



Designation: D 5226 – 98

## Standard Practice for Dissolving Polymer Materials<sup>1</sup>

This standard is issued under the fixed designation D 5226; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last approval. A superscript epsilon ( $\epsilon$ ) indicates an editorial change since the last revision or reapproval.

### 1. Scope \*

1.1 This practice outlines the parameters applicable to the preparation of a polymeric solution, such as solvent, concentration, temperature, pressure, time, agitation, and heating mode.

1.2 The proper use of this practice requires a knowledge of solvents and their effect on polymeric materials.

1.3 *This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.*

NOTE 1—There is no equivalent ISO standard.

### 2. Referenced Documents

2.1 *ASTM Standards:*

D 883 Terminology Relating to Plastics<sup>2</sup>

D 1600 Terminology for Abbreviated Terms Relating to Plastics<sup>2</sup>

2.2 *Other Document:*

Polymer Handbook<sup>3</sup>

### 3. Terminology

3.1 Definitions are in accordance with Terminology D 883.

3.2 Abbreviations are in accordance with Terminology D 1600.

### 4. Summary of Practice

4.1 A polymer solution can be described or prepared using the cell classifications listing the parameters relative to solvate the polymer. The cell classifications are listed in the following order: polymer, solvent, concentration, temperature, time, container, heating mode, and agitation.

4.1.1 A polymer and a list of suggested solvents for making a solution are listed in Annex A1.

4.1.2 Table 1 designates the parameters for container, heating mode, and type of agitation.

NOTE 2—To illustrate the use of the cell classifications with Table 1, a 2 % solution of poly(vinyl chloride) using cyclohexanone would be written as:

PVC–cyclohexanone–20–66–40–BEC

where:

PVC	= the abbreviation of the polymer from Annex A1,
cyclohexanone	= the solvent from Annex A1,
20	= the weight of polymer in tenths of a percent,
66	= the temperature in degrees Celsius,
40	= the time in tenths of an hour,
B	= glass container from Table 1,
E	= bath heater from Table 1, and
C	= random agitation from Table 1.

### 5. Significance and Use

5.1 This practice embodies the specifications to describe the preparation of a polymeric solution.

### 6. Procedure

6.1 *Polymer*—Select the applicable polymer from Annex A1 and write its abbreviation.

6.2 *Solvent*—Select the solvent applicable to the polymer from Annex A1.

6.3 *Concentration*—Write the polymer gram weight in tenths of a percent per milliliter of solvent.

6.4 *Temperature*—Write the solution temperature in degrees Celsius.

6.5 *Time*—Write the time for solution in tenths of an hour.

6.6 *Container*—Select the type of container from Table 1.

6.7 *Heating Mode*—Select the heating mode from Table 1.

6.8 *Agitation*—Select the agitation mode from Table 1.

### 7. Precision and Bias

7.1 No statement is made about the precision or bias of this practice since the procedure is descriptive with no measurements being made.

### 8. Keywords

8.1 polymer solutions; solution preparation; solutions; solvents

<sup>1</sup> This practice is under the jurisdiction of ASTM Committee D-20 on Plastics and is the direct responsibility of Subcommittee D20.70 on Analytical Methods.

Current edition approved Nov. 10, 1998. Published January 1999. Originally published as D 5226 – 92. Last previous edition D 5226 – 97.

<sup>2</sup> *Annual Book of ASTM Standards*, Vol 08.01.

<sup>3</sup> Available from John Wiley and Sons, New York, NY.

**TABLE 1 Parameters**

Designation	Container	Heater Mode	Agitation Mode
A	unspecified	unspecified	unspecified
B	glass	none	none
C	sealed glass vial	oven	random
D	metal	hot plate	magnetic stirrer
E	fluoropolymer	bath	propeller blade
F		block heater	wrist action
G		electric mantle	ultrasonic
H		microwave	

## ANNEX

### (Mandatory Information)

#### A1. SUGGESTED SOLVENTS FOR POLYMERIC SOLUTIONS

##### A1.1 Note the following:

A1.1.1 The solvents are listed in random order.

A1.1.2 An increase in polymer molecular weight reduces solubility.

A1.1.3 Branching increases the solubility compared to a linear polymer of the same molecular weight.

A1.1.4 Solubility normally increases with rising temperature.

A1.1.5 The temperature is for room temperature unless noted.

A1.1.6 The following abbreviations are used:

D.S. degree of substitution

S.C. substituent content

conc. concentrated

##### A1.2 Solvents:

A1.1.7 *Acrylonitrile-butadiene-acrylate (ABA)*—aromatic hydrocarbons, chlorinated hydrocarbons, tetrahydrofuran, esters, ketones, N,N-dimethylformamide, N,N-dimethylacetamide (if high acrylonitrile).

