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Gaseous fire-extinguishing systems — Physical properties and system design —

Part 13: IG-100 extinguishant

Systèmes d'extinction d'incendie utilisant des agents gazeux iTeh STANDARE et conception des systèmes — Partie 13: Agent extincteur IG-100 (standards.iteh.ai)

<u>ISO 14520-13:2005</u> https://standards.iteh.ai/catalog/standards/sist/2d42f4bd-a5d5-479d-9992-079afaa95710/iso-14520-13-2005



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Foreword

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International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 14520-13 was prepared by Technical Committee ISO/TC 21, Equipment for fire protection and fire fighting, Subcommittee SC 8, Gaseous media and firefighting systems using gas.

This second edition cancels and replaces the first edition (ISO 14520-13:2000), which has been technically revised. (standards.iteh.ai)

ISO 14520 consists of the following parts, under the general title *Gaseous fire-extinguishing systems* — *Physical properties and system design*: ISO 14520-13:2005 https://standards.iteh.ai/catalog/standards/sist/2d42f4bd-a5d5-479d-9992-

https://standards.iteh.a/catalog/standards/sist/2d42t4bd-a5d5-479 – Part 1: General requirements 079afaa95710/iso-14520-13-2005

- Part 2: CF₃I extinguishant
- Part 5: FK-5-1-12 extinguishant
- Part 6: HCFC Blend A extinguishant
- Part 8: HFC 125 extinguishant
- Part 9: HFC 227ea extinguishant
- Part 10: HFC 23 extinguishant
- Part 11: HFC 236fa extinguishant
- Part 12: IG-01 extinguishant
- Part 13: IG-100 extinguishant
- Part 14: IG-55 extinguishant
- Part 15: IG-541 extinguishant

Parts 3, 4 and 7, which dealt with FC-2-1-8, FC-3-1-10 and HCFC 124 extinguishants, respectively, have been withdrawn, as these types are no longer manufactured.

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Gaseous fire-extinguishing systems — Physical properties and system design —

Part 13: IG-100 extinguishant

1 Scope

This part of ISO 14520 gives specific requirements for gaseous fire-extinguishing systems, with respect to the IG-100 extinguishant. It includes details of physical properties, specification, usage and safety aspects and is applicable to systems operating at nominal pressures of 200 bar and 300 bar, both at 15 °C. This does not preclude the use of other systems; however, design data for other pressures were not available at time of publication.

2 Normative references STANDARD PREVIEW

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies 4520-13:2005

https://standards.iteh.ai/catalog/standards/sist/2d42f4bd-a5d5-479d-9992-ISO 14520-1:—¹), Gaseous fire-extinguishing/issystems₁₃₋₂₀Physical properties and system design — Part 1: General requirements

3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 14520-1 apply.

4 Characteristics and uses

4.1 General

Extinguishant IG-100 shall comply with the specification according to Table 1.

IG-100 is a colourless, odourless, electrically non-conductive gas with a density approximately the same as that of air.

The physical properties are given in Table 2.

IG-100 extinguishes fires mainly by a reduction of the oxygen concentration in the atmosphere of the hazard enclosure.

¹⁾ To be published. (Revision of ISO 14520-1:2000)

Property	Requirement	
Purity	99,6 % by volume, min.	
Water content	50×10^{-6} by mass, max.	
Oxygen	0,1 % by volume, max.	
Only principal contaminants are shown. Other measurements may includy hydrocarbons, CO, NO, NO ₂ , CO ₂ , etc. Most are $< 20 \times 10^{-6}$.		

Table 1 — Specification for IG-100

Table 2 — Physical properties of IG-100

Property	Unit	Value		
Molecular mass	—	28,02		
Boiling point at 1,013 bar (absolute) ^a	°C	-195,8		
Freezing point	°C	-210,0		
Critical temperature	°C	—		
Critical pressure	bar abs ^a	—		
Critical volume iTeh STANDA	RD pcm ³ /mot	V —		
Critical density	kg/m ³	_		
Vapour pressure 20 °C	bar abs ^a			
Liquid density 20 °C ISO 1452	0-13:2005 kg/m ³			
Saturated vapour density 20 °Cttps://standards.iteh.ai/catalog/standa	rds/sist/2d42 kg/mi³a5d5-479d-	9992-		
Specific volume of superheated vapour at 1,013 bar and 20 °C	-14520-13-2005 m³/kg	0,858		
Chemical formula	N ₂			
Chemical name	Nitrogen			
^a 1 bar = 0,1 MPa = 10 ⁵ Pa; 1 MPa = 1 N/mm ² .				

