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Standard Test Methods for Carbon Black—Sulfur Content¹

This standard is issued under the fixed designation D 1619; the number immediately following the designation indicates the year of original adoption or, in the case of revision, the year of last revision. A number in parentheses indicates the year of last reapproval. A superscript epsilon (ε) indicates an editorial change since the last revision or reapproval.

This standard has been approved for use by agencies of the Department of Defense.

1. Scope

1.1 These test methods cover the determination of the sulfur content of carbon black. The following test methods are included:

		Sections
Test Method A	High-Temperature Combustion With In-	6 to 13
	frared Absorption Detection Procedures	
Test Method B	X-Ray Fluorescence	14

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- 1.2 The values stated in SI units are to be regarded as standard. No other units of measurement are included in this standard.
- 1.3 This standard does not purport to address all of the safety concerns, if any, associated with its use. It is the responsibility of the user of this standard to establish appropriate safety and health practices and determine the applicability of regulatory limitations prior to use.

2. Referenced Documents

- 2.1 ASTM Standards: ²
- D 240 Test Method for Heat of Combustion of Liquid Hydrocarbon Fuels by Bomb Calorimeter
- D 1193 Specification for Reagent Water D1509Test Methods for Carbon Black—Heating Loss
- D 1509 Test Methods for Carbon BlackHeating Loss
- D 1799 Practice for Carbon Black—Sampling Packaged Shipments
- D 1900 Practice for Carbon Black—Sampling Bulk Shipment⁴Shipments
- D 4483 Practice for Determining Evaluating Precision for Test Method Standards in the Rubber and Carbon Black Manufacturing Industries
- E 1 Specification for ASTM Liquid-in-Glass Thermometers

3. Significance and Use

3.1 The total sulfur content of a carbon black is useful in calculations for reconstructing a rubber composition from analytical data.

4. Reagents

- 4.1 *Purity of Reagents*—Reagent grade chemicals shall be used in all tests. Unless otherwise indicated, it is intended that all reagents shall conform to the specifications of the Committee on Analytical Reagents of the American Chemical Society, where such specifications are available.³ Other grades may be used, provided it is first ascertained that the reagent is of sufficiently high purity to permit its use without lessening the accuracy of the determination.
- 4.2 *Purity of Water* Unless otherwise indicated, references to water shall be understood to mean reagent water conforming to Specification D 1193.

5. Sampling

5.1 Samples shall be taken in accordance with Practice D 1799 or Practice D 1900.

¹ These test methods are under the jurisdiction of ASTM Committee D-24D24 on Carbon Black and are the direct responsibility of Subcommittee D24.31 on Non-Carbon-Black Components of Carbon Black.

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² For referenced ASTM standards, visit the ASTM website, www.astm.org, or contact ASTM Customer Service at service@astm.org. For Annual Book of ASTM Standards, Vol 05.01.volume information, refer to the standard's Document Summary page on the ASTM website.

³ Annual Book of ASTM Standards, Vol 11.01. Reagent Chemicals, American Chemical Society Specifications, American Chemical Society, Washington, DC. For suggestions on the testing of reagents not listed by the American Chemical Society, see Analar Standards for Laboratory Chemicals, BDH Ltd., Poole, Dorset, U.K., and the United States Pharmacopeia and National Formulary, U.S. Pharmaceutical Convention, Inc. (USPC), Rockville, MD.

TEST METHOD A HIGH-TEMPERATURE COMBUSTION WITH INFRARED ABSORPTION DETECTION PROCEDURES

6. Summary of Test Method

- 6.1 The specimen is burned in a tube furnace at a minimum operating temperature of 1350°C in a stream of oxygen to oxidize the sulfur. Moisture and particulates are removed from the gas by traps filled with anhydrous magnesium perchlorate. The gas stream is passed through a cell in which sulfur dioxide is measured by an infrared (IR) absorption detector. Sulfur dioxide absorbs IR energy at a precise wavelength within the IR spectrum. Energy is absorbed as the gas passes through the cell body in which the IR energy is being transmitted. Thus, at the detector, less energy is received. All other IR energy is eliminated from reaching the detector by a precise wavelength filter. Thus, the absorption of IR energy can be attributed only to sulfur dioxide whose concentration is proportional to the change in energy at the detector. One cell is used as both a reference and a measurement chamber. Total sulfur as sulfur dioxide is detected on a continuous basis. This test method is empirical. Therefore, the apparatus must be calibrated by the use of standard reference materials (SRM).
- 6.2 This test method is for use with commercially available sulfur analyzers equipped to carry out the preceding operations automatically and must be calibrated using standard reference material (carbon black) of known sulfur content based on the range of sulfur in each carbon black specimen analyzed.

7. Apparatus

- 7.1 Measurement Apparatus—equipped to automatically combust the specimen.
- 7.2 Combustion Tube, made of mullite, porcelain, or zircon, approximately 40- to 45-mm inside diameter with a 3-mm thick wall, at least 450-mm long with provisions for routing the gasses produced by combustion through the infrared cell.
 - 7.3 Boat Puller—rod of a heat-resistant material with a bent or disk end to insert and remove boats from the combustion tube.

8. Reagents

- 8.1 Purity of Reagents—see 4.1.
- 8.2 Magnesium Perchlorate.

iTeh Standards

9. Preparation of Apparatus

9.1 Assemble the apparatus according to the manufacturer's instructions. Make a minimum of two determinations (see 10.3) to condition the equipment prior to calibrating the system.

10. Calibration

- 10.1 Select black standard reference materials (SRM) containing known sulfur values of approximately 0.5, 1.0, and 1.5% sulfur. Select standards having sulfur values of approximately 0.5, 1.0, and 1.5 % sulfur⁴.
- 10.2 Adjustment of Response of Measurement System—Weigh out approximately 0.5 g of the 1.0 % sulfur standard. Analyze the specimen (see Section 11). Repeat this procedure. Adjust instrument as recommended by the manufacturer until the absence of drift is indicated.
- 10.3 Calibration Procedure—Weigh out four specimens of the 1.0 % sulfur standard. Follow the calibration procedure recommended by the manufacturer. Confirm the calibration by analyzing the 1.0 % sulfur standard. The value should be within the allowable limits of the known value. If not, repeat the procedure. Then weigh out and analyze two specimens, each of the other calibration standards. Record the results after each analysis. Compare the results obtained to the known sulfur values of the specimens. They should be within the allowable limits of the known value of the respective specimen. If not, refer to the manufacturer's instructions for checking linearity of the analyzer.

11. Procedure

- 11.1 Stabilize and calibrate the analyzer (see 10.1 through 10.3).
- 11.2 Raise the furnace temperature as recommended by the manufacturer to at least 1350°C. Weigh the specimen not to exceed more than 0.5 g of carbon black. Spread the specimen evenly in a combustion boat and use a boat puller to position the specimen in the hot zone of the furnace for at least 2 min, or until completely combusted.
 - Note 1—The analytical cycle should begin automatically as soon as sulfur is detected.
- 11.3 When the analysis is complete, the instrument should indicate the sulfur value. Refer to the manufacturer's recommended procedure.

12. Report

12.1 The percent sulfur value is obtained directly from the apparatus.

⁴ Annual Book of ASTM Standards, Vol 09.01.

⁴ Coal standards have been found to be suitable standards and are usually available from the instrument manufacturer or may be obtained from Alpha Products for Analysis. 3090 Johnson Road, Stevensville, MI 49127, www.alpha-resources.com.