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Copper and copper alloys - Determination of aluminium content - Part 2: FAAS method

Kupfer und Kupferlegierungen - Bestimmung des Aluminiumgehaltes - Teil 2:  
Flammenatomabsorptionsspektrometrisches Verfahren (FAAS)

Cuivre et alliages de cuivre - Dosage de l'aluminium - Partie 2 : Méthode par  
spectrométrie d'absorption atomique dans la flamme (SAAF)

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**Ta slovenski standard je istoveten z: EN 14936-2:2006**

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ICS 77.120.30

English Version

## Copper and copper alloys - Determination of aluminium content - Part 2: FAAS method

Cuivre et alliages de cuivre - Dosage de l'aluminium -  
Partie 2 : Méthode par spectrométrie d'absorption atomique  
dans la flamme (SAAF)

Kupfer und Kupferlegierungen - Bestimmung des  
Aluminiumgehaltes - Teil 2:  
Flammenatomabsorptionsspektrometrisches Verfahren  
(FAAS)

This European Standard was approved by CEN on 15 May 2006.

CEN members are bound to comply with the CEN/CENELEC Internal Regulations which stipulate the conditions for giving this European Standard the status of a national standard without any alteration. Up-to-date lists and bibliographical references concerning such national standards may be obtained on application to the Central Secretariat or to any CEN member.

This European Standard exists in three official versions (English, French, German). A version in any other language made by translation under the responsibility of a CEN member into its own language and notified to the Central Secretariat has the same status as the official versions.

CEN members are the national standards bodies of Austria, Belgium, Cyprus, Czech Republic, Denmark, Estonia, Finland, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugal, Romania, Slovakia, Slovenia, Spain, Sweden, Switzerland and United Kingdom.

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EUROPEAN COMMITTEE FOR STANDARDIZATION  
COMITÉ EUROPÉEN DE NORMALISATION  
EUROPÄISCHES KOMITEE FÜR NORMUNG

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## Foreword

This document (EN 14936-2:2006) has been prepared by Technical Committee CEN/TC 133 "Copper and copper alloys", the secretariat of which is held by DIN.

This European Standard shall be given the status of a national standard, either by publication of an identical text or by endorsement, at the latest by December 2006, and conflicting national standards shall be withdrawn at the latest by December 2006.

Within its programme of work, Technical Committee CEN/TC 133 requested CEN/TC 133/WG 10 "Methods of analysis" to prepare the following standard:

EN 14936-2, *Copper and copper alloys — Determination of aluminium content — Part 2: FAAS method*

This is one of two parts of the standard for the determination of aluminium content in copper and copper alloys. The other part is:

EN 14936-1, *Copper and copper alloys — Determination of aluminium content — Part 1: Titrimetric method*

According to the CEN/CENELEC Internal Regulations, the national standards organizations of the following countries are bound to implement this European Standard: Austria, Belgium, Cyprus, Czech Republic, Denmark, Estonia, Finland, France, Germany, Greece, Hungary, Iceland, Ireland, Italy, Latvia, Lithuania, Luxembourg, Malta, Netherlands, Norway, Poland, Portugal, Romania, Slovakia, Slovenia, Spain, Sweden, Switzerland and United Kingdom.

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## 1 Scope

This part of this European Standard specifies a flame atomic absorption spectrometric method (FAAS) for the determination of the aluminium content of copper and copper alloys in the form of unwrought, wrought and cast products.

The method is applicable to products having aluminium mass fractions between 0,010 % and 2,5 %.

## 2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 1811-1, *Copper and copper alloys — Selection and preparation of samples for chemical analysis — Part 1: Sampling of cast unwrought products*

ISO 1811-2, *Copper and copper alloys — Selection and preparation of samples for chemical analysis — Part 2: Sampling of wrought products and castings*

NOTE Informative references to documents used in the preparation of this standard, and cited at the appropriate places in the text, are listed in the Bibliography.

