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**Iron ores for blast furnace  
feedstocks — Determination of  
reduction under load**

*Minerais de fer pour charges de hauts fourneaux — Détermination de  
la réduction sous charge*

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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see [www.iso.org/directives](http://www.iso.org/directives)).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see [www.iso.org/patents](http://www.iso.org/patents)).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT) see the following URL: [Foreword - Supplementary information](#)

The committee responsible for this document is ISO/TC 102, *Iron ore and direct reduced iron*, Subcommittee SC 3, *Physical testing*.

This third edition cancels and replaces the second edition (ISO 7992:2007), of which it constitutes a minor revision to contemplate the outcomes of the studies on mass definition, as well as minor editorial improvements.

## Introduction

This International Standard concerns one of a number of physical test methods that have been developed to measure various physical parameters and to evaluate the behaviour of iron ores, including reducibility, disintegration, crushing strength, apparent density, etc. This method was developed to provide a uniform procedure, validated by collaborative testing, to facilitate comparisons of tests made in different laboratories.

The results of this test have to be considered in conjunction with other tests used to evaluate the quality of iron ores as feedstocks for blast furnace processes.

This International Standard can be used to provide test results as part of a production quality control system, as a basis of a contract, or as part of a research project.

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# Iron ores for blast furnace feedstocks — Determination of reduction under load

**CAUTION** — This International Standard may involve hazardous operations and equipment. This International Standard does not purport to address all of the safety issues associated with its use. It is the responsibility of the user of this International Standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to its use.

## 1 Scope

This International Standard specifies a method to provide a relative measure for evaluating the structural stability of iron ores when reduced under conditions resembling those prevailing in the reduction zone of a blast furnace.

This International Standard is applicable to lump ores and hot-bonded pellets.

## 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 2597-1, *Iron ores — Determination of total iron content — Part 1: Titrimetric method after tin(II) chloride reduction* ISO 7992:2015

ISO 2597-2, *Iron ores — Determination of total iron content — Part 2: Titrimetric methods after titanium(III) chloride reduction* ISO 7992:2015

ISO 3082, *Iron ores — Sampling and sample preparation procedures*

ISO 9035, *Iron ores — Determination of acid-soluble iron(II) content — Titrimetric method*

ISO 11323, *Iron ore and direct reduced iron — Vocabulary*

## 3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 11323 apply.

## 4 Principle

The test portion is isothermally reduced in a fixed bed, at 1 050 °C, under static load, using a reducing gas consisting of CO, H<sub>2</sub> and N<sub>2</sub>, until a degree of reduction of 80 % is obtained. The differential gas pressure across the bed and the change in the test bed height are measured at 80 % reduction.

## 5 Sampling, sample preparation and preparation of test portions

### 5.1 Sampling and sample preparation

Sampling of a lot and preparation of a test sample shall be in accordance with ISO 3082.

The size range for pellets and lump ores shall be – 12,5 mm + 10,0 mm.

A test sample of at least 6,0 kg, on a dry basis, of the sized material shall be obtained.

Oven-dry the test sample to constant mass at  $105\text{ °C} \pm 5\text{ °C}$  and cool it to room temperature before preparation of the test portions.

NOTE Constant mass is achieved when the difference in mass between two subsequent measurements becomes less than 0,05 % of the initial mass of the test sample.

## 5.2 Preparation of test portions

Collect each test portion by taking ore particles at random.

NOTE Manual methods of division recommended in ISO 3082, such as riffing, can be applied to obtain the test portions.

At least five test portions, each of approximately 1 200 g ( $\pm$  the mass of one particle) shall be prepared from the test sample: four test portions for testing and one for chemical analysis.

Weigh the test portions to the nearest 1 g and register the mass of each test portion on its recipient label.

## 6 Apparatus

### 6.1 General

The test apparatus shall comprise the following:

- a) ordinary laboratory equipment, such as an oven, hand tools and safety equipment;
- b) a reduction-tube assembly; [ISO 7992:2015](https://standards.iteh.ai/catalog/standards/sist/a952b0ee-c627-463b-9b91-6d57fe74582e/iso-7992-2015)
- c) a furnace; <https://standards.iteh.ai/catalog/standards/sist/a952b0ee-c627-463b-9b91-6d57fe74582e/iso-7992-2015>
- d) a system to supply the gases and regulate the flow rates;
- e) a tumble drum;
- f) test sieves;
- g) a weighing device.

