



Designation: D5135 – 07

Standard Test Method for Analysis of Styrene by Capillary Gas Chromatography¹

This standard is issued under the fixed designation D5135; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ϵ) indicates an editorial change since the last revision or reapproval.

1. Scope*

1.1 This test method covers the determination of the impurities typically found in styrene and uses those values to determine the purity of styrene by difference by gas chromatography. Typical impurities in commercial styrene are found in [Table 1](#).

1.2 This test method is applicable to impurity concentrations in the range of 0.001 to 1.00 % and for styrene purities of 99 % or higher when using the internal standard procedure.

1.3 In determining the conformance of the test results using this method to applicable specifications, results shall be rounded off in accordance with the rounding-off method of Practice [E29](#).

1.4 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.

1.5 *This standard does not purport to address all the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use. For a specific hazard statement, see Section 9.*

2. Referenced Documents

2.1 ASTM Standards:²

[D3437 Practice for Sampling and Handling Liquid Cyclic Products](#)

[D4307 Practice for Preparation of Liquid Blends for Use as Analytical Standards](#)

[D4790 Terminology of Aromatic Hydrocarbons and Related Chemicals](#)

[E29 Practice for Using Significant Digits in Test Data to](#)

[Determine Conformance with Specifications](#)

[E355 Practice for Gas Chromatography Terms and Relationships](#)

[E691 Practice for Conducting an Interlaboratory Study to Determine the Precision of a Test Method](#)

[E1510 Practice for Installing Fused Silica Open Tubular Capillary Columns in Gas Chromatographs](#)

2.2 Other Document

[OSHA Regulations, 29 CFR paragraphs 1910.1000 and 1910.1200](#)³

3. Terminology

3.1 See Terminology [D4790](#) for definitions of terms used in this test method.

4. Summary of Test Method

4.1 In this test method, the chromatogram peak area for each impurity is compared to the peak area of the internal standard (*n*-heptane or other suitable known) added to the sample. From the response factors of these impurities relative to that of the internal standard and the amount of internal standard added, the concentration of the impurities are calculated. The styrene content is obtained by subtracting the total amount of all impurities from 100.00.

5. Significance and Use

5.1 This test method is designed to obtain styrene purity on the basis of impurities normally present in styrene and may be used for final product inspections and process control.

5.2 This test method will detect the impurities shown in [Table 1](#), non-aromatic hydrocarbons containing ten carbons or less, and others where specific impurity standards are available. Absolute purity cannot be determined if unknown impurities are present.

6. Interferences

6.1 The internal standard chosen must be satisfactorily resolved from any impurity as well as the product peak. A peak will be satisfactorily resolved from a neighboring peak if the

¹ This test method is under the jurisdiction of ASTM Committee D16 on Aromatic Hydrocarbons and Related Chemicals and is the direct responsibility of Subcommittee D16.07 on Styrene, Ethylbenzene and C9 and C10 Aromatic Hydrocarbons.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For *Annual Book of ASTM Standards* volume information, refer to the standard's Document Summary page on the ASTM website.

³ Available from U.S. Government Printing Office Superintendent of Documents, 732 N. Capitol St., NW, Mail Stop: SDE, Washington, DC 20401, <http://www.access.gpo.gov>.

*A Summary of Changes section appears at the end of this standard

TABLE 1 Impurities Known or Suggested to be Present in Commercial Styrene

Ethylbenzene
o-xylene
m-xylene
p-xylene
Isopropylbenzene
n-propylbenzene
m-ethyltoluene
p-ethyltoluene
α-methylstyrene
m-vinyltoluene
p-vinyltoluene
Phenylacetylene

TABLE 2 Typical Instrument Parameters

Carrier gas	helium
Carrier gas flow rate at 110°C, mL/min	1.2
Detector	flame ionization
Detector temperature, °C	240
Injection port temperature, °C	230
Hydrogen flow rate, mL/min	30
Air flow rate, mL/min	275
Make up gas	nitrogen
Make up gas flow rate, mL/min	23
Split flow, mL/min	150
Column	60 m × 0.32 mm ID × 0.5 μm bonded polyethylene glycol-fused silica capillary
Column temperature, °C	110
Chart speed, cm/min	1
Sample size, μL	0.6

distance from the valley to the baseline between the two peaks is not greater than 50 % of the peak height of the smaller of the two peaks.

7. Apparatus

7.1 Gas Chromatograph—Any instrument having a flame ionization detector that can be operated at the conditions given in **Table 2**. The system shall have sufficient sensitivity to obtain a minimum peak height response for 0.001 weight % impurity of twice the height of the background noise.

7.2 Column—A capillary column containing a stationary phase of cross-linked polyethylene glycol has been found to be satisfactory. Any column can be used that is capable of resolving all significant impurities from styrene and from the internal standard.

7.3 Recorder—Electronic integration, with tangent capabilities is recommended.

7.4 100-mL Volumetric Flask.

7.5 Microsyringes, assorted volumes.

8. Reagents and Materials

8.1 Purity of Reagent—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Commit-

tee on Analytical Reagents of the American Chemical Society,⁴ where such specifications are available.

8.2 Carrier Gas, makeup and detector gases 99.999 % purity. Oxygen in carrier gas less than 1 ppm, less than 0.5 ppm is preferred. Purify carrier, makeup and detector gases to remove oxygen, water, and hydrocarbons.

8.3 Air, purify to remove hydrocarbons and water. Air should contain less than 0.1 ppm THC.

8.4 n-Heptane, 99.0 % minimum purity, or other internal standard, such as n-octane, previously analyzed to be free of compounds coeluting with impurities in the sample.

8.5 Styrene, the highest purity available, but not less than 99.7 %.

8.6 Pure Compounds, for calibration, shall be those compounds that are typically present in commercial styrene. These should be at least 99 % pure as they are to be used for determining response factors.

9. Hazards

9.1 Consult current OSHA regulations, suppliers' Material Safety Data Sheets, and local regulations for all materials used in this procedure.

10. Sampling

10.1 Sample the material in accordance with Practice **D3437**.

11. Preparation of Apparatus

11.1 Follow manufacturer's instructions for mounting and conditioning the column into the chromatograph and adjusting the instrument conditions described in **Table 2**, allowing sufficient time for the equipment to reach equilibrium. See Practices **E355** and **E1510** for additional information on gas chromatographic practices and terminology.

12. Calibration

12.1 Prepare a calibration mixture containing approximately 99.5 weight % styrene and the expected significant impurities at their expected concentration (see Practice **D4307**). Weigh all components to the accuracy required to calculate the concentration of each to the nearest 0.001 %.

12.2 With a microsyringe, add 50 μL of internal standard to a 100-mL volumetric flask about three-fourths full of calibration mixture. Mix well. Add calibration mixture to mark and again mix well. If n-heptane is used as the internal standard,

⁴ *Reagent Chemicals, American Chemical Society Specifications*, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see *Analar Standards for Laboratory Chemicals*, BDH Ltd., Poole, Dorset, U.K., and the *United States Pharmacopeia and National Formulary*, U.S. Pharmacopeial Convention, Inc. (USPC), Rockville, MD.