



Designation: D6082 – 06

Standard Test Method for High Temperature Foaming Characteristics of Lubricating Oils¹

This standard is issued under the fixed designation D6082; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope*

1.1 This test method describes the procedure for determining the foaming characteristics of lubricating oils (specifically transmission fluid and motor oil) at 150°C.

1.2 Foaming characteristics of lubricating oils at temperatures up to 93.5°C are determined by Test Method D892 or IP 146.

1.3 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.4 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

2. Referenced Documents

2.1 *ASTM Standards:*²

D892 Test Method for Foaming Characteristics of Lubricating Oils

E1 Specification for ASTM Liquid-in-Glass Thermometers

E128 Test Method for Maximum Pore Diameter and Permeability of Rigid Porous Filters for Laboratory Use

E1272 Specification for Laboratory Glass Graduated Cylinders

2.2 *Energy Institute Standards:*³

IP 146 Standard Method of Test for Foaming Characteristics of Lubricating Oils

3. Terminology

3.1 Definitions:

¹ This test method is under the jurisdiction of ASTM Committee D02 on Petroleum Products and Lubricants and is the direct responsibility of Subcommittee D02.06 on Analysis of Lubricants.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from Energy Institute, 61 New Cavendish St., London, WIG 7AR, U.K., <http://www.energyinst.org.uk>.

3.1.1 *diffuser, n—for gas*, a device for dispersing gas into a liquid (Test Method D892).

3.1.1.1 *Discussion*—Although diffusers can be made of either metallic or non-metallic materials, in this test method the diffuser is sintered stainless steel.

3.1.2 *entrained air (or gas), n—in liquids*, a two-phase mixture of air (or gas) dispersed in a liquid in which the liquid is the major component on a volumetric basis.

3.1.2.1 *Discussion*—The air (or gas) is in the form of discrete bubbles of about 10 to 1000 μm in diameter. The bubbles are not uniformly dispersed. In time, they rise to the surface to coalesce to form larger bubbles which break or form foam. Subsurface coalescence can also occur, in which case, the bubbles will rise more rapidly.

3.1.3 *foam, n—in liquids*, a collection of bubbles formed in or on the surface of a liquid in which the air or gas is the major component on a volumetric basis.

3.1.4 *gas, n—a fluid (such as air) that has neither independent shape nor volume but tends to expand indefinitely.*

3.1.5 *lubricant, n—any material interposed between two surfaces that reduces friction or wear between them.*

3.1.5.1 *Discussion*—In this test method, the lubricant is an oil which may or may not contain additives such as foam inhibitors.

3.1.6 *maximum pore diameter, n—in gas diffusion*, the diameter of a capillary of circular cross-section which is equivalent (with respect to surface tension effects) to the largest pore of the diffuser under consideration. The pore dimension is expressed in micrometers (μm).

3.1.7 *permeability, n—in gas diffusion*, the rate of a substance that passes through a material (diffuser) under given conditions.

3.2 Definitions of Terms Specific to This Standard:

3.2.1 *bottom volume*—the volume of liquid sample, that is, sample substantially free of air, at any given time during the test.

3.2.2 *collapse time, n—in foam testing*, the time in seconds, for zero foam to appear after the air is disconnected at the end of the five minute air blowing time.

*A Summary of Changes section appears at the end of this standard.

