
**Clothing for protection against heat
and flame — Determination of heat
transmission on exposure to both
flame and radiant heat**

*Vêtements de protection contre la chaleur et la flamme —
Détermination de la transmission de chaleur lors de l'exposition
simultanée à une flamme et à une source de chaleur radiante*

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

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For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 94, *Personal safety — Personal protective equipment*, Subcommittee SC 13, *Protective clothing*.

This second edition cancels and replaces the first edition (ISO 17492:2003), which has been technically revised. It also incorporates the Technical Corrigendum ISO 17492:2003/Cor.1:2004. The main changes compared with the previous edition are as follows:

- technical modifications and rewording have been made to all clauses, including to [Annexes A](#) and [B](#);
- [Clauses 5](#) to [12](#) have been renumbered;
- modifications have been made to [Figures 1](#), [2](#) and [3](#).

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Introduction

The measurement of the thermal energy transferred from the exterior of a material to the interior when exposed to a thermal hazard can be a significant factor in determining the level of protection or insulation provided by an assembly. While full-scale test methods are a better means of determining how an assembly performs, small scale tests such as those described in ISO 6942 and ISO 9151 can be used in establishing benchmarks of performance for the materials from which these assemblies are made. These tests enable the user of a material to anticipate how the properties of a particular material could affect the performance of the assembly when exposed to a high heat flux.

The purpose of an assembly for thermal protection is to prevent or reduce the potential for skin burn injury to the wearer. The performance of a product can be determined by comparing the total exposure energy to that which is transferred through the protective material to a known point where the thermal exposure would produce a burn injury in human tissue. The total exposure energy required to cause the onset of a second-degree burn in human tissue is identified as the thermal protection index (TPI). In the TPI analysis of the data, the specimen is exposed to steady heat until the energy transferred through the specimen is equivalent to the energy that would cause the onset of a second-degree burn injury (e.g. a blister).

Other uses include comparison of the insulation from a high-temperature exposure in terms other than the response of human tissue to heat exposure. For these uses, an alternate method of evaluating the heat transfer is provided. The total energy transferred that causes a temperature rise of the copper sensor by 12 °C and 24 °C is determined as the heat transfer index (HTI). In the HTI analysis of the data, the specimen is exposed to heat until a specified amount of energy is transferred. This is a measure of the insulation performance and thermal capacity of the specimen.

Unlike what is described in ISO 6942 or ISO 9151, the heat source in this test method is approximately 50 % radiant heat and 50 % convective heat. This equalized radiant/convective output is set to a thermal energy exposure having a heat flux of 84 kW/m². The magnitude of this heat flux is intended to determine the performance of the specimen when exposed to both the high temperature radiation and hot gases that exist in actual fire situations. The level of this heat flux represents a moderately high industrial or emergency fire-fighting exposure that requires the use of a protective material.

This document can be used to measure and describe the properties of materials, products or assemblies in response to both convective and radiant heat under controlled laboratory conditions. It is not recommended to use this document to describe or appraise a fire hazard or fire risk of materials, products or assemblies under actual fire conditions. However, the results of this test method can be used as elements of a fire-risk assessment that takes into account all of the factors pertinent to an assessment of the fire hazard of a particular end use.

NOTE 1 This test method does not necessarily correlate to the heat-insulation performance of vertically oriented flame-resistant textile materials when exposed to convective and radiant heat or used in actual clothing configurations.

NOTE 2 The performance of materials made of flame-resistant fibres can be determined by the amount of heat energy transferred through the specimen and by observing any changes affected by the exposure on the specimen. The TPI and the HTI measure the accumulated thermal energy received by a sensor, which is an indication of the ability of the material to inhibit the transfer of heat.

NOTE 3 A human tissue burn (blister) is predicted to result when the total thermal energy transmitted by the material reaches the second-degree burn threshold identified by the Stoll curve.

NOTE 4 The TPI or the HTI for flame-resistant materials can be used to establish anticipated thermal performance levels for single layer or multilayer constructions or assemblies.

NOTE 5 Different specimen-mounting conditions, which are determined by the number of layers of material in the test specimen, are provided in this method. Each condition emphasizes a different thermal characteristic of the sample and represents the way in which the material is used in the end-use application.

NOTE 6 The spaced configuration, with a spacer placed between the back surface of the specimen and the sensor, reflects applications in which there is an air space or gap between the specimen and the protected surface. This spaced configuration also eliminates the cooling effect, which occurs due to specimen contact with the sensor and allows the specimen to heat to a temperature during the test the same as that which might occur in an actual fire exposure. This mounting condition gives a measure of the insulation performance and thermal capacity of the specimen and air gap as a combination.

