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**Solid biofuels — Determination of  
total content of carbon, hydrogen and  
nitrogen**

*Biocombustibles solides — Détermination de la teneur totale en  
carbone, hydrogène et azote*

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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2. [www.iso.org/directives](http://www.iso.org/directives)

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received. [www.iso.org/patents](http://www.iso.org/patents)

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT), see the following URL: [Foreword - Supplementary information](#)

The committee responsible for this document is ISO/TC 238, *Solid biofuels*.

For the purposes of research on instrumental methods for the determination of total carbon, hydrogen and nitrogen contents in solid biofuels standards, users are encouraged to share their views on ISO 16948:2015 and their priorities for changes to future editions of the document. Click on the link below to take part in the online survey:

[ISO 16948 online survey](#)

## Introduction

Instrumental methods for the analysis of carbon, hydrogen and nitrogen are now in widespread and in regular use, often in preference to formerly developed chemical methods for which International Standards exist.

The reliable determination of carbon, hydrogen and nitrogen is important for quality control and the results can be used as input parameters for calculations applied to the combustion of solid biofuels. The environmental importance of the nitrogen content is linked to emissions of NO<sub>x</sub> (formation of fuel NO<sub>x</sub>). Hydrogen content is important for calculation of the net calorific value. Carbon content is required for the determination of CO<sub>2</sub>-emissions.

It is recognized that the Kjeldahl method is most reliable for determining nitrogen contents with a concentration lower than 0,1 %. Possible suitable methods are summarized in the bibliography.

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# Solid biofuels — Determination of total content of carbon, hydrogen and nitrogen

## 1 Scope

This International Standard describes a method for the determination of total carbon, hydrogen and nitrogen contents in solid biofuels.

## 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 16559, *Solid biofuels — Terminology, definitions and descriptions*

ISO 14780<sup>1)</sup>, *Solid Biofuels — Sample preparation*

ISO 16993, *Solid biofuels — Conversion of analytical results from one basis to another*

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## 3 Terms and definitions **(standards.iteh.ai)**

For the purposes of this document the terms and definitions given in ISO 16559 and the following apply.

### 3.1

#### **reference material RM**

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material or substance one or more of whose property values are sufficiently homogeneous and well established to be used for the calibration of an apparatus, the assessment of a measurement method, or for assigning values to materials

### 3.2

#### **certified reference material CRM**

reference material, accompanied by a certificate, one or more of whose property values are certified by a procedure which establishes traceability to an accurate realisation of the unit in which the property values are expressed, and for which each certified value is accompanied by an uncertainty at a stated level of confidence

### 3.3

#### **NIST standard reference material SRM**

CRM issued by NIST that also meets additional NIST-specific certification criteria and is issued with a certificate or certificate of analysis that reports the results of its characterisations and provides information regarding the appropriate use(s) of the material

Note 1 to entry: The National Institute of Standards and Technology (NIST), known between 1901 and 1988 as the National Bureau of Standards (NBS), is a [measurement standards laboratory](#), also known as a National Metrological Institute (NMI), which is a non-regulatory agency of the [United States Department of Commerce](#)

1) To be prepared.

## 4 Principle

A known mass of sample is burnt in oxygen, or in an oxygen/carrier gas mixture, under conditions such that it is converted into ash and gaseous products of combustion. These consist mainly of carbon dioxide, water vapour, elemental nitrogen and/or oxides of nitrogen, oxides and oxyacids of sulfur and hydrogen halides. The products of combustion are treated to ensure that any hydrogen associated with sulfur or halides products of combustion are liberated as water vapour. Oxides of nitrogen are reduced to nitrogen, and those products of combustion which would interfere with the subsequent gas-analysis procedures are removed. The carbon dioxide, water vapour and nitrogen mass fractions of the gas stream are then determined quantitatively by appropriate instrumental gas analysis procedures.

## 5 Reagents and calibration substances

### 5.1 General

**WARNING — Care should be exercised when handling reagents, many of which are toxic and corrosive.**

Unless otherwise stated, use only reagents and calibration standards of recognized analytical grade for the analysis.

### 5.2 Carrier gas

The carrier gas used is helium or another suitable gas as specified by the instrument manufacturer.

### 5.3 Oxygen

Oxygen is used as specified by the instrument manufacturer.