A1.1.8 *Acrylonitrile-butadiene-styrene (ABS)*—N,N-dimethylformamide, N,N-dimethylacetamide (if high acrylonitrile), cyclohexanone (above 35°C), cyclohexanone/acetone, methylcyclohexane/acetone, decahydronaphthalene/dimethyl oxalate, benzene, toluene, ethylbenzene, styrene, lower chlorinated hydrocarbons, phenol/acetone, tetrahydrofuran, dimethyltetrahydrofuran, dioxane, methyl ethyl ketone, diisopropyl ketone, glycol formal, ethyl acetate, butyl acetate, methyl-, ethyl-, n-butyl phthalate, 1-nitropropane, carbon disulfide, tributyl phosphate, phosphorus trichloride.

A1.1.9 *Alkydes*—tetrahydrofuran.

A1.1.10 *Arylonitrile/methyl methacrylate (AMMA)*—benzene, toluene, xylene, methylene chloride, chloroform, ethylene chloride, chlorobenzene, isobutanol (hot), cyclohexanol (hot), *B*-ethoxyethanol, dioxane, methyl ethyl ketone, diisopropyl ketone, cyclohexanone, acetic acid, isobutyric acid, methyl formate, ethyl acetate, cyclohexyl acetate, isobutyl propionate, butyl lactate.

A1.1.11 *Allyl diglycol carbonate (ADC)*—benzene, chloroform, acetone.

A1.1.12 *Carboxymethyl cellulose (CMC)*:

A1.1.12.1 *S.C.* = 5 to 10 %—alkali.

A1.1.12.2 *S.C.* = 15 to 30 %—water (sodium salt).

A1.1.12.3 *S.C.* = *high*—benzene/alcohol, benzene/acetone, chloroform, pyridine, acetone, esters, tetrahydrofuran.

A1.1.13 *Cellulose Acetate (CA)*:

A1.1.13.1 *D.S.* = 0.6 to 0.8—water.

A1.1.13.2 *D.S.* = 1.3 to 1.7—2-methoxyethanol.

A1.1.13.3 *D.S.* = 2.0 to 2.3—methylene chloride/methanol @ 80:20, chloroform/methanol, benzyl alcohol, phenols, ethylene glycol ethers, dioxane, diethanolamine, pyridine, analine, acetone, cyclohexanone, formic acid, acetic acid (glacial), methyl acetate, ethyl acetate/nitrobenzene, glycol monoethyl ether acetate, nitromethane, tetrahydrofuran.

A1.1.14 *Cellulose acetate-butyrate (CAB)*:

A1.1.14.1 *D.S. (acetate)* = 0.8 and *D.S. (butyrate)* = 2.35—benzene, toluene (hot), chloroform, carbon tetrachloride, tetrachloroethane, methanol (hot), acetone, cyclohexanone, dioxane, aliphatic esters, nitroethane.

A1.1.14.2 *D.S. (acetate)* = 2.1 and *D.S. (butyrate)* = 0.7—chloroform, dichloroethane, tetrachloroethane, dioxane, acetone, cyclohexanone, methyl acetate, ethyl acetate, nitroethane.

A1.1.15 *Cellulose acetate propionate (CAP)*—benzene, dichloromethane, chlorobenzene, acetone, ethyl acetate.

A1.1.16 *Cellulose nitrate (CN)*:

A1.1.16.1 *N* = 6.8 %—water.

A1.1.16.2 *N* = 10.5 to 12 %—alcohol (lower), alcohol/diethyl ether, acetone, amyl acetate, ethylene glycol ethers, acetic acid (glacial).

A1.1.16.3 *N* = 12.7 %—halogenated hydrocarbons, ethanol/diethyl ether, acetone, methyl amyl acetone, cyclohexanone, methyl acetate, ethyl acetate, ethyl butyrate, ethyl lactate, ethylene glycol ether acetates, ethylene carbonate, furan derivatives, nitrobenzene.

A1.1.17 *Cellulose propionate (CP)*—benzene, dichloroethane, chlorobenzene, acetone, ethyl acetate.

A1.1.18 *Cellulose triacetate (CTA)*—methylene chloride, methylene chloride/ethanol @ 80:20, chloroform, chloroform/alcohol, trichloroethane, tetrahydrofuran, dioxane, acetone, acetone/water @ 80:20, methyl acetate, ethylene glycol ether acetates, ethylene carbonate.