NOTE 7 The contact configuration, with the sensor in contact with the specimen, gives a measure of the insulation performance and thermal capacity of the specimen and reflects applications in which the textile is in contact with the protected surface.

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Clothing for protection against heat and flame — Determination of heat transmission on exposure to both flame and radiant heat

1 Scope

This document specifies a test method for measuring the heat transferred through horizontally mounted flame-resistant textile materials when exposed to a combination of convective and radiant heat. The exposure conditions are adjusted to be approximately a 50/50 mixture of pure convective heat and pure radiant heat. The total exposure heat flux is 84 kW/m².

This test method is applicable to any type of sheet material used either as a single layer or in a multilayer construction when all structures or sub-assemblies are made of flame-resistant materials. It does not apply to materials that are not flame resistant.

This test method does not apply to the evaluation of materials exposed to any other type of thermal energy sources, such as radiant heat only or flame contact only. ISO 6942 is applicable when evaluating materials for exposure to radiant heat only. ISO 9151 is applicable when evaluating materials due to flame contact only.

NOTE Some, but not all, textiles materials can ignite and continue to burn after exposure to the convective and radiant heat produced by this test method.

2 Normative references

ISO 17492:2019

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The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 139, *Textiles — Standard atmospheres for conditioning and testing*

3 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <http://www.iso.org/obp>
- IEC Electropedia: available at <http://www.electropedia.org/>

3.1

break-open

formation of a hole in the material during thermal exposure

3.2

burn injury

burn damage that occurs at various levels of depth within human tissue

Note 1 to entry: Burn injury in human tissue occurs when the tissue is heated and kept at an elevated temperature for a critical period of time. The amount of burn injury (first, second or third degree) depends upon both the level of the elevated temperature and the duration. The material performance in this document is related to a second-degree burn injury and is determined by the amount of thermal energy transferred through the specimen that is sufficient to cause the onset of a second-degree burn. The onset of a second-degree burn injury involves damage to the epidermis and part of the dermis.

3.3

charring

formation of carbonaceous residue as the result of pyrolysis or incomplete combustion

3.4

dripping

material response shown by the flow of the material and formation of falling droplets

3.5

embrittlement

formation of a brittle residue as the result of pyrolysis or incomplete combustion

3.6

exposure energy

thermal energy that is incident to the test specimen

3.7

exposure time

total time over which the *exposure energy* (3.6) is applied to the test material

3.8

heat flux

thermal intensity indicated by the amount of energy transmitted divided by time and by area to the surface

Note 1 to entry: Heat flux is expressed in kilowatts per square metre (kW/m²).

3.9

heat transfer index (dual exposure) (standards.iteh.ai)

HTI(DE)

time, in whole seconds, to cause a temperature rise of the copper calorimeter by 12 °C and 24 °C from a combined convective and radiant heat exposure

Note 1 to entry: The time to cause a 12 °C temperature rise is indicated with a suffix of T12, and that for a 24 °C rise with a suffix of T24, e.g. HTI(DE)-T12 and HTI(DE)-T24. The relative value between these two indices indicates the characteristic of the energy transfer. If HTI(DE)-T24 is twice that of HTI(DE)-T12, the rate of energy transfer is constant. If HTI(DE)-T24 is less than twice that of HTI(DE)-T12, the rate of energy transfer is increasing, showing a loss in insulation performance. If HTI(DE)-T24 is greater than twice that of HTI(DE)-T12, the rate of energy transfer is decreasing, showing increasing insulation performance.

3.10

heat transfer burn intersection

time, in seconds, at which the thermal energy transferred through the material and absorbed by the copper calorimeter intersects the *Stoll curve* (3.18) where a second-degree burn injury is predicted to begin

3.11

heat transfer burn time

time from the start of the thermal exposure to *heat transfer burn intersection* (3.10)

Note 1 to entry: Heat transfer is determined from the measured temperature rise of a sensor. In this test method, a copper calorimeter is used as the sensor. The calorimeter diameter is large enough to average the heat received through the exposed specimen. The calorimeter thickness is selected so as to cause the temperature rise of the sensor to be similar to that of human tissue when exposed to heat. The sensor face is painted a dull black to cause it to absorb radiant heat similarly to human tissue.

