DRAFT INTERNATIONAL STANDARD ISO/DIS 13099-3



ISO/TC 24/SC 4

Secretariat: DIN

Voting begins on 2013-03-25

Voting terminates on 2013-06-25

INTERNATIONAL ORGANIZATION FOR STANDARDIZATION • ΜΕЖДУНАРОДНАЯ ОРГАНИЗАЦИЯ ПО СТАНДАРТИЗАЦИИ • ORGANISATION INTERNATIONALE DE NORMALISATION

## Methods for zeta potential determination —

### Part 3: Acoustic methods

Méthode de détermination du potentiel zêta ---Partie 3: Méthodes acoustiques

ICS 19.120

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### Contents

Foreword	iv
Introduction	v
1 Scope	1
2 Normative references	1
3 Terms, definitions and symbols	1
4 Symbols	6
5 Principle	7
6 Zeta potential probe design elements	8
<ul> <li>7 Determination of the dynamic electrophoretic mobility</li> <li>7.1 Subtracting background electroacoustic signal generated by ions</li> </ul>	9
<ul> <li>8 Calculation of zeta potential</li></ul>	10 10 11
8.1.4 ESA theories for concentrates with particles interaction	
<ul> <li>8.2 Overlapped double layers.</li> <li>9 Operational procedures.</li> <li>9.1 Requirements.</li> <li>9.1.1 Instrument location</li> <li>9.1.2 Sample handling.</li> <li>9.1.3 Preparation.</li> <li>9.1.4 Sample volume fraction requirement.</li> <li>9.2 Verification</li> <li>9.2 Reference materials</li> <li>9.2 Repeatability</li> <li>9.2 Intermediate precision</li> <li>9.2 Trueness.</li> <li>9.3 Sources of measurement error</li> </ul>	13 13 13 13 13 13 14 14 14 15
Annex A (informative) Electroacoustics: high frequency Electrokinetics	
Annex B (informative) Verification of electroacoustic theories	17
Bibliography	20

### Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

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ISO 13099-3 was prepared by Technical Committee ISO/TC 24, Particle characterization including sieving, Acb Subcommittee SC 4, Particle characterization. 2

Here in standards in the standard standards is the standard standards in the standard standards is the standard standard standards is the standard standard standard standards is the standard st ISO 13099 consists of the following parts, under the general title Colloidal systems - Methods for zeta potential determination:

- Part 1: Electroacoustic and electrokinetic phenomena
- Part 2: Optical methods

The following part is under preparation:

Part 3: Acoustic methods

### Introduction

Zeta potential is a parameter that can be used to predict the long term stability of suspensions and emulsions and to study surface morphology and surface adsorption of particles and other surfaces in contact with a liquid. Zeta potential is not a directly measurable parameter. It can be determined using appropriate theoretical models from experimentally determined parameters, which depend on electric charge separation at interfaces. "Electrokinetic phenomena" encompass such experimentally observed effects. A group of electrokinetic phenomena at high frequency on MHz scale is referred to as "electroacoustics" [1]. Each classical electrokinetic phenomena have been widely used to determine electrophoretic mobility of various concentrated particulates without sample dilution. The purpose of this part of the ISO standard in methods for Zeta potential determination is description of general features of such electroacoustic methods that should be common for all instrumental implementation for measuring electrophoretic mobility using electroacoustics and following calculation of zeta potential of particulates.

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# Colloidal systems — Methods for zeta potential determination — Part 3: Acoustic methods

### 1 Scope

This part of ISO 13099 describes in general electroacoustic effects that can be defined as high frequency electrokinetic phenomena.

Particular attention is given to two methods of measurement of electrophoretic mobility of particles suspended in a liquid at high concentration above 1 % vl, colloid vibration current (CVI) [2] and electric sonic amplitude (ESA) [3, 4], respectively.

Estimation of surface charge and determination of zeta potential may be achieved from measured electrophoretic mobility using proper theoretical models, which are described in detail in ISO 13099, Part 1.

### 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 13099-1:2012, Colloidal systems — Methods for zeta potential determination — Part 1: Electroacoustic and electrokinetic phenomena

ISO 13099-2:2012, Colloidal systems — Methods for zeta potential determination — Part 2: Optical methods

### 3 Terms, definitions and symbols

For the purposes of this document, the following terms and definitions apply.

### 3.1 Electric double layer (EDE)

NOTE The electric double layer (EDL) is a spatial distribution of electric charges that appears on and at the vicinity of the surface of an object when it is placed in contact with a liquid.

### 3.1.1

### Debye-Hückel approximation

model assuming small electric potentials in the electric double layer

### 3.1.2

Debye length  $\kappa^1$ 

characteristic length of the electric double layer in an electrolyte solution

NOTE 1 to entry: The Debye length is expressed in nanometers.

# 3.1.3 diffusion coefficient

D

mean squared displacement of a particle per unit time

#### 3.1.4 **Dukhin number**

### Du

dimensionless number which characterizes contribution of the surface conductivity in electrokinetic and electroacoustic phenomena, as well as in conductivity and dielectric permittivity of heterogeneous systems

### 3.1.5

### dynamic viscosity

 $\eta$ 

ratio of the applied shear stress and the rate of shear of a liquid

NOTE 1 to entry: For the purpose of this part of ISO 13099, dynamic viscosity is used as a measure of resistance of a fluid which is being deformed by shear stress.