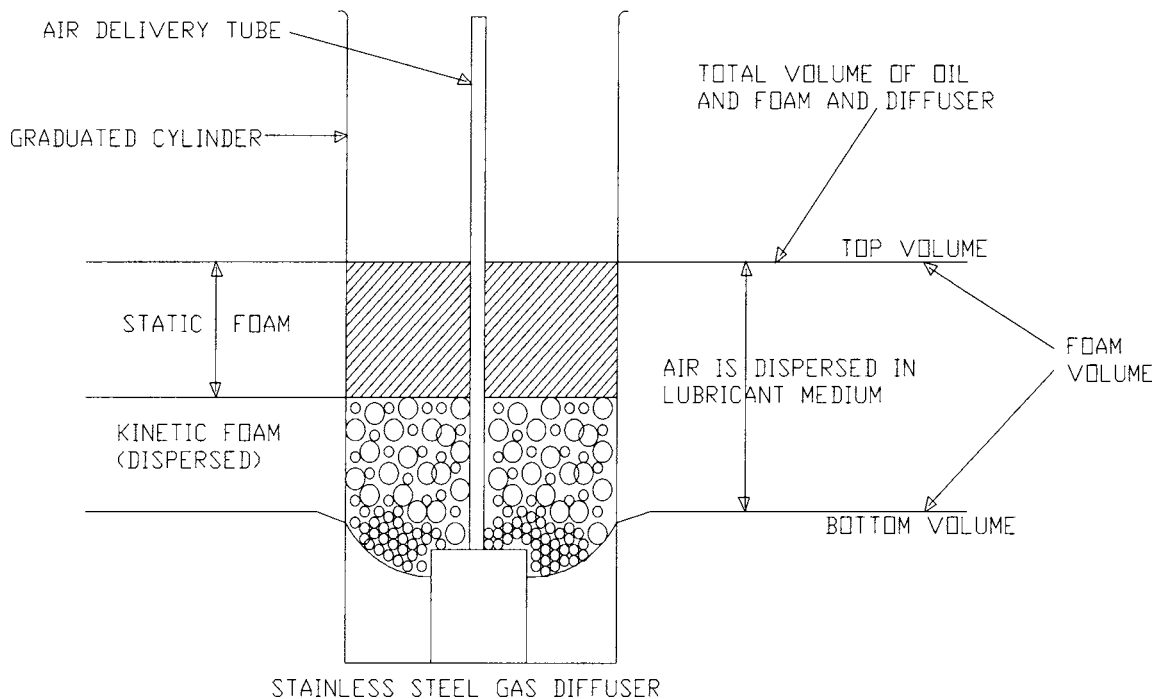


FIG. 1 Terminology Diagram

3.2.3 *dynamic bubble, n*—the first bubble to pass through and escape from the diffuser followed by a continuous succession of bubbles when testing for the pore diameter in Annex A1.

3.2.3.1 *Discussion*—When a diffuser is immersed in a liquid such as propan-2-ol, air can be trapped in the pores. It can escape eventually or as soon as a pressure is applied to the diffuser. When testing for pore diameter (Annex A1), the escape of such bubbles is to be ignored.

3.2.4 *foam stability, n—in foam testing*, the amount of static foam remaining at specified times following the disconnecting of the air supply.

3.2.4.1 *five-second foam stability*—the amount of static foam present 5 s after disconnecting the air supply.

3.2.4.2 *fifteen-second foam stability*—the amount of static foam present 15 s after disconnecting the air supply.

3.2.4.3 *one-minute foam stability*—the amount of static foam present 1 min after disconnecting the air supply.

3.2.4.4 *five-minute foam stability*—the amount of static foam present 5 min after disconnecting the air supply.

3.2.4.5 *ten-minute foam stability*—the amount of static foam present 10 min after disconnecting the air supply.

3.2.5 *foaming tendency, n—in foam testing*, the amount of static foam immediately before the cessation of air flow.

3.2.6 *kinetic foam, n*—entrained air that has been created by the passage of air through the diffuser during the test (see Fig. 1).

3.2.6.1 *Discussion*—Because the process of passing air through the diffuser and the oil sample during the test has resulted in an increase in volume and because such entrained air can be considered as foam on its way to being made, the term kinetic foam has been introduced.

3.2.7 *percent volume increase, n—in foam testing*, the increase in total volume expressed as a percentage of the initial total volume with diffuser in place at test temperature.

3.2.8 *static foam, n*—foam that has been created by the passage of air through the diffuser during the test (see Fig. 1).