3.12**human tissue heat tolerance**

amount of thermal energy transferred to human tissue that predicts a reaction in human tissue, such as a pain sensation or the onset of a second-degree burn

Note 1 to entry: The tolerance of human tissue to heat exposure was developed by Stoll et al.^[4] (see [Table 1](#)) and is referred to as the *Stoll curve* ([3.18](#)). It is used in this method as the heat transfer criteria in determining the *thermal protection index (TPI)* ([3.19](#)) value of the test material.

3.13**ignition**

initiation of combustion

3.14**melting**

liquefaction of a material when exposed to heat

3.15**response to heat exposure**

observable response of the textile to the *exposure energy* ([3.6](#)) as indicated by *break-open* ([3.1](#)), *melting* ([3.14](#)), *dripping* ([3.4](#)), *charring* ([3.3](#)), *embrittlement* ([3.5](#)), *shrinkage* ([3.16](#)), *sticking* ([3.17](#)) or *ignition* ([3.13](#))

3.16**shrinkage**

decrease in one or more dimensions of an object or material

3.17**sticking**

response evidenced by softening of a material and adherence of one material to the surface of itself or another material

3.18**Stoll curve**

relationship between the amount of thermal energy absorbed by human tissue and the time of exposure which predicts the onset of a second-degree burn in human tissue

Note 1 to entry: See [Table 1](#).

3.19**thermal protection index****TPI**

total *exposure energy* ([3.6](#)) experienced by the specimen to cause a second-degree *burn injury* ([3.2](#)) to begin which is defined by the time the measured temperature of the copper calorimeter intercepts the *Stoll curve* ([3.18](#))

Note 1 to entry: The exposure energy is expressed as energy per unit area in kJ/m².

4 Principle

A flame-resistant specimen, mounted in a static horizontal position, is placed a specific distance from a combined convective/radiant heat source and exposed to a heat flux of $(84 \pm 2) \text{ kW/m}^2$ until sufficient thermal energy passes through the specimen to cause the equivalent of the onset of a second-degree burn injury in human tissue, or to indicate a temperature rise of 24 °C in the copper calorimeter.

The specimen is mounted either in direct contact with the copper calorimeter, designated as the “contact configuration”, or with a $(6,35 \pm 0,05) \text{ mm}$ air space between the specimen and the copper calorimeter, designated as the “spaced configuration”.

The test exposure is composed of convective energy supplied by two gas burners and radiant heat from nine radiant tubes. The combined total energy of the exposure is achieved by first setting the radiant exposure and then adding the convective source. The total energy exposure is then confirmed with the

copper calorimeter. Note that the gas burner flames contribute both convective and radiant heat to the surface of the specimen.

The amount of energy transferred through and by the specimen is measured with a copper calorimeter and analysed by one of the following two methods.

- a) The thermal performance can be evaluated from the times for a 12 °C and 24 °C temperature rise in the copper calorimeter: the HTI(DE)-T12 and HTI(DE)-T24 values. The rate at which the temperature of the copper calorimeter rises is a direct measurement of the thermal energy transferred.
- b) The thermal performance can also be compared with the times for the energy transferred through and by the specimen to cause the onset of a second-degree burn: the TPI. This index is based on the human tissue heat tolerance data of Stoll et al^[4].

The effect of the exposure on the physical appearance of the specimen shall be noted as specified in [10.4](#).

5 Apparatus

5.1 Heat source, consisting of a convective heat source and a radiant heat source. The convective heat source shall consist of two Meker or Fisher burners affixed beneath the specimen holder assembly opening and subtended at an angle between 30° to 45° from the horizontal so that the flames converge at a point immediately beneath the specimen. The radiant heat source shall consist of nine 500W quartz T3 translucent (frosted) infrared lamps affixed beneath and centred between the burners as shown in [Figure 1](#). The burners shall be Meker or Fisher burners with a 40 mm diameter top and with an orifice size appropriate for propane gas.

NOTE Two different energy sources are used in this test method. The energy from the Meker burners is primarily convective due to the flowing high temperature gases. These gases do emit thermal radiation. About 1/3 of the energy from the burners is in this form^[5]. The lamps are primarily radiant heaters but, because of their location below the test specimen, convection currents will rise and strike the test specimen. The fraction of the energy from the lamps that is convection is small.

5.2 Specimen holder assembly, consisting of a steel frame (7 850 ± 200) kg/m³ that rigidly holds and positions, in a reproducible manner, the specimen support holder plate and specimen relative to the heat source.

5.3 Protective shutter, placed between the heat source and the specimen. The protective shutter shall be capable of completely dissipating the thermal load from the heat source (usually by means of water cooling) for the time period before and after each specimen exposure. A microswitch shall be connected to the shutter or manually operated to indicate the start of the exposure to the data acquisition system. The start of exposure (i.e. time zero) shall be when the following edge clears the heat source.