4.2 Use of IG-100 systems

IG-100 total flooding systems may be used for extinguishing fires of all classes within the limits specified in ISO 14520-1:— $^{2)}$, Clause 4.

The extinguishant requirements per volume of protected space are shown in Table 3 for various levels of concentration. These are based on methods given in ISO 14520-1— $^{2)}$, 7.6.

The extinguishing concentrations and design concentrations for *n*-heptane and Surface Class A hazards are given in Table 4.

²⁾ To be published. (Revision of ISO 14520-1:2000)

Temperature	Specific vapour	IG-100 volume requirements per unit volume of protected space , <i>VIV</i> (m ³ /m ³) This information refers only to IG-100, and may not represent any other products containing							
<i>T</i>	volume		nitrogen as a component.						
1	5		Design concentration (by volume)						
°C	m ³ /kg	34 %	38 %	42 %	46 %	50 %	54 %	58 %	62 %
-40	0,4790	0,522	0,601	0,685	0,775	0,872	0,976	1,091	1,217
-35	0,4893	0,511	0,588	0,671	0,758	0,853	0,956	1,068	1,191
-30	0,4996	0,501	0,576	0,657	0,743	0,836	0,936	1,046	1,167
-25	0,5098	0,491	0,565	0,644	0,728	0,819	0,917	1,025	1,143
-20	0,5201	0,481	0,554	0,631	0,714	0,803	0,899	1,005	1,120
-15	0,5304	0,472	0,543	0,619	0,700	0,787	0,882	0,985	1,099
-10	0,5406	0,463	0,533	0,607	0,686	0,772	0,865	0,966	1,078
-5	0,5509	0,454	0,523	0,596	0,674	0,758	0,849	0,948	1,058
0	0,5612	0,446	0,513	0,585	0,661	0,744	0,833	0,931	1,038
5	0,5715	0,438	0,504	0,574	0,649	0,731	0,818	0,914	1,020
10	0,5817	0,430	0,495	0,564	0,638	0,718	0,804	0,898	1,002
15	0,5920	0,423	0,486	0,554	0,627	0,705	0,790	0,883	0,984
20	0,6023	0,416	0,478	0,545	0,616	0,693	0,777	0,868	0,968
25	0,6126	0,409	0,470	0,536	0,606	0,682	0,764	0,853	0,951
30	0,6228	0,402	0,462	0,527	0,596	0,670	0,751	0,839	0,936
35	0,6331	0,395	0,455	0,518	0,586	0,659	0,739	0,825	0,920
40	0,6434	0,389	0,448	0,510	0,577	0,649	0,727	0,812	0,906
45	0,6536	0,383	0,440	0,502	0,568	/0,639	0,716	0,799	0,892
50	0,6639	0,377	0,434	0,494	0,559	0,629	0,704	0,787	0,878
55	0,6742	0,371	0,427	90,487	0,550	0,619	0,694	0,775	0,864
60	0,6845	0,366	0,421	0,479	0,542	0,610	0,683	0,763	0,851
65	0,6947	0,360	0,414	0,472	0,534	0,601	0,673	0,752	0,839
70	0,7050	0,355	0,408	0,465	0,526	0,592	0,663	0,741	0,827
75	0,7153 ^{up}	0,350	0,403	0,459	0,519	0,584	0,654	0,730	0,815
80	0,7256	0,345	0,397	0,452	0,511	0,575	0,645	0,720	0,803
85	0,7358	0,340	0,391	0,446	0,504	0,567	0,636	0,710	0,792
90	0,7461	0,335	0,386	0,440	0,497	0,560	0,627	0,700	0,781
95	0,7564	0,331	0,381	0,434	0,491	0,552	0,618	0,691	0,770
100	0,7666	0,326	0,376	0,428	0,484	0,545	0,610	0,682	0,760

Table 3 — IG-100 total flooding quantity

VIV is the agent volume requirement (in cubic metres per cubic metre); i.e. the quantity *Q* (in cubic metres) of agent required at a reference temperature of 20 °C and a pressure of 1,013 bar per cubic metre of protected volume to produce the indicated concentration at the temperature specified:

$$Q_{\mathsf{R}} = m \cdot S_{\mathsf{R}}$$

where

S_R is the specific reference volume (in cubic metres per kilogram); i.e. the specific vapour volume at the filling reference temperature for superheated IG-100 vapour at a pressure of 1,013 bar which may be approximated by the formula:

