

# SLOVENSKI STANDARD SIST ISO 602:1998

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Coal -- Determination of mineral matter

Charbon -- Détermination du taux de matières minérales EVIEW

## (standards.iteh.ai) Ta slovenski standard je istoveten z: ISO 602:1983

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73.040	Premogi	Coals	
SIST ISO 6	602:1998	en	

2003-01. Slovenski inštitut za standardizacijo. Razmnoževanje celote ali delov tega standarda ni dovoljeno.

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INTERNATIONAL ORGANIZATION FOR STANDARDIZATION MEX CHAPODHAR OPPAHUSALUR TO CTAHDAPTUSALUNO ORGANISATION INTERNATIONALE DE NORMALISATION

# **Coal** — **Determination of mineral matter**

Charbon - Détermination du taux de matières minérales

Second edition - 1983-02-01

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Descriptors : coal, chemical analysis, determination of content, mineral matter.

### SIST ISO 602:1998

## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of developing International Standards is carried out through ISO technical committees. Every member body interested in a subject for which a technical committee has been authorized has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work.

Draft International Standards adopted by the technical committees are circulated to the member bodies for approval before their acceptance as International Standards by the ISO Council.

International Standard ISO 602 was developed by Technical Committee ISO/TC 27, EVIEW Solid mineral fuels.

This second edition was submitted directly to the ISO Council, in accordance with clause 6.11.2 of part 1 of the Directives for the technical work of ISO. It cancels and replaces the first edition (i.e. ISO 602-1974), which had been approved by the member bodies of the following countries: https://standards.iteh.ai/catalog/standards/sist/aa5ce35c-f7b0-42e9-a194-

Australia	Egy
Austria	Fran
Belgium	Ger
Brazil	Indi
Canada	Italy
Chile	Kor
Colombia	Net
Czechoslovakia	Nev
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No member body had expressed disapproval of the document.

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# **Coal** — **Determination of mineral matter**

1	Scope and field of application	4.1	<b>Hydrochloric acid,</b> <i>ϱ</i> 1,18 g/ml.			
This International Standard specifies a method of determining the amount of mineral matter in all types of coal, including brown coals and lignites.		4.2	<b>Hydrochloric acid,</b> solution, $c(HCI) = 5 \text{ mol/I}$ .			
	C C	4.3	Hydrofluoric acid, $\varrho$ 1,13 g/ml.			
_	References iTeh STANDARD 157, Hard coal – Determination of forms of sulphur.		NING – Very toxic by inhalation, in contact with and if swallowed. Causes severe burns.			
ISO	331, Coal – Determination of moisture in the analysis   pple – Direct gravimetric method.	Keep container tightly closed in a well-ventilated place In case of contact with eyes, rinse immediately with oplenty of water and seek medical advice.				
ISO 348, Hard coal — Determination of moisture in the analysis ds/sis sample — Direct volumetric method. 4864983a7dfc/sist-iso-		Wear of a	5ce35c-f7b0-42e9-a194- (ear suitable protective clothing and gloves. In case f accident or feeling unwell, seek medical advice nmediately (show the label where possible).			
	) 352, Solid mineral fuels — Determination of chlorine — h temperature combustion method.	Imme				
	) 587, Solid mineral fuels — Determination of chlorine using hka mixture.	54	Apparatus			
ISO 1170, Coal and coke — Calculation of analyses to different bases.		All the apparatus listed below shall be resistant to acids, especially hydrofluoric acid. A suitable material is polyvinyl chloride (PVC).				
ISC	) 1171, Solid mineral fuels — Determination of ash.	5.1	Beaker, of capacity 200 ml, with a cover slip.			
3	Principle	5.2	Thermometer pocket: a tube, sealed at one end, to			
Partial demineralization of a sample of the coal by treatment		carry a thermometer.				
with hydrochloric and hydrofluoric acids under such conditions that the coal substance remains unaffected. Recording of the loss in mass of the coal due to the acid treatment and de-			Stirrer.			
termination of the undissolved part of the mineral matter by ashing the partially demineralized coal. Determination of the iron content of the ash so that the pyrites present in the extracted coal can be calculated. Determination of the amount of hydrochloric acid absorbed by the coal substance.		5.4	Wash-bottle.			
		<b>5.5</b> Filter, with a sintered alumina filter plate, for example a shown in the figure.				
4	Reagents	5.6	Filter flask.			
	ring the analysis use only reagents of recognized analytical Ide and only distilled water or water of equivalent purity.	5.7	Balance, accurate to 0,1 mg.			

#### Preparation of sample 6

The coal used for determination of mineral matter is the analysis sample, ground to pass a sieve of 200 µm aperture. If necessary, expose the sample in a thin layer for the minimum time required for the moisture content to reach approximate equilibrium with the laboratory atmosphere.

Before starting the determination, mix the air-dried analysis sample of coal (see the note) thoroughly for at least 1 min, preferably by mechanical means.

NOTE - Alternatively, the coal sample may be dried at 105 to 110 °C before carrying out the procedure.