5.4 Specimen mounting plate, consisting of a piece of steel (7 850 ± 200) kg/m³, 200 mm square and 3,2 mm thick, with a 100 mm square hole in its centre. A piece of 90° angle section steel (6,35 mm by 25 mm long) shall be welded to the outside edge of each corner perpendicular to the plane of the plate. The overall dimension, including the angle sections, will be about (212,7 ± 1) mm (see [Figure 2](#)).

5.5 Specimen holding plate, 200 mm × 200 mm × 3,2 mm thick steel (7 850 ± 200) kg/m³ with a 130 mm × 130 mm centred square hole. The spacer and sensor assembly shall fit without binding into the hole of the specimen holding plate (see [Figures 1](#) and [2](#)).

5.6 Spacer, 130 mm × 130 mm × (6,35 ± 0,05) mm thick steel (7 850 ± 200) kg/m³ with a 100 mm × 100 mm centred square hole (see [Figures 1](#) and [2](#)).

5.7 Sensor assembly, a copper calorimeter assembled in a mounting block with an additional weight on top.

The assembly consists of the following components.

- Copper calorimeter, consisting of a disc of copper of at least 99 % purity, weighing $(18,0 \pm 0,05)$ g and having a diameter of 40 mm and thickness of 1,6 mm with one thermocouple connected as specified in [Figure 3](#). The thermocouple shall be bonded to the copper disk by pinning or using a high melting point solder. The thermocouple wire size shall be AWG 30 or equivalent.
- Copper calorimeter mounting block consisting of a 128 mm × 128 mm square piece of asbestos-free non-combustible heat-insulating board of nominal thickness 13 mm machined as specified in [Figure 3](#). The thermal conductivity of the mounting block shall be $\leq 0,15$ W/m·K.
- The copper calorimeter shall be pinned to its mounting block at three locations around its circumference with flat head pins. An alternate means of mounting that has been found effective is to use a high-temperature epoxy for securing the copper calorimeter to its mounting block.
- The face of the copper calorimeter shall be flush with the surface of the mounting block. The face of the copper calorimeter shall also be spray-coated with a thin layer of flat black paint with an absorptivity of 0,95 or greater. See [Annex A](#) for possible suppliers of the paint.
- The weight shall be selected so that the complete sensor assembly (copper calorimeter in its mounting board plus the weight) shall weigh $(1\ 000 \pm 10)$ g in total. This weight shall produce a uniformly distributed load on the test specimen.

5.8 Data acquisition/analysis/equipment control system, which is required to control the timing of movement of parts of the test apparatus to record the data, and to calculate the time intervals. The timing functions are to open and close the shutter and to start and stop the temperature recording. The copper calorimeter temperature shall be recorded at least four times per second up to a temperature of 200 °C with a minimum resolution of 0,1 °C and an accuracy of $\pm 0,75$ °C. Timing accuracy shall be ± 2 %. It shall have a built-in cold junction correction and be capable of converting the millivolt signals from the type J thermocouple to temperature.

NOTE Other thermocouple types that meet the requirements of IEC 60584-1:1977 have been used, e.g. type K. This requires modification of the reference tables and [Table 1](#) to reflect the thermocouple output as a function of temperature.

5.9 Gas supply, propane (minimum 95 % by volume C_3H_8), with an appropriate reducer and valve arrangements to control the gas-supply pressure at (55 ± 1) kPa and capable of providing a flow equivalent to 2 l/min air at standard conditions (conditions are set for air and then an appropriate gas is used at those settings).

5.10 Gas flow control, any gas rotameter or mass flow controller with a range that gives a flow equivalent to 2 l/min air at standard conditions.

5.11 Heat flux transducer, a Schmidt-Boelter or Gardon-type radiation heat flux transducer with a diameter of 25 mm and a minimum heat flux operating range from 0 kW/m² to 20 kW/m² (see [9.1.3](#)). The heat flux transducer shall have a minimum view angle of 150° and a minimum spectral response flat within 3 % over a range of at least 1,0 µm to 3,0 µm. The unit used shall be traceable to a recognized national standards body such as NIST. The heat flux transducer shall have an accuracy of ± 3 %. If the heat flux transducer is water cooled, the cooling-water temperature shall be above the ambient dew-point temperature (for the laboratory environment).

5.12 Solvent, acetone or petroleum, to clean the sensor.

WARNING — Exercise care in using these solvents around heat sources.