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SIST ISO 182-3:1996

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INTERNATIONAL
STANDARD**ISO**
182-3First edition
1993-04-01

Plastics — Determination of the tendency of compounds and products based on vinyl chloride homopolymers and copolymers to evolve hydrogen chloride and any other acidic products at elevated temperatures —

Part 3:
Conductometric method

Plastiques — Détermination de la tendance des compositions à base d'homopolymères et copolymères du chlorure de vinyle à dégager du chlorure d'hydrogène et éventuellement d'autres produits acides à températures élevées —

Partie 3: Méthode conductimétrique

Reference number
ISO 182-3:1993(E)

ISO 182-3:1993(E)**Foreword**

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

International Standard ISO 182-3 was prepared by Technical Committee ISO/TC 61, *Plastics*, Sub-Committee SC 6, *Ageing, chemical and environmental resistance*.

Together with the three other parts of ISO 182, it cancels and replaces ISO Recommendation R 182:1970, of which the four parts of ISO 182 constitute a technical revision.

ISO 182 consists of the following parts, under the general title *Plastics — Determination of the tendency of compounds and products based on vinyl chloride homopolymers and copolymers to evolve hydrogen chloride and any other acidic products at elevated temperatures*:

- Part 1: *Congo red method*
- Part 2: *pH method*
- Part 3: *Conductometric method*
- Part 4: *Potentiometric method*

Annexes A, B and C of this part of ISO 182 are for information only.

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International Organization for Standardization
Case Postale 56 • CH-1211 Genève 20 • Switzerland

Printed in Switzerland

Plastics — Determination of the tendency of compounds and products based on vinyl chloride homopolymers and copolymers to evolve hydrogen chloride and any other acidic products at elevated temperatures —

Part 3: Conductometric method

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WARNING — The use of this part of ISO 182 may involve hazardous materials, operations and equipment. This part of ISO 182 does not purport to address all of the safety problems associated with its use. It is the responsibility of the user of this part of ISO 182 to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

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1 Scope

1.1 This part of ISO 182 specifies a method for the determination of the thermal stability at elevated temperature of compounds and products based on vinyl chloride homopolymers and copolymers (in the following text abbreviated as PVC) which undergo dehydrochlorination (the evolution of hydrogen chloride).

1.2 The method may be used as a quality control test during manufacture and conversion of PVC compounds. It may also be used for the characterization of PVC compounds and products, especially with regard to the effectiveness of their heat-stabilizing systems.

It is suitable for coloured PVC compounds and products where a discolouration test under the action of heat may be unsatisfactory.

1.3 The method is recommended for compounded PVC materials and products only, although it can be used for polymers in powder form under appropriate conditions, to be agreed upon between the interested parties. The method is not recommended for PVC

compounds in the form of dry blends, since such materials may not be sufficiently homogeneous.

1.4 PVC compounds and products may evolve other decomposition products, in addition to hydrogen chloride, at elevated temperatures. A limited number of these decomposition products may affect the conductivity of water when they are absorbed into it. Compensation for this effect is not within the scope of this part of ISO 182, and therefore care is necessary in attempting to compare results for dissimilar compounds and products. In this case a method suitable for the determination of chloride ion (Cl⁻) in the absorbing solution shall be used (see ISO 182-4).

1.5 The method may also be applied to other plastics materials which can evolve hydrogen chloride or other hydrogen halides when heated under the conditions prescribed by the relevant specifications, or as agreed upon between the interested parties.

2 Normative references

The following standards contain provisions which, through reference in this text, constitute provisions of this part of ISO 182. At the time of publication, the editions indicated were valid. All standards are subject

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to revision, and parties to agreements based on this part of ISO 182 are encouraged to investigate the possibility of applying the most recent editions of the standards indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 182-4:1993, *Plastics — Determination of the tendency of compounds and products based on vinyl chloride homopolymers and copolymers to evolve hydrogen chloride and any other acidic products at elevated temperatures — Part 4: Potentiometric method.*

ISO 565:1990, *Test sieves — Metal wire cloth, perforated metal plate and electroformed sheet — Nominal sizes of openings.*

ISO 4793:1980, *Laboratory sintered (fritted) filters — Porosity grading, classification and designation.*

ISO 6353-2:1983, *Reagents for chemical analysis — Part 2: Specifications — First series.*

3 Definition

For the purposes of this part of ISO 182, the following definition applies.