3.2.9 *top volume, n*—the volume of the foam (if any), liquid, diffuser, and the submersed portion of the delivery tube (see Fig. 1).

3.2.10 *total volume, n—in foam testing*, the volume of foam, liquid, diffuser, and submersed portion of delivery tube (see Fig. 1).

3.2.10.1 *initial total volume (V_1), n—in foam testing*, the volume of the foam, liquid, diffuser, and submersed portion of the delivery tube at test temperature prior to connecting the air supply.

3.2.10.2 *final total volume (V_2), n—in foam testing*, the volume of the foam, liquid, diffuser, and submersed portion of the delivery tube just before disconnecting the air supply.

3.2.11 *zero foam, n—in foam testing*, occurs when any portion of the top surface of the liquid is free of bubbles.

4. Summary of Test Method

4.1 A measured quantity of sample is heated to 49°C for 30 min and allowed to cool to room temperature. The sample is transferred to a 1000 mL graduated cylinder, heated to 150°C, and aerated at 200 mL/min with dry air for 5 min with a metal diffuser. The amount of foam generated before disconnecting the air, the amount of static foam at optional times after disconnecting the air (list of options in Section 10); and the time for the foam to collapse are measured and the percent increase in total volume calculated.

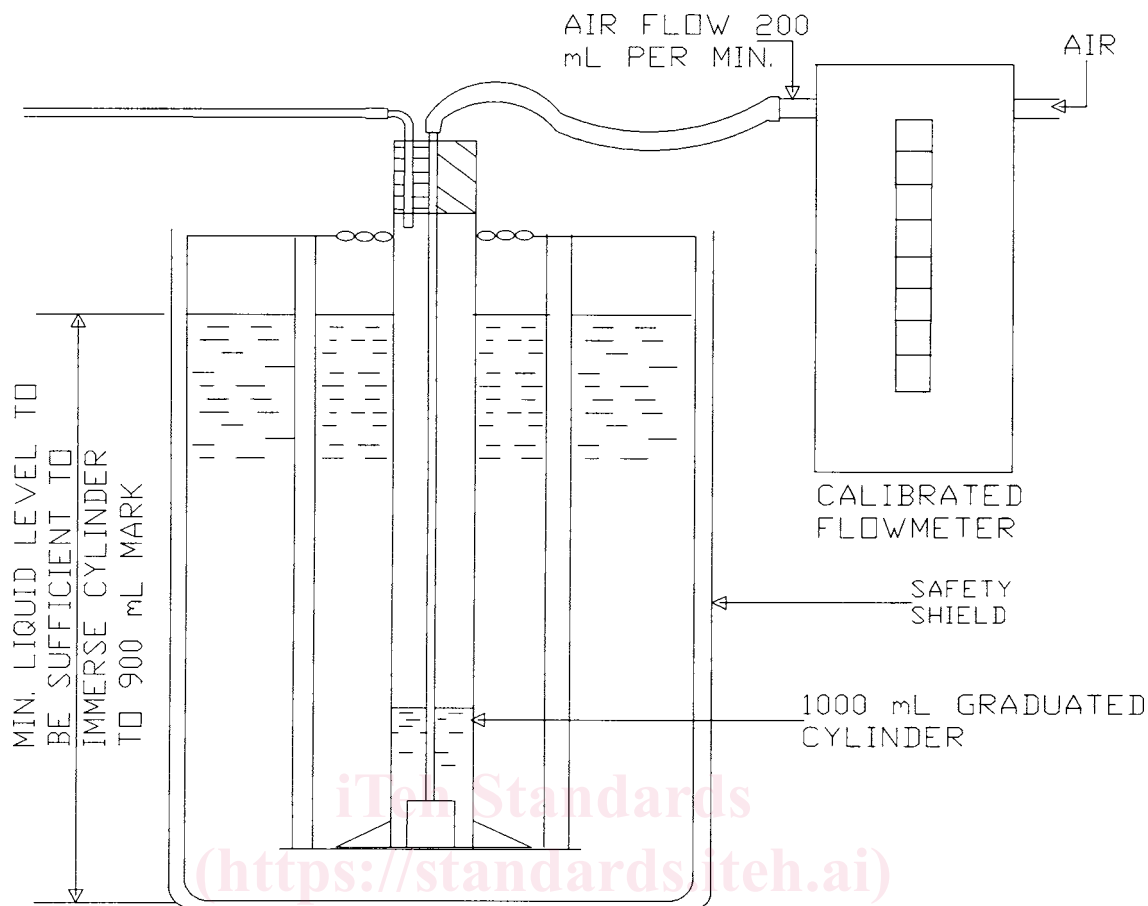


FIG. 2 Typical Liquid Bath

5. Significance and Use

5.1 The tendency of oils to foam at high temperature can be a serious problem in systems such as high-speed gearing, high volume pumping, and splash lubrication. Foaming can cause inadequate lubrication, cavitation, and loss of lubricant due to overflow, and these events can lead to mechanical failure.

