
**Naprave za gašenje s plinom - Fizikalne lastnosti in projektiranje - 2. del:
Gasilo CF3I**

Gaseous fire-extinguishing systems - Physical properties and system design - Part
2: CF3I extinguishant

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**Gaseous fire-extinguishing systems —
Physical properties and system design —**

**Part 2:
CF₃I extinguishant**

*Systèmes d'extinction d'incendie utilisant des agents gazeux —
Propriétés physiques et conception des systèmes —*

Partie 2: Agent extincteur CF₃I

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 14520-2 was prepared by Technical Committee ISO/TC 21, *Equipment for fire protection and fire fighting*, Subcommittee SC 8, *Gaseous media and firefighting systems using gas*.

This second edition cancels and replaces the first edition (ISO 14520-2:2000), which has been technically revised. It incorporates ISO 14520-2:2000/Cor 1:2001.

ISO 14520 consists of the following parts, under the general title *Gaseous fire-extinguishing systems — Physical properties and system design*:

- *Part 1: General requirements*
- *Part 2: CF₃I extinguishant*
- *Part 5: FK-5-1-12 extinguishant*
- *Part 6: HCFC Blend A extinguishant*
- *Part 8: HFC 125 extinguishant*
- *Part 9: HFC 227ea extinguishant*
- *Part 10: HFC 23 extinguishant*
- *Part 11: HFC 236fa extinguishant*
- *Part 12: IG-01 extinguishant*
- *Part 13: IG-100 extinguishant*
- *Part 14: IG-55 extinguishant*
- *Part 15: IG-541 extinguishant*

Parts 3, 4 and 7, which dealt with FC-2-1-8, FC-3-1-10 and HCFC 124 extinguishants, respectively, have been withdrawn, as these types are no longer manufactured.

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Gaseous fire-extinguishing systems — Physical properties and system design —

Part 2: CF₃I extinguishant

1 Scope

This part of ISO 14520 gives specific requirements for gaseous fire-extinguishing systems, with respect to the CF₃I extinguishant. It includes details of physical properties, specification, usage and safety aspects and is applicable to systems operating at a nominal pressure of 25 bar. This does not preclude the use of other systems.

2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 14520-1:2006, *Gaseous fire-extinguishing systems — Physical properties and system design — Part 1: General requirements*

[SIST ISO 14520-2:2006](https://standards.iteh.ai/catalog/standards/sist/73349d73-0c99-4d49-89d5-6add81b36c39/sist-iso-14520-2-2006)

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3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 14520-1 apply.

4 Characteristics and uses

4.1 General

Extinguishant CF₃I shall comply with the specification according to Table 1.

CF₃I is a colourless, almost odourless, electrically non-conductive gas with a density approximately seven times that of air.

The physical properties are given in Table 2.

CF₃I extinguishes fires mainly by chemical means, but also by some physical means.

Table 1 — Specification for CF₃I

Property	Requirement
Purity	99,9 % by mass, min.
Acidity	1×10^{-6} by mass, max.
Water content	6×10^{-6} by mass, max.
Non-volatile residue	100×10^{-6} by mass, max.
Suspended matter or sediment	None visible

Table 2 — Physical properties of CF₃I

Property	Unit	Value
Molecular mass	—	195,9
Boiling point at 1,013 bar (absolute) ^a	°C	−22,5
Freezing point	°C	−110
Critical temperature	°C	122
Critical pressure	bar abs ^a	40,4
Critical volume	cm ³ /mol	225,0
Critical density	kg/m ³	871
Vapour pressure 20 °C	bar abs ^a	4,65
Liquid density 20 °C	kg/m ³	2 096
Saturated vapour density 20 °C	kg/m ³	8,051
Specific volume of superheated vapour at 1,013 bar and 20 °C	m ³ /kg	0,124
Chemical formula	CF ₃ I	
Chemical name	Trifluoroiodomethane	

^a 1 bar = 0,1 MPa = 10⁵ Pa; 1 MPa = 1 N/mm²

4.2 Use of CF₃I systems

CF₃I total flooding systems may be used for extinguishing fires of all classes within the limits specified in ISO 14520-1:2006, Clause 4.

The extinguishant requirements per volume of protected space are given in Table 3 for various levels of concentration. These are based on methods given in ISO 14520-1:2006, 7.6.

The extinguishing concentrations and design concentrations for *n*-heptane and Surface class A hazards are given in Table 4, and inerting concentrations in Table 5.