[Figure 1](#) shows an example of the test apparatus.

**6.2 Reduction tube**, with a double wall made of non-scaling, heat-resistant metal to withstand temperatures higher than  $1\ 050\text{ °C}$  and resistant to deformation. The internal diameter of the inner reduction tube shall be  $125\text{ mm} \pm 1\text{ mm}$ . A removable perforated plate, made of non-scaling, heat-resistant metal to withstand temperatures higher than  $1\ 050\text{ °C}$ , shall be mounted in the reduction tube to support the test portion and to ensure uniform gas flow through it. The perforated plate shall be 10 mm thick, with a diameter 1 mm less than the internal diameter of the tube. The holes in the plate shall be 3 mm to 4 mm in diameter, at a pitch centre distance of 5 mm to 6 mm. The internal diameter of the outer reduction tube shall be enough to allow gas flow preheating before entering the inner reduction tube.

[Figure 2](#) shows an example of a reduction tube.

[Figure 3](#) shows the principle for oxygen flushing of thermocouples to avoid mismeasurement due to carbon deposition.

**6.3 Loading device**, capable of supplying a total static load of  $50\text{ kPa} \pm 2\text{ kPa}$  evenly to the test portion. The load shall be transferred by means of a ram with rigid perforated footplate, so as to distribute it evenly to the surface of the porcelain balls placed on top of the test portion. The footplate shall be 10 mm



thick, with a diameter 1 mm less than the internal diameter of the tube. The holes in the plate shall be 3 mm to 4 mm in diameter, at a pitch centre distance of 5 mm to 6 mm.

**6.4 Device for measuring differential gas pressure**, having a resolution of 0,01 kPa.

**6.5 Height-measuring device**, having a resolution of 0,1 mm.

**6.6 Porcelain balls**, having a size range between 10,0 mm and 12,5 mm.

**6.7 Furnace**, having a heating capacity and temperature control able to maintain the entire test portion, as well as the gas entering the bed, at  $1\ 050\text{ °C} \pm 10\text{ °C}$ .

**6.8 Balance**, capable of weighing the reduction tube assembly, including the test portion, to an accuracy of 1 g. The balance shall have an appropriate device to suspend the reduction tube assembly.

**6.9 Gas-supply system**, capable of supplying the gases and regulating gas flow rates. A frictionless connection between the gas-supply system and the reduction tube shall be ensured to not affect the weight loss determination during reduction.

**6.10 Weighing device**, capable of weighing the test sample and test portions to an accuracy of 1 g.

## 7 Test conditions

### 7.1 General

Volumes and flow rates of gases used are as measured at a reference temperature of 0 °C and at a reference atmospheric pressure of 101,325 kPa (1,013 25 bar).

### 7.2 Reducing gas

#### 7.2.1 Composition

The reducing gas shall consist of the following:

CO      40,0 %  $\pm$  0,5 % (volume fraction);

H<sub>2</sub>      2,0 %  $\pm$  0,5 % (volume fraction);

N<sub>2</sub>      58,0 %  $\pm$  0,5 % (volume fraction).

#### 7.2.2 Purity

Impurities in the reducing gas shall not exceed the following:

CO<sub>2</sub>    0,2 % (volume fraction);

O<sub>2</sub>      0,1 % (volume fraction);

H<sub>2</sub>O    0,2 % (volume fraction).

#### 7.2.3 Flow rate

The flow rate of the reducing gas, during the entire reducing period, shall be maintained at 83 L/min  $\pm$  1 L/min.

### 7.3 Heating and cooling gas

Nitrogen (N<sub>2</sub>) shall be used as the heating and cooling gas. Impurities shall not exceed 0,1 % (volume fraction).

The flow rate of N<sub>2</sub> shall be maintained at 50 L/min until the test portion reaches 1 050 °C and at 83 L/min during temperature-equilibration period. If desired, the test portion may be cooled under nitrogen to below 100 °C. During cooling, it shall be maintained at 5 L/min.

### 7.4 Temperature of the test portion

The temperature of the entire test portion shall be maintained at 1 050 °C ± 10 °C during the entire reducing period and, as such, the reducing gas shall be preheated before entering the test portion.

### 7.5 Loading of the test portion

During the entire heating and reducing periods, the test portion shall be subjected to a constant load of 50 kPa ± 2 kPa applied over the surface of the bed.