Dynamic viscosity determines the dynamics of an incompressible Newtonian fluid. NOTE 2 to entry:

NOTE 3 to entry: Dynamic viscosity is expressed in pascal seconds.

### 3.1.6

### electric surface charge density

 $\sigma$ 

charges on interface per area due to specific adsorption of ions from the liquid bulk or due to dissociation of the surface groups

Electric surface charge density is expressed in coulombs per square metre. NOTE 1 to entry:

### 3.1.7

### electric surface potential

standat  $\psi^{s}$ difference in electric potential between the surface and the bulk liquid

NOTE 1 to entry: Electric surface potential is expressed in volts -de9-9500

### 3.1.8

#### ζ-potential electrokinetic potential zeta potential

https://standards.ite difference in electric potential between that at the slipping plane and that of the bulk liquid

Electrokinetic potential is expressed in volts. NOTE 1 to entry:

### 3.1.9

### Gouy-Chapman-Stern model

model describing the electric double layer

### 3.1.10

### isoelectric point

condition of liquid medium, usually the value of pH, that corresponds to zero zeta-potential of dispersed particles

### 3.1.11

### slipping plane

shear plane

abstract plane in the vicinity of the liquid/solid interface where liquid starts to slide relative to the surface under influence of a shear stress

### 3.1.12

### Stern potential

ψd

electric potential on the external boundary of the layer of specifically adsorbed ions

NOTE 1 to entry: Stern potential is expressed in volts.

#### 3.2 Electrokinetic phenomena

Electrokinetic phenomena are associated with tangential liquid motion adjacent to a charged surface. NOTE 1 to entry:

### 3.2.1

#### electroosmosis

motion of liquid through or past a charged surface, e.g. an immobilized set of particles, a porous plug, a capillary or a membrane, in response to an applied electric field, which is the result of the force exerted by the applied field on the countercharge ions in the liquid

### 3.2.2

### electroosmotic counter-pressure

 $\Delta p_{eo}$ 

pressure difference that is applied across the system to stop the electroosmotic flow

NOTE 1 to entry: The electroosmotic counter-pressure value is positive if the high pressure is on the higher electric 1 De potential side. 20 2

Electroosmotic counter-pressure is expressed in pascals NOTE 2 to entry:

### 3.2.3

### electroosmotic velocity

veo

uniform velocity of the liquid far from the charged interface

NOTE 1 to entry: Electroosmotic velocity is expressed in metres per second. standar

### 3.2.4

#### electrophoresis

movement of charged colloidal particles or polyelectrolytes, immersed in a liquid, under the influence of an external electric field

### 3.2.5

#### electrophoretic mobility

μ

electrophoretic velocity per unit electric field strength

NOTE 1 to entry: Electrophoretic mobility is positive if the particles move toward lower potential (negative electrode) and negative in the opposite case.

NOTE 2 to entry: Electrophoretic mobility is expressed in metres squared per volt second.

#### 3.2.6 electrophoretic velocity

### $v_{e}$

particle velocity during electrophoresis

NOTE 1 to entry: Electrophoretic velocity is expressed in metres per second.

### 3.2.7

### sedimentation potential

 $U_{sed}$ 

potential difference sensed by two electrodes placed some vertical distance apart in a suspension in which particles are sedimenting under the effect of gravity

NOTE 1 to entry: When the sedimentation is produced by a centrifugal field, the phenomenon is called centrifugation potential.

NOTE 2 to entry: Sedimentation potential is expressed in volts.

### 3.2.8

### streaming current

<sup>l</sup>str

current through a porous body resulting from the motion of fluid under an applied pressure gradient

NOTE 1 to entry: Streaming current is expressed in amperes.

### 3.2.9

### streaming current density

 $J_{\rm str}$ 

streaming current per area

Streaming current density is expressed in coulombs per square metre NOTE 1 to entry: standardsisi

### 3.2.10

### streaming potential

 $U_{\rm str}$ 

1150-13099 ar potential difference at zero electric current, caused by the flow of liquid under a pressure gradient through a capillary, plug, diaphragm or membrane

NOTE 1 to entry: Streaming potentials are created by charge accumulation caused by the flow of countercharges inside volts. capillaries or pores.

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https

Streaming potential is expressed in NOTE 2 to entry:

### 3.2.11

#### surface conductivity

 $K^{\sigma}$ 

excess electrical conduction tangential to a charged surface

NOTE 1 to entry: Surface conductivity is expressed in siemens.

#### 3.3 Electroacoustic phenomena

NOTE 1 to entry: Electroacoustic phenomena arise due to the coupling between the ultrasound field and electric field in a liquid that contains ions. Either of these fields can be primary driving forces. Liquid might be a simple newtonian liquid or complex heterogeneous dispersion, emulsion or even a porous body. There are several different electroacoustic effects, depending on the nature of the liquid and type of the driving force.

### 3.3.1 colloid vibration current

CVI

A<sub>cvi</sub>

a.c. current generated between two electrodes, placed in a dispersion, if the latter is subjected to an ultrasonic field

NOTE 1 to entry: Colloid vibration current is expressed in amperes.