3.1 stability time, t_s : Time, measured by reference to a predetermined change in the conductivity of absorbing demineralized water, required for a certain amount of hydrogen chloride to be evolved when a prescribed mass of PVC compound or product is maintained at an elevated temperature under the test conditions specified in this part of ISO 182.

4 Principle

A test portion of the PVC compound or product is maintained at an agreed temperature in a nitrogen gas stream and the hydrogen chloride evolved is absorbed in a given amount of demineralized water. The amount of hydrogen chloride evolved is determined in relation to the recorded change in conductivity of the water.

5 Reagents

During the test, use only reagents of recognized analytical grade in accordance with ISO 6353-2.

5.1 Pure nitrogen, containing less than 6 ppm oxygen and less than 0,1 ppm carbon dioxide by volume. The purity shall be such that when the gas is passed through demineralized water (see 5.3) for 1 h at a rate of 7,2 l/h \pm 0,1 l/h, the conductivity of the water remains unchanged.

The gas shall be dried by passing it through a suitable drying agent, and the flow-rate through the dehydrochlorination cell adjusted by means of a needle valve and measured using a suitable flowmeter.

5.2 Hydrochloric acid, aqueous solution, $c(\text{HCl}) = 0,1 \text{ mol/l}$.

5.3 Demineralized water, with a pH of $4,0 \pm 0,1$ and a conductivity not greater than $40 \mu\text{S/cm}$, adjusted by the addition of 0,1 mol/l hydrochloric acid.

6 Apparatus

The general arrangement of the apparatus is shown in figure 1. The figure shows a re-usable dehydrochlorination cell A. This cell may be replaced by a disposable cell B.

6.1 Dehydrochlorination cells.

6.1.1 Cell A (re-usable), with shape and dimensions as shown in figure 2.

A recommended procedure for cleaning the cell is given in annex A.

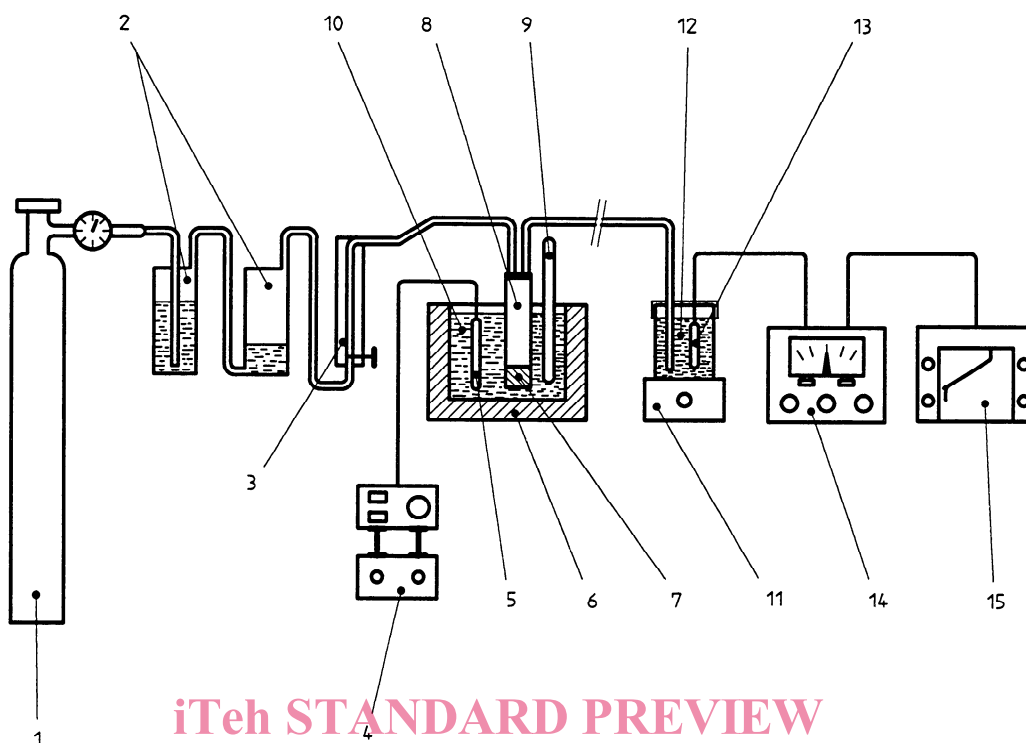
6.1.2 Cell B (disposable), with shape and dimensions as shown in figure 3.

