact with air leading to carbonation.

8.3 Determination of dry residue

The whole test sample, complying with the size criteria in 8.2 shall not be further dried. The dry residue (w_{dr}) of the test sample shall be determined on a separate test portion.

The dry residue shall be determined at 105 °C ± 5 °C according to EN 14346. The dry residue is calculated as follows:

$$w_{dr} = 100 \times \frac{m_d}{m_r} \underbrace{\text{SIST EN 14429:2015}}_{\text{https://standards.iteh.ai/catalog/standards/sist/5326967c-9ca8-49cf-9677-fb6a9f7684fc/sist-}$$
(1)

where

 w_{dr} is the dry residue of the waste, expressed as a percentage (%);

- $m_{\rm d}$ is the mass after drying, in grams (g);
- $m_{\rm r}$ is the mass before drying, in grams (g).

8.4 Preparation of the test portion

Prepare at least 8 test portions by the use of a sample splitter or by coning and quartering in accordance with EN 15002. Based on sample heterogeneity and eluate volume requirement for analysis, test portion size shall be either M_d = 15 g, 30 g or 60 g (with a tolerance of ± 10 %).

Calculate the undried mass of the test portion $M_{\rm w}$ to be used for the test as follows:

$$M_{\rm w} = \frac{M_{\rm d}}{w_{\rm dr}} \times 100 \tag{2}$$

where

 M_d is the dried mass of the test portion, in grams (g);

 $M_{\rm w}$ is the undried mass of the test portion, in grams (g);

 w_{dr} is the dry residue of the waste, expressed as a percentage (%).

9 Procedure

9.1 Contact time

The leaching procedure consists of three defined stages:

- Period A (acid/base addition) from t_0 up t_0 +4 h for acid/base addition in three steps;
- Period B (equilibration period) from t_{0+4h} up to t_0 +44 h equilibration period;
- Period C (verification period) from $t_{0+44 h}$ up to $t_0 + 48 h$ for verification of equilibrium condition.

Measure the pH in the liquid after each of these periods.

The total contact period (A+B+C) is 48 h.

9.2 pH-range

The test shall cover the range pH 2 to pH 12 (both included i.e. the lowest value \leq 2 and the highest value \geq 12) with at least 8 pH values tested including the natural pH (without acid or base addition). The maximum difference between two consecutive pH values shall not exceed 1,5 pH units.

NOTE 1 To ensure that the appropriate pH values can be obtained in one run additional bottles can be prepared of which only the ones with the desired final pH values are retained for analysis.

NOTE 2 The pH range covered by the test may be restricted to a pH range relevant for the specific material and the considered problem. The pH range to be covered can depend on the specific properties of the waste material, the available information on this material and the questions to be answered by performing the test. The number of pH levels considered can be reduced, correspondingly.

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9.3 Leaching test teh.ai/catalog/standards/sist/5326967c-9ca8-49cf-9677-fb6a9f7684fc/sist-

General

The following procedure applies for each of the chosen pH values to be tested.

9.3.2 Preparation of leachant

Identify the acid or base consumption for reaching the relevant pH values as A (mol H^+/kg dry matter) or B (mol H^+/kg dry matter).

NOTE 1 The acid or base consumption for the considered pH values may be derived from available information, from the preliminary procedures in Annex C or from information in Annex D.

Calculate the volume V of liquid to establish L/S = 10 l/kg \pm 0,2 l/kg for the actual size of test portion M_w (see 8.4) including the volume of acid or base.

$$V = 10 \times M_{\rm d} \tag{3}$$

where

9.3.1

- *V* is the total volume of liquid in the test, in ml;
- M_{d} is the dry mass of the test portion, in grams (g) (see 8.4);

NOTE 2 In relation to L/S-ratio, V in this equation is equivalent with the "L" and M_d is equivalent to the "S".