Table 3 — CF₃I total flooding quantity

Temperature <i>T</i>	Specific vapour volume <i>S</i>	CF ₃ I mass requirements per unit volume of protected space, <i>m/V</i> (kg/m ³) This information refers only to CF ₃ I, and may not represent any other products containing Trifluoroiodomethane as a component							
		Design concentration (by volume)							
°C	m ³ /kg	3 %	4 %	5 %	6 %	7 %	8 %	9 %	10 %
−25	0,1013	0,3053	0,4113	0,5196	0,6301	0,7430	0,8584	0,9763	1,0969
−20	0,1038	0,2980	0,4014	0,5070	0,6149	0,7251	0,8377	0,9528	1,0704
−15	0,1063	0,2909	0,3920	0,4851	0,6005	0,7081	0,8180	0,9304	1,0453
−10	0,1088	0,2843	0,3830	0,4837	0,5867	0,6918	0,7992	0,9090	1,0212
−5	0,1113	0,2779	0,3744	0,4729	0,5735	0,6763	0,7813	0,8886	0,9983
0	0,1138	0,2718	0,3661	0,4625	0,5609	0,6614	0,7641	0,8691	0,9764
5	0,1163	0,2659	0,3583	0,4526	0,5488	0,6472	0,7477	0,8504	0,9554
10	0,1188	0,2603	0,3507	0,4430	0,5373	0,6336	0,7320	0,8325	0,9353
15	0,1213	0,2550	0,3436	0,4339	0,5262	0,6205	0,7169	0,8153	0,9160
20	0,1238	0,2498	0,3366	0,4251	0,5156	0,6080	0,7024	0,7989	0,8975
25	0,1263	0,2449	0,3299	0,4167	0,5054	0,5960	0,6885	0,7831	0,8797
30	0,1288	0,2401	0,3235	0,4086	0,4956	0,5844	0,6751	0,7679	0,8627
35	0,1313	0,2356	0,3173	0,4008	0,4861	0,5733	0,6623	0,7532	0,8462
40	0,1338	0,2311	0,3114	0,3934	0,4771	0,5625	0,6499	0,7392	0,8304
45	0,1363	0,2269	0,3057	0,3861	0,4683	0,5522	0,6380	0,7256	0,8152
50	0,1388	0,2228	0,3002	0,3792	0,4599	0,5423	0,6265	0,7125	0,8005
55	0,1413	0,2189	0,2949	0,3725	0,4517	0,5327	0,6154	0,6999	0,7863
60	0,1438	0,2151	0,2898	0,3660	0,4439	0,5234	0,6047	0,6878	0,7727
65	0,1463	0,2114	0,2848	0,3598	0,4363	0,5145	0,5944	0,6760	0,7595
70	0,1488	0,2078	0,2800	0,3537	0,4290	0,5058	0,5844	0,6647	0,7467
75	0,1513	0,2044	0,2754	0,3479	0,4219	0,4975	0,5747	0,6537	0,7344
80	0,1538	0,2011	0,2709	0,3422	0,4150	0,4894	0,5654	0,6431	0,7224
85	0,1563	0,1979	0,2666	0,3367	0,4084	0,4816	0,5563	0,6328	0,7109
90	0,1588	0,1948	0,2624	0,3314	0,4020	0,4740	0,5476	0,6228	0,6997
95	0,1613	0,1917	0,2583	0,3263	0,3957	0,4666	0,5391	0,6132	0,6888
100	0,1638	0,1888	0,2544	0,3213	0,3897	0,4595	0,5309	0,6038	0,6783

m/V is the agent mass requirement (in kilograms per cubic metre); i.e. mass, *m*, in kilograms of agent required per cubic metre of protected volume *V* to produce the indicated concentration at the temperature specified;

V is the net volume of hazard (in cubic metres); i.e. the enclosed volume minus the fixed structures impervious to extinguishant

$$m = \left(\frac{c}{100 - c} \right) \frac{V}{S}$$

T is the temperature (in degrees Celsius); i.e. the design temperature in the hazard area;

S is the specific volume (in cubic metres per kilogram); the specific volume of superheated CF₃I vapour at a pressure of 1,013 bar may be approximated by

$$S = k_1 + k_2 T$$

where $k_1 = 0,1138$; $k_2 = 0,0005$

c is the concentration (in percent); i.e. the volumetric concentration of CF₃I in air at the temperature indicated, and a pressure of 1,013 bar absolute.