
**Plastics — Polyols for use in the
production of polyurethanes —
Determination of basicity (total amine
value), expressed as percent nitrogen**

*Plastiques — Polyalcools utilisés pour la production de
polyuréthannes — Détermination de la basicité (valeur totale d'amines)
en pourcentage d'azote*

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ISO copyright office
Case postale 56 • CH-1211 Geneva 20
Tel. + 41 22 749 01 11
Fax + 41 22 749 09 47
E-mail copyright@iso.org
Web www.iso.org

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 25761 was prepared by Technical Committee ISO/TC 61, *Plastics*, Subcommittee SC 12, *Thermosetting materials*.

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Introduction

Polyurethanes are produced by the catalysed reaction of isocyanates with polyols. The basicity of the polyol employed affects the rate of reaction and speed of cure of the product. It is therefore necessary to determine the basicity in order to predict reactivity and monitor product quality.

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Plastics — Polyols for use in the production of polyurethanes — Determination of basicity (total amine value), expressed as percent nitrogen

SAFETY STATEMENT — Persons using this document should be familiar with normal laboratory practice, if applicable. This document does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user to establish appropriate safety and health practices and to ensure compliance with any regulatory requirements.

1 Scope

1.1 The method specified in this International Standard measures the basic constituents in polyols that are soluble in glacial acetic acid and reactive with perchloric acid. Samples containing 0,3 % to 10 % of nitrogen have been evaluated by this method. The method is applicable to polyether polyols and polyether polyol blends that are used in polyurethane reactions. The results are measures of batch-to-batch uniformity and may be used to estimate reactivity in polyurethane reactions.

1.2 It is also permissible to express the results in equivalents of base per gram of sample.

1.3 This method is technically equivalent to that in ASTM D 6979-03.

2 Normative references

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The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 3696, *Water for analytical laboratory use — Specification and test methods*

ISO 6353-1, *Reagents for chemical analysis — Part 1: General test methods*

ISO 6353-2, *Reagents for chemical analysis — Part 2: Specifications — First series*

ISO 6353-3, *Reagents for chemical analysis — Part 3: Specifications — Second series*

3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

3.1

polyol

polymer based on ethylene oxide and/or propylene oxide which contains two or more hydroxyl groups

3.2

polyurethane

polymer prepared by the reaction of an organic di- or polyisocyanate with a compound containing two or more hydroxyl groups

- 3.3 percent nitrogen**
quantity of perchloric-acid-titratable base in a sample, expressed as a mass percentage of nitrogen
- 3.4 alkalinity**
quantity of perchloric-acid-titratable base in a sample, expressed as mg of KOH/g of sample.
- 3.5 total amine value**
quantity of perchloric-acid-titratable base in a sample, identified only as amines and expressed as mg of KOH/g of sample

4 Principle

A test portion of the sample is dissolved in glacial acetic acid. The resulting single-phase solution is titrated at room temperature to a potentiometric end point with a standardized solution of perchloric acid in acetic acid. The result is reported as percent nitrogen or mg of KOH/g of sample.

5 Sampling

Draw samples from a well-mixed vessel into a thoroughly cleaned and dry borosilicate-glass container (soft-glass containers are not acceptable). If sampling from a line or valve, flush the line thoroughly with the product before starting to draw the sample. Seal the sample until analysis.

6 Interference

Any acidic or basic materials inadvertently introduced into the sample will cause errors in the analysis. Any material capable of serving as a buffer may interfere with the analysis by obscuring the titration end point.

7 Reagents

Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of ISO 6353-1, ISO 6353-2 and ISO 6353-3, as applicable. Other grades may be used, however, provided that it is first determined that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

Unless otherwise indicated, references to water shall be understood to mean grade 2 water as defined in ISO 3696.

- 7.1 Glacial acetic acid.**
- 7.2 Acetic anhydride.**
- 7.3 Perchloric acid,** nominal concentration 70 %.
- 7.4 Perchloric acid in acetic acid,** concentration 0,10 mol/l.

One way of preparing the 0,10 mol/l perchloric acid in acetic acid is as follows. In a 1 000 ml volumetric flask, mix 8,7 ml of perchloric acid (7.3) with 500 ml of glacial acetic acid (7.1). Add 25 ml of acetic anhydride (7.2) and dilute to volume with glacial acetic acid.

WARNING — Perchloric acid is extremely irritating to the skin, eyes and mucous membranes, highly toxic via oral and inhalation routes, and can form explosive mixtures when mixed with carbonaceous material or allowed to dry. Concentrated perchloric acid should only be used in a hood approved for perchloric acid use. Chemically resistant gloves should be worn. In the event of skin contact, wash with soap and water. Goggles or safety glasses with side shields should be worn. In the event of eye contact, flush with copious amounts of water for 15 minutes. In the event of inhalation, move the victim to an uncontaminated area. In the event of ingestion, do not induce vomiting. For all exposures, seek professional medical advice.

It is strongly recommended that reference articles on perchloric acid safety be consulted to determine safe-handling and clean-up procedures (see References [1] to [6] and references cited therein).

8 Procedure

8.1 Weigh a test portion of the sample into a suitable container, calculating the mass of the test portion as follows:

$$M' = 2/P$$

where

M' is the mass of the test portion, in g;

P is the expected percent nitrogen content of the sample.

For test portion masses below 10,0 g, record the mass to the nearest 0,1 mg; for test portion masses greater than 10,0 g, record the mass to the nearest 0,01 g.

8.2 Add 100 ml of glacial acetic acid and stir gently until the test portion has dissolved completely.

If necessary, the mixture can be heated gently until the test portion has completely dissolved.

8.3 Titrate the test solution potentiometrically with 0,10 mol/l perchloric acid (7.4) through the end point which occurs at about 600 mV.

Colorimetric end point determination can also be used. A suitable indicator is patent blue VF. Use ten drops of indicator solution (0,3 % by mass of indicator in glacial acetic acid).

8.4 Under normal circumstances, the blank value is negligibly small and is not included in the calculation. However, a solvent blank should preferably be evaluated at suitable intervals to confirm that the solvent blank is indeed a negligible value.

NOTE Some laboratories report using this general procedure for lower levels of basicity, employing 0,01 mol/l perchloric acid as the titrant. Better results are reported at low levels with a colorimetric end point determination. No blank is needed since the solvent is neutralized to a green end point before the test solution is added. Precision data reported in this International Standard are based on the range 0,3 % to 10 % nitrogen. Precision data at lower ranges of nitrogen are not available at the current time. Therefore, precision studies will be necessary before the method is applied to ranges below 0,3 % nitrogen.

9 Expression of results

9.1 Calculate the basicity of the sample, as percent nitrogen (% N) as follows:

$$\% \text{ N} = \frac{V \times F \times 1,4}{m \times 1000} \times 100$$