

**SLOVENSKI STANDARD  
SIST EN 902:2016****01-julij-2016****Nadomešča:  
SIST EN 902:2009****Kemikalije, ki se uporabljajo za pripravo pitne vode - Vodikov peroksid**

Chemicals used for treatment of water intended for human consumption - Hydrogen peroxide

Produkte zur Aufbereitung von Wasser für den menschlichen Gebrauch -  
Wasserstoffperoxid

**iTeh STANDARD PREVIEW**

**(standards.iteh.ai)**

Produits chimiques utilisés pour le traitement de l'eau destinée à la consommation humaine - Peroxyde d'hydrogène

[SIST EN 902:2016](#)

<https://standards.iteh.ai/catalog/standards/sist/fb13cbb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>

**Ta slovenski standard je istoveten z: EN 902:2016**

**ICS:**

|           |                             |                                     |
|-----------|-----------------------------|-------------------------------------|
| 13.060.20 | Pitna voda                  | Drinking water                      |
| 71.100.80 | Kemikalije za čiščenje vode | Chemicals for purification of water |

**SIST EN 902:2016****en,fr,de**

## iTeh STANDARD PREVIEW (standards.iteh.ai)

[SIST EN 902:2016](#)

<https://standards.iteh.ai/catalog/standards/sist/fb13cbb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>

EUROPEAN STANDARD  
NORME EUROPÉENNE  
EUROPÄISCHE NORM

EN 902

May 2016

ICS 71.100.80

Supersedes EN 902:2009

English Version

**Chemicals used for treatment of water intended for human  
consumption - Hydrogen peroxide**

Produits chimiques utilisés pour le traitement de l'eau  
destinée à la consommation humaine - Peroxyde  
d'hydrogène

Produkte zur Aufbereitung von Wasser für den  
menschlichen Gebrauch - Wasserstoffperoxid

This European Standard was approved by CEN on 18 March 2016.

CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration. Up-to-date lists and bibliographical references concerning such national standards may be obtained on application to the CEN-CENELEC Management Centre or to any CEN member.

This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the CEN-CENELEC Management Centre has the same status as the official versions.

CEN members are the national standards bodies of Austria, Belgium, Bulgaria, Croatia, Cyprus, Czech Republic, Denmark, Estonia, Finland, Former Yugoslav Republic of Macedonia, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugal, Romania, Slovakia, Slovenia, Spain, Sweden, Switzerland, Turkey and United Kingdom.

SIST EN 902:2016  
<https://standards.iteh.ai/catalog/standards/sist/fb13cbb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>



EUROPEAN COMMITTEE FOR STANDARDIZATION  
COMITÉ EUROPÉEN DE NORMALISATION  
EUROPÄISCHES KOMITEE FÜR NORMUNG

**CEN-CENELEC Management Centre: Avenue Marnix 17, B-1000 Brussels**

**Contents**

|  | Page      |
|--|-----------|
| <b>European foreword.....</b>  | <b>4</b>  |
| <b>Introduction .....</b>  | <b>5</b>  |
| <b>1 Scope.....</b>  | <b>6</b>  |
| <b>2 Normative references.....</b>   | <b>6</b>  |
| <b>3 Description.....</b>  | <b>6</b>  |
| <b>3.1 Identification .....</b>  | <b>6</b>  |
| <b>3.1.1 Chemical name .....</b>   | <b>6</b>  |
| <b>3.1.2 Synonym or common name .....</b>                                    | <b>6</b>  |
| <b>3.1.3 Relative molecular mass.....</b>                                    | <b>6</b>  |
| <b>3.1.4 Empirical formula .....</b>   | <b>6</b>  |
| <b>3.1.5 Chemical formula .....</b>  | <b>6</b>  |
| <b>3.1.6 CAS Registry Number .....</b>                                       | <b>6</b>  |
| <b>3.1.7 EINECS reference .....</b>  | <b>7</b>  |
| <b>3.2 Commercial form.....</b>  | <b>7</b>  |
| <b>3.3 Physical properties.....</b>  | <b>7</b>  |
| <b>3.3.1 Appearance and odour.....</b>                                       | <b>7</b>  |
| <b>3.3.2 Density .....</b>   | <b>7</b>  |
| <b>3.3.3 Solubility in water.....</b>  | <b>7</b>  |
| <b>3.3.4 Vapour pressure.....</b>  | <b>7</b>  |
| <b>3.3.5 Boiling point at 100 kPa .....</b>                                  | <b>7</b>  |
| <b>3.3.6 Crystallization point.....</b>                                      | <b>8</b>  |
| <b>3.3.7 Specific heat .....</b>   | <b>8</b>  |
| <b>3.3.8 Viscosity, dynamic .....</b>  | <b>8</b>  |
| <b>3.3.9 Critical temperature .....</b>                                      | <b>9</b>  |
| <b>3.3.10 Critical pressure .....</b>  | <b>9</b>  |
| <b>3.3.11 Physical hardness .....</b>  | <b>9</b>  |
| <b>3.4 Chemical properties.....</b>  | <b>9</b>  |
| <b>4 Purity criteria .....</b>   | <b>9</b>  |
| <b>4.1 General.....</b>  | <b>9</b>  |
| <b>4.2 Composition of commercial product.....</b>                            | <b>10</b> |
| <b>4.3 Impurities and main by-products .....</b>                             | <b>10</b> |
| <b>4.4 Chemical parameters.....</b>  | <b>10</b> |
| <b>5 Test methods .....</b>  | <b>10</b> |
| <b>5.1 Sampling.....</b>   | <b>10</b> |
| <b>5.2 Analysis.....</b>   | <b>10</b> |
| <b>5.2.1 Determination of hydrogen peroxide content (main product) .....</b> | <b>10</b> |
| <b>5.2.2 Chemical parameters.....</b>  | <b>13</b> |
| <b>6 Labelling - Transportation - Storage .....</b>                          | <b>15</b> |
| <b>6.1 Means of delivery .....</b>   | <b>15</b> |
| <b>6.2 Labelling according to the EU Legislation .....</b>                   | <b>15</b> |
| <b>6.3 Transportation regulations and labelling.....</b>                     | <b>19</b> |
| <b>6.4 Marking.....</b>  | <b>19</b> |
| <b>6.5 Storage .....</b>   | <b>19</b> |
| <b>6.5.1 Containers.....</b>   | <b>19</b> |