#### Procedure 7

Weigh, to the nearest 0,1 mg, about 6 g of the sample into the beaker (5.1) and add 40 ml of the hydrochloric acid solution (4.2) (see note 1). Insert the stirrer (5.3) and the tube (5.2) carrying a thermometer and place the cover slip over the beaker. Place the beaker in a water-bath maintained at 55 to 60 °C. Stir the contents at 5 min intervals, remove the beaker after 45 min and allow the coal suspension to settle for 10 min. Filter the solution through the filter (5.5) under suction. (standa

 $m_5$  is the mass, in grams, of ash, less iron oxide from the Wash the coal retained on the filter with water, drain and pyrites in the extracted coal. transfer the coal back to the beaker with the aid of not more. than 5 ml of water. Care is required to avoid loss of coal by

Add 40 ml of the hydrofluoric acid (4.3) to the beaker and repeat the heat treatment and filtration as previously described. Rinse the coal retained on the filter into the beaker with not more than 5 ml of water. Add 50 ml of hydrochloric acid (4.1) to the beaker, replace it in the water-bath and repeat the heat treatment previously described. Filter and wash the coal three times, decanting each time. Transfer the coal entirely to the filter and wash 20 times with 25 ml portions of hot water each time. Recover any residual coal from the beaker by means of a rubber tipped rod and cold water. Drain the coal under suction for 5 to 10 min.

Dismantle the filter, break up the wet coal and dry the filter and coal in a vacuum oven at 50 °C and a pressure of 3,5 kPa<sup>1)</sup> absolute for about 1,5 h. Remove and allow to cool in air for about 1 h to attain equilibrium and then weigh. Recover the coal and transfer as much as possible to a glass stoppered bottle. Brush the filter top and filter paper free from coal and reweigh. Obtain the mass of extracted coal by difference.

Mix the extracted coal thoroughly and determine its moisture, ash, and chlorine, as well as the total iron content of the ash; determine also the moisture and ash content of the original sample, each determination being carried out according to the appropriate International Standards (see clause 2). Calculate the hydrochloric acid equivalent to the chlorine content and the pyrites equivalent to the total iron content.

1) 1 kPa = 10 mbar

### NOTES

1 For reactive coals, including brown coals and lignites, the acids may be placed in the beaker before adding the sample to avoid local overheating.

2 The first hydrochloric acid extraction is unnecessary for coals having a carbon dioxide content of less than 0.5 % (m/m).

#### **Expression of results** 8

All results used in the calculation shall be on the dry basis. The mineral matter, MM, expressed as a percentage by mass, is given by the formula<sup>2)</sup>

$$\frac{m_1 - m_2 + m_3 + m_4 + 1,1 m_5}{m_4} \times 100$$

where

is the mass, in grams, of the test portion taken;  $m_1$ 

 $m_2$  is the mass, in grams, of the test portion after extraction;

 $m_3$  is the mass, in grams, of pyrites in the extracted coal;

m<sub>2</sub> is the mass, in grams, of hydrochloric acid in the extracted coal; liteh.ai

catalog/standarde/minerarmatter/factor/Fally is given by the formula splashing (see note 2). 4864983a7dfe/sist-iso-602-1998

> MM A

where A is the percentage of ash in the original coal.

An example of the calculation is given in the annex.

The result (preferably the mean of two determinations - see clause 9) shall be reported to the nearest 0,1 %.

### NOTES

1 If it is required to derive the mineral matter from the percentage of ash by the application of a formula, see ISO 1170.

The factor 1,1 allows approximately for the water of hydration of the aluminium and silicon compounds in the demineralized coal. In most cases the correction is small and can be ignored.

#### 9 Precision of the method

	Maximum acceptable difference between results obtained			
Mineral	In the same laboratory	In different laboratories		
matter	(Repeatability)	(Reproducibility)		
	0,4 % absolute	(see 9.2)		

<sup>2)</sup> See the notes.

### 9.1 Repeatability

The result of duplicate determinations, carried out at different times in the same laboratory by the same operator with the same apparatus on the same analysis sample, should not differ by more than the above value.

### 9.2 Reproducibility

No value for reproducibility can be quoted for determinations carried out in different laboratories, since insufficient data are available.

### 10 Test report

The test report shall include the following information:

- a) an identification of the product tested;
- b) the reference of the method used;
- c) the results and the method of expression used;
- d) any unusual features noted during the determination;

e) any operation not included in this International Standard, or regarded as optional.



Figure - Filtering device (Left half shown in section)

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## Annex

# Example of calculation

The following example illustrates the method of calculation. All results are quoted on the dry basis.

			% ( <i>m/m</i> )	g			
1	Mass of original coal	=		6,836	( <i>m</i> <sub>1</sub> )		
2	Ash content of original coal	=	52,8		(A)		
3	Mass of extracted coal	=		3,205	( <i>m</i> <sub>2</sub> )		
4	Ash content of extracted coal	=	6,03		(A <sub>1</sub> )		
5	Amount of hydrochloric acid absorbed by the extracted coal	-	1,06		(P <sub>1</sub> )		
6	Amount of iron oxide in the ash of the extracted coal	=	4,2		(P <sub>2</sub> )		
7	Loss in mass of the sample $(m_1 - m_2)$	=		3,631			
8	Mass of hydrochloric acid absorbed ( $P_1m_2/100$ )	=		0,034	( <i>m</i> <sub>4</sub> )		
9	Mass of the pyrites in the extracted coal (1,5 $P_2m_2/100$ )	=		0,202	( <i>m</i> <sub>3</sub> )		
10	Mass of the residual ash [1,1 $m_2 (A_1 - P_2)/100$ ]	=		0,065	(1,1 <i>m</i> <sub>5</sub> )		
11	Total mass of mineral matter $(m_1 - m_2 + m_3 + m_4 + 0.1 m_5 RD PR)$		/ IE W	3,932	( <i>m</i> <sub>6</sub> )		
12							
13	Mineral matter factor $(MM/A)$		1,09		(F <sub>MM</sub> )		
	910-11-00 (00-1009)						

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