5.2 Correlation between the amount of foam created or the time for foam to collapse, or both, and actual lubrication failure has not been established. Such relations should be empirically determined for foam sensitive applications.

6. Apparatus

6.1 *Heating Bath*, any heating system capable of maintaining a sample temperature of $150 \pm 1^\circ\text{C}$ (see Fig. 2).

NOTE 1—The precision of this test method was determined using only liquid baths.

6.1.1 *Heating Transfer Fluid*, any liquid with low volatility, at the test temperature, chemical stability and sufficient low viscosity to permit stirring.

NOTE 2—A 4 to 7 mm²/s (cSt) polyalphaolefin has been found to be a suitable fluid. Silicone fluids are foam inhibitors and their use may alter the foam characteristics of the test sample and should be discouraged.

NOTE 3—A slow purge of nitrogen gas through the fluid, or as a blanket over the bath fluid, reduces darkening (oxidation) of the bath fluid.

6.2 *Foaming Test Apparatus*, 1000-mL graduated cylinder (of cylinders meeting Specification E1272 Class B tolerance

requirement of ± 6 mL and having at least graduations of 10 mL), fitted with a device to overcome buoyancy if a liquid bath is used and modified to have a circular top. It shall be capable of withstanding the extreme temperature conditions of this test method.

NOTE 4—It may be necessary to confirm the volume of the cylinder.

NOTE 5—A heavy metal ring large enough to fit over the outside and rest on the bottom of the cylinder has been found to be suitable to overcome buoyancy.

NOTE 6—Graduated cylinders having a pouring spout can be prepared for this test method by making a horizontal cut below the spout and removing the part above the cut. The cut edges of the cylinder should be fire-polished or smoothed by grinding.