|   |           |
|---|-----------|
| 6.5.2 Long term stability .....   | 19        |
| 6.5.3 Storage incompatibilities.....  | 20        |
| <b>Annex A (informative) General information on hydrogen peroxide .....</b>   | <b>21</b> |
| A.1 Origin .....  | 21        |
| A.1.1 Raw materials .....   | 21        |
| A.1.2 Manufacturing process .....   | 21        |
| A.2 Use .....   | 21        |
| A.2.1 Function .....  | 21        |
| A.2.2 Form in which it is used.....   | 21        |
| A.2.3 Treatment dose .....  | 21        |
| A.2.4 Means of application.....   | 21        |
| A.2.5 Secondary effects .....   | 21        |
| A.2.6 Removal of excess product.....  | 21        |
| A.3 Routine analyses.....   | 22        |
| A.3.1 Determination of chemical parameters .....  | 22        |
| <b>Annex B (normative) General rules relating to safety.....</b>  | <b>23</b> |
| B.1 Rules for safe handling and use.....  | 23        |
| B.2 Emergency procedures.....   | 23        |
| B.2.1 First aid .....   | 23        |
| B.2.2 Spillage .....  | 23        |
| B.2.3 Fire.....   | 23        |
| <b>Annex C (normative) Determination of arsenic, antimony and selenium (atomic absorption spectrometry hydride technique) .....</b> | <b>24</b> |
| C.1 Safety precautions.....   | 24        |
| C.2 General principle .....   | 24        |
| C.3 Interferences.....  | 24        |
| C.4 Reagents.....   | 24        |
| C.5 Apparatus .....   | 26        |
| C.6 Procedure .....   | 28        |
| C.6.1 Preparation of the apparatus .....  | 28        |
| C.6.2 Preparation of calibration solutions .....  | 29        |
| C.6.3 Preparation of test solutions and standard solutions.....   | 29        |
| C.6.4 Determination of arsenic with sodium borohydride.....   | 29        |
| C.6.5 Determination of selenium with sodium borohydride .....   | 29        |
| C.6.6 Determination of antimony with sodium borohydride .....   | 30        |
| C.7 Calculation.....  | 30        |
| <b>Bibliography .....</b>   | <b>31</b> |

## European foreword

This document (EN 902:2016) has been prepared by Technical Committee CEN/TC 164 "Water supply", the secretariat of which is held by AFNOR.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by November 2016, and conflicting national standards shall be withdrawn at the latest by November 2016.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. CEN [and/or CENELEC] shall not be held responsible for identifying any or all such patent rights.

This document supersedes EN 902:2009.

Significant technical differences between this edition and EN 902:2009 are as follows:

- a) deletion of the reference to EU Directive 67/548/EEC of June 27,1967 in order to take into account the latest Directive in force (see [1]);
- b) use of the changed classification and labelling (see [3]).

