



Designation: E252 – 06

Standard Test Method for Thickness of Foil, Thin Sheet, and Film by Mass Measurement¹

This standard is issued under the fixed designation E252; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

This standard has been approved for use by agencies of the Department of Defense.

1. Scope*

1.1 This test method covers the determination of the thickness of metallic foil and sheet 0.015 in. (0.38 mm) and less in thickness by measuring the mass of a specimen of known area and density. The test method is applicable to other sheet, foil, and film as indicated in [Annex A3](#).

1.2 The values stated in inch-pound units are to be regarded as the standard. The values given in parentheses are mathematical conversions to SI units, which are provided for information only and are not considered standard.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 The following documents of the issue in effect on the date of material purchase, unless otherwise noted, form a part of this specification to the extent referenced herein:

2.2 *ASTM Standards*:²

[D1505 Test Method for Density of Plastics by the Density-Gradient Technique](#)

[E29 Practice for Using Significant Digits in Test Data to Determine Conformance with Specifications](#)

3. Apparatus

3.1 *Precision Blanking Press*—to cut foil or sheet circles that are 8.000 ± 0.008 in.² (51.613 ± 0.051 cm²) in area or 3.1915 ± 0.0015 in. (81.06 ± 0.04 mm) in diameter. Other size

¹ This test method is under the jurisdiction of ASTM Committee B07 on Light Metals and Alloys and is the direct responsibility of Subcommittee B07.05 on Testing.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

specimens may be used with the recognition that the accuracy stated in [6.1](#) is no longer applicable. See [Annex A1](#) for the selection of other specimen sizes and the resulting change in accuracy of the test method.

3.2 *Balance*—capable of measuring to the nearest 0.1 mg of thickness for the 8.000-in.² (51.613-cm²) circle.

4. Procedure

4.1 Blank an 8.000 ± 0.008 -in.² (51.613 ± 0.051 -cm²) circle representative of the foil or sheet, swab with acetone or other suitable solvent to ensure a surface free of soil, and determine the mass of the clean, dry specimen to the nearest 0.1 mg. Use a suitable solvent to remove any coating known to exceed 0.005 mg/ft² (4.645 mg/cm²) of surface area.

5. Calculation

5.1 Determine the thickness from the relationship:

$$T = \frac{M}{A \cdot D}$$

where:

T = thickness of the foil, sheet, or film, in. (or cm),

M = mass of the circle, g,

A = area of the circle, in.² (or cm²), and

D = density of the foil, sheet, or film, g/in.³ (or Mg/m³).

5.2 *Densities of Aluminum Alloys:*

5.2.1 Calculate the density of aluminum foil or sheet from chemical composition limits of the alloy by the method described in [Annex A2](#). The densities of foil or sheet alloys determined in this manner are accurate to within ± 0.3 %.

5.2.2 Calculated densities for some of the common foil or sheet alloys can be found in [Table 1](#). A column headed "mils/g for 8.000-in.² Area" is added for convenience in determining thickness of the 8.000-in.² (51.613-cm²) specimens. The mass of the specimen in grams multiplied by this factor is equal to the thickness of the foil or sheet in mils. One mil is equal to 0.001 in. (0.0254 mm).

6. Precision and Bias

6.1 Following the procedure outlined in this test method, repeated mass measurements of the same specimen on different

*A Summary of Changes section appears at the end of this standard

TABLE 1 Densities of Aluminum Foil or Sheet Alloys Applicable to the Determination of Thickness by the Mass Measurement Method

Alloy	Density		mils/g for 8.000-in. ² Area
	g/in. ³	Mg/m ^{3A}	
1100	44.41	2.71	2.815
1145	44.24	2.700	2.826
1188	44.24	2.700	2.826
1199	44.24	2.700	2.826
1235	44.33	2.705	2.820
3003	44.74	2.73	2.794
5052	43.92	2.68	2.846
5056	43.26	2.64	2.890
8079	44.57	2.72	2.805
8111	44.41	2.71	2.815

^A Registration Record of Aluminum Association Designations and Chemical Composition Limits for Wrought Aluminum and Wrought Aluminum Alloys, Aluminum Assoc., Washington, DC.

balances should result in agreement within 1 mg. It is outside of the scope of this test method to describe maintenance and calibration procedures for balances, but disagreement larger

than 1 mg warrants attention to maintenance or recalibration of the balance.

ANNEXES

(Mandatory Information)

A1. SPECIMEN SIZE AND SHAPE AND ITS EFFECT ON ACCURACY

A1.1 General

A1.1.1 Specimens of sizes and shapes other than the 8.000-in.² (51.613-cm²) circle may be used provided consideration is given to controllable factors affecting the accuracy of the method. Specifically, the area of the specimen shall be known and controlled to an accuracy of $\pm 0.1\%$, and the minimum mass of the specimen shall be 70 mg. Specimens ranging in size from 8 to 32 in.² (52 to 206 cm²) are convenient to handle and can be prepared to meet the aforementioned requirements.

A1.2 Source of Error

A1.2.1 Inherent errors in determining thickness by the mass measurement method result from the limits on the accuracy of the density value assigned to the alloy, the accuracy with which a specimen can be cut and its area determined, and the accuracy of the mass measurement. Much time could be devoted to a discussion of refinement of errors but it shall suffice here to draw on experience as a guide for determining the accuracy of the method.

A1.3 Error from Uncertainty of the Densities of the Specimen (E_D)

A1.3.1 The density of aluminum foil or sheet alloys shall be those listed in Table 1 or it shall be determined by the method described in Annex A2. Values so obtained are accurate to $\pm 0.3\%$ of the true density. The error imposed by uncertainty of the density then is $E_D = \pm 0.3\%$ of the thickness determined.

A1.4 Error from Control of the Area of the Specimen (E_A)

A1.4.1 A precision blanking press can cut a specimen whose area is known and reproducible to an accuracy of $\pm 0.1\%$. If d is the specific diameter required to provide the area used in the thickness computation, then the error in area resulting from a small error, Δd , in the diameter is $200 \Delta d/d\%$. It follows then that to maintain an area accurate to $\pm 0.1\%$, the tolerance on the diameter of the blanked circle shall be ± 0.0005 times the circle diameter. The fact that the tolerance on diameter decreases in direct proportion to the diameter is a factor to consider in selecting the specimen size to use in the method. Compliance with this tolerance limits the area error to $E_A = \pm 0.1\%$ of the thickness determined.

A1.5 Error from Measuring the Mass of the Specimen (E_M)

A1.5.1 The accuracy of measuring the mass of a foil or sheet specimen has been found to be 0.7 mg. This imposes a maximum error on the method of $\pm 0.07/(T \cdot A \cdot D)\%$ of the thickness determined. Since D , density of the foil or sheet, is fixed, it is seen that the magnitude of the mass measurement error is a function of the thickness, T , of the foil or sheet and the area, A , of the specimen. The area, A , is a controllable factor in the method, and the importance of selecting a large area to minimize the overall percentage error in the method for thin foil or sheet is apparent from a few simple calculations. The product $T \cdot A \cdot D$ is the mass of the specimen in grams, so to prevent the mass measurement error from introducing errors in