### 5.4 Additional reagents

Additional reagents are of types and qualities as specified by the instrument manufacturer.

### 5.5 Calibration substances

Examples of pure organic substances suitable for calibration are given in [Table 1](#).

**Table 1 — Examples of suitable calibration substances and their theoretical C, H and N contents**

Name	Formula	% C	% H	% N
Acetanilide	C <sub>8</sub> H <sub>9</sub> NO	71,1	6,7	10,4
Atropin	C <sub>17</sub> H <sub>23</sub> NO <sub>3</sub>	70,6	8,0	4,8
Benzoic acid	C <sub>7</sub> H <sub>6</sub> O <sub>2</sub>	68,8	5,0	0,0
Cystine	C <sub>6</sub> H <sub>12</sub> N <sub>2</sub> O <sub>4</sub> S <sub>2</sub>	30,0	5,0	11,7
Diphenyl amine	C <sub>12</sub> H <sub>11</sub> N	85,2	6,6	8,3
EDTA	C <sub>10</sub> H <sub>16</sub> N <sub>2</sub> O <sub>8</sub>	41,1	5,5	9,6
Phenylalanine	C <sub>9</sub> H <sub>11</sub> NO <sub>2</sub>	65,4	6,7	8,5
Sulfanil amide	C <sub>6</sub> H <sub>8</sub> N <sub>2</sub> O <sub>2</sub> S	41,8	4,7	16,3
Sulfanilic acid	C <sub>6</sub> H <sub>7</sub> NO <sub>3</sub> S	41,6	4,1	8,1
TRIS	C <sub>4</sub> H <sub>11</sub> NO <sub>3</sub>	39,7	9,2	11,6

The materials shall be dry and of high purity, i.e. more than 99,9 %. For calibration purposes, the contents of C, H and N according to the certificate of the materials shall be used, not the theoretical contents. Other pure materials can be used provided that they meet the requirement of this standard.

## 5.6 Use of Certified Reference Materials (CRM or SRM)

Use certified reference materials, issued by an internationally recognized authority, to check if the accuracy of the calibration meets the required performance characteristics. Examples of certified reference materials are: NBS 1573 tomato leaves and NBS 1575 pine needles.

When, due to matrix effects or concentration range limitations, no good recoveries for the certified reference materials can be obtained, calibration with at least two CRM or SRM materials, may solve these problems. In that case CRM or SRM materials other than used for the calibration shall be used for verification purposes.

NOTE A CRM or SRM is prepared and used for three main purposes:

- a) to help develop accurate methods of analysis;
- b) to calibrate measurement systems used to facilitate exchange of goods, institute quality control, determine performance characteristics, or measure a property at the state-of-the-art limit;
- c) to ensure the long-term adequacy and integrity of measurement quality assurance programs.

## 6 Apparatus

No specific design of systems is presented here because there are a range of components and configurations available, which can be used to carry out the test method satisfactorily.

The apparatus shall, however, meet the following functional requirements:

- a) The conditions of combustion of the sample shall be such that all of the carbon (including that in mineral carbonates), the hydrogen (including that in the water of constitution of the minerals), and the nitrogen present, shall be converted into carbon dioxide, water vapour (except for hydrogen associated with oxyacids of sulfur and volatile halides), and gaseous nitrogen and/or oxides of nitrogen respectively.
- b) The combustion gases, or a representative aliquot, shall be treated to remove and/or separate out any components which would subsequently interfere with the detection and measurement of the carbon dioxide, water vapour or nitrogen in the gas stream.
- c) Hydrogen present as hydrogen halides or sulfur oxyacids shall be liberated, as water vapour, into the gas stream prior to determination of water vapour content.
- d) Any nitrogen oxides produced by the combustion process shall be reduced to nitrogen prior to presentation to the detection system.
- e) The detection systems shall provide responses that correlate directly with the concentrations of the combustion gases, over the full range applicable and preferably in a linear manner.
- f) If a nonlinear response is provided by a detection system, it shall include provisions for evaluating that response in a manner which correlates accurately with the concentration of the combustion gas.
- g) It shall include a means of displaying the detector responses or of calculating and presenting the concentrations of carbon, hydrogen and nitrogen in the sample following the input of other appropriate data as necessary.