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# Hardmetals — Determination of total carbon content — Gravimetric method

*Métaux-durs — Dosage du carbone total — Méthode gravimétrique*

[Revision of second edition (ISO 3907:1985)]

ICS 77.160

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## Foreword

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ISO 3907 was prepared by Technical Committee ISO/TC 119, *Powder Metallurgy*, Subcommittee SC 4, *Sampling and testing methods for hardmetals*.

This third edition cancels and replaces the second edition (ISO 3907:1985) which has been technically revised.

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# Hardmetals — Determination of total carbon content — Gravimetric method

## 1 Scope

This International Standard specifies a gravimetric method for determination of the total carbon content of carbides and hardmetals.

## 2 Field of application

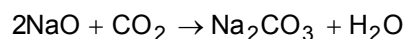
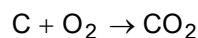
This method is applicable to

- carbides of chromium, hafnium, molybdenum, niobium, tantalum, titanium, vanadium, tungsten and zirconium,
- mixtures of these carbides and binder metals, free of lubricant,
- all grade of presintered or sintered hardmetals, produced from these carbides, having a total carbon content exceeding 4 % (m/m).

## 3 Principle

Oxidation of carbon to carbon dioxide at high temperature in a stream of pure oxygen, with the addition of a flux if necessary.

Absorption of the carbon dioxide, carried by oxygen, by ascarite in a tared bulb. Determination of the increase in mass of the ascarite, which corresponds to the quantity of carbon dioxide formed.



## 4 Reagents

During the analysis, use only reagents of recognized analytical grade, and only distilled water or water of equivalent purity.

**4.1 Oxygen**, with a limitation of carbon-containing impurities  $\leq 0,6$  ml of carbon per cubic metre of oxygen.

**4.2 Magnesium perchlorate**, anhydrous.

**CAUTION — To prevent any possibility of explosion, contact of this reagent with organic materials should be avoided, especially when discarding it.**

4.3 Flux, for example tin metal, copper metal or oxide, iron metal.

4.4 Ascarite.

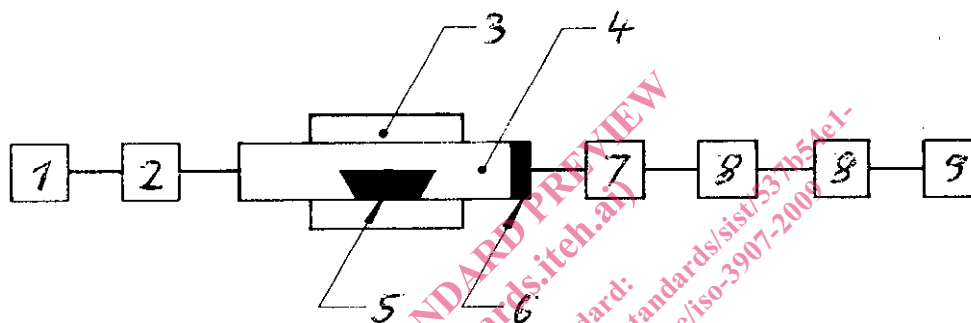
## 5 Apparatus

Ordinary laboratory apparatus and

5.1 **Apparatus**, consisting of an electric furnace with a combustion tube, a purification train and a system to absorb carbon dioxide. If necessary to obtain oxygen of adequate purity, an oxygen purification train may also be used.

Successive parts of the apparatus shall be joined together with connecting tubes forming an airtight seal.

The apparatus is shown schematically in Figure 1.



### Key

- 1 — **Source of oxygen** (see 4.1), with pressure-regulating valve.
- 2 — **Flow meter**.
- 3 — **Electric furnace**, capable of operation at up to 1 350 °C, with a suitable device for temperature control.
- 4 — **Combustion tube**, made of a non-porous refractory material. The internal diameter of the tube should be 18 to 30 mm and its length at least 650 mm, so that the ends of the tube do not reach a temperature higher than 60 °C during the operation.
- 5 — **Boat**, made of a refractory material, pretreated in an oxygen stream at the test temperature for 10 min or alternatively, at 800 to 1 000 °C for 1 h.

The boat shall be of suitable dimensions, for example length 80 to 100 mm, width 12 to 14 mm and depth 8 to 9 mm.