## 3 Principle

Dissolution of a test portion in aqua regia followed, after suitable dilution, by aspiration into a nitrous oxide/acetylene flame of an atomic absorption spectrometer. Measurement of the absorption of the 309,3 nm line emitted by an aluminium hollow-cathode lamp.

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## 4 Reagents and materials

### 4.1 General

During the analysis, use only reagents of recognized analytical grade and only distilled water or water of equivalent purity.

### 4.2 Hydrochloric acid, HCl ( $\rho = 1,19$ g/ml).

### 4.3 Hydrochloric acid solution, 1 + 1

Dilute 500 ml of hydrochloric acid (4.2) in 500 ml of water.

### 4.4 Nitric acid, HNO<sub>3</sub> ( $\rho = 1,40$ g/ml).

### 4.5 Nitric acid solution, 1 + 1

Dilute 500 ml of nitric acid (4.4) in 500 ml of water.

### 4.6 Aluminium stock solution, 1,000 g/l Al

Weigh ( $1 \pm 0,001$ ) g of aluminium (Al  $\geq 99,9$  %) and dissolve it in 50 ml of the hydrochloric acid solution (4.3). Transfer the solution into a 1 000 ml one-mark volumetric flask. Dilute to the mark with water and mix well.

1 ml of this solution contains 1,000 mg of Al.

#### 4.7 Aluminium standard solution, 0,050 g/l Al

Transfer 10,0 ml of the aluminium stock solution (4.6) into a 200 ml one-mark volumetric flask. Dilute to the mark with water and mix well.

Prepare this solution immediately prior to use.

1 ml of this solution contains 0,050 mg of Al.

#### 4.8 Aluminium standard solution, 0,200 g/l Al

Transfer 20,0 ml of the aluminium stock solution (4.6) into a 100 ml one-mark volumetric flask. Dilute to the mark with water and mix well.

Prepare this solution immediately prior to use.

1 ml of this solution contains 0,200 mg of Al.

#### 4.9 Copper base solution, 50 g/l Cu

Weigh 25,0 g of aluminium-free copper ( $\text{Cu} \geq 99,95\%$ ) and transfer it into an 800 ml beaker. Add 125 ml of hydrochloric acid (4.2) and, cautiously, 250 ml of nitric acid solution (4.5). Cover with a watch glass and heat gently until the copper is completely dissolved, then continue heating to the boiling point. Cool to room temperature. Wash the beaker cover and the sides of the beaker with water. Transfer the solution quantitatively into a 500 ml one-mark volumetric flask, dilute to the mark with water and mix well.

#### 4.10 Copper base solution, 10 g/l Cu

Transfer 40,0 ml of the copper base solution (4.9) into a 200 ml one-mark volumetric flask. Add 45 ml of hydrochloric acid (4.2) and 90 ml of nitric acid solution (4.5). Dilute to the mark with water and mix well.

## 5 Apparatus

### 5.1 Ordinary laboratory apparatus.

### 5.2 Atomic absorption spectrometer, fitted with a nitrous oxide/acetylene burner.

### 5.3 Aluminium hollow-cathode lamp.

## 6 Sampling

Sampling shall be carried out in accordance with ISO 1811-1 or ISO 1811-2, as appropriate.

Test samples shall be in the form of fine drillings, chips or millings with a maximum thickness of 0,5 mm.

## 7 Procedure

### 7.1 Preparation of the test portion solution

#### 7.1.1 Test portion

Weigh ( $1 \pm 0,001$ ) g of the test sample.

### 7.1.2 Aluminium mass fractions between 0,01 % and 0,10 %

Transfer the test portion (7.1.1) into a 250 ml beaker. Add 5 ml of the hydrochloric acid (4.2) and 10 ml of the nitric acid solution (4.5). Cover and heat gently until the test portion is completely dissolved, then continue heating to the boiling point. Cool to room temperature. Wash the beaker cover and the sides of the beaker with water.

Transfer the dissolved test portion into a 100 ml one-mark volumetric flask. Dilute to the mark with water and mix well.