According to the CEN/CENELEC Internal Regulations, the national standards organizations of the following countries are bound to implement this European Standard: Austria, Belgium, Bulgaria, Croatia, Cyprus, Czech Republic, Denmark, Estonia, Finland, Former Yugoslav Republic of Macedonia, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugal, Romania, Slovakia, Slovenia, Spain, Sweden, Switzerland, Turkey and the United Kingdom.  
<https://standards.iteh.ai/catalog/standards/sist-en-902-2016-412c-a874-329fd3b8753b/sist-en-902-2016>

## Introduction

In respect of potential adverse effects on the quality of water intended for human consumption, caused by the product covered by this European Standard:

- 1) this European Standard provides no information as to whether the product may be used without restriction in any of the Member States of the EU or EFTA;
- 2) it should be noted that, while awaiting the adoption of verifiable European criteria, existing national regulations concerning the use and/or the characteristics of this product remain in force.

NOTE 1 Conformity with this standard does not confer or imply acceptance or approval of the product in any of the Member States of the EU or EFTA. The use of the product covered by this European Standard is subject to regulation or control by National Authorities.

NOTE 2 This product is a biocide and needs to comply with the relevant legislation in force. In the European Union, at the time of publication, this legislation is REGULATION (EU) No 528/2012 [1].

## iTeh STANDARD PREVIEW (standards.iteh.ai)

[SIST EN 902:2016](#)

<https://standards.iteh.ai/catalog/standards/sist/fb13ccb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>

## 1 Scope

This European Standard is applicable only to hydrogen peroxide and not to mixtures with other chemicals used for treatment of water intended for human consumption. It describes the characteristics of hydrogen peroxide and specifies the requirements and the corresponding test methods for hydrogen peroxide. It gives information on its use in water treatment. It also determines the rules relating to safe handling and use (see Annex B).

## 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

EN ISO 3696, *Water for analytical laboratory use — Specification and test methods (ISO 3696)*

EN ISO 12846, *Water quality — Determination of mercury — Method using atomic absorption spectrometry (AAS) with and without enrichment (ISO 12846)*

ISO 3165, *Sampling of chemical products for industrial use — Safety in sampling*

ISO 6206, *Chemical products for industrial use — Sampling — Vocabulary*

ISO 8288, *Water quality — Determination of cobalt, nickel, copper, zinc, cadmium and lead — Flame atomic absorption spectrometric methods*

ISO 9174, *Water quality — Determination of chromium — Atomic absorption spectrometric methods*

## 3 Description

SIST EN 902:2016

<https://standards.iteh.ai/catalog/standards/sist/fb13ccb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>

### 3.1 Identification

#### 3.1.1 Chemical name

Hydrogen peroxide.

#### 3.1.2 Synonym or common name

None.

#### 3.1.3 Relative molecular mass

34,02.

#### 3.1.4 Empirical formula

H<sub>2</sub>O<sub>2</sub>.

#### 3.1.5 Chemical formula

H-O-O-H.

#### 3.1.6 CAS Registry Number<sup>1)</sup>

7722-84-1.

1) Chemical Abstracts Service Registry Number.

### 3.1.7 EINECS reference<sup>2)</sup>

231-765-0.

### 3.2 Commercial form

The product is supplied as an aqueous solution.

### 3.3 Physical properties

#### 3.3.1 Appearance and odour

The product is colourless liquid, slightly pungent odour.

#### 3.3.2 Density

The density of hydrogen peroxide is given in Table 1.

**Table 1 — Density**

| Solution concentration<br>Mass fraction in% | Density<br>g/ml at 20 °C |
|---|--------------------------|
| 20  | 1,075                    |
| 30  | 1,114                    |
| 35  | 1,132                    |
| 50  | 1,195                    |
| 60  | 1,241                    |
| 70  | 1,289                    |

#### 3.3.3 Solubility in water

<https://standards.iteh.ai/catalog/standards/sist/fb13ccb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>

The product is miscible with water in all proportions.

#### 3.3.4 Vapour pressure

The vapour pressure of hydrogen peroxide depending on concentration is given in Table 2.

**Table 2 — Vapour pressure**

| Solution concentration<br>Mass fraction in % | Vapour pressure<br>kPa at 20 °C |
|--|---------------------------------|
| 20   | 2,0                             |
| 30   | 1,8                             |
| 35   | 1,7                             |
| 50   | 1,3                             |
| 60   | 1,1                             |
| 70   | 0,8                             |

#### 3.3.5 Boiling point at 100 kPa<sup>3)</sup>

The boiling point of hydrogen peroxide depending on concentration is given in Table 3.

2) European Inventory of Existing Commercial Chemical Substances.

3) 100 kPa = 1 bar.

**Table 3 — Boiling point**

| Solution concentration<br>Mass fraction in % | Boiling point<br>°C at 100 kPa |
|--|--------------------------------|
| 20   | 103                            |
| 30   | 106                            |
| 35   | 108                            |
| 50   | 114                            |
| 60   | 119                            |
| 70   | 125                            |

**3.3.6 Crystallization point**

The crystallization point of hydrogen peroxide depending on concentration is given in Table 4.

