

Designation: F1088 – 04a^{ε1}

Standard Specification for Beta-Tricalcium Phosphate for Surgical Implantation¹

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 ε^1 Note—Subsection 3.2 was editorially corrected in June 2006.

1. Scope

1.1 This specification covers chemical and crystallographic requirements for biocompatible beta-tricalcium phosphate (β-TCP) for surgical implant applications. For a material to be identified as medical grade beta-tricalcium phosphate, it must conform to this specification (see Appendix X1).

2. Referenced Documents

2.1 ASTM Standards:²

F748 Practice for Selecting Generic Biological Test Methods for Materials and Devices

F981 Practice for Assessment of Compatibility of Biomaterials for Surgical Implants with Respect to Effect of Materials on Muscle and Bone

2.2 American Society for Quality (ASQ) Document:

C1 Specification of General Requirements for a Quality Program³

2.3 International Organization for Standardization Document:

ISO 10993 Biological Evaluation of Medical Devices⁴

2.4 United States Pharmacopeia (USP) Documents:⁵ 04Inductively coupled plasma/mass spectroscopy (ICP/MS), Identification Tests for Calcium and Phosphate <191> Lead <252> in 2.4 and 2.5 shall be used. Mercury <261>

Arsenic <211>

Heavy Metals <231> Method 1

U.S. Geological Survey Method⁶

3. Chemical Requirements

3.1 Elemental analysis for calcium and phosphorus will be consistent with the expected stoichiometry of beta-tricalcium phosphate $(Ca_3(PO_4)_2)$. The calcium and phosphorus content shall be determined using a suitable method such as USP <191> (see 2.4) or X-ray fluorescence.

3.2 A quantitative X-ray diffraction analysis shall indicate a minimum beta-tricalcium phosphate content of 95 % as determined using Powder Diffraction File #5508987 and a method equivalent to Forman⁸ or Rietveld.^{9,10}

3.3 For beta-tricalcium phosphate, the concentration of trace elements shall be limited as follows:

Other Metals	ppm, max
Pb,	30
	5
As	3
Cd	5

atomic absorption spectroscopy (AAS), or the methods listed

3.3.1 The analysis of other trace elements may be required, based on the conditions, apparatus, or environments specific to the manufacturing techniques and raw materials.

3.4 The maximum allowable limit of all heavy metals determined as lead will be 50 ppm as described in 2.4 or

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For Annual Book of ASTM Standards volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from American Society for Quality (ASQ), 600 N. Plankinton Ave., Milwaukee, WI 53203-3005.

⁴ Available from American National Standards Institute (ANSI), 25 W. 43rd St., 4th Floor, New York, NY 10036.

⁵ Available from U.S. Pharmacopeia (USP), 12601 Twinbrook Pkwy., Rockville, MD 20852.

^{2.5} Other Reference:

⁶ Crock, J. G., Felichte, F. E., and Briggs, P. H., "Determination of Elements in National Bureaus of Standards Geological Reference Materials SRM 278 Obsidian and SRM 688 Basalt by Inductively Coupled Plasma-Atomic Emission Spectrometry," Geostandards Newsletter, Vol 7, 1983, pp. 335-340.

⁷ International Centre for Diffraction Data, 12 Campus Blvd, Newtown Square, PA 19073-3273.

⁸ Forman, D. W. and Metsger, D. S., "The Determination of Phase Composition of Calcium Phosphate Ceramics by X-Ray Diffraction," Transactions of the Seventh Annual Meeting of the American Society for Bone and Mineral Research, Kelseyville, CA, 1985 p. 391.

⁹ Jackson, L. E., Barralet, J. E., and Wright, A. J., "Rietveld Analysis in Sintering Studies of Ca-Deficient Hydrxyapatite," Bioceramics 16, Key Engineering Materials, Vols 254-256, 2004, pp. 297-300.

¹⁰ Rietveld, H. M., Acta Crystallogr., Vol 22, 1967, p. 151.