NOTE 1 Other types of cell may be employed if it has been proved that the results obtained are equivalent to those obtained with one of the cells described in 6.1.1 and 6.1.2.

6.2 Test portion holder, for use with cell A. The test portion is supported on a porous sintered-glass disc (grade P 100, ISO 4793), 10 mm in diameter.

To prevent blocking of the porous disc, it is advisable to place a thin, soft layer of glass wool between it and the test portion.

A recommended procedure for cleaning the disc support is given in annex A.



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|-------------------------------------------------------------|---------------------------------------|
| 1 N ₂ cylinder | 9 Thermometer (scale division 0,1 °C) |
| 2 Purification train | 10 Silicone oil |
| 3 Ball flowmeter | 11 Magnetic stirrer |
| 4 Electronic temperature controller (scale division 0,1 °C) | 12 Absorbing solution |
| 5 Temperature sensor | 13 Measuring electrode(s) |
| 6 Heating bath | 14 Conductance meter |
| 7 PVC test portion | 15 Recorder |
| 8 Dehydrochlorination cell | |

Figure 1 — General arrangement of apparatus

Dimensions in millimetres

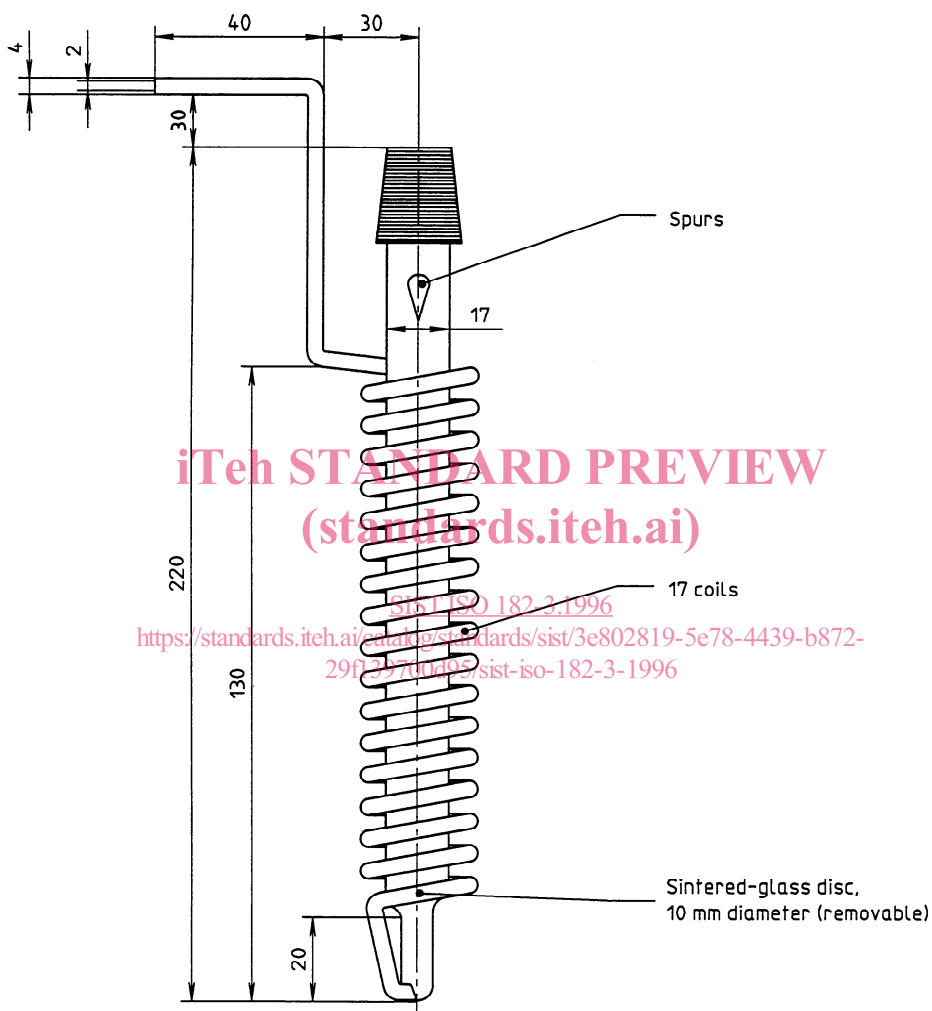


Figure 2 — Cell A (re-usable) for dehydrochlorination of PVC samples

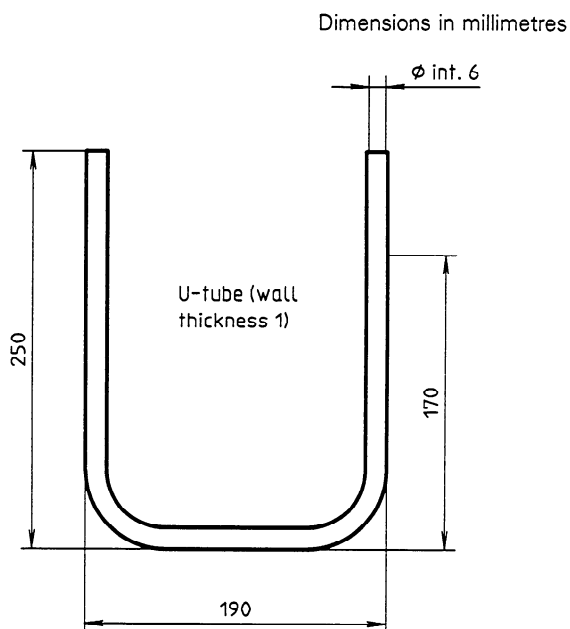


Figure 3 — Cell B (disposable) for dehydrochlorination of PVC samples

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6.3 Glass connection tube, for use with cell A, having dimensions as shown in figure 4. The connection tube is secured to cell A by two springs fixed to hooks on the ground-glass joints. The tube shall be provided with an insulating jacket.

6.4 Expansion joints and cell connections, for use with cell B. Cell B is connected to the apparatus through flexible polytetrafluoroethylene (PTFE) and silicone rubber tubes. Special joints allow for thermal expansion. The complete joint arrangement is shown in figure 5.

6.5 Oil bath, with a capacity of at least 10 l. The bath shall be capable of operating in the temperature range 170 °C to 210 °C and of maintaining the test temperature with an accuracy of $\pm 0,1$ °C.