**Table 4 — Crystallization point**

| Solution concentration<br>Mass fraction in % | Crystallization point<br>°C |
|--|-----------------------------|
| 20   | - 14,6                      |
| 30   | - 25,7                      |
| 35   | - 32,5                      |
| 50   | - 51                        |
| 60   | - 55                        |
| 70   | - 37                        |

**3.3.7 Specific heat**

<https://standards.iteh.ai/catalog/standards/sist/fb13ccb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>

The specific heat of hydrogen peroxide depending on concentration is given in Table 5.

**Table 5 — Specific heat**

| Solution concentration<br>Mass fraction in % | Specific heat<br>kJ/(kg.K) |
|--|----------------------------|
| 20   | 3,82 at 20 °C              |
| 50   | 3,32 at 20 °C              |
| 100  | 2,63 at 25 °C              |

**3.3.8 Viscosity, dynamic**

The viscosity of hydrogen peroxide depending on concentration is given in Table 6.

**Table 6 — Viscosity**

| Solution concentration<br>Mass fraction in % | Viscosity<br>MPa.s at 20 °C |
|--|-----------------------------|
| 20   | 1,04                        |
| 30   | 1,07                        |
| 35   | 1,10                        |
| 50   | 1,17                        |
| 60   | 1,20                        |
| 70   | 1,24                        |
| 100  | 1,25                        |

### 3.3.9 Critical temperature

The critical temperature of pure hydrogen peroxide is 457 °C.

### 3.3.10 Critical pressure

The critical pressure of pure hydrogen peroxide is 21,7 mPa.

### 3.3.11 Physical hardness

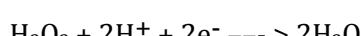
Not applicable.

## iTeh STANDARD PREVIEW (standards.iteh.ai)

Hydrogen peroxide is a weak acid.

According to species in solution, it ~~is an oxidizing agent (E° = 1,776 V)~~ or a reducing agent (E° = 0,682 V). <https://standards.iteh.ai/catalog/standards/sist/fb13cbb8-ba76-4d2c-a874-329fd3b8753b/sist-en-902-2016>

Oxidizing agent:



Reducing agent:



NOTE 1 It can be activated by ultraviolet light, ozone or metals to generate free radicals.

NOTE 2 Singlet oxygen can be obtained by reaction of hydrogen peroxide with hypochlorite ion.

## 4 Purity criteria

### 4.1 General

This European Standard specifies the minimum purity requirements for hydrogen peroxide used for the treatment of water intended for human consumption. Limits are given for impurities commonly present in the product. Depending on the raw material and the manufacturing process other impurities may be present and, if so, this shall be notified to the user and when necessary to relevant authorities.

Users of this product should check the national regulations in order to clarify whether it is of appropriate purity for treatment of water intended for human consumption, taking into account raw water quality, required dosage, contents of other impurities and additives used in the product not stated in this product standard.

Limits have been given for impurities and chemicals parameters where these are likely to be present in significant quantities from the current production process and raw materials. If the production process or raw materials lead to significant quantities of impurities, by-products or additives being present, this shall be notified to the user.

## 4.2 Composition of commercial product

The hydrogen peroxide is usually available in concentrated solution with concentration within the range of mass fraction of 20 % to 70 %. Diluted products are also available.

The concentration of hydrogen peroxide shall be equal to or greater than the manufacturer specified value.

## 4.3 Impurities and main by-products

Not applicable.

## 4.4 Chemical parameters

The product shall conform to the requirements specified in Table 7.

**Table 7 — Chemical parameters**

| Parameter             | Limit in mg/kg of product<br>(mass fraction 100 %) |        |
|-----------------------|--|--------|
|                       | Type 1   | Type 2 |
| Arsenic (As)<br>max.  | 0,5  | 1      |
| Cadmium (Cd)<br>max.  | 0,5  | 1      |
| Chromium (Cr)<br>max. | 0,5  | 1      |
| Mercury (Hg)<br>max.  | 0,5  | 1      |
| Nickel (Ni)<br>max.   | 1  | 5      |
| Lead (Pb)<br>max.     | 0,5  | 1      |
| Antimony (Sb)<br>max. | 0,5  | 1      |
| Selenium (Se)<br>max. | 0,5  | 1      |

NOTE Cyanide which does not exist in a strong oxidizing medium such as hydrogen peroxide is not a relevant chemical parameter. Pesticides and polycyclic aromatic hydrocarbons are not by-products of the manufacturing process. For parametric values of chemical parameters in drinking water, see [2].

## 5 Test methods

### 5.1 Sampling

Observe the general recommendations of ISO 3165 and take ISO 6206 into account.

### 5.2 Analysis

#### 5.2.1 Determination of hydrogen peroxide content (main product)

##### 5.2.1.1 Principle

Titration of a test portion of hydrogen peroxide in acidic medium with a potassium permanganate standard volumetric solution.