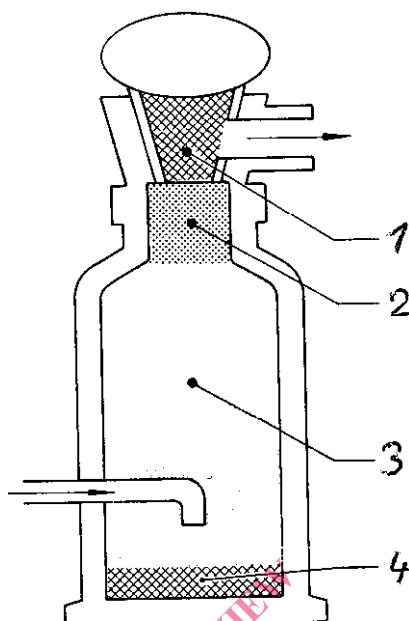
The pretreated boats shall be kept in a desiccator. The ground surfaces of the desiccator and its lid shall not be greased.

- 6 — **Plug of silica wool**.
- 7 — **Drying bulb**, containing anhydrous magnesium perchlorate (see 4.2).
- 8 — **Absorption bulbs**, containing ascarite (see 4.4) and a small amount of anhydrous magnesium perchlorate (see 4.2).

An example of an absorption bulb is shown in Figure 2.

- 9 — **Absorption bulb**, facing the opposite way to H to avoid introduction of carbon dioxide and moisture from the air.

Figure 1 — Apparatus

**Key**

- 1 Glass wool in stopper
- 2 Anhydrous magnesium perchlorate (see 4.2)
- 3 Ascarite (see 4.4), not too compressed
- 4 Bottom layer of glass wool to protect inside of tube

Figure 2 — Absorption bulb

**5.2 Hook**, made from heat-resisting metal wire with a carbon content less than 0,05 % (m/m). Its diameter should be approximately 3 mm and its length 500 to 600 mm.

**6 Sampling**

**6.1** The sample shall be crushed to a powder in a mortar made of a materials which does not alter the sample composition. The powder shall pass a 180  $\mu\text{m}$  sieve.

**6.2** The analysis shall be carried out on two or three test portions.

**7 Procedure****7.1 General**

Check the temperature in the combustion zone (1 200 to 1 350  $^{\circ}\text{C}$  and not less than 1 300  $^{\circ}\text{C}$  if chromium carbide is present), the gastightness of the apparatus and the efficiency of the oxygen purification. Pass oxygen through the apparatus for 10 min to 15 min at a rate of 300 to 500  $\text{cm}^3/\text{min}$  depending on the diameter of the tube used. Then disconnect the absorption bulbs (H), weigh them at ambient temperature and replace them in position.

**7.2 Test portion**

The mass of the test portion ( $m_0$ ) shall be such that it contains approximately 0,03 g of carbon, and shall be determined to the nearest 0,000 1 g.

If necessary, add to the test portion 0,2 g to 1 g of the flux (see 4.3).

**7.3 Blank test**

Carry out the blank test by combustion (proceed as outlined in 7.4 and 7.5) in the presence of the quantity of flux used in the analysis, and carefully determine the increase in mass ( $m_1$ ) of the absorption bulbs.

**7.4 Combustion**

Open the combustion tube at the oxygen inlet end and, using the hook (see 5.2), place the boat (E) containing the test portion (see 7.2) in the centre of the heated zone of the tube. Quickly close the tube and immediately pass a stream of oxygen at a rate of 300 cm<sup>3</sup>/min to 500 cm<sup>3</sup>/min, depending on the diameter of the tube used. Continue to pass oxygen for 10 min to 20 min so that the carbon dioxide is completely removed from the combustion tube and the purifying bulb.

**7.5 Determination**

Close the tap of the absorption bulbs (HH) and immediately remove the bulbs from the apparatus. After 5 min weigh the bulbs to the nearest 0,000 1 g. It is recommended that the fused mass in the boat be visually examined to verify that combustion has been completed. The increase in mass represents the carbon dioxide absorbed ( $m_2$ ).

**8 Expression of results**

**8.1 Calculation**

The carbon content, expressed as a percentage by mass, is given by the formula

$$27,29 \frac{m_2 - m_1}{m_0}$$

where

- $m_0$  is the mass, in grams, of the test portion;
- $m_1$  is the mass, in grams, of carbon dioxide obtained from the blank test;
- $m_2$  is the mass, in grams, of the carbon dioxide obtained from the combustion of the test portion;
- 27,29 is the carbon dioxide to carbon conversion factor, multiplied by 100.

**8.2 Tolerances**

The deviation between two or three independent determinations shall not exceed the values shown in the table.

Total carbon content % (m/m)	Range for two determinations % (m/m)	Range for three determinations % (m/m)
from 4 to 10	0,05	0,06
over 10	0,07	0,08

**8.3 Final result**

Report the arithmetical mean of acceptable determinations rounded to the nearest 0,01 % (m/m).