 $S_{\mathsf{R}} = k_1 + k_2 \cdot T_{\mathsf{R}}$

where $k_1 = 0,799$ 68; $k_2 = 0,002$ 93; T_R is the reference temperature (in degrees Celsius); i.e. filling temperature (20 °C in the table);

$$m = \frac{V}{S} \cdot \ln\left(\frac{100}{100 - c}\right)$$

- *V* is the net volume of hazard (in cubic metres); i.e. the enclosed volume minus the fixed structures impervious to extinguishant;
- *T* is the temperature (in degrees Celsius); i.e. the design temperature in the hazard area;
- *S* is the specific volume (in cubic metres per kilogram); the specific volume of superheated IG-100 vapour at a pressure of 1,013 bar may be approximated by

$$S = k_1 + k_2 \cdot T$$

c is the concentration (in percent); i.e. the volumetric concentration of IG-100 in air at the temperature indicated, and a pressure of 1,013 bar absolute.

Fuel	Extinguishment % by volume	Minimum design % by volume
Class B		
Heptane (cup burner) Heptane (room test)	33,6 33,6	43,7
Surface Class A		
Wood crib PMMA PP ABS	30,0 28,8 30,0 31,0	40,3
Higher Hazard Class A	a	41,5

Table 4 — IG-100 reference extinguishing and design concentrations

The extinguishment values for the Class B and the Surface Class A fuels are determined by testing in accordance with ISO 14520-1:— ³⁾, Annexes B and C.

The minimum design concentration for the Class B fuel is the higher value of the heptane cup burner or room test heptane extinguishment concentration multiplied by 1,3.

The minimum design concentration for Surface Class A fuel is the highest value of the wood crib, PMMA, PP or ABS extinguishment concentrations multiplied by 1,3. In the absence of any of the 4 extinguishment values, the minimum design concentration for Surface Class A shall be that of Higher Hazard Class A.

See ISO 14520-1:- ³⁾, 7.5.1.3, for guidance on Class A fuels.

The extinguishing and design concentrations for room-scale test fires are for informational purposes only. Lower and higher extinguishing concentrations than those shown for room-scale test fires may be achieved and allowed when validated by test reports from internationally recognized laboratories.

^a The minimum design concentration for Higher Hazard Class A fuels shall be the higher of the surface Class A or 95 % of the Class B minimum design concentration.

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5 Safety of personnel

Any hazard to personnel created by the discharge of IG-100 shall be considered in the design of the system.

Potential hazards can arise from the following:

- a) the extinguishant itself, by oxygen reduction;
- b) the combustion products of the fire.

For minimum safety requirements, see ISO 14250-1:— ³⁾, Clause 5.

Physiological information for IG-100 is given in Table 5.

³⁾ To be published. (Revision of ISO 14520-1:2000)

Property	Value % by volume		
No observed adverse effect level (NOAEL)	43		
Lowest observed adverse effect level (LOAEL)	52		
These values are based on physiological effects in human subjects of hypoxic atmospheres. These values are the functional equivalents of NOAEL and LOAEL values, and correspond to 12 % minimum oxygen for the no-effect level and 10 % minimum oxygen for the low-effect level.			

Table 5 — Physiological information for IG-100

6 System design

6.1 Fill pressure

The fill pressure of the container shall not exceed the values given in Tables 6 and 7, for systems operating at 200 bar at 15 $^{\circ}$ C and 300 bar at 15 $^{\circ}$ C, respectively.

Other pressures may be used and the minimum design pressure specified accordingly.

The relationships between pressure and temperature are shown in Figure 1.

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Table 6 — Storage container characteristics for IG-100 — 200 bar

Property 14520-13:2005	Unit	Value		
Fillingpressure at 1512 @/catalog/standards/sist/2d42f4bd-at	5d5-4 5a rdª9992	- 200		
Maximum container working pressure at 50 °C	bar ^a	240		
Reference should be made to Figure 1 for further data on pressure/temperature relationships.				
^a 1 bar = 0,1 MPa = 10 ⁵ Pa; 1 MPa = 1 N/mm ² .				

Table 7 — Storage container characteristics for IG-100 — 300 bar

Property	Unit	Value		
Filling pressure at 15 °C	bar ^a	300		
Maximum container working pressure at 50 °C	bar ^a	360		
Reference should be made to Figure 1 for further data on pressure/temperature relationships.				
^a 1 bar = 0,1 MPa = 10 ⁵ Pa; 1 MPa = 1 N/mm ² .				