The bath shall be designed so that the temperature distribution is even throughout, and shall have a thermal capacity sufficient to avoid temperature change when the dehydrochlorination cell is immersed in it.

6.6 Thermometer, with a scale suitable for reading the heating bath temperature in the range 170 °C to 210 °C and with a scale division of 0,1 °C.

6.7 Balance, with a scale division of 1 mg.

6.8 Measurement cell.

A suitable measurement cell is shown in figure 6. If the diameters of the measurement electrode and the feed tube for the gases from decomposition of the test portion are sufficiently small, a 300 ml Erlenmeyer flask is a suitable alternative measurement cell.

A recommended procedure for cleaning the cell is given in annex A.

6.9 Magnetic stirrer, capable of providing gentle agitation within the measurement cell.

6.10 Conductance meter, with a conductivity probe constant of 1 cm^{-1} , giving direct readings in $\mu\text{S/cm}$, and preferably equipped with an automatic temperature compensating device and an output for a chart recorder.

NOTE 2 Graphite electrodes have been shown to operate in a more trouble-free manner than platinized platinum electrodes.

6.11 Stopclock, or other suitable timing device, if not included in the recorder.

6.12 Flowmeter, for example a rotameter, or other suitable device capable of measuring a gas flow-rate within the range $120 \text{ cm}^3/\text{min} \pm 4 \text{ cm}^3/\text{min}$.

7 Preparation of test samples

The measured stability times t_s depend to some extent on the surface area of the prepared test portions as well as on their thermal history. Any cutting or grinding of a material necessary to produce the test portions shall be conducted in a uniform manner, avoiding heating of the material.

NOTE 3 Cryogenic grinding is recommended.

7.1 PVC plastisols

Spread these materials on glass plates and gel them in an oven at an agreed temperature so that sheets 0,5 mm thick are formed. Cut these sheets into squares approximately 2 mm on a side.