## 8 Procedure

### 8.1 Number of determinations for the test

Carry out the test as many times as required by the procedure in [Annex A](#).

### 8.2 Chemical analysis

Take, at random, one of the test portions prepared in [5.2](#) and use it for the determination of the iron(II) oxide content ( $w_1$ ) in accordance with [ISO 9035](#) and the total iron content ( $w_2$ ) in accordance with [ISO 2597-1](#) or [ISO 2597-2](#).

### 8.3 Reduction

In order to achieve a more uniform gas flow, place a double-layer bed of porcelain balls ([6.6](#)) in the reduction tube ([6.2](#)) on the perforated plate and level its surface. Measure the height of the top surface of the porcelain layer.

Take, at random, another test portion prepared in [5.2](#) and record its mass ( $m_0$ ). Place it on the bed of porcelain balls and level its surface.

Place a further double layer of the porcelain balls upon the test portion and level its surface. Measure the height of the top surface of this porcelain layer.

Close the top of the reduction tube by connecting the heating assembly containing the loading device ([6.3](#)) to the reduction tube. Insert the reduction tube assembly into the furnace ([6.7](#)) and suspend it centrally from the balance ([6.8](#)), ensuring that there is no contact with the furnace wall or heating elements.

Connect the thermocouple, ensuring that its tip is at the central position, as shown in [Figure 2](#). Connect the measurement devices for the differential pressure ([6.4](#)) and for the change in the height of the test bed ([6.5](#)).

Connect the gas-supply system ([6.9](#)), the discharge line and the compressed air to the loading device. Apply a load of 50 kPa ± 2 kPa.

Pass a flow of N<sub>2</sub> through the test portion at a rate of 50 L/min and commence heating. When the temperature of the test portion approaches 1 050°C, increase the flow rate to 83 L/min. Continue

heating while maintaining the flow of N<sub>2</sub> until the balance reading is constant and the temperature is constant at 1 050 °C ± 10 °C for 10 min.

**DANGER — Carbon monoxide and the reducing gas, which contains carbon monoxide, are toxic and therefore hazardous. Testing shall be carried out in a well-ventilated area or under a hood. Precautions should be taken for the safety of the operator, in accordance with the safety codes of each country.**

Tare the balance, start the time control device and immediately introduce the reducing gas at a flow rate of 83 L/min to replace the N<sub>2</sub>. Record the differential pressure across the test bed, the height of the test bed and the mass loss of the test portion ( $\Delta m_t$ ) continuously or at least every 5 min for the first 30 min, and thereafter at 10 min intervals.

Calculate the degree of reduction,  $R_t$ , relative to the iron(III) state, after  $t$  min, as shown in Formula (1)<sup>1)</sup>:

$$R_t = \left( \frac{0,111 w_1}{0,430 w_2} + \frac{\Delta m_t}{m_0 \times 0,430 w_2} \times 100 \right) \times 100 \quad (1)$$

where

$m_0$  is the mass, in grams, of the test portion;

$\Delta m_t$  is the mass loss, in grams, of the test portion after reduction time  $t$ ;

$w_1$  is the iron(II) oxide content, as a percentage by mass, of the test sample prior to the test, determined in accordance with ISO 9035, it is calculated from the iron(II) content by multiplying it by the oxide conversion factor FeO/Fe(II) = 1,286;

$w_2$  is the total iron content, as a percentage by mass, of the test portion prior to the test, determined in accordance with ISO 2597-1 or ISO 2597-2.

When the degree of reduction reaches 80 %, turn off the power and stop the flow of the reducing gas. Remove the load and record the time.

If, after 4 h, 80 % of reduction has not been achieved, the reduction may be stopped.

If any further evaluations are to be performed on the reduced test portion, introduce N<sub>2</sub> at a flow rate of 5 L/min until the test portion reaches room temperature.

## 9 Expression of results

### 9.1 Preparation of the reduction curve

Prepare the reduction curve by plotting the degree of reduction  $R_t$  against time  $t$ .

### 9.2 Calculation of the differential pressure at 80 % reduction ( $\Delta p_{80}$ )

The differential pressure at 80 % reduction,  $\Delta p_{80}$ , expressed in kPa, is calculated as follows.

Plot the differential gas pressure against the degree of reduction and read off, from the curve, the differential pressure ( $\Delta p_{80}$ ) corresponding to 80 % reduction.

Record the result to two decimal places.

1) The derivation of Formula (1) is given in [Annex B](#).