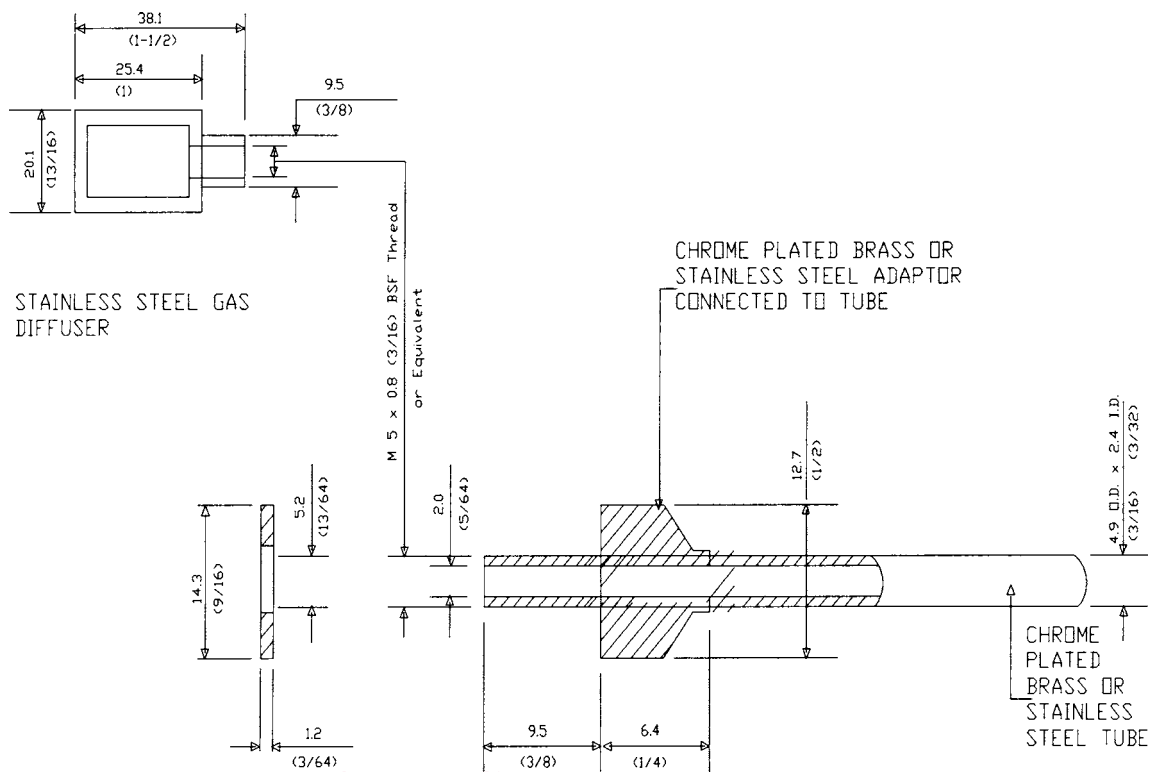
6.3 *Flow Meter and Regulator*, calibrated and capable of maintaining air flow volume of 200 ± 5 mL/min.

6.3.1 A gas volume meter graduated in hundredths of a litre, or a technically equivalent flow measuring device, with sufficient capacity to measure a flow rate of at least 6000 mL/min, while generating a back pressure of no more than 10 mm of water, is required.

6.4 *Laboratory Oven*, without fan, capable of maintaining $49 \pm 1^\circ\text{C}$.

6.5 *Stirrer*, capable of 500 rpm \pm 100 rpm, fitted with a 1½-in., 3-blade, 1.5-pitch marine propeller.

6.6 *Stainless Steel Diffuser*, 5 µm stainless steel diffuser meeting the specifications of pore size 15 to 60 µm and



WASHER TYPICAL AIR TUBE ASSEMBLY

Dimensions are Millimeters

(Inches)

FIG. 3 Gas Diffuser and Air Tube Dimensions

permeability between 3000 to 6000 mL/min when tested according to the method given in Annex A1 (see Fig. 3).

6.6.1 Calibration verification of new diffusers and calibration after each set (10 samples or less) of evaluations is required.

6.7 *Thermometer*, or other temperature sensing device capable of measuring $150 \pm 0.2^\circ\text{C}$. An ASTM thermometer Specification E1 No. 41C-86, or equivalent, is suitable.

6.8 *Timing Device*, capable of measuring minutes and seconds (± 1 s).

6.9 *High Speed Blender*, with a container capacity of one litre, capable of maintaining a speed of 22 000 rpm (± 2000).

7. Reagents

7.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society where

such specifications are available.⁴ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.

7.2 *Acetone*, (**Warning**—Extremely flammable. Vapors may cause flash fire.)

7.3 *Compressed Air*, hydrocarbon free and dry to a dew point of -60°C or lower.

7.4 *Heptane*, (**Warning**—Flammable. Vapor harmful.)

7.5 *Propan-2-ol*, (**Warning**—Flammable. Vapor harmful.)

7.6 *Toluene*, (**Warning**—Flammable. Vapor harmful.)

⁴ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.