### 7.1.3 Aluminium mass fractions between 0,10 % and 2,5 %

Transfer the test portion (7.1.1) into a 250 ml beaker. Add 5 ml of the hydrochloric acid (4.2) and 10 ml of the nitric acid solution (4.5). Cover and heat gently until the test portion is completely dissolved, then continue heating to the boiling point. Cool to room temperature. Wash the beaker cover and the sides of the beaker with water.

Transfer the dissolved test portion into a 500 ml one-mark volumetric flask. Dilute to the mark with water and mix well.

## 7.2 Blank test

Carry out a blank test simultaneously with the determination, following the same procedure and using the same quantities of all reagents as used for the determination, but omitting the test portion.

## 7.3 Check test

Make a preliminary check of the apparatus by preparing a solution of a standard material or a synthetic sample containing a known amount of aluminium and of composition similar to the material to be analysed. Carry out the procedure specified in 7.5.

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## 7.4 Establishment of the calibration curve

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### 7.4.1 Preparation of the calibration solutions

#### 7.4.1.1 General

In all cases, copper, chloride and nitrate concentrations and acidity in the calibration solutions shall be similar to those of the test portion solutions.

The presence of copper in the calibration solutions compensates for chemical interaction effects of copper in the test portion solution. Normally no similar additions are required to compensate for the effect of alloying elements. If an alloying element is present in the material to be analysed in mass fraction > 10 %, an appropriate mass of this element shall be added to the calibration solutions.

The aluminium concentration of the calibration solutions shall be adjusted to suit the sensitivity of the spectrometer used, so that the curve of absorbance as a function of concentration is a straight line.

#### 7.4.1.2 Aluminium mass fractions between 0,01 % and 0,10 %

Into each of a series of six 100 ml one-mark volumetric flasks, introduce the volumes of the aluminium standard solution (4.7) and of the copper base solution (4.9) shown in Table 1. Dilute to the mark with water and mix well.



Table 1 — Calibration for aluminium mass fractions between 0,01 % and 0,10 %

Aluminium standard solution volume (4.7)	Corresponding aluminium mass	Corresponding aluminium concentration after final dilution	Copper base solution volume (4.9)	Corresponding copper mass	Corresponding aluminium mass fraction of sample
ml	mg	mg/ml	ml	g	%
0 <sup>a</sup>	0	0	20	1,000	0
2	0,10	0,001 0	20	1,000	0,010
5	0,25	0,002 5	20	1,000	0,025
10	0,50	0,005 0	20	1,000	0,050
15	0,75	0,007 5	20	1,000	0,075
20	1,00	0,010	20	1,000	0,100

<sup>a</sup> Blank test on reagents for calibration curve.

## 7.4.1.3 Aluminium mass fractions between 0,10 % and 2,5 %

Into each of a series of fifteen 100 ml one-mark volumetric flasks, introduce the volumes of the aluminium standard solution (4.8) and of the copper base solution (4.10) shown in Table 2. Dilute to the mark with water and mix well.

Table 2 — Calibration for aluminium mass fractions between 0,10 % and 2,5 %

Aluminium standard solution volume (4.8)	Corresponding aluminium mass	Corresponding aluminium concentration after final dilution	Copper base solution volume (4.10)	Corresponding copper mass	Corresponding aluminium mass fraction of sample
ml	mg	mg/ml	ml	g	%
0 <sup>a</sup>	0	0	20	0,200	0
1	0,20	0,002 0	20	0,200	0,10
2	0,40	0,004 0	20	0,200	0,20
4	0,80	0,008 0	20	0,200	0,40
6	1,2	0,012	20	0,200	0,60
8	1,6	0,016	20	0,200	0,80
10	2,0	0,020	20	0,200	1,00
12	2,4	0,024	20	0,200	1,20
14	2,8	0,028	20	0,200	1,40
16	3,2	0,032	20	0,200	1,60
18	3,6	0,036	20	0,200	1,80
20	4,0	0,040	20	0,200	2,00
22	4,4	0,044	20	0,200	2,20
24	4,8	0,048	20	0,200	2,40
25	5,0	0,050	20	0,200	2,50

<sup>a</sup> Blank test on reagents for calibration curve.